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Volatile Organic Compound (VOC) Species Data Manual

For EPA-450/3-78-119, Volatile Organic Compound (VOC) Species Data Manual PLEASE NOTE

In the production process, this volume underwent changes that make the text inconsistent with its index. What is Chapter 6 in the text is referred to in the index as Chapter 9. Properly, Chapter 6 should be changed to Chapter 9, because chapter numbers should reflect Source Classification Code (SCC) categories. Pages 6-01-1 through 6-49-10 should be changed to 9-01-1 through 9-49-10. Profile table numbers and SCC numbers in this chapter should also be changed, with the initial number of each being 9, not 6.

It should be noted that SCCs beginning with 9 were artifically derived as a miscellaneous category associated with this project and are not associated with the National Emissions Data System (NEDS).

Volatile Organic Compound (VOC) Species Data Manual

by

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VOLATILE ORGANIC COMPOUND (VOC) SPECIES DATA MANUAL

INTRODUCTION

The purpose of this study was to prepare a manual that (1) summarizes the available species information on Volatile Organic Compounds (VOC) from stationary and mobile sources in a format useful to the air pollution control community in general, and especially those preparing emission inventories for photochemical models, and (2) provides documentation on the derivation of each of the specie profile tables included herein.

KVB has reviewed the VOC species material prepared for the California Air Resources Board (CARB), as well as mobile source data from reports supplied by the EPA in order to tabulate and reference these data. A data table format has been developed that depicts the species data for each source in a manner to facilitate the allocation of all VOC's into reactivity classes as required by various photochemical models. This table also includes the following information for each VOC identified: SAROAD code, chemical name and classification, molecular weight, and percent concentration by weight and volume.

SUMMARY

MANUAL ORGANIZATION AND SUMMARY INDEX

This manual contains approximately 175 unique VOC emission profiles for controlled and uncontrolled point and area stationary sources. Also included are profiles for mobile sources (a composite for gasoline, one for diesel fuel and another composite profile based on gasoline and diesel fueled vehicles).

The 175 profiles for stationary and area sources have been organized using the National Emission Data System (NEDS) Source Classification Codes (SCC) category numbering system and descriptive headings as listed in Volume V of the Aeros Manual, and as also contained in Appendix C of AP-42, as well as Appendix I of this manual. Where more than one SCC could be assigned to a specific profile, the lowest applicable SCC was used rather than duplicate the profile for each applicable SCC.

A profile table/summary index has been developed that includes the following information:

- a) SCC Category Section Numbers
- b) Descriptive Titles
- c) KVB Profile Key Numbers
- d) Table Numbers
- e) Page Numbers for Sections and Profiles
- f) AP-42 Section References
- g) Applicable SCC Numbers for each Profile

For convenience to the user, this summary index has also been organized separately into the following:

- a) Alphabetically by Profile Title
- b) Numerically by Applicable SCC Number

Report Format Summary

Each emission profile or series of related profiles is preceded by a brief description for proper utilization of the profile(s). The documentation of these profiles is in the form of a mini-report for each SCC category for which profiles were available. These mini-reports are similar to the sections found in AP-42 and each report has been organized as follows:

- a) Process Description--A brief description of the process involved but enough information to include limits and applications.
- b) Emissions--Reference to the applicable section in AP-42, or inclusion of emission data and factors when they were known.
- c) Controls or Process Modification—A brief description of control equipment or process modification and their effect on the profile.
- d) Profile Basis--An explanation of how the profile was developed; how many and what kind of tests, questionnaire data, literature data, engineering judgement, assumptions, etc.
- e) Data Qualification--A description of any limitations or restrictions on profile use.
- f) References--Careful documentation of reports, published data and names and titles of personal contacts.

Items a) through e) above precede the profiles in the mini-report. Item f) "references" will be found after the profiles for each category or, in some cases, at the end of a series of similar categories.

Data Table Format Summary

Data tables have been developed that summarize the most important emission profile data for each source category. The data table format has been organized to present the data so that they can be readily utilized by the air pollution control community in general as well as the photochemical modeler. Each data table contains the following information:

- 1. General Information
 - a) Date
 - b) Table number
 - c) Descriptive title
 - d) Data confidence level

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- e) Control device information
- f) Process modification when appropriate
- g) Method of obtaining profile data
- h) References used to develop profile data
- i) Applicable SCC categories
- 2. Specific Information for Each Species
 - a) SAROAD code
 - b) Chemical name
 - c) Molecular weight
 - d) Percent concentration by weight
 - e) Percent concentration by volume
 - f) Chemical classification
- 3. Specific Information for Each Chemical Classification
 - a) Number of compounds in each classification
 - b) Average molecular weight for each classification
 - c) Average weight percent for each classification
 - d) Average volume percent for each classification
 - e) Average molecular weight of composite compound

Chemical Classification Summary

The chemical classification for each specie listed in the profile data tables has been categorized according to recommendations by John E. Summerhays of the EPA's Source Receptor Analysis Branch, at Research Triangle Park. The chemical classification for each species is as follows:

- 1) Paraffin
- 2) Olefin
- 3) Aromatic
- 4) Carbonyl (aldehydes and ketones)
- 5) Miscellaneous such as:

Esters Acetylene Nitriles
Alcohols Amines Etc.
Ethers Amides
Acids

- 6) Methane
- 7) Non-reactive other than methane (Reference Federal Register July 8, 1977)

Ethane
Trichloromethane (chloroform)

Trichlorotrifluoroethane (Freon 113) Carbon Tetrachloride

Diethylene dibromide

Acetonitrile
Methyl chloride
1,1,1-trichloroethane (methyl chloroform)
1,1,2-trichloroethane
Ethylene dichloride
Benzene

Data Confidence Level Summary

Data confidence levels for each profile table have been expressed using Roman Numerals I through V as follows:

- I. High Degree of Confidence--Based on a composite of many tests so that the data are highly representative of the population.
- II. Above Average Confidence—
 Based on a moderate number of tests or questionnaires which indicate that the data are reasonably representative of the population.
- III. Average Confidence-
 Based on data which seems reasonable and should be more or less representative of the population.
- IV. Below Average Confidence-Based on a little data but not sufficient enough to necessarily
 be representative of the population.
 - V. Low Degree of Confidence--Results are highly judgemental and could vary significantly from source to source.

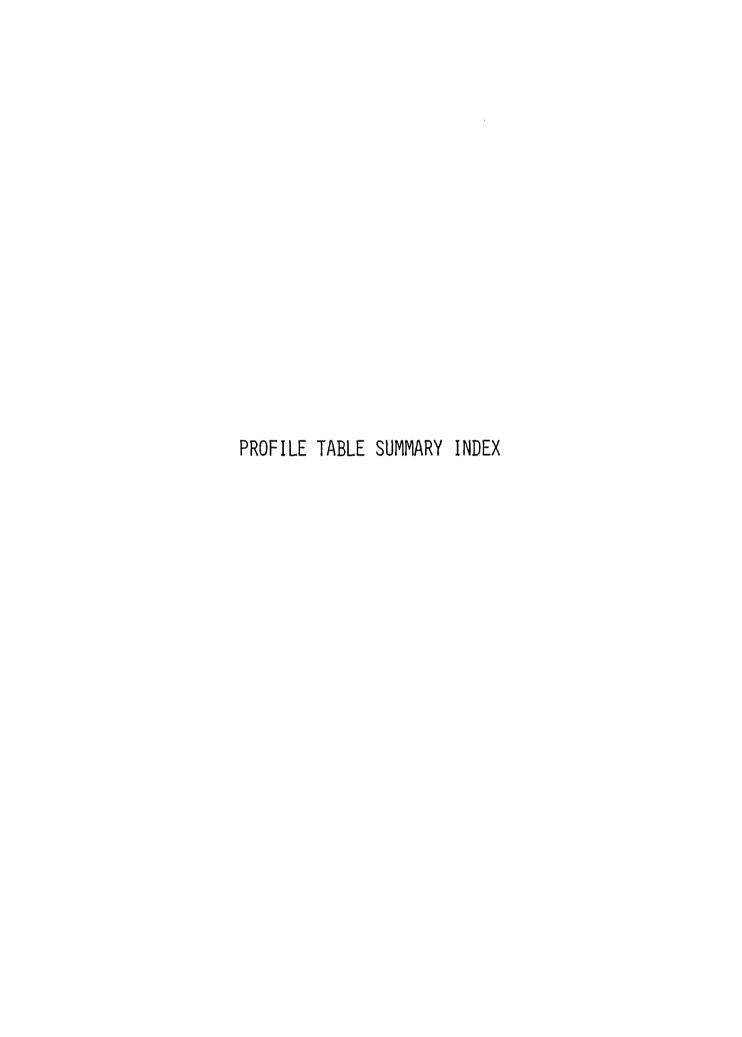
Appendices

There are three appendices at the end of this manual which the user may wish to refer to. Appendix I is a reference data section that contains the following useful information:

- A) Chemical File Sorted by SAROAD Code
- B) Chemical File Sorted by Chemical Classification
- C) Organic Species by Sources
- D) NEDS Source Classification Codes
- E) Area Source Categories
- F) Boiling Point Range Compounds

Appendix II is also a reference section but which describes emission profile development by means of field tests, industry questionnaires and literature data.

Appendix III describes by stepwise calculations how emission profiles were developed for volume percents and average molecular weights from weight percent data.



Section No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCC
2-01	INTERNAL COMBUSTION ENGINE			2.01-1		
	NATURAL GAS TURBINE	0007	2-01-002	2.01-3	3.3	2-01-002-01, 2-02-002-01
	DIESEL FUEL, RECIPROCATING	8000	2-01-003	2.01-4	3.3	2-01-003-01, 2-02-004-01
	DISTILLATE OIL, RECIPROCATING	0009	2-02-001	2.01-5	3.3	2~02~001~02
	NATURAL GAS, RECIPROCATING	0010	2-02-002 A	2.01-6	3.3	2-02-002-02
	NATURAL GAS, 30 HP RECIP.	0308	2-02-002 B	2.01-7	3.3	2~02-002~02
3-01	INDUSTRIAL PROCESS, CHEMICAL MFG.			3.01-1		
	VARNISH MFGBODYING OIL	0066	3-01-015	3.01-3	5.10	3-01-015-01
	PLASTICS, PVC-GENERAL	0067	3-01-018 A	3.01-7	5.13	3-01-018-01
	PLASTICS, POLYPROPYLENE-GENERAL	0068	3-01-018 B	3.01-8	5.13	3-01-018-02
	PHTHALIC ANHYDRIDE-WASTE SUMP	0071	3-01-019 A	3.01-12	5.12	3-01-019-03
	PHTHALIC ANHYDRIDE-CONTROLLED	0276	3-01-019 B	3.01-13	5.12	3-01-019-03
	PRINTING INK COOKING, GENERAL	0072	3-01-020	3.01-17	5.14	3-01-020-01, -99
	AUTOMOTIVE TIRES, TUBER ADHESIVE	0272	3-01-026 A	3.01-21	N/A	3-01-026-20
	AUTO TIRES, TUBER ADHESIVE, WHITE SIDEWALL	0273	3-01-026 B	3.01-22	N/A	
	SYNTHETIC RUBBER, AUTO TIRE PRODUCTION	0274	3-01-026 C	3.01-23	N/A	3-01-026-20
	ETHYLENE DICHLORIDE, DIRECT CHLORINATION	0078	3-01-125	3.01-28	N/A	3-01-125-02, -99
	OTHER, FLARES	0079	3-01-999	3.01-32	N/A	3-01-999-99
	OTHER, FLARES	0079	3-01-999	3.01-32	N/A	3-01-999-99

No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCC
3-02	INDUSTRIAL PROCESS, FOOD/AGRICULTURE			3.02-1		
	FERMENTATION-BEER	0211	3-02-009	3.02-3	6.5	3-02-009-03
3-03	INDUSTRIAL PROCESS, PRIMARY METALS			3.03-1		
	METALLURGICAL COKE MFG., BY-PRODUCT PROCESS, COKE OVEN STACK GAS	0011	3-03-003	3.03-4	7.2	3-03-003-01, -02, -03 3-03-003-04, -05, -06
	IRON PRODUCTION, BLAST FURNACE ORE CHARGING & AGGLOMERATE CHARGING	0012	3-03-006 A	3.03-8	7.5	3-03-008-01, -02
	IRON SINTERING	0013	3-03-008 B	3.03-9	7.5	3-03-008-03
	STEEL PRODUCTION, OPEN HEARTH WITH OXYGEN LANCE	0 306	3-03-009 A	3.03-13	7.5	3-03-009-01
	STEEL PRODUCTION, OPEN HEARTH WITH OXYGEN LANCE-CONTROLLED	0014	3-03-009 B	3.03-14	7.5	3-03-009-01
	STEEL PRODUCTION, BASIC OXYGEN FURNACE	0016	3-03-009 C	3.03-15	7.5	3-03-009-03
3-05	INDUSTRIAL PROCESS, MINERAL PRODUCTS			3.05-1		
	ASPHALT ROOFING, BLOWING OPERATION	0021	3-05-001 A	3.05-4	8.2	3-05-001-01
	ASPHALT ROOFING, DIPPING	0022	3-05 - 001 B	3.05-5	8.2	3-05-001-02, -04
	ASPHALT ROOFING, SPRAYING	0023	3-05-001 C	3.05-6	8.2	3-05-001-03
	ASPHALT ROOFING, TAR KETTLE	00 24	3-05-001 D	3.05-7	N/A	3-05-001-99
	ASPHALT CONCRETE, ROTARY DRYER NATURAL GAS FIRED	0025	3-05-002 A	3.05-11	8.1	3-05-002-01
	ASPHALT CONCRETE, IN-PLACE ROAD ASPHALT	0026	3-05-002 B	3.05-12	8.1	3-05-002-02

Section No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCG
3-06	INDUSTRIAL PROCESS, PETROLEUM INDUSTRY			3.06-1		
	REFINERY, FCC, CO BOILER	0029	3-06-002	3.06-4	9.1	3-06-002-01
	REFINERY, FUGITIVE EMISSIONS, DRAINAGE/SEPARATION PITS, COVERED, CRUDE OIL & GAS	0031	3-06-005	3.06-7	9.1	3-06-005-01
	REFINERY FUGITIVE EMISSIONS COOLING TOWERS	0035	3-06-007	3.06-10	9.1	3-06-007-01
	REFINERY MISCELLANLOUS, PIPE/VALVE FLANGES, COMPOSITE	0316	3-06-008 A	3.06-13	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, GASOLINE	0317	3-06-008 B	3.06-14	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, CRACKED GASOLINE	0319	3-06-008 C	3.06-15	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, GAS-OIL STOCK	0322	3-06-008 D	3.06-16	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, REFORMATE STOCK	0309	3-06-008 E	3.06-17	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, DISTILLATE	0318	3-06-008 F	3.06-18	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, NAPHTHA	0 320	3-06-008 G	3.06-19	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE REFINERY GAS	0324	3-06-008 н	3.06-20	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, PIPE/VALVE FLANGES, NATURAL GAS	0323	3-06-008 J	3.06-21	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, VALVES & FLANGES, WET & DRY NATURAL GAS	0041	3-06-008 K	3.06-22	9.1	3-06-008-01
	REFINERY MISCELLANEOUS, VALVES & FLANGES, GAS PLANT	0036	3-06-008 L	3.06-23	9.1	3-06-008-01
	MISCELLANEOUS, WET GAS VALVE FROM TRAPS, WET NATURAL GAS, COMPOSITE	0042	3-06-008 M	3.06-24	9.1	3-06-008-01
	MARKETING, FUGITIVE EMISSIONS, RELIEF VALVES, LIQUIFIED PETROLEUM GAS	0047	3-06-008 N	3.06-25	9.1	3-06-008-02

Section No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCC
3-06	INDUSTRIAL PROCESS, PETROLEUM INDUSTRY (CONTINUED)					
	REFINERY MISCELLANEOUS, PUMP SEALS, COMPOSITE	0321	3-06-008 P	3.06-28	9.1	3-06-008-03
	REFINERY MISCELLANEOUS, PUMP SEALS, STRAIGHT RUN GASOLINE	0312	3-06-008 Q	3.06-29	9.1	3-06-008-03
	REFINERY MISCELLANEOUS, PUMP SEALS, CRACKED GASOLINE	0310	3-06-008 R	3.06-30	9.1	3-06-008-03
	REFINERY MISCELLANEOUS, PUMP SEALS, GAS-OIL STOCK	0313	3-06-008 S	3.06-31	9.1	3-06-008-03
	REFINERY MISCELLANEOUS, PUMP SEALS, REFORMATE STOCK	0314	3-06-008 Т	3.06-32	9.1	3-06-008-03
	REFINERY MISCELLANEOUS, PUMP SEALS, DISTILLATE	0311	3-06-008 U	3.06-33	9.1	3-06-008-03
	REFINERY MISCELLANEOUS, PUMP SEALS, NAPHTHA	0315	3-06-008 V	3.06-34	9.1	3-06-008-03
	MISCELLANEOUS, PUMP SEALS, NATURAL GASOLINE	0043	3-06-008 W	3.06-35	9.1	3-06-008-03
	MISCELLANEOUS, PUMP SEALS, GASOLINE	0038	3-06 - 008 x	3.06-36	9.1	3-06-008-03
	MISCELLANEOUS, COMPRESSOR SEALS, WET AND DRY NATURAL GAS	0044	3-06-008 Y	3.06-37	9.1	3-06-008-04
	MISCELLANEOUS, COMPRESSOR SEALS, REFINERY GAS	0039	3-06 - 008 Z	3.06-38	9.1	3-06-008-04
	REFINERY FLARES, NATURAL GAS	0051	3-06-009	3.06-41	9.1	3-06-009-01
	REFINERY, CATALYTIC REFORMER, GENERAL FUGITIVE EMISSIONS	0053	3-06-013	3.06-43	9.1	3-06-013-01

No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCC
3~ 30	INDUSTRIAL PROCESS, TEXTILE MANUFACTURING			3.30-1		
	FABRIC DYEING, GENERAL	0060	3-30-001	3.30-2	N/A	3-30-001-99
3-90	INDUSTRIAL PROCESS, IN PROCESS FUEL			3.90-1		
	PROCESS GAS, COKE OVEN BLAST FURNACE GAS	0217	3-90-007	3.90-3	N/A	3-90-007-01
4-01	POINT SOURCE EVAPORATION, CLEANING SOLVENT			4.01-1		
	DRY CLEANING, 1, 1, 1-TRICHLOROETHANE	0087	4-01-001 A	4.01-4	4.1	4-01-001-99 4-01-002-02
	DRY CLEANING, STODDARD SOLVENT	0086	4-01-001 B	4.01-5	4.1	4-01-001-02 4-01-002-01
	DRY CLEANING, PERCHLOROETHYLENE	0085	4-01-001 C	4.01-6	4.1	4-01-001-01 4-01-002-03
	DEGREASING, TRICHLOROETHANE	0088	4-01-002 A	4.01-10	H/A	4-01-002-02
	DEGREASING, DICHLOROMETHANE	0275	4-01-002 B	4.01-11	N/A	4-01-002-04
	DEGREASING, TRICHLOROETHYLENE	0271	4-01-002 C	4.01-12	N/A	4-01-002-05
	DEGREASING, TOLUENE	0090	4-01-002 D	4.01-13	N/A	4-01-002-06
	DEGREASING, TRICHLOROFLUOROMETHANE	0088	4-01-002 E	4.01-14	N/A	4-01-002-99
	DEGREASING, TRICHLOROTRIFLUOROETHANE (FREON 113)	0277	4-01-002 F	4.01-15	N/A	4-01-002-99
4 -02	POINT SOURCE EVAPORATION, SURFACE COATING			4.02-1		
	PAINT, POLYMERIC (HOT AIR DRIED)	0125	4-02-001 A	4. 0 2 -5	4.2	4-02-001-01
	PAINT SOLVENT, ACETONE	0219	4-02-001 B	4.02-6	4.2	4-02-001-02 4-02-003-02 4-02-004-02 4-02-009-02
	PAINT SOLVENT, ETHYL ACETATE	0220	4-02-901 C	4.02-7	4.2	4-02-001-03 4-02-003-03 4-02-004-03
	PAINT SOLVENT, METHYL ETHYL KETONE	0221	4-02-001 D	4.02-8	4.2	4-02-001-04 4-02-005-03 4-02-007-02 4-02-009-18
	VARNISH/SHELLAC, GENERAL	0127	4-02-003 A	4.02-9	4.2	4-02-003-01

Section No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCC
4-02	POINT SOURCE EVAPORATION, SURFACE COATING (CONTINUED)					
	VARNISH/SHELLAC, VARNISH RESIN	0132	4-02-003 B	4.02-10	4.2	4-02-003-01
	VARNISH/SHELLAC, LXH-221, AIR CONVERTING VARNISH	0278	4-02-003 C	4.02-11	4.2	4-02-003-01
	COMPOSITE, VARNISH/SHELLAC, A. BROWN CO., BROLITE, MIL-V-173B, TT-V-109B	0133	4- 02-003 D	4.02-12	4.2	4-02-003-01
	VARNISH/SHEILAC, XYLENE	0223	4-02-003 E	4.02-13	4.2	4-02-003-05 4-02-009-24
	LACQUER, PAPERBOARD PRODUCTS AND CONTAINERS	0149	4-02-004 A	4.02-14	4.2	4-02-004-01, -99
	LACQUER, METAL FURNITURE	0148	4-02-004 B	4.02-15	4.2	4-02-004-01, -99
	LACQUER, LXB-472-E SEMIGLOSS TOP COAT	0147	4 -02-004 C	4.02-16	4.2	4-02-004-01, -99
	LACQUER, AIRCRAFT COATING	0155	4-02-004 D	4.02-17	4.2	4-02-004-01, -99
	LACQUER, AIRCRAFT PARTS	0146	4-02-004 E	4.02-18	4.2	4-02-004-01 4-02-006-99
	LACQUER, PLASTIC COATING	0150	4-02-004 F	4.02-19	4.2	4-02-004-01, -99
	ENAMEL, GENERAL COMPOSITE	0156	4-02-005 A	4.02-20	4.2	4-02-005-01, -99
	ENAMEL, POLYESTER, MODIFIED ACRYLIC	0159	4-02-005 B	4.02-21	4.2	4-02-005-01, -99
	ENAMEL, COMPOSITE FOR WOOD FURNITURE	0157	4- 02-005 C	4.02-22	4.2	4-02-005-01, -99
	ENAMEL, AIRCRAFT INDUSTRY	0164	4-02-005 D	4.02-23	4.2	4-02-005-01, -99
	ENAMEL, CELLOSOLVE ACETATE	0222	4 - 02-005 E	4.02-24	4.2	4-02-005-02 4-02-009-07
	PRIMER, GENERAL	0134	4-02-006 A	4.02-25	4.2	4-02-006-01, -99
	PRIMER, PAPERBOARD PRODUCTS & CONTAINERS	0137	4-02-006 B	4.02-26	4.2	4-02-006-01
	PRIMER, METAL FURNITURE	0136	4-02-006 C	4.02-27	4.2	4-02-006-01

No.	Title	Profile Key No.	Table No.	Page No.	AP-42 Section	Applicable SCC
4-02	POINT SOURCE EVAPORATION, SURFACE COATING (CONTINUED)					
	PRIMER, RED OXIDE SHOP COAT, KOPPERS P-470-A-66	0331	4- 02-306 D	4.02-28	4.2	4-02-036-01
	PRIMER, WATER BASED AUTOMOTIVE PAINT SPRAY BOOTH	0280	4-02-006 E	4.02-29	4.2	4-02-006-01
	PRIMER, BLACK KOPPERS A-1131-66	0281	4-02-006 F	4.02-30	4.2	4-02-006-01
	PRIMER NAPHTHA	0 2 8 2	4-02-906 G	4.02-31	4.2	4-02-006-02 4-02-007-05 4-02-009-21 4-05-003-07
	PRIMER, MINERAL SPIRITS	0283	4-02-006 ห	4.02-32	4.2	4-02-006-04 4-02-009-20 4-05-002-03 4-05-004-02 4-05-005-08
	PRIMER, MINERAL SPIRITS, SPECIATION	0225	4- 02-006 J	4.02-33	4.2	4-02-006-04
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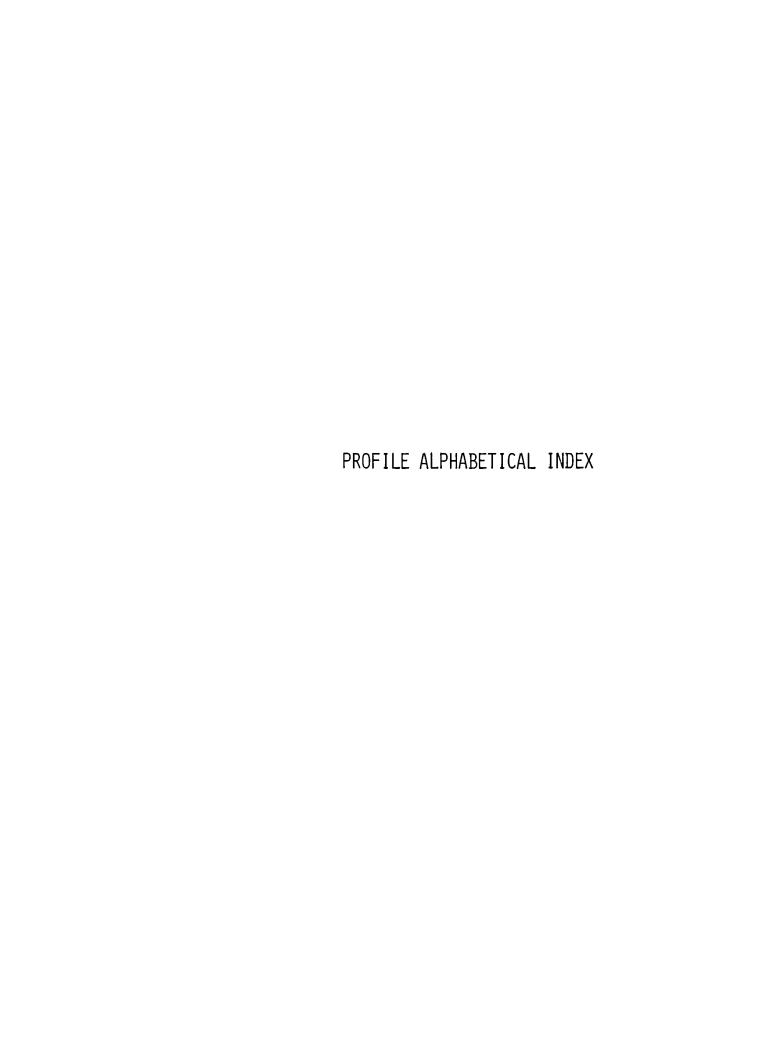
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COMPOSITE GASOLINE/DIESEL	9.11-2	0325	9-11-061
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LIGHT & HEAVY VEHICLE	9.06-4	0329	9-06-021 A

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
DIESEL FUEL			
RECIPROCATING	2.01-4	8000	2-01-003
LIGHT & HEAVY VEHICLE	9.07-3	0330	9-07-021
DISTILLATE OIL, RECIPROCATING	2.01-5	0009	2-02-001
NATURAL GAS, RECIPROCATING			
INDUSTRIAL	2.01-6	0010	2-02-002 A
30 HP BUDA	2.01-7	0308	2-02-002 B
NATURAL GAS TURBINE	2.01-3	0007	2-01-002
IRON PRODUCTION			
BLAST FURNACE CHARGING	3.03-8	0012	3-03-008 A
IRON SINTERING	3.03-9	0013	3-03-008 B
LETTERPRESS (see Printing Press)			
LITHOGRAPHY (see Printing Press)			
OPEN HEARTH FURNACE (see Steel Production)			
OVENS (see Surface Coating-Coating Oven)			
PAINT (see Surface Coating)			
PESTICIDES, DOMESTIC & COMMERCIAL	9.35-11	0076	9-35-705
PETROLEUM SEEPS	9.47-8	0205	9-47-411

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
PETROLEUM STORAGE			
FIXED ROOF TANK			
BENZENE	4.03-15	0298	4-03-001 F
CRUDE OIL, COMPOSITE, MARINE TERMINAL	4.03-26	0305	4-06-002
CRUDE OIL, COMPOSITE, PRODUCTION	4.03-12	0296	4-03-001 0
CRUDE OIL, COMPOSITE, REFINERY	4.03-13	0297	Г
CRUDE OIL & WASTE WATER	4.03-24	0188	Ç
CYCLOHEXANE	4.03-16	0299	
CYCLOPENTANE	4.03-17	0300	H
GASOLINE COMPOSITE	4.03-10	0098	P
GASOLINE, FCCU & REFORMER BLEND	4.03-11	0187	В
HEPTANE	4.03-18	0301	J
HEXANE	4.03-19	0230	K
ISO-OCTANE	4.03-20	0302	L
ISO-PENTANE	4.03-21	0231	M
JET A	4.03-14	0100	E
PENTANE	4.03-22	0303	N
TOLUENE	4.03-23	0185	P
FLOATING ROOF TANK (see Fixed Roof Tank)			
VARIABLE VAPOR SPACE			
LIQUIFIED PETROLEUM GAS	4.03-25	0232	4-03-003

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
PHTHALIC ANHYDRIDE			
CONTROLLED	3.01-13	0276	3-01-019 B
WASTE SUMP	3.01-12	0071	3-01-019 A
PIPE/VALVES (see Refinery)			
PLASTICS			
PROPYLENE, GENERAL	3.01-8	0068	3-01-018 B
PVC, GENERAL	3.01-8	0067	3-01-018 A
PRIMER (see Surface Coating)			
PRINTING INK, COOKING, GENERAL	3.01-17	0072	3-01-020
PRINTING PRESS			
FLEXOGRAPHIC			
ALCOHOL BASED SOLVENT, COMPOSITE	4.05-9	0172	4-05-003 A
N-PROPYL ALCOHOL	4.05-10	0304	4.05-003 B
GRAVURE			
PAPERBOARD PRINTING	4.05-19	0181	4-05-005 A
SOLVENT,			
COMMERCIAL PRINTING	4.05 -2. L	0184	С
GENERAL PRINTING	4.05-22	0182	D
PERIODICAL PRINTING	4.05-20	0183	В
LETTERPRESS			
INKING & DRYING	4.05-4	0334	4-05-002 A
INKING PROCESS	4.05-5	0166	4-05-002 B

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
LITHOGRAPHY			
INKING & DRYING	4.05-14	0333	4-05-004 A
INKING & DRYING, CONTROLLED	4.05-15	0332	4-05-004 B
PUMP SEALS (see Refinery)			
REFINERY			
CATALYTIC REFORMER	3.06-43	0053	3-06-013
COMPRESSOR SEALS			
NATURAL GAS	3.06-37	0044	3-06 - 008 Y
REFINERY GAS	3.06-38	0039	3 - 06-008 Z
COOLING TOWERS	3.06-10	0035	3-06-007
DRAINAGE/SEPARATION PITS	3.06-7	0031	3 - 06-005
FCC BOILER	3.06-4	0029	3-06-002
FLARES, NATURAL GAS	3.06-41	0051	3-06-009
PIPE/VALVE FLANGES			
COMPOSITE	3.06-13	0316	3-06-008 A
DISTILLATE	3.06-18	0318	3-06-008 F
GASOLINE	3.06-14	0317	В
GASOLINE, CRACKED	3.06-15	0319	С
GAS-OIL STOCK	3.06-16	0322	D
GAS PLANT	3.06-23	0036	L
NAPHTHA	3.06-19	0320	G
NATURAL GAS	3.06-21	0323	y J

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
NATURAL GAS, WET & DRY	3.06-22	0041	3-06-008 K
REFINERY GAS	3.06-20	0324	н
REFORMATE STOCK	3.06-17	0309	E
PUMP SEALS			
COMPOSITE	3.06-28	0321	3-06-008 P
DISTILLATE	3.06-33	0311	υ
GASOLINE	3.06-36	0038	х
GASOLINE, CRACKED	3.06-30	0310	R
GASOLINE, NATURAL	3.06-35	0043	w
GASOLINE, STRAIGHT RUN	3.06-29	0312	Q
GAS-OIL STOCK	3.06-31	0313	s
NAPHTHA	3.06-34	0315	v
REFORMATE STOCK	3.06-32	0314	Т
RELIEF VALVES, LPG	3.06-25	0047	N
TRAPS, WET NATURAL GAS COMPOSITE	3.06-24	0042	↓ M
RESIDENTIAL FUEL			
NATURAL GAS	9.01-3	0195	9-01-005
SHELLAC (see Surface Coating)			
SINTERING (see Iron Production)			
SOLID WASTE			
ANIMAL WASTE DECOMPOSITION	9.49-9	0203	9-49-999 B
INCINERATOR, BAR SCREEN WASTE	5.01-8	0122	5-01-005

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.	
LANDFILL SITE, CLASS II	9.49-5	0202	9-49-999 A	
OPEN BURNING DUMP, LANDSCAPE/PRUNING	5.01-4	0121	5-01-002	
SOLVENT STORAGE (see: Surface Coating; Petroleum S	torage; Clean	ing Solvents)		
SOLVENT USE (see also Surface Coating)				
DOMESTIC SOLVENTS, GENERAL	9.35-7	0197	9-35-702	
PESTICIDES, DOMESTIC & COMMERCIAL	9.35-11	0076	9-35-705	
STEEL PRODUCTION				
BASIC OXYGEN FURNACE	3.03-15	0016	3-03-009 C	
OPEN HEARTH W/OXYGEN LANCE	3.03-13	0306	3-03-009 A	
OPEN HEARTH W/OXYGEN LANCE, CONTROLLED	3.03-14	0014	3-03-009 в	
SURFACE COATING				
ADHESIVE				
AUTOMOTIVE VINYL TOP, SPRAY BOOTH	4.02-37	0285	4-02-007 C	
BENZENE	4.02-39	0287	F	
FOSTER BOND SEAL NO. 107	4.02-38	0145	E	
LABEL	4.02-35	0141	A	
METAL FURNITURE	4.02-36	0142	В	
ARCHITECTURAL COATINGS	9.35-3	0196	9-35-103	

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
CITRUS COATING WAX			
BROGDEX 502	4.02-40	0293	4-02-999
FLAVORSEAL 115-1800	4.02-42	0295	4-02-999
FLAVORSEAL 320-1820	4.02-41	0294	4-02-999
COATING OVEN			
AUTOMOTIVE			
ENAMEL	4.02-51	0162	4-02-008
LACQUER	4.02-52	0154	4-02-008
PRIMER	4.02-48	0279	4-02-008
VINYL TOP ADHESIVE	4.02-50	0286	4-02-008
METAL PARTS, GENERAL	4.02-47	0092	4-02-008
ZINC CHROMATE, INFRARED DRYER	4.02-49	0056	4-02-008
ENAMEL			
AIRCRAFT INDUSTRY	4.02-23	0164	4-02-005
CELLOSOLVE ACETATE	4.02-24	0222	
GENERAL, COMPOSITE	4.02-20	0156	i
POLYESTER, MODIFIED ACRYLIC	4.02-21	0159] 1
WOOD FURNITURE COMPOSITE	4.02-22	0157	. ↓
LACQUER			
AIRCRAFT COATING	4.02-17	0155	4-02-004 1
AIRCRAFT PARTS	4.02-18	0146	1

DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
LXB-472-E SEMIGLOSS TOP COAT	4.02-16	0147	4÷02−004 C
METAL FURNITURE	4.02-15	0148	В
PAPER PRODUCTS	4.02-14	0149	A
PLASTIC COATING	4.02-19	0150	↓ G
PAINT			
POLYMERIC, HOT AIR DRIED	4.02-5	0125	4-02-001 A
SOLVENT			
ACETONE	4.02-6	0219	В
ETHYL ACETATE	4.02-7	0220	С
METHYL ETHYL KETONE	4.02-8	0221	↓ D
PRIMER			
AUTOMOTIVE, WATER BASED SPRAY BOOTH	4.02-29	0280	4-02-006 E
GENERAL	4.02-25	0134	A
KOPPERS, A-1131-66, BLACK	4.02-32	. 0281	F
KOPPERS, P-470-A-66, RED OXIDE	4.02-30	0331	D
METAL FURNITURE	4.02-27	0136	c
PAPERBOARD PRODUCTS	4.02-26	0137	В
SHELL M-75	4.02-34	0284	K
SOLVENT			
MINERAL SPIRITS	4.02-32	0283	н
MINERAL SPIRITS, SPECIATION	4.02-33	0225	J
NAPHTHA	4.02-31	0282	G

			
DESCRIPTION	PAGE NO.	PROFILE KEY NO.	TABLE NO.
SOLVENT			
BUTYL ACETATE	4.02-57	0288	4-02-009 B
BUTYL ALCOHOL	4.02-58	0289	С
CELLOSOLVE	4.02-59	0290	D
DIMETHYL FORMAMIDE	4.02-60	0292	E
ETHYL ALCOHOL	4.02-61	0226	F
GENERAL	4.02-56	0096	A
ISOPROPYL ACETATE	4.02-63	0228	Н
ISOPROPYL ALCOHOL	4.02-62	0227	G
LACTOL SPIRITS	4.02-64	0229	J
METHYL ALCOHOL	4.02-65	0291	V K
VARNISH MANUFACTURE, BODYING OIL	3.01-15	0066	3-01-015
VARNISH/SHELLAC			
AIR CONVERTING VARNISH, LXH-221	4.02-11	0278	4-02-003 C
BROLITE MIL-V-173B	4.02-12	0133	D
GENERAL	4.02-9	0127	A
RESIN, VARNISH	4.02-10	0132	В
SOLVENT			
XYLENE	4.02-13	0223	E
SYNTHETIC RUBBER			▼
AUTO TIRE PRODUCTION	3.01-23	0274	3-01-026 C
VARNISH (see Surface Coating)			



1-01 EXTERNAL COMBUSTION BOILERS--RESIDUAL OIL

--DISTILLATE OIL

--NATURAL GAS

--REFINERY GAS

-- COKE OVEN GAS

Process Description 1

External combustion sources include utility, industrial, commercial and institutional boilers; commercial and domestic combustion units; process heaters, furnaces, kilns, etc. Coal, oil and natural gas are the major fossil fuels used by these sources.

Emissions2,3

As a rule, very small concentrations of hydrocarbons will be produced during coal, oil and gas combustion. If a unit is operated improperly or not maintained, as is the case with small, often unattended units, then the resulting concentrations of these pollutants may increase by several orders of magnitude.

Emission factors for various types of fossil fuels and boiler sizes will be found in Sections 1.3, 1.4 and 1.5 of the EPA document AP-42 (Ref. 3).

Controls

The normal procedure followed in controlling unburned or partially burned hydrocarbon pollutants is to increase the combustion efficiency of the unit rather than treatment at the exhaust stack. Because of the low concentrations of VOC found in the exhaust gas of a properly tuned device control equipment is generally not necessary or practical.

Profile Basis²

Flue gas samples were taken from the exhaust stacks of external combustion boilers burning different fossil fuels using the portable sampling train and procedure described in Appendix 2. Most tests involved taking one sample, however, on selected sources two independent trains were used to provide duplicate field samples. All organic species contributing at least 1% of the total organic composition are identified in the external combustion profiles.

Data Qualification

Although only one test was performed for each of the fuels described in the following profiles, the test locations were carefully selected on the basis of the representative nature of their emissions to all other devices of that particular type and fuel. As a result, these profiles should be applicable to other non-tested sources as indicated by the applicable SCC categories listed in the notes for each profile.

DECEMBER 14, 1978 TABLE 1-01-004

EXTERNAL COMBUSTION BOILER

RESIDUAL OIL

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0001

LINE NO.	SARGAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43212	N-BUTANE	58. 12	14.00	8. 41	1	PARAFFIN	
2	43231	N-HEXANE	86. 17	5. 00	2. 02	1	PARAFFIN	
3	43502	FORMALDEHYDE	30. 03	42.00	48. 81	4	CARBONYL	
4	43551	ACETONE	58 . 08	28.00	16. 82	4	CARBONYL	
5	43201	METHANE	16. 04	11.00	23. 94	6	METHANE	
		TOTAL		100. 00	100. 00			
	2 COMP	DUNDS OF CLASSIFICATION 1	63. 55	19. 00	10. 43			
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
		COUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 4	37. 22	70.00	65. 63			
		OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 6	16. 04	11.00	23. 94			
		OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	5 COMP	OUND COMPOSITE	34. 90	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA OC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA, AP-42 SECTION 1.3

C. APPLICABLE SCC CATEGORIES: 1-01-004-01, -02, -03 (REFER ALSO TO SUMMARY INDEX)

EXTERNAL COMBUSTION BOILER

DISTILLATE OIL

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0002

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	5. 20	2. 58	1	PARAFFIN	
2	43106	ISOMERS OF HEPTANE	100. 20	2. 60	1. 12	1	PARAFFIN	
3	43107	ISOMERS OF OCTANE	114. 23	4. 70	1. 76	1	PARAFFIN	
4	43122	ISOMERS OF PENTANE	72. 15	5. 50	3. 27	1	PARAFFIN	
5	43204	PROPANE	44. 09	1. 20	1. 16	1	PARAFFIN	
6	43212	N-BUTANE	58. 12	12. 20	9. 03	1	PARAFFIN	
7	43214	ISOBUTANE	58. 12	4. 10	3. 05	1	PARAFFIN	
8	43220	N-PENTANE	72. 1 5	4. 70	2. 79	1	PARAFFIN	
9	43231	N-HEXANE	86. 17	10. 80	5, 37	1	PARAFFIN	
10	43232	N-HEPTANE	100. 20	. 30	. 13	1	PARAFFIN	
11	43502	FORMALDEHYDE	30. 03	48 . 70	69, 73	4	CARBONYL	
		TOTAL		100. 00	99. 99			
	10 COMP	DUNDS OF CLASSIFICATION 1	72. 76	51. 30	30. 26			
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	1 COMP	DUNDS OF CLASSIFICATION 4	30. 03	48. 70	69. 73			
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	11 COMP	DUND COMPOSITE	42. 96	100.00	99, 99			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA, AP-42 SECTION 1.3

C. APPLICABLE SCC CATEGORIES: 1-01-005-01, -02, -03 (REFER ALSO TO SUMMARY INDEX)

DECEMBER 14, 1978 TABLE 1-01-006

EXTERNAL COMBUSTION BOILER

NATURAL GAS

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0003

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	, -	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATI ON
1	43105	ISOMERS OF HEXANE	86. 17	1.00	. 28	1	PARAFFIN
2	43122	ISOMERS OF PENTANE	72. 15	9. 00	2. 90	1	PARAFFIN
3	43204	PROPANE	44. 09	4. 00	2. 11	1	PARAFFIN
4	43212	N-BUTANE	58. 12	9. 00	3. 60	1	PARAFFIN
5	43220	N-PENTANE	72. 15	6. 00	1. 93	1	PARAFFIN
6	43248	CYCLOHEXANE	84. 16	1.00	. 28	1	PAR A FFIN
7	45202	TOLVENE	92 . 13	2. 00	. 51	3	AROMATIC
8	43502	FORMALDEHYDE	30. 03	8.00	6. 17	4	CARBONYL
9	43201	METHANE	16. 04	56. 00	81.04	6	METHANE
10	45201	BENZENE	78. 11	4. 00	1. 18	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	6 COMP	DUNDS OF CLASSIFICATION 1	62. 92	3 0. 00	11. 10		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	92, 13	2. 00	. 51		
		OUNDS OF CLASSIFICATION 4	30. 03	B. 00	6, 17		
	O COMP	GUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	56. 00	81. 04		
	1 COMP	OUNDS OF CLASSIFICATION 7	78. 11	4. 00	1. 18		
	10 COMP	OUND COMPOSITE	23. 23	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA, AP-42 SECTION 1.4

C. APPLICABLE SCC CATEGORIES: 1-01-006-01, -02, -03 (REFER ALSO TO SUMMARY INDEX)

EXTERNAL COMBUSTION BOILER

REFINERY GAS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0004

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44. 09	18. 90	15. 65	1	PARAFFIN
2	43212	N-BUTANE	58. 12	23. 10	14. 52	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	4. 40	2. 77	1	PARAFFIN
4	43205	PROPYLENE	42. 08	17. 50	15. 18	2	OLEFIN
5	43502	FORMALDEHYDE	3 0. 0 3	7. 60	9. 23	4	CARBONYL
6	43201	METHANE	16. 04	7. 60	17. 29	6	METHANE
7	43202	ETHANE	30. 07	20. 90	25. 36	7	NON-REACTIVE
		TOTAL		100.00	100. 00		
	O COMP	Character of a contraction a	E4 45	44 40	22.04		
		DUNDS OF CLASSIFICATION 1	51. 45	46. 40 17. 50	32. 94 15. 18		
		DUNDS OF CLASSIFICATION 2	42. 08	+			
		DUNDS OF CLASSIFICATION 3 DUNDS OF CLASSIFICATION 4	. 00 30. 03	. 00	. 00 9. 23		
				7. 60			
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		BUNDS OF CLASSIFICATION 6	16. 04	7. 60	17. 29		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	20. 90	25. 36		
	7 COMP	DUND COMPOSITE	36. 51	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA QC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA, AP-42 SECTION NONE

C. APPLICABLE SCC CATEGORIES: 1-01-007-01, -02, 1-02-007-01, -02, -03

EXTERNAL COMBUSTION BOILER

COKE OVEN GAS

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0005

_INE NO.	SARDAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	Ci	CHEMICAL ASSIFICATION
	OUDL	TATALLE.	WE 10111	WE 20111	1020112	-	
1	43203	ETHYLENE	28. 05	11. 70	7. 29	2	OLEFIN
2	43205	PROPYLENE	42. 08	. 30	. 12	2	OLEFIN
3	43206	ACETYLENE	26. 04	. 80	. 54	5	MISCELLANEOUS
4	43201	METHANE	16. 04	82. 80	90. 18	6	METHANE
5	43202	ETHANE	30. 07	2. 50	1. 45	7	NON-REACTIVE
6	45201	BENZENE	78. 11	1. 90	. 42	7	NON-REACTIVE
		TOTAL		100.00	100.00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	2 COMP	OUNDS OF CLASSIFICATION 2	28. 28	12.00	7. 41		
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 0 0	00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 5	26. 04	. 80	. 54		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	82. 80	90. 18		
	2 COMP	OUNDS OF CLASSIFICATION 7	40. 86	4. 40	1.87		
	- COMP	OUND COMPOSITE	17. 46	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

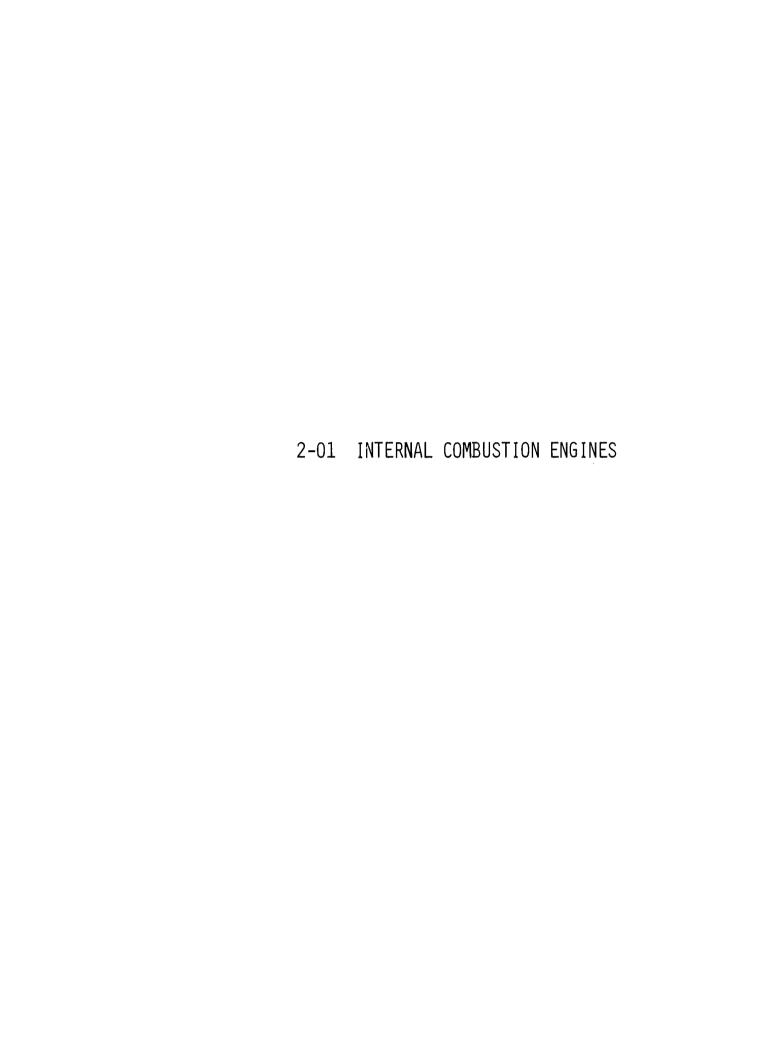
B. REFERENCES: KVB TEST DATA, AP-42 SECTION NONE

C. APPLICABLE SCC CATEGORIES: 1-02-008-02, 3-90-007-02, 3-90-008-01,-99

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for the Environmental Protection Agency, Durham, NC, EPA-340/1-78-004, April 1978.
- 2. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I & II, KVB, Inc., Tustin, CA, June 1978.
- 3. "Compilation of Air Pollutant Emission Factors," Environmental Protection Agency, Research Trinagle Park, NC, AP-42, August 1977.

1.01-8 26200-795



Process Description 1

Engines included in this category are internal combustion (IC) engines used in applications similar to those associated with external combustion sources. The major engines in this category are gas turbines and large, heavy duty, general utility reciprocating engines. Stationary internal combustion engines find applications in electrical power generators, in gas pipeline pump and compressor drives and in various process industries. The majority of gas turbines are used in electrical generation for continuous, peaking or stand-by power. The primary fuels used are natural gas and No. 2 (distillate) fuel oil, although residual oil is used in a few applications.

Emissions^{1,2}

The organic emission factor data presented in Section 3.3, "Off-Highway, Stationary Sources" in AP-42 (Ref. 1) are for very large stationary engines in the 800 hp category which are much larger than those typically found in the California South Coast Air Basin. The results of tests conducted by KVB, Inc. on IC engines also resulted in large emission rates (Ref. 2), however, insufficient data were obtained to generalize an emission factor for these engines. The AP-42 values appear reasonable, although it should be recognized that emission rates for the smaller IC engines may be much greater.

Controls²

Although stationary IC engines are a significant source of organic emissions, control equipment to reduce these emissions is not yet required. Hydrocarbon emissions from IC engines can be reduced by: (a) improved operating practice, (b) proper maintenance, (c) improved equipment design, (d) fuel substitution, and (e) add-on devices such as catalytic converters.

Profile Basis

Field test samples of fuel combustion exhaust from IC engines burning different fuels were taken using the portable sampling train and procedures described in Appendix 2. Profile 2-02-002A was determined from one test on a natural gas burning IC engine driving a compressor at a refinery. Flue gas flow rate was 1679 SCFM and gas temperature was 660 °F. Profile 2-02-002B is a composite of four tests on separate well pump engines at an oil field. These were six cylinder, four cycle, 30 hp, Buda engines operating on natural gas.

Engineering evaluation of literature data was used to develop profiles 2-01-003, diesel fuel turbine and 2-02-001, distillate oil reciprocating engine (Ref. 3).

Data Qualification

The source tests and literature data used were carefully selected on the basis of the representative nature of their emissions to other devices of that particular type and fuel. Therefore, the following profiles for IC engines may be correctly applied to other sources as evidenced by the applicable SCC categories listed in the notes for each profile.

2.01-2

26 200 – 79 5

TABLE 2-01-002

INTERNAL COMBUSTINE ENGINE, ELECTRICAL GENERATION

NATURAL GAS TURBINE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0007

INE NO.	SARDAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43502	FORMALDEHYDE	30. 03	30. 00	18. 63	4	CARBONYL
2	43201	METHANE	16. 04	70.00	81. 37	6	METHANE
		TOTAL		100.00	100. 00		
	O COMP	POUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	POUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 4	30. 03	30.00	18. 63		
	O COMP	POUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	POUNDS OF CLASSIFICATION 6	16. 04	70. 00	81.37		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	2 COMP	POUND COMPOSITE	18. 65	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA, AP-42 SECTION 3.3

C. APPLICABLE SCC CATEGORIES: 2-01-002-01, 2-02-002-01

INTERNAL COMBUSTION ENGINE, ELECTRICAL GENERATION

DIESEL FUEL, RECIPROCATING DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0008

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43203	ETHYLENE	28. 05	28. 70	32. 45	2	OLEFIN
2	43205	PROPYLENE	42. 08	17. 30	13. 04	2	OLEFIN
3	43213	BUTENE	56 . 10	13. 40	7. 5 8	2	OLEFIN
4	43218	1,3-BUTADIENE	54. 09	7. 00	4. 09	2	OLEFIN
5	43206	ACETYLENE	26. 04	11. 30	13. 76	5	MISCELLANEOUS
6	43201	METHANE	16. 04	11.60	22. 93	6	METHANE
7	43202	ETHANE	30 . 07	2. 80	2. 95	7	NON-REACTIVE
8	45201	BENZENE	78. 11	7. 90	3. 20	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	4 COMP	OUNDS OF CLASSIFICATION 2	36. 83	66. 4 0	57. 16		
	O COMP	OUNDS OF CLASSIFICATION 3	. 0 0	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	26. 04	11. 30	13. 76		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	11.60	22. 93		
		OUNDS OF CLASSIFICATION 7	55. 07	10. 70	6. 15		
	8 COMP	DUND COMPOSITE	31.70	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER, AP-42 SECTION 3.3

C. APPLICABLE SCC CATEGORIES: 2-01-003-01, 2-02-004-01

INTERNAL COMBUSTION ENGINE, INDUSTRIAL DISTILLATE OIL, RECIPROCATING DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0009

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43203	ETHYLENE	28. 05	28. 70	32, 45	2	OLEFIN
2	43205	PROPYLENE	42. 08	17. 30	13. 04	2	OLEFIN
3	43213	BUTENE	56. 10	13. 40	7. 58	2	OLEFIN
4	43218	1,3-BUTADIENE	54. 09	7. 00	4. 09	2	OLEFIN
5	43206	ACETYLENE	26. 04	11.30	13. 76	5	MISCELLANEOUS
6	43201	METHANE	16. 04	11. 60	22. 93	6	METHANE
7	43202	ETHANE	30. 07	2. 80	2. 95	7	NON-REACTIVE
8	45201	BENZENE	78. 11	7. 9 0	3. 20	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	4 COMP	OUNDS OF CLASSIFICATION 2	36, 83	66, 40	57. 16		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	QUNDS OF CLASSIFICATION 5	26, 04	11.30	13. 76		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	11.60	22. 93		
		OUNDS OF CLASSIFICATION 7	55. 07	10. 70	6. 15		
	B COMP	OUND COMPOSITE	31. 70	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA LITERATURE DATA.

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER, AP-42 SECTION 3.3

C. APPLICABLE SCC CATEGORIES: 2-02-001-02

TABLE 2-02-002 A

INTERNAL COMBUSTION ENGINE, INDUSTRIAL

NATURAL GAS, RECIPROCATING DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0010

LINE SAROAD CHEMICAL MOLECULAR PERCENT PERCENT CHEMICAL NO. CODE NAME WEIGHT WEIGHT VOLUME CLASSIFICATION 1 43204 PROPANE 44. 09 10.00 4. 20 PARAFF IN 2 43212 . 31 N-BUTANE 58. 12 1.00 **PARAFFIN** 1 3 43214 ISOBUTANE 58. 12 1.00 . 31 PARAFFIN 1 43203 ETHYLENE 28. 05 1.00 . 67 OLEFIN 43502 FORMALDEHYDE 30. 03 1.00 CARBONYL . 61 6 43201 METHANE 16.04 76.00 87. 72 METHANE 43202 **ETHANE** 30. 07 10.00 6. 17 NON-REACTIVE TOTAL 100.00 99. 99 3 COMPOUNDS OF CLASSIFICATION 1 45.89 12.00 4.82 1 COMPOUNDS OF CLASSIFICATION 2 28. 05 1.00 . 67 O COMPOUNDS OF CLASSIFICATION 3 . 00 . 00 . 00 1 COMPOUNDS OF CLASSIFICATION 4 30, 03 1.00 . 61 O COMPOUNDS OF CLASSIFICATION 5 . 00 . 00 . 00 87. 72 1 COMPOUNDS OF CLASSIFICATION 6 16.04 76.00 1 COMPOUNDS OF CLASSIFICATION 7 30.07 10.00 6. 17 7 COMPOUND COMPOSITE 100.00 99, 99 18. 51

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA, AP-42 SECTION 3.3

C. APPLICABLE SCC CATEGORIES: 2-02-002-02

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DECEMBER 14, 1978 TABLE 2-02-002B

INTERNAL COMBUSTION ENGINE, INDUSTRIAL, COMPOSITE NATURAL CAS, RECIPROCATING, 30 HP, 4 CYCLE, 6 CYL., BUDA

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 030B

LINE	SARDAD	CHEMICAL	MDLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44, 09	. 40	. 15	1	PARAFFIN
2	43203	ETHYLENE	28. 05	2. 50	1.47	2	OLEFIN
3	43205	PROPYLENE	42. OB	. 10	. 03	2	OLEFIN
4	43206	ACETYLENE	26. 04	. 90	. 58	5	MISCELLANEOUS
5	43201	METHANE	16. 04	93. 50	96. 33	6	METHANE
6	43202	ETHANE	30. 07	2. 60	1.44	7	NON-REACTIVE
		TOTAL		100.00	100. 00		
	1 COMP	OUNDS OF CLASSIFICATION 1	44. 09	. 40	. 15		
	2 COMP	OUNDS OF CLASSIFICATION 2	29. 33	2. 60	1. 50		
	O COMP	GUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 5	26, 04	. 90	. 58		
	1 COMP	OUNDS OF CLASSIFICATION 6	16, 04	93. 50	96. 33		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	2. 60	1. 44		
	6 COMP	DUND COMPOSITE	16. 53	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF MULTIPLE SAMPLING TRAINS WITH STACK EXTENSION

B. REFERENCES: KVB TEST DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 2-02-002-02

REFERENCES

- 1. "Compilation of Air Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 2. Taback, H. J., et al., "Control of Hydrocarbons from Stationary Sources in the California South Coast Air Basin, Vol. I and II, KVB, Inc., June 1978.
- 3. Sonnichsen, T. W., KVB Engineer.

2.01-8 26200-795

INDUSTRIAL PROCESSES

- 3-01 CHEMICAL MANUFACTURING
- 3-02 FOOD/AGRICULTURE
- 3-03 PRIMARY METALS
- 3-05 MINERAL PRODUCTS
- 3-06 PETROLEUM INDUSTRY
- 3-30 TEXTILE MANUFACTURING
- 3-90 IN-PROCESS FUEL

3-01-015 VARNISH--BODYING OIL, GENERAL

Process Description 1,2

Varnish is a clear coating produced by chemical reactions at elevated temperatures. Varnish is generally defined as an unpigmented coating consisting of resins, oils, thinners, and dryers which forms a film by evaporation of the solvents and by oxidation and polymerization of the remaining constituents. The two basic types of varnish are spirit varnishes and oleoresinous varnishes.

The cooking process is the most important step in any varnish-making operation. Varnish is cooked in open or enclosed gas-fired kettles for periods of 4 to 16 hours at temperatures of 200 to 650 °F (93 to 340 °C) depending upon the particular batch being processed. The average plant produces 280 tons of varnish per year. For further process details, consult AP-40 (Ref. 2).

Emissions^{1,2}

The cooking and thinning operations are the major sources of hydrocarbon emissions in the varnish manufacturing process. The average batch starts to release vapors at about 350 °F (177 °C) and reaches its maximum rate of release at approximately the same time the maximum cooking temperature is reached. Obviously, the open kettle allows the vaporized material to be emitted to the atmosphere more than the closed kettle operation. The addition of solvents and thinners during the cooking process also results in the emissions of hydrocarbons to the atmosphere, especially if the thinning process is carried out in open tanks.

The quantity, composition and rate of emissions depend upon:

- . ingredients in the cooker
- . maximum kettle temperature level
- . method of blending in additives
- . degree of stirring
- . cooking time, and
- . extent of air or inert gas blowing

The average uncontrolled and controlled hydrocarbon emissions for varnish manufacturing are shown below.

TABLE OF HYDROCARBON EMISSIONS FROM VARNISH MANUFACTURING

	%	_		n Emis 280 Ton	
Type of Operation and Control	Control	lb/ton	kg/mt	lb/hr	kg/hr
Mixing and cooking, uncontrolled	0	370	185	11.8	5.35
Mixing and cooking, with incinerator	99	3.7	1.85	.12	0.05

Ref. 1

Profile 3-01-015 presents an estimation of the volatile organic species emitted from a varnish manufacturing process.

Controls

The varnish industry controls emissions mainly for economic reasons. Equipment used by the industry to reduce process emissions include scrubbers, absorbers, adsorbers, and afterburners. Sublimation and solvent reformulation are also practiced. Incineration of organic gases is one certain method for elimination of organic compounds and their associated odors. Catalytic oxidation has also been used with some success in controlling hydrocarbon emissions from varnish-making operations. Consult AP-40 for specific information on control equipment and procedures (Ref. 2).

Profile Basis 3,4

Data contained within Profile 3-01-015 were developed from engineering evaluation of survey data provided in Reference 3.

Data Qualification

The previously mentioned emission factors may be used to estimate the volatile organic emissions from a varnish-manufacturing operation. Emission factor application and limitation information is discussed in Reference 1. Profile 3-01-015 may be used to characterize the volatile organic emissions from a bodying oil type mixing and cooking operation.

POINT SOURCE EVAPORATION, CHEMICAL MANUFACTURING VARNISH MANUFACTURER, BODYING DIL, CENERAL DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE KVB PROFILE KEY 0066

LINE	SAROAD	CHEMICAL.	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43551	ACETONE	58. 08	38. 70	45. 68	4	CARBONYL
2	43552	METHYL ETHYL KETONE	72. 10	41.60	39 , 5 7	4	CARBONYL
3	43560	METHYL ISOBUTYL KETONE	100. 16	16. 70	11. 45	4	CARBONYL
4	43367	GLYCOL ETHER	62. 07	3. 00	3. 29	5	MISCELLANEOUS
		TOTAL		100. 00	99. 99		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	= ==•···	COUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	3 COMP	OUNDS OF CLASSIFICATION 4	68. 80	97. 00	96, 70		
	1 COMP	OUNDS OF CLASSIFICATION 5	62. 07	3. 00	3, 29		
	O COMP	OUNDS OF CLASSIFICATION 6	00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	4 COMP	OUND COMPOSITE	68. 58	100. 00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-01-015-01

3.01-3

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA 340/1-78-004, April 1978.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. MRC-DA-487, "Source Assessment: Prioritization of Air Pollution from Industrial Surface Coating Operations," EPA-650/2-75-019-a), PB 243-243/1BA.
- 4. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.

3-01-018 PLASTICS--PVC, GENERAL --POLYPROPYLENE, GENERAL

Process Description

The manufacture of most resins or plastics begins with the polymerization or linking of the basic compound (monomer), usually a gas or liquid, into high molecular weight noncrystalline solids. The manufacture of the basic monomer is not considered part of the plastics industry and is usually accomplished at a chemical or petroleum plant.

The manufacture of most plastics involves an enclosed reaction or polymerization step, a drying step, and a final treating and forming step. These plastics are polymerized or otherwise combined in completely enclosed stainless steel or glass-lined vessels. Treatment of the resin after polymerization varies with the proposed use. Resins for moldings are dried and crushed or ground into molding powder. Resins such as the alkyl resins that are to be used for protective coatings are normally transferred to an agitated thinning tank, where they are thinned with solvent and then stored in large steel tanks equipped with water-cooled condensers to prevent loss of solvent to the atmosphere. Still other resins are stored in latex form as they come from the reaction kettle.

Emissions¹

The major sources of air contamination in plastics manufacturing are the emissions of raw materials or monomers, emissions of solvents or other volatile liquids during the reaction, emissions of sublimed solids such as phthalic anhydride in alkyd production, and emissions of solvents during storage and handling of thinned resins. Emission factors for the manufacture of plastics as reported in AP-42 (Ref. 1) are shown below.

EMISSION FACTORS FOR PLASTICS MANUFACTURING WITHOUT CONTROLS EMISSION FACTOR RATING: E

		Parti	culate	е	Gas	ses
Type of plastic	1	b/ton	kg/1	мт	lb/ton	kg/MT
Polyvinyl chloride		35 ^a	17.	5 ^a	17 ^b	8.5 ^b
Polypropylene		3	1.	5	0.7°	0.35
General	5	10	2.5	5		

^a Usually controlled with a fabric filter efficiency of 98 to 99 percent.

b As vinyl chloride.

c As propylene.

The VOC's emitted during the storage of solvents and thinned resins can be calculated based on the information contained within API Bulletins 2517 and 2523 (Refs 2 & 3). Profiles 3-01-018 A and B characterize the VOC emissions from polyvinyl chloride and polypropylene plastic manufacturing, respectively (Refs. 4 & 5).

Controls

The control equipment used in the plastics industry is usually a basic part of the system and serves to recover a reactant or product. These controls generally include:

- . floating roof tanks for solvent and thinned resin storage
- . vapor recovery systems on volatile material
 - adsorption
 - condensers
- . storage units
- . purge lines that vent to a flare system, and
- . vapor recovery systems on vacuum exhaust lines

Profile Basis 4,5

Profiles 3-01-018 A and B are based on an engineering evaluation of the process material handled for polyvinyl chloride and polypropylene production.

Data Qualification

AP-42 should be consulted for further information on the development of the presented emission factors. Profiles 3-01-018 A and B may be used to characterize the VOC emissions from a polyvinyl chloride or polypropylene plastics manufacturing facility.

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING

PLASTICS, PVC, GENERAL DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0067

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43860	VINYL CHLORIDE TOTAL	62. 50	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION. 5	62. 50	100.00	100.00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COMPOSITE	62. 50	100.00	100, 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: PVC RESIN MANUFACTURERS

C. APPLICABLE SCC CATEGORIES: 3-01-018-01

DECEMBER 14, 1978

TABLE 3-01-018B

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING

PLASTICS, POLYPROPYLENE, GENERAL

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0068

LINE NO.	SARQAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43205	PROPYLE	NE	42. 08	100.00	100.00	2	OLEFIN	 -
			TOTAL		100. 00	100. 00			
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 2	42. 08	100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	DUND COM	POSITE	42. 08	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: POLYPROPYLENE MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 3-01-018-02

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 2. Petrochemical Evaporation Loss from Storage Tanks, American Petroleum Institute, Bulletin 2523, November 1969.
- 3. Evaporation Loss from Floating-Roof Tanks, American Petroleum Institute, Bulletin 2517, February 1962.
- 4. Sonnichsen, T. W., Engineer, KVB, Inc.
- 5. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

3.01-9 26200-795

3-01 INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING

3-01-019 PHTHALIC ANHYDRIDE

Process Description 1-3

Phthalic anhydride (PAN) is produced by the vapor phase oxidation of naphthalene or o-xylene with excess air in fixed or fluid bed catalytic converters using some form of vanadium pentoxide as a catalyst. Regardless of which chemical is used as feedstock, the processes are similar. Air and a raw material, either o-xylene or naphthalene, are fed to the reactor as a heated vaporized mixture. After the oxidation process takes place, the process vapors pass through gas coolers and condensers where the anhydride is separated from the process air stream. The condensed phthalic anhydride is melted and purified by fractionation and then stored. The process air stream is generally passed through a wet scrubber or thermal incinerator before venting to the atmosphere. The average phthalic anhydride plant produces approximately 20,700 tons of finished product yearly.

Emissions 1,2,4

The greatest contributor of VOC emissions is the reactor and condenser effluent which is vented from the condenser unit. Particulates, sulfur oxides (from o-xylene-based production), and carbon monoxide constitute the major pollutants found in the process gas stream. In addition to this source, there are four minor sources of organic emissions which include:

- 1. Feed and product storage tanks.
- 2. Process refining vents.
- 3. Flaking and bagging operation.
- 4. Loss of heat transfer medium (Dowtherm A).

The uncontrolled and controlled hydrocarbon emissions from phthalic anhydride manufacturing as reported in Reference 1 are shown below:

HYDROCARBON EMISSIONS FROM PHTHALIC ANHYDRIDE MANUFACTURING

Type of	*				4 tons/hr)
Operation and Control	Control	lbs/ton	kg/MT	lbs/hr	kg/hr
Reactor & condenser effluent, uncontrolled	0	130	65	312	142
Reactor & condenser effluent, incinerator	99	1.3	0.65	3.1	1.9
Reactor & condenser effluent, scrubber	95	6	3	14.4	6.5

Profile 3-01-019-B presents the VOC emissions measured downstream of a thermal incinerator treating the PAN reactor and condenser effluent for o-xylene feedstock based PAN Process (Ref. 4). Effluent flow rate was measured at 28,000 SCFM. PAN production was reported to be 2.1 tons/hr. Profile 3-01-019-A presents the VOC emissions from a PAN liquid waste sump vent. A flow rate of 63 SCFM was measured using an anemometer (Ref. 4).

Controls

Controls

Controls designed to reduce or eliminate VOC pollutants contained in the main process stream (reactor and condenser effluent) consist basically of:

- . Wet scrubbers
- . Incineration--direct-flame --catalytic
- . Combination of above

Further information on control equipment and its limitations can be found in AP-40 (Ref. 5).

Profile Basis 4

Profile 3-01-019-A is based on a grab sample of the gas coming off of a PAN liquid waste sump. Flow rate was determined using an anemometer.

Profile 3-01-019-B is based on a sampling train catch of a controlled (thermal incinerator), o-xylene feedstock based PAN process stream. Samples were collected by means of glass gas-collecting bottles plus NIOSH type charcoal tubes downstream of a thermal incinerator. Stack gas flow rate was determined through the use of an "S" type pitot tube and thermometer.

Data Qualification

The above mentioned emission factors may be used to estimate the total VOC emissions from either of the two basic PAN processes. Profile 3-01-019-B may be used to characterize the controlled (thermal incineration) process stream VOC emissions from a PAN process using o-xylene feedstock. Profile 3-01-019A may be used to characterize the VOC emissions from a PAN liquid waste sump.

3.01-11

DECEMBER 14, 1978 TABLE 3-01-019A

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING PHTHALIC ANHYDRIDE, PROCESS WASTE SUMP DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0071

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43122	ISOMERS OF PENTANE	72. 15	22. 60	20. 55	1	PARAFFIN	
2	43204	PROPANE	44. 09	10.80	16. 09	1	PARAFFIN	
3	43212	N-BUTANE	58. 12	33. 90	38. 28	1	PARAFFIN	
4	43231	N-HEXANE	86.17	21. 40	16. 28	1	PARAFFIN	
5	43248	CYCLOHEXANE	84. 16	11. 30	8. 80	1	PARAFFIN	
		TOTAL		100. 00	100.00			
	5 COMP	DUNDS OF CLASSIFICATION 1	65. 60	100. 00	100. 00			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 0 0	00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
•	5 COMP	OUND COMPOSITE	65. 60	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 4)

C. APPLICABLE SCC CATEGORIES: 3-01-019-03

26200-795

DECEMBER 14, 1978 TABLE 3-01-0198

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING PHTHALIC ANHYDRIDE, PROCESS INCINERATOR DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: INCINERATOR PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0276

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43204	PROPANE	44. 09	. 90	. 37	1	PARAFFIN
2	43212	N-BUTANE	58. 12	2. 20	. 70	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	. 40	. 13	1	PARAFF IN
4	43205	PROPYLENE	42. 08	3. 10	1. 36	2	OLEFIN
5	43551	ACETONE	58. 08	B. 60	2. 71	4	CARBONYL
6	43206	ACETYLENE	26. 04	4. 40	3. 10	5	MISCELLANEOUS
7	43201	METHANE	16. 04	BQ. QQ	91. 41	6	METHANE
8	43202	ETHANE	30 . 07	. 40	. 24	7	NON-REACTIVE
		TOTAL		100.00	100. 02		
	з соме	OUNDS OF CLASSIFICATION 1	53. 79	3. 50	1. 20		
		DUNDS OF CLASSIFICATION 2	42. 08	3. 10	1. 36		
		OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	58. QB	8. 60	2. 71		
		OUNDS OF CLASSIFICATION 5	26. 04	4. 40	3, 10		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. Q4	80.00	91. 41		
		OUNDS OF CLASSIFICATION 7	30. 07	. 40	. 24		
	B COMP	DUND COMPOSITE	18. 33	100.00	100. 02		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA OC-MS ANALYSIS OF SAMPLING TRAIN CATCH AT INCINERATOR OUTLET

B. REFERENCES: KVB TEST DATA (REF. 4)

C. APPLICABLE SCC CATEGORIES: 3-01-019-03

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA 340/1-78-004, April 1978,
- 2. Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Shreve, R. N., and Brink, J. A., Jr., "Chemical Process Industries," Fourth Ed., McGraw-Hill Book Co., 1977.
- 4. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 5. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.

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26200-795

3-01 INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING

3-01-020 PRINTING INKS--COOKING, GENERAL

Process Description 1-3

Printing inks consist of a fine dispersion of pigments or dyes in a vehicle which may be a drying oil with or without natural or synthetic resins and added driers or thinners.

The major classes of printing ink are:

- . letterpress
- . lithographic
- . flexographic
- . rotogravure

The major steps involved in the manufacturing of printing inks are:

- . cooking the vehicle and adding the dyes
- . grinding the pigment into the vehicle using a roller mill
- . replacing water in the wet pigment pulp by an ink vehicle (commonly known as the flushing process)

Emissions 1

Vehicle preparation by heating is the largest source of hydrocarbon emissions from ink manufacturing. At 350°F (175°C) the resins, drying oils, petroleum oils and solvents decompose, and the decomposition products are emitted from the cooking vessel. The emissions continue throughout the cooking process, reaching a maximum just after the maximum temperature has been reached.

The quantity, composition, and rate of VOC emissions from ink manufacturing depend upon the following process variables:

- . cooking time and temperature
- . ingredients
- . method of introducing additives
- . degree of stirring, and
- . extent of air and gas blowing

The estimated organic emissions from a typical ink manufacturing process are listed below.

HYDROCARBON EMISSIONS FROM PRINTING INK MANUFACTURE

		Hydrocarbon Emissions (based on 924 tons/y)					
Type of Operation and Control	Control	lb/ton	kg/MT	lb/hr	kg/hr		
General Vehicle Cooking, uncontrolled General Vehicle Cooking with Scrubber	0	120	60	12.0	5.4		
& Afterburner	90	12	5.4	1.2	0.54		
Oil Vehicle Cooking, uncontrolled Oil Vehicle Cooking with Scrubber	0	40	20	4.0	1.8		
& Afterburner	90	4	1.8	0.4	0.18		
Oleoresinous Vehicle Cooking, uncontrolled Oleoresinous Vehicle Cooking with Scrubber	0	150	75	15.0	6.8		
& Afterburner	90	15	6.8	1.5	0.68		
Cooking of Alkyds, uncontrolled Cooking of Alkyds with Scrubber	0	160	80	16.0	7.3		
& Afterburner	90	16	7.3	1.6	0.73		

Source: Ref. 1

Profile 3-01-020 presents the estimated VOC emissions applicable to a typical printing ink manufacturer (Ref. 3).

Control^{1,4}

Volatile hydrocarbon emissions from vehicle cooking can generally be reduced by 90% through the use of scrubbers or condensers followed by an afterburner. AP-40 offers a thorough explanation of the advantages and limitations of each type of control device (Ref. 4).

Profile Basis 5,6

Profile 3-01-020 is based on an engineering evaluation of an organic emissions control strategy report by Trijonas (Ref. 6).

Data Qualification

Reference 1 should be consulted for further information on the development and limitations of the presented hydrocarbon emission factors.

Profile 3-01-020 may be used to characterize the VOC emissions from a typical ink manufacturing process.

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING PRINTING INK COOKING, GENERAL

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0072

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	25. 50	15. 80	1	PARAFF IN
2	43248	CYCLOHEXANE	84. 16	5. 00	4. 16	1	PARAFFIN
3	45106	ISOMERS OF DIETHYLBENZENE	134. 21	3. 50	1. 83	3	ARUMATIC
4	45203	ETHYLBENZENE	106. 16	3. 00	1. 97	3	AROMATIC
5	43551	ACETONE	58. 08	5. 50	6 . 70	4	CARBONYL
6	43552	METHYL ETHYL KETONE	72. 10	5. 00	4. 87	4	CARBONYL
7	43301	METHYL ALCOHOL	32. 04	5. 00	11.00	5	MISCELLANEOUS
8	43302	ETHYL ALCOHOL	46. 07	2. 50	3. 81	5	MISCELLANEOUS
9	43304	ISOPROPYL ALCOHOL	60. 0 9	38.00	44. 57	5	MISCELLANEOUS
10	43305	N-BUTYL ALCOHOL	74. 12	3. 00	2. 89	5	MISCELLANEOUS
11	43435	N-BUTYL ACETATE	116. 16	4. 00	2. 40	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	2 COMP	OUNDS OF CLASSIFICATION 1	107. 78	30. 50	19. 96		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	2 COMP	OUNDS OF CLASSIFICATION 3	119. 67	6. 50	3. 80		
	2 COMP	OUNDS OF CLASSIFICATION 4	6 3. 98	10. 50	11. 57		
	5 COMP	OUNDS OF CLASSIFICATION 5	57. 20	52. 50	64. 67		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
•	11 COMP	DUND COMPOSITE	70. 45	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: PRINTING INK MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 3-01-020-01,-99

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- 2. Shreve, R. N., and Brink, J. A., Jr., "Chemical Process Industries," Fourth Ed., McGraw-Hill Book Co., 1977.
- 3. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 4. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 5. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 6. Trijonas, J. C. and Arledge, K. W., "Impact of Reactivity Criteria on Organics Emission Control Strategies in the Metropolitan Los Angeles AQCR," EPA-600/3-76-091, August 1976.

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3-01

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING

3-01-026 SYNTHETIC RUBBER--AUTO TIRE PRODUCTION, GENERAL

Process Description 1

The automobile tire is initially built up as a cylinder on a collapsible, round rotating drum. Layers of cords embedded in a proper compound are applied, one layer tying the beads together in one direction and another layer in another direction. The beads--wire cables embedded in a tough, hard rubber-are "stitched" to the tire by folding the end of the cord fabric over. Last, the tread formed by extrusion is laid on, and the ends lapped together. The tire-building drum is collapsed, and the cylindrical tire removed and placed on a press. Here an inflatable rubber bag, usually made from butyl rubber and on a movable stem, is blown up inside the tire; the press mold is simultaneously closed, giving the tire a doughnut shape. Heat is applied through the mold and by steam inside the bag. Excess rubber escapes through weep holes and, after a timed cure at preselected temperature, the tire is formed. Present-day tires may be tubeless, with the air-retentive layer built in, or an inner tube extruded from butyl rubber may be used. Butyl rubber, although a very "dead" rubber, has outstanding resistance to passage of air.

Radial tires require a slightly different method of assembly from the traditional bias cord tires. Belts of rayon, polyester, fiberglass and steel are all being used.

Emissions²

Emissions of VOC's to the atmosphere can occur during the manufacturing of automotive tires where the evaporation of a solvent, curing of an adhesive, tire painting, tire molding, or the thermal breakdown of rubber occurs.

Quantitative and qualitative tests (Ref. 2) were performed at a typical automotive tire manufacturing plant. The tire processing devices and their associated emission rates are presented below. Cas samples were collected from the exhaust vents of specified processing equipment downstream of any control device.

Device	Control		Reported Solvent	Emissions			
Description		and/or Adhesive	lb/hr	g/scf	Other		
Tuber Cement Unit	None	A	Std 200 Thinner + Shell Sol M-48	51.6	4.59	0.10 lb/ tire (a)	
White Sidewall Tuber Unit	None	В	n-hexane	62.2	7.72	(b)	
Bead Dip Tank	None	С	Std 200 Thinner + Shell Sol M-48	49.3	5.98	(c)	

^(a)Average size tire

Controls²

Control of gaseous hydrocarbons resulting from the manufacture of automobile tires generally take the form of charcoal adsorbers, direct-flame and catalytic incineration, chemical absorbers, vapor condensation, process and material changes, and improved maintenance. For a thorough discussion of the above-mentioned control devices, refer to References 2 and 3.

Profile Basis²

Gaseous hydrocarbon samples were taken in the exhaust ducts downstream of any control equipment. Vapor samples were collected using a gas
collecting bottle followed by two NIOSH type charcoal tubes. Identification
and quantification of the organic compound species was made using a gas chromatograph (Poropak column). The results are presented in profiles 3-01-026-A,
B and C. Exhaust mass flow rate determinations were made using an "S"
type pitot tube; temperature measurements were by thermometer.

Data Qualification

The following profiles only apply to the specified tire manufacturing equipment employing the specified solvents and/or adhesive. Any alteration as to process rate, emissions control, and process material such as adhesive composition, would invalidate the presented emission factors and profiles.

⁽b) Adhesive is applied to a variety of tire components in ribbon form; therefore emission rates cannot be related to the quantity of tires processed.

⁽c) Hydrocarbons continuously evaporating from tank; not valid to relate to tires or beads processed.

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING, SYNTHETIC RUBBER AUTOMOTIVE TIRES, TUBER ADHESIVE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0272

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION	
1	43106	ISOMERS OF HEPTANE	100. 20	4. 70	4. 34	1 PARAFFIN	
2	43107	ISOMERS OF OCTANE	114. 23	80	. 65	1 PARAFFIN	
3	43115	C-7 CYCLOPARAFFINS	9 8. 19	43. 20	40. 67	1 PARAFFIN	
4	43116	C-8 CYCLOPARAFFINS	112. 23	6. 00	4. 99	1 PARAFFIN	
5	43231	N-HEXANE	86. 17	5. BO	6. 19	1 PARAFFIN	
6	43232	N-HEPTANE	100. 20	1. 90	1.76	1 PARAFFIN	
7	43233	NDCTANE	114. 23	. 40	. 37	1 PARAFFIN	
8	43248	CYCLOHEXANE	84. 16	16. 30	17. 93	1 PARAFFIN	
9	43262	METHYLCYCLOPENTANE	84. 16	17. 70	19. 41	1 PARAFFIN	
10	45202	TOLVENE	92 . 13	. 40	. 37	3 AROMATIC	
11	45201	BENZENE	78. 11	2. 80	3. 33	7 NON-REACTIVE	
		TOTAL		100. 00	100. 01		
	9 COMP	OUNDS OF CLASSIFICATION 1	93. 00	96. BO	96. 3 1		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	92. 13	. 40	. 37		
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 0 0		
		OUNDS OF CLASSIFICATION 5	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 7	78. 11	2, 80	3. 33		
•	11 COMP	DUND COMPOSITE	92. 50	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 3-01-026-20

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING, SYNTHETIC RUBBER AUTOMOBILE TIRES, TUBER ADHESIVE, WHITE SIDEWALL

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0273

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	24. 20	23, 81	1 PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	. 20	. 17	1 PARAFFIN
3	43231	N-HEXANE	86. 17	33. 60	33. 05	1 PARAFFIN
4	43242	CYCLOPENTANE	70. 14	. 20	. 25	1 PARAFFIN
5	43248	CYCLOHEXANE	B4. 16	. 40	. 42	1 PARAFFIN
6	43262	METHYLCYCLOPENTANE	84. 16	33. 70	33. 90	1 PARAFFIN
7	45201	BENZENE	78. 11	7. 70	8. 39	7 NON-REACTIVE
		TOTAL		100. 00	99. 99	
	6 COMP	OUNDS OF CLASSIFICATION 1	85. 40	92. 30	91. 60	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 7	78. 11	7. 70	8. 39	
•	7 COMP	DUND COMPOSITE	84. 79	100.00	99. 99	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 3-01-026-20

DECEMBER 14, 1978 TABLE 3-01-026C INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING, SYNTHETIC RUBBER

AUTOMOBILE TIRE PRODUCTION

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0274

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43115	C-7 CYCLOPARAFFINS	78. 19	99. 84	99. 71	1	PARAFFIN	
2	43116	C-8 CYCLOPARAFFINS	112. 23	. 08	. 10	1	PARAFFIN	
3	43248	CYCLOHEXANE	84. 16	. 04	. 10	1	PARAFFIN	
4	43262	METHYLCYCLOPENTANE	84. 16	. 04	. 10	1	PARAFFIN	
		TOTAL		100. 00	100. 01			
	4 COMP	DUNDS OF CLASSIFICATION 1	98. 18	100. 00	100. 01			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 4	, 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 6	. 0 0	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	4 COMP	DUND COMPOSITE	98 . 18	100.00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 3-01-026-20

REFERENCES

- 1. Shreve, R. N. and Brink, J. A., "Chemical Process Industries," Fourth Edition, McGraw-Hill Book Co., 1977.
- Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.

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3-01 INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING

3-01-125 ETHYLENE DICHLORIDE

Process Description 1,2

The preparation of ethylene dichloride almost always centers around large plants using a balanced combination of two processes: direct chlorination of ethylene with chlorine; oxychlorination in which ethylene, hydrogen chloride, and oxygen react to form the same product. The annual production of a typical plant is 208,000 tons per year.

In the direct chlorination process, chlorine and ethyelene are fed into a reactor where the reaction takes place under 100-120°F (38-49°C) and 10-20 PSIG. Crude ethylene dichloride emerges from the reactor in liquid form and is purified by passage through a series of condensers, separators, and wash towers.

In the oxychlorination process, ethylene, oxygen, and hydrochloric acid are fed to a fixed or fluid bed reactor where crude ethylene dichloride is absorbed from the gas stream and the non-condensible gases are vented to the atmosphere. The crude product is refined in a finishing system.

Emissions 1

The quantity of hydrocarbons released to the atmosphere is considerably lower for the direct chlorination process than for the oxychlorination process. The major source of emissions from the direct chlorination processes is the gas vented from the scrubbing column. This gas stream contains small amounts of ethylene, ethylene dichloride, vinyl chloride, and impurities in the feed. The vent gas from the oxychlorination process is also a key source of atmospheric emissions. In both cases, emission rates may vary due to significant differences in product recovery systems. Ethyelene dichloride may also be

released by storage tanks. Controlled and uncontrolled hydrocarbon emissions from a typical ethylene dichloride plant are presented below.

HYDROCARBON EMISSIONS FROM ETHYLENE DICHLORIDE MANUFACTURE

			n Emissions 24 tons/hr)		
	% Control	lbs/ Ton of Product	kg/ M Ton of Product	lbs/hr	kg/hr
Direct Chlorination with	0	5-8	2.5-4	119-190	60-95
Incineration of Vent Gases	80	1-1.6	0.5-0.8	24-38	12-19
	90	0.5-0.8	0.3-0.4	12-19	6-9.5
	99	0.05-0.08	0.03-0.04	1.2-1.9	0.6-1
Oxychlorination with	0	50-140	25-70	1190-3330	600-1670
Incineration of Vent Gases	80	10-28	5-14	240-660	11 9- 330
	90	5-14	2.5-7	119-333	60-167
	99	0.05-1.4	0.25-0.7	12-33	6-16.6
Storage	0	1.2	0.6	28.6	14.3

Source: Ref. 1

Profile 3-01-125 presents the VOC emissions from an ethylene dichloride storage facility (Ref. 3).

Controls 1,3

No emissions control for the ethylene dichloride industry has yet been demonstrated. The producers of this chemical use various methods of product recovery and the emissions from each process vary. Possible hydrocarbon emission control devices would include thermal or catalytic incineration, having control efficiencies approaching 100 percent. The preceding table presents emission rates that could be attained with incineration of vent gases. AP-40 presents a thorough explanation of the advantages and limitations of each control method (Ref. 3).

Profile Basis

Profile 3-01-125 is based on an inspection of the process material and fluid stored.

Data Qualification

Reference 1 should be consulted for information on the applications and limitations of the presented emission factors.

Profile 3-01-125 may be used to characterize the VOC emissions from an ethylene dichloride storage facility. This profile may also be applied to the VOC emissions estimate for the entire plant taking into consideration that ethylene dichloride would account for the majority of VOC pollutants emitted during processing.

DECEMBER 14, 1978 TABLE 3-01-125

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING ETHYLENE DICHLORIDE, DIRECT CHLORINATION

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0078

ND.	SARDAD		CHEMICAL NAME	MOLECULA WEIGHT	R PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43815	ETHYLENE	DICHLORIDE TOTAL	99. 00	100. 00 100. 00	100. 00 100. 00	7	NON-REACTIVE	
	O COMP	OUNDS OF	CLASSIFICATION	1 .00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	2 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	3 .00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	4 . 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	5 . 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION	6 . 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION	7 99 . 00	100. 00	100. 00			
	1 COMP	DUND COMP	DSITE	99. 00	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF CHEMICAL FORMULATION

B. REFERENCES: ETHYLENE DICHLORIDE MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 3-01-125-02, -99

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA 340/1-78-004, April 1978.
- 2. Shreve, R. N., and Brink, J. A., Jr., "Chemical Process Industries," Fourth Ed., McGraw-Hill Book Co., 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 4. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

3.01-29 26200-795

3-01 INDUSTRIAL PROCESS--CHEMICAL MANUFACTURING

3-01-999 OTHER/NOT CLASSIFIED--WASTE GAS FLARES

Process Description 1,2

Occasionally modern chemical processes produce excess quantities of waste gases. Every chemical plant must be equipped to handle excess gas production which is produced intermittently, and large volumes of hydrocarbon vapors produced very rapidly from process units during emergencies. A number of devices are utilized to recover these gases but facilities for the ultimate disposal of excess vapors are usually necessary. This is generally accomplished by combustion in waste gas flares although small amounts may be vented to the fireboxes of heaters or boilers.

The objective of combustion in a waste gas flare is the oxidation of the hydrocarbon vapors to carbon dioxide and water without the production of smoke and objectionable odors.

The three common types of waste gas flares in use are: elevated flares, ground-level flares and burning pits. The two major types of elevated smokeless, waste gas flares in current use are the air-aspirating venturi flare and the steam-injected flare.

1,2 Emissions

The smokeless flare when properly designed attain a high degree of combustion efficiency. Actual field testing of flares is generally not feasible because of safety considerations, the erratic nature of the gas flow to the flare and the inaccessibility of most flare tips.

A hydrocarbon emission factor of 5 $lbs/10^3$ bbl refinery capacity for refinery flares is reported in AP-42 (Ref. 1). Data on chemical industry waste gas flares is not presently available.

Profile 3-01-999 presents the estimated VOC emissions from a chemical process waste gas flare (Ref. 2 & 3).

Controls²

Based on the design of smokeless flares essentially complete combustion does occur. Control therefore, would amount to the conversion of a smoking type flare to a smokeless type flare air-aspirating venturi flare or steaminjected flare followed by proper maintenance.

AP-40 offers a thorough explanation of the design, application and limitations of various waste gas flare systems (Ref. 2). Another good reference is the APRI Manual on Disposal of Refinery Wastes (Ref. 5).

Profile Basis^{3,4}

Profile 3-01-999 was based on a survey and engineering evaluation of pertinent literature. This profile represents a composite of many chemical industry waste-gas flares.

Data Qualification

Profile 3-01-999 may be used with discretion to characterize the volatile organic emissions from a chemical industry waste gas flare.

In lieu of a more applicable emission factor, the AP-42 value of $5~{\rm lbs/10}^3~{\rm bbl}$ plant capacity refinery flare emission factor may also be applied with discretion.

DECEMBER 14, 1978

TABLE 3-01-999

INDUSTRIAL PROCESS, CHEMICAL MANUFACTURING OTHER, FLARES DATA CONFIDENCE LEVEL. IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0079

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NŪ.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43248	CYCLOHEXANE	84. 16	1.80	. 96	1 PARAFFIN
2	43120	ISOMERS OF BUTENE	56. 10	8. 90	7. 28	2 OLEFIN
^C 3	43203	ETHYLENE	28. 05	21.60	35, 27	2 OLEFIN
4	43205	PROPYLENE	42.08	9. 00	9. BO	2 OLEFIN
5	43213	BUTENE	56. 10	4. 10	3. 34	2 OLEFIN
6	45102	ISOMERS OF XYLENE	106. 16	1.30	. 55	3 AROMATIC
7	45202	TOLUENE	92 . 13	4. 10	2. 06	3 AROMATIC
8	45220	STYRENE	104. 14	3. 40	1. 51	3 AROMATIC
9	43502	FORMALDEHYDE	30. 03	1.70	2. 61	4 CARBONYL
10	43503	ACETALDEHYDE	44. 05	. 70	. 73	4 CARBONYL
11	43510	BUTYRALDEHYDE	72. 12	. BO	. 50	4 CARBONYL
12	43206	ACETYLENE	26. 04	1.00	1. 74	5 MISCELLANEOUS
13	43301	METHYL ALCOHOL	32. 04	5. 40	7. 74	5 MISCELLANEOUS
14	43302	ETHYL ALCOHOL	46. 07	1.40	1. 37	5 MISCELLANEOUS
15	43304	ISOPROPYL ALCOHOL	60. 09	2. 50	1. 92	5 MISCELLANEOUS
16	43305	N-BUTYL ALCOHOL	74. 12	. 50	32	5 MISCELLANEOUS
17	43438	ETHYL ACRYLATE	100. 11	. 80	. 37	5 MISCELLANEOUS
16	43601	ETHYLENE OXIDE	44. 05	4. 60	4. 76	5 MISCELLANEOUS
19	43602	PROPYLENE OXIDE	58. 08	1. 40	1.10	5 MISCELLANEOUS
20	43704	ACRYLONITRILE	55. 00	3.00	2. 52	5 MISCELLANEOUS
21	43812	ETHYL CHLORIDE	64. 52	7. 20	5. 13	5 MISCELLANEOUS
22	43860	VINYL CHLORIDE	62. 50	. 60	46	5 MISCELLANEDUS
23	45300	PHENOLS	94. 11	2. 00	. 96	5 MISCELLANEOUS
24	45401	XYLENE BASE ACIDS	230.00	1.30	. 27	5 MISCELLANEOUS
25	43801	METHYL CHLORIDE	50. 49	. 9 0	. 82	7 NON-REACTIVE
26	45201	BENZENE	78. 11	10.00	5.86	7 NON-REACTIVE
	_	TOTAL		100. 00	99. 95	
	1 COMP	DUNDS OF CLASSIFICATION 1	84. 16	1. 80	. 9 6	
	4 COMP	OUNDS OF CLASSIFICATION 2	35. 87	43. 60	55. 69	
	3 COMP	DUNDS OF CLASSIFICATION 3	98. 40	8. 80	4, 12	
	3 COMP	DUNDS OF CLASSIFICATION 4	38. 18	3. 20	3. 84	
	13 COMP	OUNDS OF CLASSIFICATION 5	50. B 3	31.70	28. 66	
		DUNDS OF CLASSIFICATION 6	. 00	00	. 00	
		DUNDS OF CLASSIFICATION 7	74. 72	10. 90	6. 68	
	26 COMP	DUND COMPOSITE	45. 89	100.00	99. 95	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

3.01-32

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

C. APPLICABLE SCC CATEGORIES: 3-01-999-99

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A., (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. Sonnichsen, T. W., KVB Engineer.
- 4. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 5. "Manual on Disposal of Refinery Wastes," Vol. II, American Petroleum Institute, Division of Refinery, Washington, D.C., 5th Ed., 1957.

3.01-33 26200-795

3-02 INDUSTRIAL PROCESS, FOOD/AGRICULTURE

3-02-009 FERMENTATION--BEER

Process Description 1

The manufacture of beer from grain is a multiple-step process. From the time the grain is harvested until the beer manufacturing process is complete, the following events take place at the brewery:

- 1. Malting of the barley (germinating barley by soaking in water followed by kiln drying),
- 2. addition of corn, grit, rice,
- conversion of starch to maltose by enzymatic processes,
- separation of wort (liquid to be fermented) from grain,
- hopping (addition of cones of the hop plant) and boiling of wort,
- 6. cooling of wort,
- 7. addition of yeast,
- 8. fermentation
- 9. removal of settled yeast,
- 10. filtration,
- 11. carbonation,
- 12. aging, and
- 13. packaging.

Emissions^{1,2}

Emissions from fermentation processes are nearly all gases and primarily consist of carbon dioxide, hydrogen, oxygen, and water vapor; none of which present an air pollution problem. Gaseous hydrocarbons are also emitted from the drying of spent grains and yeast in beer.

The hydrocrabon emission rate may be approximated by assuming that one percent by weight of spent grain is emitted as hydrocarbon (Ref. 1).

Assuming the grain loses 20 percent of its weight during the manufacturing process, for every pound of spent grain, 1.25 pounds of raw grain are required. Therefore, each 1.25 pounds of input discharges 0.01 pounds of hydrocarbons. Based on the above, hydrocarbon emissions from beer processing are detailed below:

TABLE OF VOC EMISSIONS FROM BEER PROCESSING

		H	ydrocarb	on Emissi	ons
	8			(16.1 tons/hour 1b/hr kg/hr 42.3 19.2	ns/hour)
Type of Operation and Control	Control	lb/ton	kg/ton	lb/hr	tons/hour) kg/hr
Beer processing, uncontrolled	0	2.63	1.32	42.3	19.2
Beer processing, incineration	99	0.0263	0.0132	0.42	0.19

(Reference 1)

Control

The major VOC species emitted during processing is ethyl alcohol which can be effectively controlled by incineration or absorption.

There is a limited quantity of ethyl alcohol from a typical processing plant. Incineration is accomplished by introducing ethyl alcohol fumes into a boiler air supply or by passing the fumes through a direct-flame after-burner. Absorption is accomplished by dissolving ethyl alcohol vapors in a selective liquid solvent. Consult AP-40 for further information on either control method (Ref. 3).

Profile Basis

The basis for Profile 3-02-009 was a survey of pertinent literature (Refs. 1, 2 and 4).

Data Qualification

The following profile may be applied to typical beer processing operations wherever grain fermentation occurs.

DECEMBER 14, 1978 TABLE 3-02-009

INDUSTRIAL PROCESS, FOOD/AGRICULTURE

FERMENTATION, BEER

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0211

.INE NO.	SARUAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHÉMICAL ASSIFICATION
1	43302	ETHYL A	NLCOHOL TOTAL	46. 07	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
						200. 00		
	O COMP	CUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 2	. 0 0	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 5	46. 07	100.00	100.00		
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	DUND COM	1POSITE	46. 07	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 1, 2 AND 4)

C. APPLICABLE SCC CATEGORIES: 3-02-009-03

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA 340/1-78-004, April 1978.
- 2. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environ-mental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 4. Considine, D. M., (ed.), "Chemical and Process Technology Encyclopedia," McGraw-Hill Book Co., 1974.

Introduction

Primary metal operations refers to the production of metals from their ores. The primary metals industries include primary aluminum production, copper smelters, lead smelters, zinc smelters, iron and steel mills, ferroalloy production, and metallurgical coke manufacturers.

Metallurgical processes are some of the largest stationary point sources of pollution. However, compared to carbon monoxide, particulate and sulfur emissions, these sources generally emit small concentrations of VOC's. Coke manufacturing is an exception to this. VOC's are emitted from stacks as exhaust gas, from fuel combustion and as fugitive emissions, and from openings or leaks in material processing retorts.

3-03-003 COKE METALLURGICAL BYPRODUCTS--GENERAL

Process Description 1

Coking is the process of heating coal in an atmosphere of low oxygen content, i.e., destructive distillation. During this process, organic compounds in the coal break down to yield gases and a residue of relatively nonvolatile nature. Two processes are used for the manufacture of metallurgical coke, the by-product process and the beehive process. The by-product process accounts for more than 98% of the coke produced.

The by-product process is oriented toward the recovery of the gases produced during the coking cycle, whereas the volatile matter is vented to the atmosphere in the beehive process. The rectangular by-product ovens are grouped together in a series, alternately interspersed with heating flues, called a coke battery. Coal is charged to the ovens through ports in the top, which are then sealed. Heat is supplied to the ovens by burning some of the coke gas produced. Coking is largely accomplished at temperatures of 2000° to 2100° F (1100° to 1150° C) for a period of about 16 to 20 hours. At the end of the coking period, the coke is pushed from the oven by a ram and quenched with water.

Emissions 1

Particulates, hydrocarbons, carbon monoxide, and other emissions originate from the following by-product coking operations: 1) charging of the coal into the incandescent ovens, 2) oven leakage during the coking period, 3) pushing the coke out of the ovens, and 4) quenching the hot coke. Gaseous emissions from the by-product ovens are drawn off to a collecting main and are subjected to various operations for separating ammonia, coke-oven gas, tar, phenol, light oil (benzene, toluene, xylene), and pyridine. unit operations are potential sources of hydrocarbon emissions.

Oven-charging operations and leakage around poorly sealed coke-oven doors and lids are major sources of gaseous emissions from by-product ovens. The hydrocarbon emission factors for coking operations are summarized in Section 7.2 "Metallurgical Coke Manufacturing" of AP-42. (Ref. 2)

Controls³

Coke oven gas is produced as a by-product or co-product and is drawn off, collected and treated. Control schemes involve hoods and vents to collect fugitive emissions for adsorption or incineration depending on concentration. Coke oven gas is largely used for fuel in the many steel plant operations.

References 4 and 5 contain a description of the coking process and details of emission sources and control measures.

Profile Basis⁶

Field test samples were taken from two different coke oven stacks and a main header using gas collection bottles for grab samples. Gas analysis was by GC-MS technique.

Data Qualification

The following profile may be correctly applied to all by-product or co-product coke manufacturing processes to determine hydrocarbon emissions that may result from oven charging, oven pushing, quenching, unloading, underfiring or leakage.

3.03 - 3

INDUSTRIAL PROCESS, PRIMARY METALS, METALLURGICAL COKE MFG. BY PRODUCT PROCESS, COKE OVEN STACK GAS DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0011

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43204	PROPANE	44. 09	. 50	. 25	1	PARAFFIN
2	43203	ETHYLENE	28. 05	27. 70	22. 56	2	OLEFIN
3	43205	PROPYLENE	42. 08	1. 90	1.03	2	OLEFIN
4	43213	BUTENE	56. 10	. 10	. 05	2	OLEFIN
5	43218	1,3-BUTADIENE	54. 09	. 50	. 21	2	DLEF IN
6	45202	TOLUENE	92. 13	. 70	. 18	3	AROMATIC
7	43206	ACETYLENE	26. 04	1. 20	1. 05	5	HISCELLANEOUS
8	43201	METHANE	16. 04	45. 30	64, 47	6	METHANE
9	43202	ETHANE	30. 07	8. 00	6, 07	7	NON-REACTIVE
10	45201	BENZENE	78. 11	14. 10	4. 13	7	NON-REACTIVE
		TOTAL		100. 00	100.00		
	1 COMP	DUNDS OF CLASSIFICATION 1	44. 09	. 50	. 25		
	4 COMP	DUNDS OF CLASSIFICATION 2	28. 94	30, 20	23, 85		
		DUNDS OF CLASSIFICATION 3	72. 13	. 70	. 18		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 5	26. 04	1. 20	1. 05		
		DUNDS OF CLASSIFICATION 6	16. 04	45. 30	64. 47		
		DUNDS OF CLASSIFICATION 7	49. 52	22. 10	10. 20		
	10 COMP	DUND COMPOSITE	22. 84	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA OC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 6), AP-42 SECTION 7.2

C. APPLICABLE SCC CATEGORIES: 3-03-003-01, -02, -03, -04, -05, -06

REFERENCES

- 1. "Air Pollutant Emission Factors," Final Report. Resources Research, Inc., Reston, VA. Prepared for National Pollution Control Administration, Durham, NC, under Contract No. CPA-22-69-119. April 1970.
- 2. "Compilation of Air Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Shreve, R. N. and Brask, J. A., Jr., "Chemical Process Industries," Fourth Edition, McGraw-Hill Book Co., 1977.
- 4. California Air Resources Board, "Coke Oven Emissions, Miscellaneous Emissions, and their Control at Kaiser Steel Corporation's Fontana Steel Making Facility," Report L&E-76-11, November 1976.
- 5. Barnes, T. M., et al., "Evaluation of Process Alternatives to Improve Control of Air Pollution from Production of Coke," Battelle Memorial Institute, Columbus, OH, PB 189 266, January 1970.
- 6. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I & II, KVB, Inc., Tustin, CA, June 1978.

3.03-5 26200-795

3-03

INDUSTRIAL PROCESS, PRIMARY METALS

3-03-008 IRON PRODUCTION-BLAST FURNACE
--IRON SINTERING

Process Description 1,2

Blast furnaces are tall refractory-lined furnaces using the countercurrent flow principle to achieve high efficiency. The raw materials which include iron ore as sinter or pellets, are charged at the top along with coke as a reducing and thermal agent and limestone for fluxing and gangue material. A hot-air blast introduced at the bottom end burns the coke to heat, reduce, and melt the charge as it descends. The liquid iron and slag collect on the furnace hearth and are tapped at regular intervals through separate tap holes.

The waste-blast furnace gas from the combustion of coke, which contains about 28% carbon monoxide and has a heating value of about 90 Btu/ft³, is collected from the top of the furnace by a downcomer pipe, cleaned to remove the dust particles and used as a fuel. This waste gas is used with coke oven gas or oil for heating hot-blast stoves. The hot-blast stoves are vented to a common exhaust stack.

Emissions^{3,4}

Compared to particulate and carbon monoxide, the VOC emission from the blast-furnace hot-blast stoves is very small (0.7 lb/hr of total hydrocarbon for a stack gas flow of 55,000 SCFM at 550° F).

Hydrocarbon emissions from iron sintering, however, are considerably more, 15 lb/hr of total hydrocarbons for a stack gas flow of 152,400 SCFM at 320°F.

Emission factors for particulates and carbon monoxide are contained in Section 7.5 "Iron and Steel Mills," of AP-42. Section 7.5 does not have emissions factors for VOC's.

Controls 3

Because of the low VOC concentration in the exhaust gases of blast furnaces, control equipment is not practical.

Profile Basis³

Profiles in Tables 3-03-008 A and B were based on one grab sample each taken from the exhaust of a blast-furnace and sintering plant respectively. Both samples were analyzed by GC-MS analysis (Ref. 3).

Data Qualification

The following profiles may be used for iron production processes. One profile is for blast furnace ore charging and agglomerate charging, and the other profile is for iron sintering operations.

3.03-7 26200-795

INDUSTRIAL PROCESS, PRIMARY METALS, IRON PRODUCTION BLAST FURNACE ORE CHARGING AND AGGLOMERATE CHARGING DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0012

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43822	TRIMETHYLFLUOROSILANE	92. 00	84, 20	48.16		MISCELLANEOUS
2	43201	METHANE	16. 04	15. 80	51.84	6	METHANE
		TOTAL		100. 00	100, 00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	92.00	84. 20	48. 16		
	1 COMP	OUNDS OF CLASSIFICATION 6	16.04	15. BO	51.84		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	2 COMP	OUND COMPOSITE	52. 62	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-03-008-01, -02

DECEMBER 14, 1978 TABLE 3-03-008B

INDUSTRIAL PROCESS, PRIMARY METALS

IRON SINTERING

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0013

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43203	ETHYLENE	28. 05	5. 90	3. 81	2	OLEFIN
2	43205	PROPYLENE	42. 08	3. 00	1. 29	2	OLEFIN
3	43206	ACETYLENE	26. 04	14. 80	10. 29	5	MISCELLANEOUS
4	43201	METHANE	16.04	73. 30	82. 80	6	METHANE
5	43202	ETHANE	30. 07	3. 00	1. 81	7	NON-REACTIVE
		TOTAL		100.00	100.00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 0 0	. 00	. 00		
		OUNDS OF CLASSIFICATION 2	31, 60	8. 90	5. 10		
		OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	26. 04	14. 80	10. 29		
		OUNDS OF CLASSIFICATION 6	16. 04	73. 30	82. 80		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	3. 00	1. 81		
	5 COMP	OUND COMPOSITE	18. 12	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLES

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-03-008-03

REFERENCES

- 1. Considine, D. M., editor, "Chemical and Process Technology Encyclopedia," McGraw-Hill Book Co., 1974.
- 2. "Ferroalloys, Steels, All-purpose Additives," The Magazine of Metals Producing, February 1967.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I & II, KVB, Inc., Tustin, CA, June 1978.
- 4. "Compilation of Air Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.

26200-795

3-03 INDUSTRIAL PROCESS, PRIMARY METALS

3-03-009 STEEL PRODUCTION--OPEN HEARTH FURNACE
--BASIC OXYGEN FURNACE

Process Description 1,2

Steel can be made by using: 1) liquid iron as the main constituent of the charge, 2) steel scrap as the main constituent of the charge, 3) pre-reduced sponge iron, and 4) a mixture of liquid iron, scrap and sponge iron.

A number of processes are available using external fuel in the form of gas, oil, electric power or chemical heat produced by exothermic reactions of metalloids (e.g., C, Si, Mn and P) contained in the charge material to make steel.

Open-hearth furnaces--In the open hearth process, a mixture of scrap iron, steel and pig iron is melted in a shallow rectangular basin, or "Hearth" for which various liquid gaseous fuels provide the heat. The basic open hearth can handle almost any type of metallic charge. The Ajax process, which uses modified tilting-type open-hearth furnaces, can refine 100% hot metal charges with oxygen.

Basic Oxygen Furnaces (BOF) -- The basic oxygen process, also called the Linz-Donawitz (LD) process is employed to produce steel from a furnace charge of approximately 70% molten blast-furnace metal and 30% scrap metal by use of a stream of commercially pure oxygen to oxidize the impurities, principally carbon and silicon.

Emissions³

Air contaminants are emitted from both the open-hearth furnace and the basic oxygen furnace throughout the process, or heat. Hydrocarbon emissions result from steel scrap, which contains grease, oil or other combustible material and from the furnace fuel.

Total hydrocarbon emissions for an open-hearth furnace at the precipitator outlet were 3.02 lbs/hour for a gas flow rate of 33,000 SCFM at 490°F.

On a basic oxygen furnace, the total hydrocarbon emissions at the precipitator outlet were 6.42 lbs/hour for a gas flow rate of 240,900 SCFM at 200 - 275°F.

Emission factors for particulates, carbon monoxide and fluorides are contained in Section 7.5 "Iron and Steel Mills," of AP-42. Section 7.5 does not have emission factors for hydrocarbons.

Controls 3

Electrostatic precipitators are installed on both open-hearth and basic oxygen furnaces for particulate control purposes. For the open-hearth furnace, the precipitator control efficiency for hydrocarbons was 25%. Control efficiency for hydrocarbons was not determined for the BOF precipitator installation.

Profile Basis³

For the open-hearth furnace, one grab sample by gas collection bottle was made of both the precipitator inlet and outlet. On the BOF, only one grab sample was made at the precipitator outlet.

Plant operation for both furnaces was 24 hours/day, 360 days/year.

Data Qualification

The following profiles may be used for steel making open-hearth and BOF processes. There are two profiles for the open-hearth furnace, one without control and one with control.

INDUSTRIAL PROCESS, PRIMARY METALS, STEEL PRODUCTION OPEN HEARTH WITH OXYGEN LANCE DATA CONFIDENCE LEVEL: 111

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0306

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43231	N-HEXANE	B6. 17	1. 67	1. 74	1	PARAFFIN
2	43232	N-HEPTANE	100. 20	6. 8 6	6. 34	1	PARAFFIN
3	45202	TOLUENE	92. 13	4. 56	4. 57	3	AROMATIC
4	43822	TRIMETHYLFLUOROSILANE	92. 00	63. 86	83. 75	5	MISCELLANEOUS
5	45201	BENZENE	78. 11	3. 05	3. 58	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP	OUNDS OF CLASSIFICATION 1	97. 18	8. 53	8, 08		
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	CUNDS OF CLASSIFICATION 3	92. 13	4. 56	4, 59		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 5	92. 00	83. 86	83. 75		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 7	78. 11	3. 05	3. 58		
	5 COMP	OUND COMPOSITE	91. 93	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-03-009-01

INDUSTRIAL PROCESS, PRIMARY METALS, STEEL PRODUCTION OPEN HEARTH WITH OXYGEN LANCE, CONTROLLED

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: ELECTROSTATIC PRECIPITATOR

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0014

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43231	N-HEXANE	86. 17	7. 00	7. 46	1	PARAFFIN
2	43232	N-HEPTANE	100. 20	35. 80	32. 87	1	PARAFFIN
3	45202	TOLUENE	92. 13	3. 60	3. 59	3	AROMATIC
4	43822	TRIMETHYLFLUOROSILANE	92. 00	40. 00	40. 06	5	MISCELLANEOUS
5	45201	BENZENE	78. 11	13. 60	16. 02	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP	DUNDS OF CLASSIFICATION 1	97. 60	42. 80	40. 33		
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 3	92. 13	3. 60	3. 59		
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 5	92. 00	40.00	40. 06		
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 7	78. 11	13. 60	16. 02		
	5 COMP	DUND COMPOSITE	92. 04	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA QC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-03-009-01

DECEMBER 14, 1978 TABLE 3-03-009C

INDUSTRIAL PROCESS, PRIMARY METALS, STEEL PRODUCTION BASIC DXYGEN FURNACE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: ELECTROSTATIC PRECIPITATOR

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0016

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43204	PROPANE	44. 09	39. 90	42, 49	1	PARAFFIN
2	43822	TRIMETHYLFLUOROSILANE	92.00	49.00	25. 02	5	MISCELLANEOUS
3	43201	METHANE	16.04	11.10	32. 49	6	METHANE
		TOTAL		100. 00	100. 00		
	1 COMP	DUNDS OF CLASSIFICATION 1	44. 09	39, 90	42. 49		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	9 2. 00	49. 00	25. 02		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	11, 10	32. 49		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	3 СОМР	OUND COMPOSITE	46. 96	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA QC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-03-009-03

REFERENCES

- 1. Considine, D. M., editor, "Chemical and Process Technology Encyclopedia," McGraw Hill Book Co., 1974.
- 2. "Compilation of Air Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Taback, J. H., et al., "Control of Hydrocarbon Emisions from Stationary Sources in the California South Coast Air Basin," Vol. I & II, KVB, Inc., Tustin, CA, June 1978.

3-05 INDUSTRIAL PROCESS, MINERAL PRODUCTS

3-05-001 ASPHALT ROOFING--BLOWING

--FELT SATURATORS: SPRAYING AND DIPPING

--HOT ASPHALT OIL STORAGE

--TAR KETTLES

Process Description 1

The manufacture of asphalt roofing felts and shingles involves saturating fiber media with hot asphalt (400-450 °F) by means of dipping and/or spraying. Although it is not always done at the same site, preparation of the asphalt saturant is an integral part of the operation. This preparation, called "blowing," consists of oxidizing the asphalt by bubbling air through the liquid asphalt for 8 to 16 hours. The saturant is then transported to the saturation tank or spray area. The saturation of the felts is accomplished by dipping, high-pressure sprays, or both. The final felts are made in various weights: 15, 30, and 55 pounds per 100 square feet (0.72, 1.5, and 2.7 kg/m²). Regardless of the weight of the final product, the makeup is approximately 40 percent dry felt and 60 percent asphalt saturant.

AP 40 presents a thorough discussion of a typical asphalt roofing line (Ref. 2).

Emissions

Candidate process areas of hydrocarbon emissions from asphalt roofing operations are:

- . Felt Saturators: Spraying and Dipping
- . Air Blowing
- . Hot Asphalt Storage
- . Tar Kettles (portable field type)

VOC emissions generally increase directly with increasing line speed, felt moisture content, air temperature and humidity, number of spray headers and gates in operation and the asphalt temperature (Ref. 2). Profiles 3-05-001B and C present the estimated organic species emanating from uncontrolled operations (Ref. 4).

Air Blowing², ³--VOC emissions from the asphalt air blowing stills include oxygen, nitrogen and its compounds, water vapor, sulfur compounds, and hydrocarbons as gases, odors, and aerosols.

The following table presents uncontrolled and controlled emissions and limitations from asphalt roofing manufacturing.

HYDROCARBON EMISSIONS FROM ASPHALT ROOFING MANUFACTURE

Type of Operation			rocarbon Emi sed on 210,0)
and Control	% Control	lbs/ton*	kg/mt	lbs/hr	kg/hr
Asphalt Blowing, uncontrolled	0	2.5	1.25	60.0	27.22
Asphalt Blowing, with afterburner	99	0.025	0.0125	0.60	0.27

*Ton of Asphalt blown Source: Reference 3

Profile 3-05-001A presents the estimated organic species eminating from an uncontrolled, asphalt blowing still.

Hot Asphalt Oil Storage--Hydrocarbon emission rates from the storage of hot asphalt oil can be calculated based on the information contained in the API Bulletin on "Petrochemical Evaporation Loss from Storage Tanks," (Ref. API Bulletin 2523) and Section 4.3 of AP-42. The VOC species for asphalt storage emissions is estimated to be identical to that of Profile 3-05-001B.

Roofing Tar Kettles--Emissions from portable roofing tar kettles are the direct result of the heating and subsequent thermal cracking and vaporization of low-boiling-point hydrocarbon oils. Emission rates of 1.9x10³ lbs HC/hour have been estimated (Ref. 4). Speciation of the volatile organic emissions is presented in Profile 3-05-001D.

Controls^{2,4}

Felt Saturators: Spraying and Dipping--As stated earlier, a reduction of those items directly related to increasing emission rates is one form of control which may, however, by uneconomical. The most common form of control 3.05-2 26200-795

is to equip the saturator machinery with a canopy type hood or room enclosure. The contaminated air which is ventilated from the canopy or room enclosure is then vented through a direct-fired afterburner capable of reaching an exhaust temperature to $1500~{}^{\circ}F$.

The recent concern for energy conservation and the periodic unavailability of natural gas required to operate an afterburner has resulted in the usage of glass fiber mats. Effluent from the roofing plant saturators pass through a slowly moving unrolled mat of glass fiber. The exhaust gases cool and coalesce as they go across the mat and impinge on the mat surface. For a thorough explanation of control methods, consult AP-40, Reference 2.

Asphalt Blowing²--Control of effluent organic vapors from asphalt airblowing stills has been accomplished mainly by wet scrubbing (Venturi-type) and/or by direct-fired incineration. For a thorough explanation of control methods, consult AP-40, Reference 2.

Tar Kettle⁴--Emission control is difficult due to the portable-in-field nature of operation.

Three possible areas for control are:

- . Improved temperature control to prevent unnecessary asphalt heating
- . Improved loading devices to cut down on the time the hood is opened
- . General maintenance to keep hood in a good closing (sealing) condition.

Profile Basis 4,5

The organic species data presented in Profiles 3-05-001A, B and C represent an engineering evaluation (Ref. 5) of similar asphalt operations from field tests conducted by KVB. Profile 3-05-001D represents actual field tests data. A gas sample was taken by means of a gas collecting bottle and NIOSH type charcoal tubes on a roofing tar kettle.

Data Qualification

Profiles 3-05-001A, B and C may be used to characterize emissions from asphalt felt saturators and asphalt blowing stills less controls. Profile 3-05-001D represents actual test data on a roofing tar kettle and is designed to be used as such.

INDUSTRIAL PROCESS, MINERAL PRODUCTS ASPHALT ROOFING, BLOWING OPERATION DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0021

INE	SARGAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	SSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	1. 80	2. 16	1	PARAFFIN	
2	43106	ISOMERS OF HEPTANE	100. 20	3. 60	3. 71	1	PARAFFIN	
3	43231	N-HEXANE	86. 17	1.80	2. 16	1	PARAFFIN	
4	43232	N-HEPTANE	100. 20	15. 80	16. 27	1	PARAFFIN	
5	43233	N-OCTANE	114. 23	74. 50	67. 15	1	PARAFFIN	
6	43502	FORMALDEHYDE	30. 03	2. 50	8. 55	4	CARBONYL	
		TOTAL		100. 00	100. 00			
	5 COMP	OUNDS OF CLASSIFICATION 1	109. 84	97. 50	91. 45			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	1 COMP	OUNDS OF CLASSIFICATION 4	30. 03	2, 50	8. 55			
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	6 COMP	OUND COMPOSITE	103. 02	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

C. APPLICABLE SCC CATEGORIES: 3-05-001-01

DECEMBER 14, 1978 TABLE 3-05-0018

INDUSTRIAL PROCESS, MINERAL PRODUCTS

ASPHALT ROOFING, DIPPING DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0022

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	12. 00	9. 16	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	11.00	7. 25	1	PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	1.00	. 59	1	PARAFFIN
4	43204	PROPANE	44. 09	13.00	19. 43	1	PARAFFIN
5	43212	N-BUTANE	58. 12	18.00	20. 42	1	PARAFFIN
6	43214	ISOBUTANE	5B. 12	8.00	9 . 09	1	PARAFFIN
7	43220	N-PENTANE	72. 15	18.00	16. 47	1	PARAFF IN
8	43231	N-HEXANE	86. 17	2.00	1. 52	1	PARAFFIN
9	43232	N-HEPTANE	100. 20	14.00	9. 22	1	PARAFFIN
10	43203	ETHYLENE	28. 05	2. 00	4. 68	2	OLEFIN
11	43202	ETHANE	30. 07	1.00	2. 17	7	NON-REACTIVE
_		TOTAL		100. 00	100.00		
	9 COMPOUNDS OF CLASSIFICATION 1		68. 69	97. 00	93. 15		
	1 COMP	DUNDS OF CLASSIFICATION 2	28. 05	2. 00	4. 68		
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMPOUNDS OF CLASSIFICATION 4		. 00	. 00	. 00		
	O COMPOUNDS OF CLASSIFICATION 5		. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMPOUNDS OF CLASSIFICATION 7		30. 07	1.00	2. 17		
	11 COMPOUND COMPOSITE		65. 95	100, 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 4)

C. APPLICABLE SCC CATEGORIES: 3-05-001-02,-04

INDUSTRIAL PROCESS, MINERAL PRODUCTS ASPHALT ROOFING, SPRAYING DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0023

INE	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION		
NO.								
1	43105	ISOMERS OF HEXANE	86. 17	12. 00	9. 16	1 PARAFFIN		
2	43106	ISOMERS OF HEPTANE	100. 20	11.00	7. 25	1 PARAFFIN		
3	43107	ISOMERS OF OCTANE	114. 23	1.00	. 59	1 PARAFFIN		
4	43204	PROPANE	44. 09	13.00	19. 43	1 PARAFFIN		
5	43212	N-BUTANE	58. 12	18.00	20. 42	1 PARAFFIN		
6	43214	ISOBUTANE	58. 12	8. 00	9. 09	1 PARAFFIN		
7	43220	N-PENTANE	72. 15	18.00	16. 47	1 PARAFFIN		
8	43231	N-HEXANE	86. 17	2. 00	1. 52	1 PARAFFIN		
9	43232	N-HEPTANE	100. 20	14.00	9. 22	1 PARAFFIN		
10	43203	ETHYLENE	28. 05	2. 00	4. 68	2 OLEFIN		
11	43202	ETHANE	30. 07	1.00	2. 17	7 NON-REACTIVE		
		TOTAL		100. 00	100. 00			
	9 COMPOUNDS OF CLASSIFICATION 1		68. 69	97. 0 0	93, 15			
		OUNDS OF CLASSIFICATION 2	28. 05	2.00	4. 68			
	O COMPOUNDS OF CLASSIFICATION 3		. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMPOUNDS OF CLASSIFICATION 5		. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 7	30. 07	1. 00	2. 17			
•	11 COMPOUND COMPOSITE		65. 95	100. 00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 4)

C. APPLICABLE SCC CATEGORIES: 3-05-001-03

INDUSTRIAL PROCESS, MINERAL PRODUCTS ASPHALT ROOFING, TAR KETTLE DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0024

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	3. 40	1. 53	1	PARAFFIN	
2	43107	ISOMERS OF OCTANE	114. 23	7. 40	2. 49	1	PARAFFIN	
3	43115	C-7 CYCLOPARAFFINS	98. 19	2. 90	1. 15	1	PARAFFIN	
4	43116	C-8 CYCLOPARAFFINS	112. 23	. 40	. 15	1	PARAFFIN	
5	43117	C-9 CYCLOPARAFFINS	126. 26	1. 50	. 46	1	PARAFFIN	
6	43122	ISOMERS OF PENTANE	72. 15	1.10	. 57	1	PARAFFIN	
7	43204	PROPANE	44. 09	10. 20	8. 85	1	PARAFFIN	
8	43212	N-BUTANE	58. 12	11.60	7. 67	1	PARAFFIN	
9	43214	ISOBUTANE	58. 12	. 70	. 46	1	PARAFFIN	
10	43220	N-PENTANE	72. 15	6. 30	3. 33	1	PARAFFIN	
11	43231	N-HEXANE	86. 17	4. 90	2. 18	1	PARAFFIN	
12	43232	N-HEPTANE	100. 20	2. 00	. 77	1	PARAFFIN	
13	43233	N-OCTANE	114. 23	2. 70	. 92	1	PARAFFIN	
14	43242	CYCLOPENTANE	70. 14	2. 50	1.38	1	PARAFFIN	
15	43121	ISOMERS OF PENTENE	70. 13	. 50	. 27	2	OLEFIN	
16	43203	ETHYLENE	28. 05	. 30	. 42	2	OLEFIN	
17	43205	PROPYLENE	42. 08	2. 00	1. 84	2	OLEFIN	
18	43213	BUTENE	56. 10	7. 00	4. 79	2	OLEFIN	
19	43224	1-PENTENE	70. 13	3. 20	1. 76	2	OLEFIN	
50	45202	TOLUENE	92 . 13	1. 90	. 80	3	AROMATIC	
21	43201	METHANE	16. 04	21.30	50 . 90	6	METHANE	
22	43202	ETHANE	30. 07	5 . 40	6. 90	7	NON-REACTIVE	
23	45201	BENZENE	78. 11	. 80	. 3 8	7	NON-REACTIVE	
		TOTAL		100. 00	99 . 97			
	14 COMP	DUNDS OF CLASSIFICATION 1	69. 4 2	57. 60	31. 91			
		DUNDS OF CLASSIFICATION 2	55. 10	13.00	9. 08			
		DUNDS OF CLASSIFICATION 3	92. 13	1. 90	. 80			
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	1 COMPOUNDS OF CLASSIFICATION 6		16. 04	21. 30	50. 90			
		OUNDS OF CLASSIFICATION 7	32. 58	6. 20	7. 28			
-	23 COMP(OUND COMPOSITE	38. 44	100.00	99. 97			

3.05-7

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH IN HOOD

B. REFERENCES: KVB TEST DATA (REF. 4)

C. APPLICABLE SCC CATEGORIES: 3-05-001-99

REFERENCES

- 1. "Compilation of Air Pollution Emission Factors," Supplements 1-8, Publication AP-42, EPA, April 1977.
- 2. "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," Prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA 340/1-78-004, April 1978.
- 4. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin,"
 Vols. I and II, KVB, Inc., Tustin, CA, June 1978.
- 5. Sonnichsen, T. W., KVB Engineer.

3.05-8 26200-795

INDUSTRIAL PROCESS, MINERAL PRODUCTS

3-05-002 ASPHALTIC CONCRETE--ROTARY DRYER
--MIXING-LOADING
--ASPHALT OIL STORAGE

Process Description1,2

3-05

Hot-mix asphalt paving consists of a combination of aggregates uniformly mixed and coated with asphaltic cement. A typical hot-mix paving plant generally consists of an oil- or gas-fired rotary dryer, a screening and classifying system, a weighing system for asphalt cement and aggregate, a mixer, and the necessary material handling equipment.

Asphaltic cement is normally used in amounts of 3 to 12 percent by weight of refined petroleum.

Emissions^{1,2}

Candidate process areas for the emission of hydrocarbon vapors to the atmosphere are:

- . Rotary Dryer
- . Asphalt-Aggregate Mixer
- . Truck Loading Area
- . Asphalt Oil Storage

Hydrocarbon emissions from the rotary gas- or oil-fired dryer are the result of the incomplete combustion of the fuel fired. Emission rates are estimated to be equivalent to those listed in AP-42 under, Combustion of Fuels. Profile 3-05-002A presents the estimated organic species breakdown of the flue gas for a natural gas fired unit. For an oil fired unit, refer to Profiles 1-01-004 and 1-01-005.

Hydrocarbon emissions from the asphalt-aggregate mixer and truck loading area are the result of the vaporization of the low boiling point hydrocarbon oils, however, their emission rates were not available at the time this report was written. A sample of the vapor coming off of a pile of compacted road asphalt was taken and analyzed for hydrocarbon species. The results are presented in Profile 3-05-002B.

Hydrocarbon emission rates for hot asphalt storage tanks can be computed from information contained within API Bulletin 2523. Speciation is similar to that presented in Profile 3-05-001B.

Controls 1,2

Hydrocarbon emissions from hot asphalt batch plants are not generally considered to be a major air pollution problem. Unburned or partially burned hydrocarbons emanating from the rotary dryer can usually be reduced considerably by improving combustion efficiency. Hydrocarbon emissions from asphalt oil storage, mixing and asphaltic cement loading can be vented into a venturi type wet scrubber, direct-fired afterburner, or introduced in with the rotary dryer's combustion air supply. AP-40 offers an excellent explanation of the various ways gaseous pollutants can be controlled (Ref. 2)

Profile Basis 3

The organic species data presented in Profile 3-05-002A represents an engineering evaluation of similar test data on a natural gas fired boiler using gas collecting bottles and NIOSH type charcoal tubes. Profile 3-05-002B represents a laboratory test performed on a sample of road asphalt. A gas collecting bottle and NIOSH type charcoal tubes were also used to obtain a gas sample for analysis.

Data Qualification

Profiles 3-05-002A, 1-01-004 and 1-01-005 represent similar situations regarding fuel combustion data for dryers. Profile 3-05-002B represents a specific situation found at a hot asphalt batch plant. The introduction of a control device affecting hydrocarbon emissions could also effect the emitted species as well as rate and should be taken into account. All presented profiles here represent uncontrolled emissions.

DECEMBER 14, 1978 TABLE 3-05-002A INDUSTRIAL PROCESS, MINERAL PRODUCTS

ASPHALT CONCRETE, ROTARY DRYER, NATURAL CAS FIRED

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0025

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL.
NO.	CODE	NAME	WEICHT	MEICHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	B6. 17	1.00	. 28	1	PARAFFIN
2	43122	ISOMERS OF PENTANE	72. 15	9. 00	2. 90	1	PARAFFIN
3	43204	PROPANE	44. 07	4. 00	2. 11	1	PARAFFIN
4	43212	N-BUTANE	58. 12	9 . 00	3. 60	1	PARAFFIN
5	43220	N-PENTANE	72. 15	6. 00	1. 93	1	PARAFFIN
6	43248	CYCLOHEXANE	84. 16	1, 00	. 28	1	PARAFFIN
7	45202	TOLUENE	9 2. 13	2. 00	. 51	3	AROMATIC
8	43502	FORMALDEHYDE	30. 03	8. 00	6. 17	4	CARBONYL
9	43201	METHANE	16. 04	56.00	B1. 04	6	METHANE
10	45201	BENZENE	78. 11	4. 00	1. 18	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	6 COMP	OUNDS OF CLASSIFICATION 1	62. 92	30. 00	11. 10		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COHP	OUNDS OF CLASSIFICATION 3	92 . 13	2. 00	. 51		
	1 COMP	CUNDS OF CLASSIFICATION 4	30. 03	8. 00	6. 17		
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	56. 00	B1. 04		
	1 COMP	OUNDS OF CLASSIFICATION 7	78. 11	4. 00	1. 18		
	10 COMP	OUND COMPOSITE	23. 23	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF SIMILAR TEST DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

INDUSTRIAL PROCESS, MINERAL PRODUCTS, ASPHALTIC CONCRETE

IN PLACE ROAD ASPHALT

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0026

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	(CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLAS	SSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	8. 10	3. 91	1	PARAFFIN	
2	43115	C-7 CYCLOPARAFFINS	98. 19	3. 70	1. 58	1	PARAFFIN	
3	43122	ISOMERS OF PENTANE	72. 15	5. 70	3. 29	1	PARAFFIN	
4	43204	PROPANE	44. 09	5. 50	5. 20	1	PARAFFIN	
5	43212	N-BUTANE	58. 12	10. 10	7. 24	1	PARAFFIN	
6	43214	ISOBUTANE	5 8. 12	11.20	8. 03	1	PARAFFIN	
7	43220	N-PENTANE	72. 15	5. 30	3. 08	1	PARAFFIN	
8	43231	N-HEXANE	86. 17	8. 80	4. 25	1	PARAFFIN	
9	43203	ETHYLENE	28. 05	2. 00	2. 96	2	OLEFIN	
0	43205	PROPYLENE	42. 08	3. 90	3. 87	2	OLEFIN	
1	43213	BUTENE	56. 10	5. 90	4. 37	2	OLEFIN	
2	43201	METHANE	16. 04	15. 70	40. 76	6	METHANE	
3	43202	ETHANE	30. 07	4. 60	6. 37	7	NON-REACTIVE	
4	45201	BENZENE	78. 11	9. 50	5. 08	7	NON-REACTIVE	
		TOTAL		100.00	99. 99			
	8 COMP	DUNDS OF CLASSIFICATION 1	66. 56	58. 40	36. 58			
	3 COMP	DUNDS OF CLASSIFICATION 2	43. 84	11.80	11, 20			
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	O COMPI	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMPI	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 6	16. 04	15. 70	40. 76			
		DUNDS OF CLASSIFICATION 7	51. 38	14. 10	11. 45			
	14 COMP	OUND COMPOSITE	41.68	100.00	99. 99			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF LABORATORY TEST SAMPLES

B. REFERENCES: KVB TEST DATA (REF. 3)

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.

3.05-13 26200-795

3-06 INDUSTRIAL PROCESS, PETROLEUM INDUSTRY
REFINERY OPERATIONS --FUGITIVE EMISSIONS

Introduction¹

Although a modern refinery is a complex system of many processes, the entire operation can be divided into four major steps: separating, converting, treating, and blending. The crude oil is first separated into selected fractions (e.g., gasoline, kerosene, fuel, oil, etc.). Because the relative volumes of each fraction produced by merely separating the crude may not conform to the market demands for each fraction, some of the less valuable products, such as heavy naphtha, are converted to products with a greater sale value, such as gasoline. This conversion is accomplished by splitting (cracking), uniting (polymerization), or rearranging (reforming) the original molecules. The final step is the blending of the refined base stocks with each other and with various additives to meet final product specifications.

To accomplish this, thousands of valves, fittings, flanges and pumps help to contain and transfer petroleum fluid to the various process units within a refinery.

The potential sources of hydrocarbon fugitive emissions from oil refining operations are discussed in the following sections as listed below:

- . Fluid Catalytic Cracker Unit (FCCU)
- . Catalytic Reformer
- . Process Drains
- . Cooling Towers
- . Pipeline Valves/Flanges/Fittings
- . Pressure Relief Valves
- . Pump/Compressor Seals

3-06 INDUSTRIAL PROCESS, PETROLEUM INDUSTRY

3-06-002 FLUID CATALYTIC CRACKING UNITS (FCCU) -- GENERAL

Process Description 1,2

Catalytic cracking involves the decomposition of large molecules into smaller, lower-boiling molecules by the addition of heat and pressure with the proper catalyst. Some of these molecules may polymerize to form large molecules. Finely powdered catalyst is lifted into the reactor by the incoming heated oil charge which vaporizes upon contact with the hot catalyst. Spent catalyst settles out in the reactor, is drawn off at a controlled rate, purged with steam, and lifted by an air stream into the regenerator where the deposited coke is burned off under controlled combustion conditions.

Emissions^{2,3}

The resulting exhaust gases from the combustion of the coke deposits on the catalyst and may contain catalyst dust, hydrocarbons, and other impurities originating in the charging stock, as well as the products of combustion.

Catalyst dust is generally the pollutant of major concern. Hydrocarbons, carbon monoxide, ammonia, and organic acids are effectively controlled by incineration in carbon monoxide (CO) waste-heat boilers (Ref. 2).

AP-42 lists an uncontrolled hydrocarbon emission factor (e.g. without CO waste-heat boilers) of 220 $1b/10^3$ bbl of fresh feed. Studies (Ref. 3) conducted on FCC units with CO waste-heat boilers reported average hydrocarbon emission rates of 1.6 $1b/10^3$ bbl of fresh feed. Profile 3-06-002 presents the VOC emissions measured downstream of a FCC unit CO waste-heat boiler (Ref. 3)

Controls

As previously mentioned, the CO waste-heat boiler offers a secondary control feature for the unburned hydrocarbons, carbon monoxide, ammonia and organic acids. Therefore additional controls for these pollutants are not necessary. Electrostatic precipitators, however, are usually installed to control particulate emissions to acceptable limits.

Profile Basis³

Data presented in Profile 3-06-002 are based on field tests conducted at a 50,000 bbl/day FCC unit employing a CO waste-heat boiler and electrostatic precipitator. Samples were collected in glass gas collection bottles followed by NIOSH type charcoal tubes. The sample point was located downstream of the CO boiler in the exhaust stack.

Data Qualification

Profile 3-06-002 may be used to characterize the VOC emissions from a typical FCC unit employing a CO waste-heat boiler and electrostatic precipitator. Emission rates could vary, depending on the degree of catalyst coking and concentration of hydrocarbons entering the CO boiler.

3.06-3 26200-795

DECEMBER 14, 1978

TABLE 3-06-002

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY

FLUID CATALYTIC CRACKER, CO BOILER

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0029

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	13. 00	3. 69	1	PARAFFIN	
Ž	43502	FORMALDEHYDE	30. 03	51.00	41. 49	4	CARBONYL	
3	43201	METHANE	16. 04	36, 00	54. 83	6	METHANE	
		TOTAL		100.00	100. 01			
	1 COMP	OUNDS OF CLASSIFICATION 1	86. 17	13. 00	3. 69			
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	1 COMP	DUNDS OF CLASSIFICATION 4	30. 03	51.00	41. 49			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	36. 00	54. 83			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 0 0	. 00			
•	3 COMP	OUND COMPOSITE	24. 43	100.00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH TAKEN WITH STACK B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-002-01

EXTENSION

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.

3-06-005 PROCESS DRAINS

Process Description 1,3

In refinery operations, condensate water and flushing water must be drained from the process equipment. These process drains also remove process liquid, leakage or spills and water used to cool pump glands.

Each major unit in a refinery is usually drained by a network of small lines which are generally trapped at their inlets and flow into junction or collection boxes, some of which are opened to the atmosphere.

The liquid waste then generally proceeds from the collection boxes into a network of large trunk lines. These connect with interceptor lines which carry the process water to API separators and water treatment plants.

Emissions

Common sources of liquid hydrocarbons entering a drainage system are: process liquid sampling, turnarounds, blind changing, process equipment leaks, and spills. As the hydrocarbon-water mixture flows through the drainage system, VOC's are evaporated from the surface and may escape to the atmosphere through vents (Ref. 1).

Hydrocarbon emission rates as found in AP-42 are: uncontrolled (210 $1b/10^3$ bbl waste); vapor recovery or separator control (8 $1b/10^3$ bbl waste).

Estimated volatile organic specie emissions data are presented in Profile 3-06-005.

Controls

Modern refining designs provide waste-water-effluent systems with running-liquid-sealed traps and liquid-sealed and covered junction boxes. These seals keep the amount of liquid hydrocarbons exposed to the air at a minimum and thereby reduce hydrocarbon losses.

Another form of control is to connect vapor recovery equipment to the drainage system.

Profile Basis²

Profile 3-06-005 is a composite of five separate refinery API separators and process drain hydrocarbon emissions tests. Controls other than inlet liquid traps were not used. This was decided to best estimate the volatile organic specie emissions from a typical refinery drainage system.

Samples were collected by means of gas collecting bottles and NIOSH type charcoal tubes. Process flow rates were estimated using engineering judgment.

Data Qualification

Profile 3-06-005 may be used to characterize the organic specie emissions from a typical refinery drainage system.

3.06-6 26200-795

INDUSTIAL PROCESS, PETROLEUM INDUSTRY, REFINERY, FUGITIVE EMISSIONS, DRAINAGE/SEPARATION PITS, COVERED, CRUDE OIL AND GAS DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0031

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	12. 20	9. 28	1 PARAFFIN	
2	43115	C-7 CYCLOPARAFFINS	98. 19	16. 90	11. 24	1 PARAFFIN	
3	43116	C-B CYCLOPARAFFINS	112. 23	5. 20	3. 01	1 PARAFFIN	
4	43122	ISOMERS OF PENTANE	72. 15	10. 10	9. 15	1 PARAFFIN	
5	43204	PROPANE	44. 09	5. 90	B. 76	1 PARAFFIN	
6	43212	N-BUTANE	58. 12	14. 30	16. OB	1 PARAFFIN	
7	43214	ISOBUTANE	5 8. 12	4. 50	5. 03	1 PARAFFIN	
8	43220	N-PENTANE	72. 15	12.00	10. 85	1 PARAFFIN	
9	43231	N-HEXANE	86. 17	11. 90	9. 02	1 PARAFFIN	
10	43201	METHANE	16. 04	2. 9 0	11.83	6 METHANE	
11	43202	ETHANE	30. 07	1. 70	3. 73	7 NON-REACTIVE	
12	45201	BENZENE	78. 11	2. 40	2. 03	7 NON-REACTIVE	
		TOTAL		100. 00	100. 01		
	9 COMP	OUNDS OF CLASSIFICATION 1	73. 70	93. 00	82. 42		
		OUNDS OF CLASSIFICATION 2		. 00	. 00		
		OUNDS OF CLASSIFICATION 3		00	. 00		
		OUNDS OF CLASSIFICATION 4		. 00	. 00		
		DUNDS OF CLASSIFICATION 5		. 00	00		
		DUNDS OF CLASSIFICATION 6		2. 90	11. 83		
		DUNDS OF CLASSIFICATION 7		4. 10	5. 76		
-	12 COMP	OUND COMPOSITE	65. 34	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 2)

REFERENCES

- 1. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 2. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.
- 3. "Joint District, Federal and State Project for the Evaluation of Refinery Emissions," Report No. 8, June 1958, Los Angeles APCD.

3.06-7a 26200-795

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY

3-06-007 COOLING TOWERS

Process Description²

Refineries use large amounts of water in the cooling of certain operations (heat exchangers) and process equipment. The large amounts of water used for cooling are conserved by recooling the water in wooden towers. Cooling is accomplished by evaporating part of this water.

Emissions

3-06

Hydrocarbons that might be entrained or dissolved in the water as a result of leaking heat exchange equipment or other process equipment where process cooling water has come into contact with a hydrocarbon stream are readily discharged to the atmosphere in the cooling tower. Improperly designed and/or maintained heat exchange equipment account for the majority of process leaks entering the water being recirculated through the tower. AP-42 (Ref. 1) reports an emission factor of 6.0 lb of HC/10 gallons of cooling water for cooling towers.

Profile 3-06-007 presents the estimated volatile organic species being emitted from a typical refinery cooling tower (Ref. 3).

Controls²

Control generally takes the form of not allowing (if possible without shutting a leaky unit down) process water that has come into contact with a hydrocarbon stream or has otherwise been contaminated with odorous material from entering the cooling tower water supply. This can be accomplished through the proper design and maintenance of heat exchange equipment which would then minimize the majority of process fluid losses.

Another form of control has been made possible with advancements in fan cooling of finned equipment, which in some instances has replaced the need for conventional cooling towers.

Profile Basis³

Profile 3-06-007 represents test data obtained on a large forced air refinery cooling tower serving an FCC unit, the gas plant for that FCC unit and a reformer unit. The water circulation rate was reported as 42,500 gal/min.

Water samples were taken at the inlet and outlet of the tower and analyzed (xylene extraction and GC analysis) for organic content. The organic content was identified as 100% isopentane.

Data Qualification

Profile 3-06-007 may be used to characterize the volatile organic emissions (as isomers of pentane) from a typical refinery cooling tower serving the above mentioned equipment.

3.06-10

DECEMBER 14, 1978 TABLE 3-06-007

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY

COOLING TOWERS, FUGITIVE EMISSIONS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0035

.INE NO.	SARCIAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43122	ISOMERS	OF PENTANE TOTAL		100. 00 100. 00	100. 00 100. 00	1 PARAFFIN
	1 COMP	OUNDS OF	CLASSIFICATION 1	72. 15	100. 00	100.00	
			CLASSIFICATION 2	. 00	. 00	. 00	
			CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	DUND COMP	POSITE	72. 15	100.00	100, 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE TAKEN FROM TOWER EXTERIOR

B. REFERENCES: KVB TEST DATA (REF. 3)

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY

3-06-008A MISCELLANEOUS PIPELINE/VALVES--FLANGES

--FITTINGS

-- RELIEF VALVES

Process Descriptions

Pipeline valves, flanges, fittings and relief valves are a necessary part of any pressurized liquid handling system and their functions are self-explanatory.

Emissions 3

3-06

Liquid and vapor leaks can develop at valve stems, flanges, and fittings as a result of heat, pressure, friction, corrosion, and vibration.

Liquid and vapor leaks can develop at pressure relief valves as a result of corrosion and failure of relief valves to reseat properly after blowoff.

The inaccessible nature of most pressure relief valves accounts for poor maintenance which could allow substantial leakage before repair.

Emission rates and their associated hydrocarbon composition are a function of leak size for rates, and process material for speciation.

A recent study (Ref. 3) conducted on refinery emissions reported the following emission rates:

		Emission Factor
Device Type	Product	(lb/day device)
Valves & Fittings		
Valves	Gas	0.4
Metal connections	Gas	0.003
Valves	Liquid	0.02
Metal connections	Liquid	0.003

Composite Emission Factor for Valves - gas/liquid 0.15 lb/day·valve*

^{*}Previous inventories (Ref. 4,5) have reported that the ratio of valves in liquid service to those in gas service in refinery operations was approximately 3 to 1. Applying this ratio to the emission factor for valves listed in the above table resulted in a composite emission factor of 0.15 lb/day valve which agrees with AP-42's 0.15 lb/day valve emission factor (Ref. 1).

Pressure relief valves have an emission factor of 11 lb/day·valve as reported in AP-42 (Ref. 1).

Details on pipeline and valve emissions are extensive and beyond the scope of this report, and the reader is advised to consult Reference 3 for further details.

The volatile organic species being emitted per device type is a function of the contained process fluid. Profile 3-06-008A through N characterize these emissions for various process fluids. Profile 3-06-008N applies to pressure relief valves for liquified petroleum gas service.

Controls 3

Control generally takes the form of:

- . proper valve or fitting selection,
- improved maintenance scheduled leak checking, tightening, greasing or replacement when possible, and
- . improved materials e.g., valve packing, glands, etc.

Profile Basis³

Over 18,000 valves and miscellaneous fittings were checked for leaks. Leaks were checked for through KVB's spray and sniff (soap sprayer and TLV Explosion meter) method which is explained in Reference 3. The organic composition of a selected group of leaking valves and flanges was determined through the use of grab samples or gas collecting bottles plus NIOSH type charcoal tubes. Profiles 3-06-008A through N are the direct result of this effort.

Data Qualification

The above mentioned hydrocarbon emission factors and profiles may be used to characterize the volatile organic emissions from a refinery. Each profile identifies the process fluid associated with the leaking valves, flanges or fittings.

3.06-**1**2

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANOUS DECEMBER 14, 1978 TABLE 3-06-008A

PIPE/VALVE FLANGES, COMPOSITE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE KVB PROFILE KEY 0316

CHEMICAL PERCENT PERCENT LINE SAROAD CHEMICAL MOLECULAR CLASSIFICATION VOLUME CODE WEIGHT WEIGHT NO. NAME PARAFFIN 1 . 63 43105 86.17 1.60 ISOMERS OF HEXANE 100.20 . 80 . 26 PARAFFIN 2 43106 ISOMERS OF HEPTANE PARAFFIN . 13 3 43107 ISOMERS OF OCTANE 114, 23 . 40 PARAFFIN 128. 25 . 50 . 13 1 43108 ISOMERS OF NONANE . 07 . 30 PARAFFIN ISOMERS OF DECANE 142, 28 43109 . 07 PARAFFIN 43115 C-7 CYCLOPARAFFINS 98.19 . 20 PARAFFIN . 03 126. 26 . 10 7 43117 C-9 CYCLOPARAFFINS PARAFFIN 8 43122 ISOMERS OF PENTANE 72. 15 7.80 3. 57 PARAFFIN 44.09 11.50 B. 63 9 43204 PROPANE 18.30 10.41 1 PARAFFIN 10 43212 N-BUTANE 58.12 4. 20 PARAFFIN 58. 12 7. 40 43214 ISOBUTANE 11 PARAFFIN 72. 15 7. 70 3. 54 12 43220 N-PENTANE 1.32 PARAFFIN 86.17 3.40 43231 N-HEXANE 13 **PARAFFIN** 100.20 1.40 . 46 14 43232 N-HEPTANE 114.23 1.80 . 53 PARAFFIN 15 43233 N-OCTANE . 60 . 17 PARAFFIN 43235 128. 25 1 16 N-NONANE 142. 28 . 80 . 20 PARAFFIN 17 43238 N-DECANE PARAFFIN 84.16 03 43248 CYCLOHEXANE . 10 1 18 19 43205 PROPYLENE 42.08 . 10 . 07 OLEFIN 45102 ISOMERS OF XYLENE 106. 16 . 20 . 07 3 AROMATIC 20 21 45202 TOLUENE 92.13 . 50 . 17 3 AROMATIC 16.04 28. 60 METHANE 22 43201 METHANE 58. 92 30.07 NON-REACTIVE 23 43202 ETHANE 5.80 6. 38 7 78.11 NON-REACTIVE 24 45201 BENZENE 10 . 03 TOTAL 100.00 100.02 18 COMPOUNDS OF CLASSIFICATION 1 62.49 64.70 34.38 1 COMPOUNDS OF CLASSIFICATION 2 42.08 . 10 . 07 96. 22 2 COMPOUNDS OF CLASSIFICATION 3 . 70 . 24 . 00 . 00 O COMPOUNDS OF CLASSIFICATION 4 . 00 O COMPOUNDS OF CLASSIFICATION 5 . 00 00 . 00 1 COMPOUNDS OF CLASSIFICATION 6 16.04 28. 60 58. 92 2 COMPOUNDS OF CLASSIFICATION 7 30. 29 5. 90 6.41 100.00 100.02 33.13 24 COMPOUND COMPOSITE

METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA NOTES:

REFERENCES: KVB TEST DATA (REF 3)

APPLICABLE SCC CATEGORIES: 3-06-008-01

PIPE/VALVE FLANGES, GASOLINE DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0317

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	1. 10	1. 42	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	3. 40	3. 72	1	PAR AF FIN
3	43107	ISOMERS OF OCTANE	114. 23	4. 70	4. 49	1	PARAFFIN
4	43108	ISOMERS OF NONANE	128. 25	8. 50	7. 23	1	PARAFFIN
5	43109	ISOMERS OF DECANE	142. 28	2. 60	1. 9 7	1	PARAFFIN
6	43115	C-7 CYCLOPARAFFINS	98. 19	1.00	1. 10	1	PARAFFIN
7	43116	C-8 CYCLOPARAFFINS	112. 23	. 20	. 22	1	PARAFFIN
8	43117	C-9 CYCLOPARAFFINS	126. 26	2. 60	2. 30	1	PARAFFIN
9	43204	PROPANE	44. 09	. 20	. 55	1	PARAFFIN
0	43212	N-BUTANE	58. 12	. 70	1. 31	1	PARAFFIN
. 1	43214	ISOBUTANE	58 . 12	. 10	. 22	1	PARAFFIN
.2	43220	N-PENTANE	72 . 1 5	1. 30	1. 97	1	PARAFFIN
.3	43231	N-HEXANE	86. 17	3. 20	4. 05	1	PARAFFIN
4	43232	N-HEPTANE	100. 20	1. 40	1. 53	1	PARAFFIN
5	43233	N-OCTANE	114. 23	5 8. 10	55. 75	1	PARAFFIN
6	43235	N-NONANE	128. 25	2. 80	2. 41	1	PARAFFIN
.7	43248	CYCLOHEXANE	84. 16	. 90	1. 20	1	PARAFFIN
8	45102	ISOMERS OF XYLENE	106. 16	2. 60	2. 74	3	AROMATIC
9	45202	TOLUENE	92 . 13	3. 20	3. 83	3	ARDMATIC
20	45201	BENZENE	78. 11	1. 40	1. 97	7	NON-REACTIVE
		TOTAL		100. 00	99. 98		
	17 COMP	DUNDS OF CLASSIFICATION 1	111. 27	92. 8 0	91. 44		
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	97. 98	5. 80	6. 57		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 7	78. 11	1. 40	1. 97		
•	20 COMP	DUND COMPOSITE	109. 74	100.00	99, 98		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PIPE/VALVE FLANGES, CRACKED GASOLINE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0319

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	12. 80	12, 21	1 PARAFFIN
2	43107	ISOMERS OF OCTANE	114. 23	4. 10	2. 95	1 PARAFFIN
3	43108	ISOMERS OF NONANE	128. 25	. 20	. 16	1 PARAFFIN
4	43115	C-7 CYCLOPARAFFINS	98 . 19	3. 90	3. 28	1 PARAFFIN
5	43117	C-9 CYCLOPARAFFINS	126. 26	. 30	. 16	1 PARAFFIN
6	43122	ISOMERS OF PENTANE	72. 1 5	21. 40	24. 34	1 PARAFFIN
7	43212	N-BUTANE	58 . 12	1. 10	1. 56	1 PARAFFIN
8	43220	N-PENTANE	72. 15	19. 70	22. 38	1 PARAFFIN
9	43231	N-HEXANE	86. 17	15. 40	14. 67	1 PARAFFIN
10	43232	N-HEPTANE	100. 20	14. 90	12, 21	1 PARAFFIN
11	43213	BUTENE	56. 10	1. 10	1. 64	2 OLEFIN
12	45102	ISOMERS OF XYLENE	106, 16	1. 60	1, 23	3 AROMATIC
13	45202	TOLUENE	92. 13	3. 00	2. 70	3 ARDMATIC
14	45201	BENZENE	78. 11	. 50	. 49	7 NON-REACTIVE
		TOTAL		100. 00	99. 98	
	10 COMP	OUNDS OF CLASSIFICATION 1	82. 00	93. 80	93. 92	
		DUNDS OF CLASSIFICATION 2	56, 10	1. 10	1. 64	
		DUNDS OF CLASSIFICATION 3	96. 52	4. 60	3. 93	
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
		DUNDS OF CLASSIFICATION 7	78. 11	. 50	. 49	
-	14 COMP	DUND COMPOSITE	82. 12	100.00	99. 98	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANOUS

PIPE/VALVE FLANGES, GAS-OIL STOCK

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0322

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	6. 00	4. 00	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	16.00	9 . 15	1	PARAFF IN
3	43122	ISOMERS OF PENTANE	72. 15	3. 80	3. 03	1	PARAFFIN
4	43204	PROPANE	44. 09	3. 30	4. 29	1	PARAFFIN
5	43212	N-BUTANE	58. 12	6. 50	6. 40	1	PARAFFIN
6	43220	N-PENTANE	72. 15	7. 40	5 . 89	1	PARAFFIN
7	43231	N-HEXANE	86. 1 7	11.80	7. 83	1	PARAFFIN
8	43232	N-HEPTANE	100. 20	8. 30	4. 75	1	PARAFFIN
7	43233	N-OCTANE	114. 23	7. 9 0	3, 75	1	PARAFFIN
10	43235	N-NONANE	128. 25	4. 40	1. 94	1	PARAFFIN
11	43238	N-DECANE	142. 28	3. 50	1. 43	1	PARAFFIN
12	45202	TOLUENE	9 2. 13	7. 30	4. 52	3	AROMATIC
13	43201	METHANE	16. 04	10.00	35. 62	6	METHANE
14	43202	ETHANE	30. 07	3. 80	7. 20	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	11 COMP	DUNDS OF CLASSIFICATION 1	85. 84	78. 9 0	52. 66		
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	92. 13	7. 30	4, 52		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 6	16. 04	10.00	35. 62		
		DUNDS OF CLASSIFICATION 7	30. 07	3. 80	7. 20		
•	14 COMP	DUND COMPOSITE	57. 25	100, 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

TABLE 3-06-008E

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MICELLANEOUS

PIPE/VALVE FLANGES, REFORMATE STOCK

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0309

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43204	PROPANE	44. 09	13. 90	21. 33	1 PARAFFIN
2	43212	N-BUTANE	58. 12	24. 70	28. 77	1 PARAFFIN
3	43220	N-PENTANE	72. 15	21. 50	20. 18	1 PARAFFIN
4	43231	N-HEXANE	86. 17	19. 30	15. 17	1 PARAFFIN
5	43232	N-HEPTANE	100. 20	10.00	6. 77	1 PARAFFIN
6	43233	N-OCTANE	114. 23	9. 70	5. 75	1 PARAFFIN
7	43202	ETHANE	30. 07	. 90	2. 03	7 NON-REACTIVE
		TOTAL		100. 00	100. 00	
	6 COMP	OUNDS OF CLASSIFICATION 1	68. 50	99. 10	97. 97	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
		OUNDS OF CLASSIFICATION 7	30. 07	. 90	2. 03	
	7 COMP	OUND COMPOSITE	67. 72	100. 00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PIPE/VALVE FLANGES, DISTILLATE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0318

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	3. 00	2. 45	1 PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	2. 00	1. 40	1 PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	3.00	1.82	1 PARAFFIN
4	43108	ISOMERS OF NONANE	128. 25	50	28	1 PARAFFIN
5	43122	ISOMERS OF PENTANE	72. 15	5. 00	4. B3	1 PARAFFIN
6	43204	PROPANE	44. 09	5. 30	8. 40	1 PARAFFIN
7	43212	N-BUTANE	58. 12	12. 70	15. 34	1 PARAFFIN
8	43214	ISOBUTANE	58. 12	2. 50	3. 01	1 PARAFFIN
9	43220	N-PENTANE	72. 15	11.00	10. 71	1 PARAFFIN
10	43231	N-HEXANE	86. 17	9 . 00	7. 28	1 PARAFFIN
11	43232	N-HEPTANE	100. 20	B. 60	6. 02	1 PARAFFIN
12	43233	N-OCTANE	114. 23	9. 90	6. 09	1 PARAFFIN
13	43235	N-NONANE	128. 25	6. 50	3. 5 7	1 PARAFFIN
14	43238	N-DECANE	142. 28	14. 80	7. 28	1 PARAFFIN
15	43248	CYCLOHEXANE	84 . 16	1. 00	. 84	1 PARAFFIN
16	43201	METHANE	16. 04	4. 20	18. 35	6 METHANE
17	43202	ETHANE	30. 07	1.00	2. 31	7 NON-REACTIVE
		TOTAL		100. 00	99. 9 8	
	15 COMP	DUNDS OF CLASSIFICATION 1	83. 76	94. 80	79. 32	
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	00	. 00	
	O COMPI	DUNDS OF CLASSIFICATION 3	00	00	. 00	
		DUNDS OF CLASSIFICATION 4	. 00	00	. 00	
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
		DUNDS OF CLASSIFICATION 6	16. 04	4, 20	18. 35	
		DUNDS OF CLASSIFICATION 7	30. 07	1. 00	2. 31	
-	17 COMP	DUND COMPOSITE	70. 09	100.00	99. 98	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PIPE/VALVE FLANGES, NAPHTHA

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0320

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	6. 60	8. 43	1	PARAFFIN
2	43107	ISDMERS OF OCTANE	114. 23	5. 60	5. 37	1	PARAFFIN
3	43108	ISOMERS OF NONANE	128. 25	20. 60	17. 63	1	PARAFFIN
4	43109	ISOMERS OF DECANE	142. 28	16. 60	12. 81	1	PARAFFIN
5	43115	C-7 CYCLOPARAFFINS	98. 19	2. 40	2. 63	1	PARAFFIN
6	43116	C-B CYCLOPARAFFINS	112. 23	. 60	. 55	1	PARAFFIN
7	43117	C-9 CYCLOPARAFFINS	126. 26	4. 80	4. 16	1	PARAFFIN
8	43122	ISOMERS OF PENTANE	72. 15	40	. 66	1	PARAFFIN
9	43204	PROPANE	44. 09	. 60	1. 53	1	PARAFFIN
0	43212	N-BUTANE	58. 12	. 60	1.10	1	PARAFFIN
1	43220	N-PENTANE	72. 15	1.30	1. 97	1	PARAFFIN
2	43231	N-HEXANE	86. 17	6. 20	7. 89	1	PARAFFIN
3	43232	N-HEPTANE	100. 20	1.80	1. 9 7	1	PARAFFIN
4	43233	N-OCTANE	114. 23	6. 40	6. 13	1	PARAFFIN
5	43235	N-NONANE	128. 25	8. 30	7. 12	1	PARAFFIN
6	43248	CYCLOHEXANE	84. 16	1. 50	1. 9 7	1	PARAFFIN
7	45102	ISOMERS OF XYLENE	106. 16	7. 00	7. 23	3	AROMATIC
3	45202	TOLUENE	92 . 13	6. 20	7. 34	3	ARDMATIC
9	45201	BENZENE	78. 11	2. 50	3. 50	7	NON-REACTIVE
		TOTAL		100.00	99. 99		
	16 COMP	DUNDS OF CLASSIFICATION 1	112. 87	84. 30	81. 92		
	O COMPO	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	2 COMP	OUNDS OF CLASSIFICATION 3	99. 09	13. 20	14. 57		
	O COMPO	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMPO	OUNDS OF CLASSIFICATION 5	. 00	00	00		
	O COMPO	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMPO	OUNDS OF CLASSIFICATION 7	78. 11	2. 50	3. 50		
-	19 COMPO	DUND COMPOSITE	109. 65	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-008-01

DECEMBER 14, 1978 TABL

TABLE 3-06-008#

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PIPE/VALVE FLANGES, REFINERY GAS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0324

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	1. 30	. 59	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	. 20	. 08	1	PARAFFIN
3	43122	ISOMERS OF PENTANE	72. 15	10. 70	5. 83	1	PARAFFIN
4	43204	PROPANE	44. 09	15. 90	14. 23	1	PARAFFIN
5	43212	N-BUTANE	58. 12	26. 80	18. 17	1	PARAFFIN
6	43214	ISOBUTANE	58. 12	11. 30	7. 65	1	PARAFFIN
7	43220	N-PENTANE	72. 15	9. 40	5. 12	1	PARAFFIN
8	43231	N-HEXANE	86. 17	2. 80	1. 30	1	PARAFFIN
9	43232	N-HEPTANE	100, 20	. 10	. 04	1	PARAFF IN
10	43201	METHANE	16. 04	16. 40	40. 28	6	METHANE
11	43202	ETHANE	30. 07	5. 10	6. 70	7	NON-REACTIVE
		TOTAL		100. 00	99. 99		
	9 COMP	OUNDS OF CLASSIFICATION 1	58. 35	78. 50	5 3. 01		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	16. 40	40. 28		
		OUNDS OF CLASSIFICATION 7	30. 07	5. 10	6. 70		
	11 COMP	DUND COMPOSITE	39, 41	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PIPE/VALVE FLANGES, NATURAL GAS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0323

LINE	SARDAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CI	CHEMICAL ASSIFICATION
NO.	CODE	NAME	METALLI	MEIGHI	VOLOTIL	VL	HODEL EQUITION
1	43204	PROPANE	44. 09	3. 60	1. 43	1	PARAFFIN
2	43212	N-BUTANE	58. 12	. 50	. 16	1	PARAFFIN
3	43203	ETHYLENE	28. 05	. 40	24	2	OLEFIN
4	43201	METHANE	16. 04	84. 50	91.79	6	METHANE
5	43202	ETHANE	30. 07	11.00	6. 38	7	NON-REACTIVE
-		TOTAL		100. 00	100.00		
	2 COMP	DUNDS OF CLASSIFICATION 1	45. 50	4. 10	1. 59		
		OUNDS OF CLASSIFICATION 2	28. 05	. 40	. 24		
		OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 4	. 00	00	. 00		
		OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 6	16. 04	84. 50	91.79		
		OUNDS OF CLASSIFICATION 7	30. 07	11.00	6. 38		
	5 COMP	OUND COMPOSITE	17. 43	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

DECEMBER 14, 1978 TABLE 3-06-00BK INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS

VALVES AND FLANGES, WET AND DRY NATURAL GAS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0041

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME		CHEMICAL SSIFICATION
NU.	CODE	NAME	WEIGHT	WEIGHI	VULUNE	CLM	SSIFICATION
1	43122	ISOMERS OF PENTANE	72. 15	1. 10	. 31	1	PARAFFIN
2	43204	PROPANE	44. 09	11.10	5. 19	1	PARAFFIN
3	43212	N-BUTANE	58. 12	4. 40	1. 56	1	PARAFF IN
4	43214	ISCRUTANE	58. 12	2. 90	1.03	1	PARAFFIN
5	43220	N-PENTANE	72. 15	. 70	. 21	1	PARAFFIN
6	43201	METHANE	16.04	62.00	79. 53	6	METHANE
7	43202	ETHANE	30. 07	17. 80	12. 18	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	_						
		OUNDS OF CLASSIFICATION 1	50. 23	20. 20	8. 30		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	62. 00	79. 53		
	1 COMP	DUNDS OF CLASSIFICATION 7	30. 07	17. 80	12. 18		
	7 COMP	OUND COMPOSITE	20. 59	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE ADJACENT TO LEAK AREA

B. REFERENCES: KVB TEST DATA (REF. 3)

DECEMBER 14, 1978 TABLE 3-06-008L INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS

VALVES AND FLANGES, GAS PLANT DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0036

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	1. 60	. 63	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	. 80	. 26	1	PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	. 40	. 13	1	PARAFFIN
4	43109	ISOMERS OF DECANE	142. 28	. 30	. 07	1	PARAFFIN
5	43115	C-7 CYCLOPARAFFINS	98 . 19	. 20	. 07	1	PARAFFIN
6	43117	C-9 CYCLOPARAFFINS	126. 26	. 10	. 03	1	PARAFFIN
7	43122	ISOMERS OF PENTANE	72. 15	7. 80	3. 57	1	PARAFFIN
8	43204	PROPANE	44. 09	11. 50	8. 62	1	PARAFFIN
9	43212	N-BUTANE	58 . 12	18. 30	10. 40	1	PARAFFIN
10	43214	ISOBUTANE	58. 12	7. 40	4. 19	1	PARAFFIN
11	43220	N-PENTANE	72 . 15	7. 70	3. 5 3	1	PARAFFIN
12	43231	N-HEXANE	86. 17	3. 40	1. 32	1	PARAFFIN
13	43232	N-HEPTANE	100. 20	1. 40	. 46	1	PARAFFIN
14	43233	N-DCTANE	114. 23	1.80	. 53	1	PARAFFIN
15	43235	N-NON AN E	128. 25	. 60	. 17	1	PARAFFIN
16	43238	N-DECANE	142. 28	. 80	. 20	1.	PARAFFIN
17	43242	CYCLOPENTANE	70. 14	. 50	. 23	1	PARAFFIN
18	43248	CYCLOHEXANE	84. 16	. 10	. 03	1	PARAFFIN
19	43205	PROPYLENE	42. 08	. 10	. 07	2	OLEFIN
20	45102	ISOMERS OF XYLENE	106. 16	. 20	. 07	3	AROMATIC
21	45202	TOLUENE	9 2. 13	. 50	. 17	3	AROMATIC
2 2	43201	METHANE	16. 04	28. 60	58. 86	6	METHANE
23	43202	ETHANE	30. 07	5. 80	6. 37	7	NON-REACTIVE
24	45201	BENZENE	78. 11	. 10	. 03	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	18 COMPO	DUNDS OF CLASSIFICATION 1	62. 29	64. 70	34. 44		
	1 COMPO	OUNDS OF CLASSIFICATION 2	42. 08	. 10	. 07		
	2 COMPO	DUNDS OF CLASSIFICATION 3	96. 2 2	. 70	. 24		
	O COMPO	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 6	16. 04	28. 60	58. 86		
		OUNDS OF CLASSIFICATION 7	30. 30	5. 90	6. 40		
-	24 COMPO	OUND COMPOSITE	33. 09	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

^{8.} REFERENCES: KVB TEST DATA (REF 3)

C. APPLICABLE SCC CATEGORIES: 3-06-008-01

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, MISCELLANEOUS WET GAS VALVE FROM TRAPS, WET NATURAL GAS, COMPOSITE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0042

LINE NO.	SAROAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43122	ISOMERS OF PENTANE	72. 15	1. 20	. 40	1	PARAFFIN
2	43204	PROPANE	44. 07	18. 60	9. 92	1	PARAFFIN
3	43212	N-BUTANE	58. 12	8. 30	3. 36	1	PARAFFIN
4	43214	ISOBUTANE	58. 12	4. 30	1. 74	1	PARAFFIN
5	43220	N-PENTANE	72. 15	. 90	. 31	1	PARAFFIN
6	43201	METHANE	16. 04	47. 00	68. 88	6	METHANE
7	43202	ETHANE	3 0. 0 7	19. 70	15. 40	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	5 COMP	OUNDS OF CLASSIFICATION 1	49. 91	33. 30	15, 73		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMPI	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	47. 00	68. 88		
		DUNDS OF CLASSIFICATION 7	30. 07	19. 70	15. 40		
	2 30111	THE WITTERS OF THE PROPERTY OF	20.0,				
-	7 COMP	DUND COMPOSITE	23. 53	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLES

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, MARKETING, FUGITIVE EMISSIONS, RELIEF VALVES, LIQUIFIED PETROLEUM GAS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0047

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEICHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44. 09	90. 40	88. 59	1	PARAFFIN
2	43214	ISOBUTANE	58. 12	. 40	. 30	1	PARAFFIN
3	43205	PROPYLENE	42. 08	5. 10	5. 23	2	OLEFIN
4	43202	ETHANE	30. 07	4. 10	5. 88	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP	OUNDS OF CLASSIFICATION 1	44, 14	90. 80	88. 89		
	1 COMP	OUNDS OF CLASSIFICATION 2	42. 08	5. 10	5. 23		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	4. 10	5. 88		
	4 COMP	OUND COMPOSITE	43. 20	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 2. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environ-mental Protection Agency, Research Triangle Park, NC AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. "Joint District, Federal and State Project for the Evaluation of Refinery Emissions," Report No. 8, June 1958, Los Angeles APCD.
- 5. "Joint District, Federal and State Project for the Evaluation of Refinery Emissions, Report No. 2, March 1957, Los Angeles APCD.

3-06-008B MISCELLANEOUS--PUMP SEALS -- COMPRESSOR SEALS

Process Description

There are many different types of pumps and compressors used to transfer liquids and gases in a modern refinery, the scope of which is beyond this report. Respective equipment manufacturers should be consulted for information on specific types of equipment.

Emissions

During normal operation, the only source of emissions from centrifugal pumps and compressors is where the drive shaft passes through the impeller casing. On reciprocating units leakage is possible at the entrance of the connecting rod into the cylinder or fluid chamber. These losses may be vapor or liquid and generally occur when shafts become scarred or move eccentrically, or through failure of the packing or seal faces (Ref. 4).

The rate at which this destruction of seal efficiency progresses depends upon the abrasive and corrosive properties of the product handled and the type of maintenance practiced (Ref. 4).

The emission factors listed below are from a recent refinery hydrocarbon emission study (Ref. 3).

			Composite					
		Emission Factor	Emission Factor					
Device Type	Product	(lb/day·device)	(lb/seal·day)					
Pump Seals								
Mechanical	<26 RVP	0.3	1.5					
Mechanical	>26 RVP	7.0	1.3					
Packed	<26 RVP	0.4	11.0					
Packed	>26 RVP	40.0	11.0					
Pumps (Composite		-	3.0					
Compressor (comp	28.0 lb/unit'day)							
RVP - Reid Vapor Pressure								

^{*}Based on a petroleum production gas plant compressor site.

The organic species estimated to be emanating from these leaks are presented in Profiles 3-06-008P through X for pumps and Profiles 3-06-008Y and Z for compressors.

Controls

Controls generally take the form of:

- . proper choice of sealing mechanism
- . selection of improved sealing materials
- . proper maintenance, and
- . venting of compressor glands to a vapor recovery system or flare

Profile Basis 3

Profiles are based on data from tests performed on approximately 80 pumps and 10 compressors. Gas samples were taken on a select few based on leak rate and process fluid composition. Samples were taken by grab samples (gas collecting bottle) or by gas collecting bottles plus NIOSH type charcoal tubes.

Emission rates were determined through the use of soap sprayers and tenting techniques. Consult Reference 3 for further details.

Data Qualification

The above mentioned profiles and emission rates may be used to characterize the fugitive emissions resulting from leaky pump and compressor seals in a refinery. Each profile identifies the applicable process fluid associated with the leaking pump or compressor seal.

TABLE 3-06-008P

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PUMP SEALS, COMPOPSITE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0321

LINE SAROAD MOLECULAR PERCENT CHEMICAL CHEMICAL PERCENT NO. CODE NAME WEIGHT WEIGHT VOLUME CLASSIFICATION 1 43105 ISOMERS OF HEXANE **B6.17** 5. 50 4. 72 PARAFFIN 2 43106 ISOMERS OF HEPTANE 100.20 3.03 4. 10 1 PARAFFIN 3 43107 ISOMERS OF OCTANE 114.23 2. 80 1.85 PARAFFIN 1 43108 ISOMERS OF NONANE 128, 25 3. 10 1.77 PARAFFIN 1 5 43109 ISOMERS OF DECANE 142.28 1.90 . 96 PARAFFIN 1 43115 C-7 CYCLOPARAFFINS 98.19 1.10 . 81 1 PARAFFIN 7 43116 C-8 CYCLOPARAFFINS 112.23 . 10 . 07 PARAFFIN 1 8 43117 C-9 CYCLOPARAFFINS . 44 126, 26 . 80 PARAFFIN 7 43122 6. 60 ISOMERS OF PENTANE 72. 15 6.79 **PARAFFIN** 10 43204 PROPANE 44.09 3.70 6. 20 PARAFFIN 11 43212 N-BUTANE 58. 12 7. 90 10.04 **PARAFFIN** 1 43212 N-BUTANE 12 58. 12 . 20 . 22 **PARAFFIN** 13 43214 ISOBUTANE 58. 12 . 80 1.03 1 PARAFFIN 14 43220 N-PENTANE 72. 15 11.10 11.37 PARAFFIN 15 43231 N-HEXANE 86.17 11.00 9, 45 **PARAFFIN** 1 43232 8. 50 6. 27 16 N-HEPTANE 100.20 PARAFFIN 17 43233 114, 23 7. 75 PARAFFIN N-OCTANE 12.00 1 43235 3. 90 2. 21 **PARAFFIN** 18 N-NONANE 128. 25 19 43238 142.28 PARAFFIN N-DECANE 5. 10 2.66 1 20 43248 CYCLOHEXANE 84. 16 . 50 . 44 1 PARAFFIN 21 45102 ISOMERS OF XYLENE 106. 16 1.30 . 89 3 AROMATIC 22 45202 92. 13 3.00 2.44 3 AROMATIC TOLUENE 23 15.20 43201 **METHANE** 16.04 3.30 6 METHANE 24 43202 ETHANE 30.07 1.20 2. 95 NON-REACTIVE 25 45201 BENZENE 78.11 . 50 . 44 NON-REACTIVE 100.00 TOTAL 100.00 85.74 90.70 78.08 20 COMPOUNDS OF CLASSIFICATION 1 O COMPOUNDS OF CLASSIFICATION 2 . 00 . 00 . 00 2 COMPOUNDS OF CLASSIFICATION 3 95.88 4. 30 3. 33

NOTES: METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

. 00

. 00

16.04

36, 31

73. 81

. 00

. 00

3. 30

1.70

100.00

. 00

. 00

15. 20

100.00

3.39

REFERENCES: KVB TEST DATA (REF. 3)

O COMPOUNDS OF CLASSIFICATION 4

O COMPOUNDS OF CLASSIFICATION 5

1 COMPOUNDS OF CLASSIFICATION 6

2 COMPOUNDS OF CLASSIFICATION 7

25 COMPOUND COMPOSITE

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PUMP SEALS, STRAIGHT RUN GASOLINE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0312

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	1. 10	1. 42	1 PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	3. 40	3. 72	1 PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	4. 70	4. 49	1 PARAFFIN
4	43108	ISOMERS OF NONANE	128. 25	8. 50	7. 23	1 PARAFFIN
5	43109	ISOMERS OF DECANE	142. 28	2. 60	1. 97	1 PARAFFIN
6	43115	C-7 CYCLOPARAFFINS	98 . 19	1.00	1. 10	1 PARAFFIN
7	43116	C-8 CYCLOPARAFFINS	112. 23	. 20	. 22	1 PARAFFIN
8	43117	C-9 CYCLOPARAFFINS	126. 26	2. 60	2. 30	1 PARAFFIN
9	43204	PROPANE	44. 09	. 20	. 55	1 PARAFFIN
10	43212	N-BUTANE	58. 12	70	1. 31	1 PARAFFIN
11	43214	ISOBUTANE	58. 12	. 10	. 22	1 PARAFFIN
12	43220	N-PENTANE	72. 15	1. 30	1. 97	1 PARAFFIN
13	43231	N-HEXANE	86. 17	3. 20	4. 05	1 PARAFFIN
14	43232	N-HEPTANE	100. 20	1.40	1. 53	1 PARAFFIN
15	43233	N-OCTANE	114. 23	58. 10	55. 75	1 PARAFFIN
16	43235	N-NONANE	128. 25	2. BO	2. 41	1 PARAFFIN
17	43248	CYCLOHEXANE	B4. 16	. 9 0	1. 20	1 PARAFFIN
18	45102	ISOMERS OF XYLENE	106. 16	2. 60	2. 74	3 AROMATIC
19	45202	TOLUENE	<i>9</i> 2. 13	3. 20	3. 83	3 AROMATIC
20	45201	BENZENE	78. 11	1. 40	1. 97	7 NON-REACTIVE
		TOTAL		100. 00	99. 98	
	17 COMP	DUNDS OF CLASSIFICATION 1	111. 27	92. 80	91. 44	
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	2 COMP	DUNDS OF CLASSIFICATION 3	97. 9 8	5. 80	6. 57	
	O COMPO	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMPO	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMPO	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	1 COMPO	DUNDS OF CLASSIFICATION 7	78. 11	1. 40	1. 97	
-	20 COMPO	DUND COMPOSITE	109. 74	100.00	99. 98	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KYB TEST DATA

B. REFERENCES: KVB TEST DATA (REF 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS

PUMP SEALS, CRACKED GASOLINE DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0310

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	B6. 17	12. 80	12. 21	1	PARAFFIN
2	43107	ISOMERS OF OCTANE	114. 23	4. 10	2. 95	1	PARAFFIN
3	43108	ISOMERS OF NONANE	128. 25	. 20	. 16	1	PARAFFIN
4	43115	C-7 CYCLOPARAFFINS	98. 19	3. 90	3. 28	1	PARAFFIN
5	43117	C-9 CYCLOPARAFFINS	126. 26	. 30	. 16	1	PARAFFIN
6	43122	ISOMERS OF PENTANE	72 . 15	21.40	24. 34	1	PARAFFIN
7	43212	N-BUTANE	58. 12	1. 10	1. 56	1	PARAFFIN
8	43220	N-PENTANE	72. 15	19. 70	22. 38	1	PARAFFIN
9	43231	N-HEXANE	86. 17	15. 40	14. 67	1	PARAFFIN
10	43232	N-HEPTANE	100. 20	14. 90	12. 21	1	PARAFFIN
11	43213	BUTENE	56. 10	1.10	1. 64	2	OLEFIN
12	45102	ISOMERS OF XYLENE	106. 16	1. 60	1. 23	3	AROMATIC
13	45202	TOLUENE	<i>9</i> 2. 13	3. 00	2. 70	3	AROMATIC
14	45201	BENZENE	78. 11	. 50	. 49	7	NON-REACTIVE
		TOTAL		100. 00	99. 98		
	10 COMP	DUNDS OF CLASSIFICATION 1	82. 00	93. 80	93. 92		
	1 COMP	DUNDS OF CLASSIFICATION 2	56. 10	1. 10	1. 64		
	2 COMP	DUNDS OF CLASSIFICATION 3	96. 52	4. 60	3. 93		
	O COMPI	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	O COMPI	DUNDS OF CLASSIFICATION 6	. 00	00	. 00		
		DUNDS OF CLASSIFICATION 7	78. 11	. 50	. 49		
•	14 COMP	DUND COMPOSITE	82. 12	100.00	99. 98		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS PUMP SEALS, GAS-DIL STOCK

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE KVB PROFILE KEY 0313

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	MEICHT	VOLUME	CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	6. 00	4. 00	1 PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	16. 00	9, 15	1 PARAFFIN
3	43122	ISOMERS OF PENTANE	72. 15	3. 80	3, 03	1 PARAFFIN
4	43204	PROPANE	44. 09	3. 30	4. 29	1 PARAFFIN
5	43212	N-BUTANE	58. 12	6. 50	6. 40	1 PARAFFIN
6	43220	N-PENTANE	72. 15	7. 40	5. 89	1 PARAFFIN
7	43231	N-HEXANE	86. 17	11.80	7. 83	1 PARAFFIN
8	43232	N-HEPTANE	100. 20	8. 30	4. 75	1 PARAFFIN
9	43233	N-OCTANÉ	114. 23	7. 90	3. 95	1 PARAFFIN
10	43235	N-NONANE	128. 25	4. 40	1. 94	1 PARAFFIN
11	43238	N-DECANE	142. 28	3. 50	1. 43	1 PARAFFIN
12	45202	TOLUENE	92. 13	7. 30	4. 52	3 AROMATIC
13	43201	METHANE	16. 04	10.00	35. 62	6 METHANE
14	43202	ETHANE	30. 07	3. 80	7. 20	7 NON-REACTIVE
		TOTAL		100. 00	100.00	
	11 COMP	DUNDS OF CLASSIFICATION 1	85. 84	78. 90	52, 66	
	O COMP	DUNDS OF CLASSIFICATION 2	00	. 00	. 00	
	1 COMP	DUNDS OF CLASSIFICATION 3	92. 13	7. 30	4, 52	
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	10.00	35, 62	
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	3. 80	7. 20	
-	14 COMPO	OUND COMPOSITE	57. 25	100.00	100.00	

NOTES: A. METHOD CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES KVB TEST DATA (REF. 3)

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS

PUMP SEALS, REFORMATE STOCK DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0314

.INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43204	PROPANE	44. 09	13. 90	21. 33	1 PARAFFIN
2	43212	N-BUTANE	58. 12	24. 70	28. 77	1 PARAFFIN
3	43220	N-PENTANE	72. 15	21. 50	20. 18	1 PARAFFIN
4	43231	N-HEXANE	86. 17	19. 30	15. 17	1 PARAFFIN
5	43232	N-HEPTANE	100. 20	10.00	6. 77	1 PARAFFIN
6	43233	N-OCTANE	114. 23	9. 70	5. 75	1 PARAFFIN
7	43202	ETHANE	30. 07	. 90	2. 03	7 NON-REACTIVE
		TOTAL		100. 00	100. 00	
	6 COMP	OUNDS OF CLASSIFICATION 1	68. 50	99 . 10	97. 97	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	. 90	2. 03	
•	7 COMP	DUND COMPOSITE	67. 72	100.00	100. 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

DECEMBER 14, 1978 TABLE 3-06-008U INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY MISCELLANEOUS

PUMP SEALS, DISTILLATE
DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0311

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	3. 00	2. 45	1 PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	2. 00	1.40	1 PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	3. 00	1. 82	1 PARAFFIN
4	43108	ISOMERS OF NONANE	128. 25	. 50	. 28	1 PARAFFIN
5	43122	ISOMERS OF PENTANE	72. 15	5. 00	4. 83	1 PARAFFIN
6	43204	PROPANE	44. 09	5. 30	B. 40	1 PARAFFIN
7	43212	N-BUTANE	5 8. 12	12. 70	15. 34	1 PARAFFIN
8	43214	ISOBUTANE	58. 12	2. 5 0	3. 01	1 PARAFFIN
9	43220	N-PENTANE	72. 15	11.00	10. 71	1 PARAFFIN
0	43231	N-HEXANE	86. 17	9. 00	7. 28	1 PARAFFIN
1	43232	N-HEPTANE	100. 20	8. 60	6. 02	1 PARAFFIN
.2	43233	N-OCTANE	114. 23	9. 9 0	6. 0 9	1 PARAFFIN
.3	43235	N-NONANE	128. 25	6. 50	3. 57	1 PARAFFIN
4	43238	N-DECANE	142. 28	14. 80	7. 28	1 PARAFFIN
5	43248	CYCLOHEXANE	84. 16	1.00	. 84	1 PARAFFIN
6	43201	METHANE	16. 04	4. 20	18. 35	6 METHANE
7	43202	ETHANE	30. 07	1.00	2. 31	7 NON-REACTIVE
		TOTAL		100. 00	99. 98	
	15 COMP	DUNDS OF CLASSIFICATION 1	83. 76	94. 80	79. 32	
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMPO	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMPO	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	1 COMPO	OUNDS OF CLASSIFICATION 6	16. 04	4. 20	18, 35	
	1 COMPO	DUNDS OF CLASSIFICATION 7	30. 07	1. 00	2. 31	
-	17 COMPO	DUND COMPOSITE	70. 09	100.00	79. 98	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-008-03

INDUSTRIAL PROCESS, PETROLEUM INDUSTY, REFINERY MISCELLANEOUS PUMP SEALS, NAPHTHA

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0315

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEICHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	6. 60	8. 43	1	PARAFFIN
2	43107	ISOMERS OF OCTANE	114. 23	5. 60	5. 37	1	PARAFFIN
3	43108	ISOMERS OF NONANE	128. 25	20. 60	17. 63	1	PARAFFIN
4	43109	ISOMERS OF DECANE	142. 28	16. 60	12. 81	1	PARAFFIN
5	43115	C-7 CYCLOPARAFFINS	98. 19	2. 40	2. 63	1	PARAFFIN
6	43116	C-8 CYCLOPARAFFINS	112. 23	. 60	. 55	1	PARAFFIN
7	43117	C-9 CYCLOPARAFFINS	126. 26	4. 80	4. 16	1	PARAFFIN
8	43122	ISOMERS OF PENTANE	72. 15	. 40	. 66	1	PARAFFIN
9	43204	PROPANE	44. 09	60	1. 53	1	PARAFFIN
0	43212	N-BUTANE	58 . 1 <i>2</i>	. 60	1. 10	1	PARAFFIN
1	43220	N-PENTANE	72. 15	1.30	1. 9 7	1	PARAFFIN
2	43231	N-HEXANE	86. 17	6. 20	7. 8 9	1	PARAFFIN
3	43232	N-HEPTANE	100. 20	1.80	1. 97	1	PARAFFIN
4	43233	N-OCTANE	114. 23	6. 40	6. 13	1	PARAFFIN
5	43235	N-NONANE	128. 25	8. 30	7. 12	1	PARAFFIN
6	43248	CYCLOHEXANE	84. 16	1. 50	1. 97	1	PARAFFIN
フ	45102	ISOMERS OF XYLENE	106. 16	7. 00	7. 23	3	AROMATIC
8	45202	TOLUENE	92. 13	გ. 20	7. 34	3	AROMATIC
9	45201	BENZENE	78. 11	2. 50	3. 50	7	NON-REACTIVE
		TOTAL		100.00	9 9. 99		
	16 COMP	DUNDS OF CLASSIFICATION 1	112. 87	84. 30	81. 92		
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	99. 09	13. 20	14. 57		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 7	78. 11	2. 50	3. 50		
-	19 COMP	DUND COMPOSITE	109. 65	100, 00	99. 99		

26200-795

3.06-34

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-008-03

DECEMBER 14, 1978 TABLE 3-06-008W INDUSTRIAL P

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, MISCELLANEOUS

PUMP SEALS, NATURAL GASOLINE DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0043

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CLA	SSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	12. 80	9. 68	1	PARAFFIN	_
2	43115	C-7 CYCLOPARAFFINS	98 . 19	17. 60	11.63	1	PARAFFIN	
3	43116	C-B CYCLOPARAFFINS	112. 23	5. 40	3. 12	1	PARAFFIN	
4	43122	ISOMERS OF PENTANE	72. 15	10. 50	9. 49	1	PARAFFIN	
5	43204	PROPANE	44. 09	6. 20	9. 16	1	PARAFFIN	
6	43212	N-BUTANE	58. 12	14. 90	16. 63	1	PARAFFIN	
7	43214	ISOBUTANE	58. 12	4. 70	5. 26	1	PARAFFIN	
8	43220	N-PENTANE	72. 15	10. 70	9. 62	1	PARAFFIN	
9	43231	N-HEXANE	86. 17	12. 40	9. 36	1	PARAFFIN	
10	43201	METHANE	16. 04	3. 00	12. 15	6	METHANE	
11	43202	ETHANE	30. 07	1.80	3. 90	7	NON-REACTIVE	
		TOTAL		100. 00	100. 00			
	9 COMP	OUNDS OF CLASSIFICATION 1	73. 71	95. 20	83. 95			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	3. 00	12. 15			
	1 COMP	DUNDS OF CLASSIFICATION 7	30. 07	1. 80	3. 90			
	11 COMP	OUND COMPOSITE	65. 00	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE ADJACENT TO LEAKING SEAL

B. REFERENCES: KVB TEST DATA (REF. 3)

PUMP SEALS, CASOLINE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0038

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	5. 50	4. 72	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	4. 10	3. 02	1	PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	2. BO	1. 84	1	PARAFF IN
4	43108	ISOMERS OF NONANE	128. 25	3. 10	1. 77	1	PARAFFIN
5	43109	ISOMERS OF DECANE	142. 28	1. 90	. 9 6	1	PARAFFIN
6	43115	C-7 CYCLOPARAFFINS	98 . 19	1. 10	. 81	1	PARAFFIN
7	43116	C-8 CYCLOPARAFFINS	112. 23	. 10	. 07	1	PARAFFIN
8	43117	C-9 CYCLOPARAFFINS	126. 26	. 80	. 44	1	PARAFFIN
9	43122	ISOMERS OF PENTANE	72. 15	6. 60	6. 78	1	PARAFFIN
10	43204	PROPANE	44. 09	3. 70	6. 19	1	PARAFFIN
11	43212	N-BUTANE	58. 12	7. 9 0	10. 03	1	PARAFFIN
12	43214	ISOBUTANE	58 . 12	. 80	1.03	1	PARAFFIN
13	43220	N-PENTANE	72. 15	11. 10	11. 36	1	PARAFFIN
14	43231	N-HEXANE	86. 17	11.00	9. 44	1	PARAFFIN
15	43232	N-HEPTANE	100.,20	8. 50	6. 27	1	PARAFFIN
16	43233	N-OCTANE	114. 23	12.00	7. 74	1	PARAFFIN
17	43235	N-NONANE	128. 25	3. 90	2. 21	1	PARAFFIN
18	43238	N-DECANE	142. 28	5. 10	2. 65	1	PARAFFIN
19	43248	CYCLOHEXANE	84. 16	. 50	. 44	1	PARAFFIN
20	43213	BUTENE	5 6. 10	. 20	. 29	2	OLEFIN
21	45102	ISOMERS OF XYLENE	106. 16	1. 30	. 88	3	AROMATIC
22	45202	TOLUENE	9 2. 13	3. 00	2. 43	3	AROMATIC
23	43201	METHANE	16. 04	3. 30	15. 19	6	METHANE
24	43202	ETHANE	30. 07	1. 20	2. 95	7	NON-REACTIVE
25	45201	BENZENE	78. 11	. 50	. 44	7	NON-REACTIVE
		TOTAL		100. 00	99 . 9 5		
	19 COMP	DUNDS OF CLASSIFICATION 1	85. 82	90. 50	77. 77		
		DUNDS OF CLASSIFICATION 2	56. 10	. 20	. 29		
		DUNDS OF CLASSIFICATION 3	95. 86	4. 30	3. 31		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 6	16. 04	3. 30	15. 19		
		DUNDS OF CLASSIFICATION 7	36. 31	1. 70	3. 39		
	25 COMP	DUND COMPOSITE	73. 78	100, 00	99, 95		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE TAKEN AT PUMP SEAL

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-008-03

DECEMBER 14, 1978 TABLE 3-06-008Y INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, MISCELLANEOUS

COMPRESSOR SEALS, WET AND DRY NATURAL GAS

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0044

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43122	ISOMERS OF PENTANE	72. 15	2. 50	. 83	1	PARAFFIN
2	43204	PROPANE	44. 09	18.00	9. 63	1	PARAFFIN
3	43212	N-BUTANE	58. 12	8. 20	3. 33	1	PARAFFIN
4	43214	ISOBUTANE	58. 12	4. 50	1.82	1	PARAFFIN
5	43220	N-PENTANE	72. 15	1.80	. 59	1	PARAFFIN
6	43231	N-HEXANE	86. 17	1.00	. 28	1	PARAFFIN
7	43201	METHANE	16. 04	48. 40	71. 26	6	METHANE
8	43202	ETHANE	30. 07	15. 60	12. 26	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	6 COMP	OUNDS OF CLASSIFICATION 1	51. 61	36. 00	16. 48		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	48. 40	71. 26		
	1 COMP	DUNDS OF CLASSIFICATION 7	30. 07	15. 60	12. 26		
•	8 COMP	DUND COMPOSITE	23. 62	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE TAKEN AT LEAKING SEAL

B. REFERENCES: KVB TEST DATA (REF. 3)

26200-795

DECEMBER 14, 1978 INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, MISCELLANEOUS TABLE 3-06-0082

COMPRESSOR SEALS, REFINERY GAS

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0039

INE	SARGAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	MEIGHT	WEICHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	1.00	. 49	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	. 10	. 04	1	PARAFFIN
3	43122	ISOMERS OF PENTANE	72. 15	8. 60	4. 82	1	PARAFFIN
4	43204	PROPANE	44. 09	16.00	14.70	1	PARAFFIN
5	43212	N-BUTANE	58. 12	23. 20	16. 16	1	PARAFFIN
6	43214	ISOBUTANE	58. 12	10.00	6. 97	1	PARAFF1N
7	43220	N-PENTANE	72. 15	7. 60	4. 25	1	PARAFFIN
8	43231	N-HEXANE	86. 17	4. 60	2. 15	1	PARAFFIN
9	43205	PROPYLENE	42. 08	8. 80	8. 46	2	OLEFIN
0	43213	BUTENE	56. 10	1. 20	. 8 5	2	OLEFIN
1	43201	METHANE	16. 04	13. 30	33. 58	6	METHANE
2	43202	ETHANE	30. 07	5. 60	7. 53	7	NON-REACTIVE
		TOTAL		100.00	100. 00		
	8 COMP	OUNDS OF CLASSIFICATION 1	58. 05	71. 10	49. 58		
		OUNDS OF CLASSIFICATION 2	43. 36	10.00	9. 31		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	13. 30	33. 58		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	5. 60	7. 53		
	12 COMP	OUND COMPOSITE	40. 47	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE TAKEN AT LEAKING SEAL

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-008-04

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. "Joint District, Federal and State Project for the Evaluation of Refinery Emissions," Report No. 6, March 1957, Los Angeles APCD.

3-06-009 WASTE GAS FLARES

Process Description 4

Modern refining processes produce large quantities of hydrocarbon gases. Every refinery must be equipped to handle excess gas production which is produced intermittently, and large volumes of hydrocarbon vapors produced very rapidly from process units during emergenices. A number of devices are utilized to recover these gases but facilities for the ultimate disposal of excess vapors are usually necessary. This is generally accomplished by combustion in waste gas flares although small amounts may be vented to the fireboxes of heaters or boilers.

The objective of combustion in a waste gas flare is the oxidation of the hydrocarbon vapors to carbon dioxide and water without the production of smoke and objectionable odors. The three common types of waste gas flares are: elevated flares, ground level flares, and open pit flares. The two major types of elevated waste gas flares used by refineries located in Los Angeles County are the air-aspirating venturi flare and the steam-injection flare, both of which are smokeless.

Emissions

The smokeless flares now in operation at most refineries appear to attain highly efficient combustion of waste gas. Actual field testing of flares is generally not feasible because of safety considerations, the erratic nature of the gas flow to flares and the inaccessibility of most flare tips. A hydrocarbon emission factor of 5 lb/10^3 bbl refinery capacity is reported in AP-42 (Ref. 1).

An estimate of the organic species emitted from a typical refinery waste flare is presented in Profile 3-06-009.

Controls

Based on the design of smokeless flares, essentially complete combustion does occur. Control for flares would amount to the conversion of a smoking type flare to a smokeless type flare, air-aspirating venturi flare or steam injected flare - followed by proper maintenance.

AP40 (Ref. 2) offers a detailed explanation of the design, application limitations of various waste gas flares. Another reference is the API Manual on Disposal of Refinery Wastes (Ref. 5).

Profile Basis 3

Profile 3-06-009 was based on a survey and evaluation of pertinent literature.

Data Qualification

AP-42's 6.0 $1b/10^3$ bbl refinery capacity may be used with discretion.

Profile 3-06-009 may be used with discretion to characterize the VOC's emitted from a refinery waste-gas flare.

DECEMBER 14, 1978 TABLE 3-06-009

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY

FLARES, NATURAL GAS

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0051

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43204	PROPANE	44. 09	30. 00	18. 94	1	PARAFFIN
2	43502	FORMALDEHYDE	30. 03	20.00	18. 55	4	CARBONYL
3	43201	METHANE	16. 04	20.00	34, 73	6	METHANE
4	43202	ETHANE	30. 07	30.00	27. 7 9	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	1 COMP	OUNDS OF CLASSIFICATION 1	44. 09	30. 00	18. 94		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 4	30. 03	20.00	18, 55		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	20, 00	34, 73		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	30. 00	27. 7 9		
	4 COMP	OUND COMPOSITE	27. 85	100. 00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. "Joint District, Federal and State Project for the Evaluation of Refinery Emissions," Report No. 8, June 1958, Los Angeles APCD.
- 5. "Manual on Disposal of Refinery Wastes," Volume II, American Petroleum Institute, Div. of Refining, Washington, D.C., 5th Ed., 1957.

3-06 INDUSTRIAL PROCESS, PETROLEUM INDUSTRY

3-06-013 CATALYTIC REFORMING.

Process Description

In reforming processes, a feed stock of gasoline undergoes molecular rearrangement by means of catalysis (usually including hydrogen removal) to produce a gasoline of higher quality and higher octane number. In various fixed-bed and fluidized-bed processes, the catalyst is regenerated continuously in a manner similar to that used with cracking units.

Emissions

Hydrocarbon emissions emanating from fixed bed catalytic reforming units result mainly from leaking peripheral equipment such as valves, fittings, flanges and pump seals.

Emission rates for valves, flanges, pumps and fittings are discussed in Section 3-06-008A and B.

The hydrocarbon specie emissions data applicable to the above mentioned equipment associated with a catalytic reforming unit are presented in Profile 3-06-013.

Controls

See Section 3-06-008A and B.

Profile Basis

Data contained within profile 3-06-013 were developed through an engineering evaluation of similar test data--leaky catalytic reformer unit valve and a leaky crude heater pump seal.

Data Qualification

Profile 3-06-013 may be used to charaterize the hydrocarbon emissions data from a catalytic reformer and associated peripheral equipment.

INDUSTRIAL PROCESS, PETROLEUM INDUSTRY, REFINERY CATALYTIC REFORMER, GENERAL FUGITIVE EMISSIONS DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0053

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	. 80	. 44	1	PARAFFIN
2	43107	ISOMERS OF OCTANE	114. 23	70	. 29	1	PARAFFIN
3	43108	ISOMERS OF NONAME	128. 25	2. 50	. 97	1	PARAFFIN
4	43109	ISOMERS OF DECANE	142. 28	2. 00	. 68	1	PARAFFIN
5	43115	C-7 CYCLOPARAFFINS	9 8. 19	. 30	. 15	1	PARAFFIN
5	43117	C-9 CYCLOPARAFFINS	126. 26	. 60	. 24	1	PARAFFIN
7	43122	ISOMERS OF PENTANE	72. 15	90	. 63	1	PARAFFIN
8	43204	PROPANE	44. 09	26. 70	29. 30	1	PARAFFIN
9	43212	N-BUTANE	58. 12	22. 60	18. 81	1	PARAFFIN
10	43214	ISOBUTANE	58. 12	20. 70	17. 21	1	PARAFFIN
11	43220	N-PENTANE	72. 15	20	. 15	1	PARAFFIN
12	43231	N-HEXANE	86. 17	. 80	. 44	1	PARAFFIN
13	43232	N-HEPTANE	100. 20	. 20	. 10	1	PARAFFIN
14	43233	N-OCTANE	114. 23	. 80	. 34	1	PARAFFIN
15	43235	N-NONANE	128. 25	1.00	. 39	1	PARAFFIN
16	43248	CYCLOHEXANE	84. 16	. 20	. 10	1	PARAFFIN
17	45102	ISOMERS OF XYLENE	106. 16	. 80	. 39	3	AROMATIC
18	45202	TOLUENE	<i>9</i> 2. 13	. 80	. 44	3	ARDMATIC
19	43201	METHANE	16. 04	. 90	2. 71	6	METHANE
20	43202	ETHANE	30. 07	16. 20	26. 06	7	NON-REACTIVE
21	45201	BENZENE	78. 11	. 30	. 19	7	NON-REACTIVE
		TOTAL		100.00	100. 03		
	16 COMP	DUNDS OF CLASSIFICATION 1	55. 87	81. 00	70. 24		
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	98. 72	1. 60	. 83		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 6	16, 04	. 90	2.71		
		OUNDS OF CLASSIFICATION 7	30. 42	16. 50	26. 2 5		
-	21 COMP	DUND COMPOSITE	48. 46	100.00	100. 03		

26200-795

3.06-43

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 3-06-013-01

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.

3.06-44 26200-795

3-30 INDUSTRIAL PROCESS, TEXTILE MANUFACTURING

3-30-001 GENERAL FABRICS--YARN DYEING

Process Description

Fabric dyeing involves the chemical bonding of a dyeing agent to the fabric to produce a durable color change. In general, fabrics are dyed in vats or spray chambers.

Emissions 1

Evaporation of the low boiling point solvents in the chemical dye formulation may result in the emission of volatile organic vapors to the atmosphere if left uncontrolled. A thorough discussion of the methods employable to calculate the VOC emission rates from various solvent based operations is presented in Section 12 of AP-40 (Ref. 1).

Profile 3-30-001 presents the estimated volatile organic specie emissions from a textile dyeing operation.

<u>Controls</u>¹

As in all solvent vapor control situations, control generally takes the form of condensers, air curtains or lids, charcoal adsorbers, or afterburners. Each situation and allowable emission limits dictate generally what control avenue to pursue. A thorough discussion of solvent vapor control equipment is offered in Section 12 of AP-40 (Section 1).

Profile Basis²

Profile 3-30-001 is based on an engineering evaluation of a textile dyeing manufacturer's solvent use and control questionnaire.

Data Qualifications

Profile 3-30-001 may be used with discretion to characterize the volatile organic emissions from an uncontrolled textile dyeing operation.

INDUSTRIAL PROCESS, TEXTILE MANUFACTURING

FABRIC DYEING, GENERAL DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0060

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43551	ACETONE	58. 08	20. 00	21. 64	4 CARBONYL
2	43552	METHYL ETHYL KETONE	72. 10	21. 40	18. 68	4 CARBONYL
3	43560	METHYL ISOBUTYL KETONE	100. 16	8. 60	5. 41	4 CARBONYL
4	43301	METHYL ALCOHOL	32. 04	5. 40	11.01	5 MISCELLANEOUS
5	43302	ETHYL ALCOHOL	46. 07	4. 50	6. 16	5 MISCELLANEOUS
6	43304	ISOPROPYL ALCOHOL	60. 09	16. 40	17. 17	5 MISCELLANEOUS
7	43305	N-BUTYL ALCOHOL	74. 12	23. 50	19. 94	5 MISCELLANEOUS
		TOTAL		100. 00	100. 01	
	O COMP	DUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	3 COMP	DUNDS OF CLASSIFICATION 4	68. 79	50.00	45. 73	
	4 COMP	DUNDS OF CLASSIFICATION 5	57. 96	50. 00	54. 28	
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
_						
	7 COMP	DUND COMPOSITE	62. 91	100. 00	100. 01	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: FABRIC DYE MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 3-30-001-99

3.30-2

REFERENCES

- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- Taback. H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978

3.30-3 26200-795

3-90 INDUSTRIAL PROCESS, IN PROCESS FUEL

3-90-007 PROCESS--COKE OVEN GAS

Process Description 1

Approximately 40% by volume of the byproduct given off during the destructive distillation of the coal in a coproduct type coking oven is reused as fuel. The byproduct gas is first stripped of its coproducts and then returned and burned for the under firing of the coke oven batteries. (See Section 3-03-003.) Coproduct ovens generally hold from 16 to 24 tons of coal and are built in batteries of 10 to 100 ovens. Once the coke ovens are fired, they are not allowed to cool down unless their replacement is required.

Emissions²

Hydrocarbon emissions result from the burning of the stripped coke oven gas for the under firing of the coke batteries. The combustion exhaust gases from each oven are manifolded together and vented through a common stack.

An average hydrocarbon emission rate of 0.6 $1b/10^3$ ACF coke oven gas fired at a HHV of 550 Btu/ft³ was reported. The coke oven battery tested consisted of 45 individual coke ovens with a reported fuel rate of 150,000 CFH, a stack gas flow rate of 2.4×10^6 SCFM and a HC emissions rate of 8.8 1b/hr. Approximately 83% by weight of this is methane (Ref. 2).

Consult Reference 3 for further emissions information.

Profile 3-90-007 presents the hydrocarbon emission species detected (Ref. 2).

Controls 3

Controls associated with coking operations are generally concerned with the visible and invisible emissions connected with the destructive distillation of the coal and not from the under firing of the coke batteries. Improvement of combustion efficiency would be a proper method of control.

Profile Basis²

The hydrocarbon species data presented in Profile 3-90-007 represents test data obtained from a typical coproduct type coke oven battery. Gas samples were taken from within the exhaust stack servicing 45 ovens under fired with stripped coke oven gas. The gas collection train consisted of a glass collecting bottle followed by NIOSH type charcoal tubes.

Process information such as fuel flow and exhaust gas flow rate were obtained from the steel manufacturer.

Data Qualification

This profile may be used to characterize the hydrocarbon emissions from a typical coproduct type coke oven using stripped coke oven gas for under firing of the coke batteries. This profile does not represent the fugitive unstripped coke oven gas emissions that occur in a coproduct type battery -- leaky door and lid seals, leaky valves, fittings, etc. For this type of specie data, refer to Profile 3-03-003.

INDUSTRIAL PROCESS, INPROCESS FUEL, PROCESS GAS

COKE OVEN BLAST-FURNACE GAS DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0217

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WE IGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43203	ETHYLENE	28. 05	2. 80	2. 86	2	OLEFIN
2	43205	PROPYLENE	42. 08	5. 50	3. 75	2	OLEFIN
3	43213	BUTENE	56. 10	6. 40	3. 26	2	OLEFIN
4	43201	METHANE	16. 04	40. 90	73. 00	6	METHANE
5	43202	ETHANE	30. 07	1. 40	1. 35	7	NON-REACTIVE
6	45201	BENZENE	78. 11	43. 00	15 . 77	7	NON-REACTIVE
		TOTAL		100. 00	99. 99		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	3 COMP	OUNDS OF CLASSIFICATION 2	42, 65	14, 70	9. 87		
	O COMP	DUNDS OF CLASSIFICATION 3	. 0 0	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16.04	40. 90	73. 00		
	2 COMP	OUNDS OF CLASSIFICATION 7	74. 32	44. 40	17. 12		
	6 COMP	OUND COMPOSITE	28. 65	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 2)

REFERENCES

- 1. Shreve, R. N. and Brink, J. A., Jr., "Chemical Process Industries," 4th Ed., McGraw-Hill Book Co., 1977.
- 2. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vols. I and II, KVB, Inc., Tustin, CA, June 1978.
- 3. Coke Oven Emissions, Miscellaneous Emissions and Their Control at Kaiser Steel Corporation's Fontana Steel Making Facility, State of California, Air Resources Board, Report No. L&E-76-11.
- 4. Danielson, J. A. (ed.), "Air Pollution Engineering Manual, Environ-mental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.

POINT SOURCE EVAPORATION

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	r (/ L	- U.	_L_/ 111	1110	JUL		

- 4-02 SURFACE COATING
- 4-03 PETROLEUM PRODUCT STORAGE
 INCLUDING MARINE TERMINAL 4-06
- 4-05 PRINTING PRESS

4-01 POINT SOURCE EVAPORATION, CLEANING SOLVENT

4-01-001 DRY CLEANING--TRICHLOROETHANE

--STODDARD SOLVENT

--PERCHLOROETHYLENE

Process Description

Dry cleaning involves the cleaning of fabrics with non-aqueous organic solvents. The dry cleaning process requires three steps: (1) washing the fabric in solvent, (2) spinning to extract excess solvent, and (3) drying by tumbling in a hot airstream.

Two general types of cleaning fluids are used in the industry: petroleum solvents and synthetic solvents. Petroleum solvents, such as Stoddard or 140-F, are inexpensive, combustible hydrocarbon mixtures similar to kerosene. Operations using petroleum solvents are known as petroleum plants. Synthetic solvents are nonflammable but more expensive halogenated hydrocarbons. Perchloroethylene and trichlorotrifluoroethane are the two synthetic dry cleaning solvents presently in use. Operations using these synthetic solvents are called "perc" plants and fluorocarbon plants, respectively.

There are two basic types of dry cleaning machines: transfer and dry-to-dry. Transfer machines accomplish washing and drying in separate machines. Usually the washer extracts excess solvent from the clothes before they are transferred to the dryer, however, some older petroleum plants have separate extractors for this purpose. Dry-to-dry machines are single units that perform all of the washing, extraction, and drying operations. All petroleum solvent machines are the transfer type, but synthetic solvent plants can be either type.

For further information on the dry cleaning industry and a typical dry cleaning cycle, refer to References 1 and 2.

Emissions

The solvent material itself is the primary emission of concern from dry cleaning operations. Solvent is given off by the washer, dryer, solvent still, muck cooker, still residue, and filter muck storage areas, as well as leaky pipes, flanges, and pumps.

Typical emission factors for dry cleaning plants are presented in Section 4.1-1 of AP-42 (Ref. 1). Respective profiles are presented at the end of this section.

Controls

Petroleum plants have generally not employed solvent recovery because of the low cost of petroleum solvents and the fire hazards associated with collecting vapors. Some emission control, however, can be obtained by maintaining all equipment in good condition (e.g., preventing lint accumulation, preventing solvent leakage, etc.) and by using good operating practices (e.g., not overloading machinery). Both carbon adsorption and incineration appear to be technically feasible controls for petroleum plants, but costs are high.

Solvent recovery is necessary in "perc" plants due to the higher cost of perchloroethylene. Recovery is effected on the washer, dryer, still, and muck cooker through the use of condensers, water/solvent separators, and carbon adsorption units. Periodically (typically once a day), solvent collected in the carbon adsorption unit is desorbed with steam, condensed, separated from the condensed water, and returned to the pure solvent storage tank. Residual solvent emitted from treated distillation bottoms and muck is not recovered. As in petroleum plants, good emission control can be obtained by good housekeeping practices (maintaining all equipment in good condition and using good operating practices).

All fluorocarbon machines are of the dry-to-dry variety to conserve solvent vapor, and all are closed systems with built-in solvent recovery. High emissions can occur, however, as a result of poor maintenance and operation of equipment. Refrigeration systems are installed on newer machines to recover solvent from the washer/dryer exhaust gases.

Profile Basis³

As mentioned earlier, the cleaning solvent material itself is the primary emission pollutant. The data contained in the profile for stoddard solvent represents actual test results obtained from a steam heated, tumble dryer at 180°F. The vapor collection train was composed of a gas collecting bottle followed by NIOSH type charcoal sampling tubes. Mass flow rates were measured using a standard pitot tube and thermometer.

Profiles for the three common dry cleaning solvents in use are presented at the end of this section. Refer to Profile 4-01-002 F for emissions using trichlorotrifluoroethane (Freon 113).

Data Qualification

The emission factors referred to in Section 4.1-1 of AP-42 specify the conditions under which profile applicability is valid.

DECEMBER 14, 1978 TABLE 4-01-001A

POINT SOURCE EVAPORATION, CLEANING SOLVENT DRY CLEANING, 1, 1, 1-TRICHLOROETHANE DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0087

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43814	1, 1, 1-TRICHLOROETHANE TOTAL	133. 42	100. 00 100. 00	100. 00 100. 00	7	NON-REACTIVE	
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	1 COMP	DUNDS OF CLASSIFICATION 7	133. 42	100. 00	100. 00			
	1 COMP	DUND COMPOSITE	133. 42	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-01-001-99, 4-01-002-02

DECEMBER 14, 1978 TABLE 4-01-001B

POINT SOURCE EVAPORATION, CLEANING SOLVENT

DRY CLEANING, STODDARD SOLVENT DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0086

LINE	SAROAD		CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE		NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATI ON	
1	43107	ISOMERS	OF OCTANE	114. 23	. 80	. 97	1	PARAFFIN	
2	43108	ISOMERS	OF NONANE	128. 25	27. 30	29. 42	1	PARAFFIN	
3	43109	ISOMERS	OF DECANE	142. 28	69. 3 0	67. 27	1	PARAFFIN	
4	43110	ISOMERS	OF UNDECANE	156. 30	2. 60	2. 35	1	PARAFFIN	
			TOTAL		100. 00	100. 01			
	4 COMP	OUNDS OF	CLASSIFICATION 1	138. 21	100. 00	100. 01			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 0 0	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	4 COMP	OUND COM	POSITE	138. 21	100. 00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-01-001-02, 4-02-002-01

D. See Appendix I-F

4.01-5

DECEMBER 14, 1978 TABLE 4-01-001C

POINT SOURCE EVAPORATION, CLEANING SOLVENT DRY CLEANING, PERCHLOROETHYLENE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0085

LINE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43817	PERCHLORO	DETHYLENE TOTAL	165. 83	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF C	CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF C	LASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF C	LASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF C	LASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF C	LASSIFICATION 5	165. 83	100. 00	100.00		
	O COMP	OUNDS OF C	LASSIFICATION 6	. 00	. 00	. 00		
	O COMP	DUNDS OF C	LASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	DUND COMPO	SITE	165. 83	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-01-001-01, 4-01-002-03

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (Ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC., AP-40, May 1973.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II., KVB, Inc., Tustin, CA, June 1978.

4-01 POINT SOURCE EVAPORATION, CLEANING SOLVENT

4-01-002 DEGREASING--TRICHLOROETHANE

--DICHLOROMETHANE

--TRICHLOROETHYLENE

--TOLUENE

--FREON 11

--FREON 113

Process Description 1

During the fabrication of many metal products, surfaces are lubricated with oils, greases, or stearates to facilitate various drawing, forming and machining operations. Degreasing operations are designed to remove this foreign material from the product's surface so that the applied surface coating can adhere properly.

Solvent degreasers vary in size from simple unheated wash basins to large heated conveyorized units in which articles are washed in hot solvent vapors. Solvent is usually vaporized in one portion of a tank either by electric, steam or gas heat. The vapors diffuse and fill that portion of the tank below the water-cooled condenser. At the condenser level, a definite interface between the vapor and air can be observed from the top of the tank. Solvent condensed at this level runs into the collection trough and from there to the clean-solvent receptacle in the other portion of the tank. Articles to be degreased are lowered in baskets into the vapor space of the tank. Solvent vapors condense on the cooler metal parts, and the hot condensate washes oil and grease from the parts. The contaminated condensate drains back into the heated tank from which it can be revaporized. When necessary, dirty parts are hand sprayed with hot solvent by means of a flexible hose and spray pump to aid in cleaning.

Emissions 1-3

Solvent emissions from vapor degreasing occur primarily during loading and unloading of the degreaser. Some solvent escapes from the vapor zone during idling conditions. Daily emissions of a single spray

degreasing booth may vary from a few pounds to 1300 pounds per day. A typical metal cleaning operation using a vapor degreaser can clean 200,000 lbs of metal in one day. The table below presents controlled and uncontrolled hydrocarbon emissions from degreasing operations.

HYDROCARBON EMISSIONS FROM DEGREASING OPERATIONS

Type of		al Cleaned	Metal	200,000 lbs of Cleaned/Day
Operation & Control % Con	trol lbs/ton	kg/m ton	lbs/hr	kg/hr
Degreasing, Uncontrolled	0 1.5	0.75	6.3	2.8
Degreasing, Refrigerated Cooling Coils 30	-60 1.0-0.6	0.5-0.3	4.2-2.5	1.9-1.1
Degreasing, Use of Covers 25	-40 1.1-0.9	0.5-0.05	4.6-3.8	2.1-1.7
Degreasing, Carbon Adsorption 40	-70 0.9-0.5	0.5-0.3	3.8-2.1	1.7-1.0

Source: Ref. 1

Many types of halogenated hydrocarbons are used in various degreasing operations. Profiles 4-01-002A through F characterize many of these solvent emissions (Ref. 2,3).

Profile Basis^{2,3}

Profiles 4-01-002A through F are based on an engineering inspection of the degreasing solvent formulations. The VOC's being emitted are a direct function of the degreasing solvent's formulation.

Data Qualification

Reference 1 should be consulted for additional information on the development and limitations when using the above mentioned emission factors.

Profiles 4-01-002A through F may be used to characterize the VOC emissions from various degreasing operations. Profile selection should be based on the degreasing solvent formulation specified for each profile.

26200-795

DECEMBER 14, 1978 TABLE 4-01-002A

POINT SOURCE EVAPORATION, CLEANING SOLVENT

DEGREASING, TRICHLOROETHANE DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0089

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43820	1, 1, 2-TRICHLORDETHANE TOTAL	131. 66	100. 00 100. 00	100. 00 100. 00	7 NON-REACTIVE
	O COMP	OUNDS OF CLASSIFICATION 1	.00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMPI	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	1 COMP	DUNDS OF CLASSIFICATION 7	131. 66	100. 00	100. 00	
	1 COMP	DUND COMPOSITE	131. 66	100.00	100. 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

POINT SOURCE EVAPORATION, CLEANING SOLVENT

DEGREASING, DICHLOROMETHANE DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0275

INE NO.	SARDAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43802	DICHLOROMETHANE TOTAL	84. 94	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	84. 94	100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COMPOSITE	84. 94	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

DECEMBER 14, 1978

TABLE 4-01-002C

POINT SOURCE EVAPORATION, CLEANING SOLVENT

DEGREASING, TRICHLOROETHYLENE DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0271

.INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43824	TRICHLORDETHYLENE TOTAL	131. 40	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	131.40	100.00	100.00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
•	1 COMP	DUND COMPOSITE	131. 40	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

4.01-13

DECEMBER 14, 1978 TABLE 4-01-002D

POINT SOURCE EVAPORATION, CLEANING SOLVENT

DEGREASING, TOLUENE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0090

LINE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	45202	TOLUENE		92. 13	100.00	100.00	3	AROMATIC	
			TOTAL		100. 00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 3	92 . 13	100.00	100. 00			
	O COMP	CUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	DUND COME	POSITE	92. 13	100.00	100. 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-01-002-06 (REFER ALSO TO SUMMARY INDEX)

DECEMBER 14, 1978 TABLE 4-01-002E

POINT SOURCE EVAPORATION, CLEANING SOLVENT DEGREASING, TRICHLOROFLUOROMETHANE (FREON 11)

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0088

NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43811	TRICHLO	ROFLUOROMETHANE TOTAL	137. 37	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	DUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	DUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	DUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	DUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	DUNDS OF	CLASSIFICATION 5	137. 37	100.00	100.00		
	O COMP	DUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	DUND COM	POSITE	137. 37	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

DECEMBER 14, 1978 TABLE 4-01-002F

POINT SOURCE EVAPORATON, SURFACE COATING DEGREASING, TRICHLOROTRIFLUOROETHANE (FREON 113) DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0277

.INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43821	TRICHLOROTRIFLUOROETHANE TOTAL	187. 38	100. 00 100. 00	100. 00 100. 00	7 NON-REACTIVE
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 7	187. 38	100. 00	100. 00	
	1 COMP	OUND COMPOSITE	187. 38	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- 2. Sonnichsen, T. W., KVB Engineer.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol I and II, KVB, Inc., Tustin, CA, June 1978.

4-02 POINT SOURCE EVAPORATIONS, SURFACE COATINGS

4-02-001 PAINTS

003 VARNISH/SHELLAC

004 LACQUER

005 ENAMEL

006 PRIMER

007 ADHESIVE

999 WAX, CITRUS COATING

Process Description 1,2

Surface coating operations primarily involve the application of paint, varnish/shellac, lacquer, enamel, or paint primer for surface decoration and/or protection before being marketed. Adhesives as the name implies involve the application of a bonding agent for the purpose of adhering materials together. Coating waxes are applied (sprayed) to various citrus fruits to retard shrivelling, weight loss, and spoilage. A number of basic industrial coating operations are utilized for this purpose, including spraying, flowcoating, roller coating, dipping and electro-coating. There are variations and combinations of these operations, each designed for a special task. The coatings applied in these operations vary widely as to composition and physical properties. A brief discussion of each method of application is presented below. For forther information, consult AP-40 (Ref. 1).

In order to accelerate the drying of the solvents used in the surface coatings, a drying or baking operation is usually an integral part of the basic coating process. This drying process is described in Section 4-02-008.

Spraying—In spraying operations, a coating from a supply tank is forced, usually by compressed air, through a "gun" which is used to direct the coating as a spray upon the article to be coated. Many spraying operations are conducted in a booth or enclosure vented by a fan to protect the health and safety of the spray gun operator by ensuring that explosive and toxic concentration levels of solvent vapors do not develop.

Flowcoating—In flowcoating operations, a coating is fed through overhead nozzles so as to flow in a steady stream over the article to be coated, which is suspended from a conveyor line. Excess paint drains from the article to a catch basin from which it is recirculated by a pump back to the flow nozzles. Impinging heated air jets aid in the removal of superfluous coating and solvent from the coated article prior to its entering an oven for baking.

Flowcoating is used on articles which cannot be dipped because of their buoyancy, such as fuel-oil tanks, gas cylinders, pressure bottles, etc.

Roller Coating—Roller coating machines are similar to printing presses in principle. The machines usually have three or more power-driven rollers. One roller runs partially immersed in the coating and transfers the coating to a second, parallel roller. The strip or sheet to be coated is run between the second and third roller and is coated by transfer of coating from the second roller. The quantity of coating applied to the sheet or strip is established by the distance between the rollers.

Dipping-Dip tanks are simple vessels which contain a working supply of coating material. They usually are equipped with a close-off lid and a drainage reservoir, which are activated in case of fire. The object to be coated is immersed in the coating material long enough to be coated completely and then removed from the tank. Provision is made to drain the excess coating from the object back to the tank, either by suspending the work over the tank or by using drain boards that return the paint to the dip tank.

Electro-coating--Electrocoating, a variation of the ordinary dip tank process of coating, is the electrodeposition of resinous materials on surfaces. This operation is accomplished in water solutions, suspensions, or dispersions. In the electrocoating process, the object being coated is the anode and the tank containing the dilute solution, suspension or dispersion of film-forming materials usually is the cathode.

Emissions

Spraying--Paint Spray Booths, Citrus Coating Wax--The discharge from a paint spray booth consists of particulate matter and organic solvent vapors. The organic solvent vapors are from the organic solvent, diluent, or thinner which is used with the coating and they evaporate from coating suspended in the airstream, from the target of the spraying, or from the inside surfaces of the booth and its accessories. The choice of the spraying method, air atomization, electrostatic, or other, is a factor in determining the amount of overspray which is the amount of sprayed coating which misses the article being coated.

Solvent concentrations in spray booth effluents generally range from 100 to 200 ppm. Solvent emissions from the spray booth stacks vary widely with extent of operation, from less than 1 to over 3,000 pounds per day.

VOC emission rates can be estimated based on the fact that although organic solvents have different evaporation rates, VOC emissions by flash-off can be estimated at various times during the spraying operation from the specific composite chemical composition. Details of this method are contained in AP-40 (Ref. 1).

Flowcoating, Roller Coating, Dipping and Electrocoating—Air pollutants from flowcoating, roller coating, dipping and electrocoating exist only in the form of organic solvent vapors since no particulate matter is formed. Solvent emission rates from these operations may also be estimated by the methods mentioned in AP-40 for surface coating operations (Ref. 1).

VOC Specie Emissions, Surface Coatings--The VOC specie emissions per application method as previously mentioned, are derived from the organic solvent, diluent, or thinner used with the coating. Presented below is an itemized list of surface coating profiles and their respective surface coating applications:

- . Profile 4-02-001 A through D: Paints
- . Profile 4-02-003 A through E: Varnish/Shellac
- . Profile 4-02-004 A through F: Lacquer
- . Profile 4-02-005 A through E: Enamel
- . Profile 4-02-006 A through K: Primer
- . Profile 4-02-007 A through E: Adhesive
- . Profile 4-02-999 A through C: Citrus Coating Wax

4.02 - 3

Controls 1,2,3

Organic solvents used in coatings and thinners are not controllable by filters, baffles, or water curtains. Solvent vapors can be controlled or recovered by the application of condensation, compression, absorption, adsorption, or combustion principles, when necessary for either economic or regulatory requirements.

Control efficiencies of 90% or greater are possible by adsorption using activated carbon, provided particulates are removed from the contaminated airstream by filtration before the airstream enters the carbon bed. Incineration is necessitated when an organic vapor is determined to be photochemically reactive and solvent recovery is not desired.

Process Modification

Appropriate substitutes for organic solvent-borne coatings exist in the increasingly popular water-borne coatings. Generally known as water-based paints or latexes, they have lower organic solvent content than the high organic solvent coatings. In the case of citrus wax coatings, one manufacturer reported that by increasing the solids content of the coating solvent a reduction in total organic emissions was achieved.

Profile Basis³

A majority of the above mentioned profiles are based on an engineering evaluation of questionnaires sent out to respective surface coaters in the Los Angeles area. Those based on test data involved the use of gas collecting bottles plus NIOSH type charcoal tubes. The particular method employed in developing the surface coating profile is so stated in the reference information at the bottom of each profile.

Data Qualification

AP-40 should be consulted for the application and limitations involved when evaluating emission factors for surface coating operations. The above mentioned profiles may be used to characterize the VOC emissions from the specified surface coatings as identified on each profile data table.

POINT SOURCE EVAPORATION, SURFACE COATING PAINT, POLYMERIC (HOT AIR DRIED)

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0125

INE NO.	SARDAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CI A	CHEMICAL ASSIFICATION	
	0022	NA IL	W_13.77	WL 19171	VOLUME	U	10011 10/// 10//	
1	45102	ISOMERS OF XYLENE	106. 16	22. 30	18. 07	3	AROMATIC	
2	45202	TOLUENE	92 . 13	55. 10	51. 46	3	AROMATIC	
3	43551	ACETONE	58. 08	12. 20	18. 07	4	CARBONYL	
4	43552	METHYL ETHYL KETONE	72. 10	10. 40	12. 39	4	CARBONYL	
		TOTAL		100. 00	99. 99			
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	2 COMP	OUNDS OF CLASSIFICATION 3	95. 78	77. 40	69. 53			
	2 COMP	OUNDS OF CLASSIFICATION 4	63. 78	22. 60	30. 46			
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 0 0	. 00			
	4 COMP	OUND COMPOSITE	86. 03	100.00	99. 99			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: PAPERBOARD MANUFACTURER

POINT SOURCE EVAPORATION, SURFACE COATING PAINT SOLVENT, ACETONE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0219

NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43551	ACETONE		58. 08	100. 00	100. 00	4 CARBONYL
			TOTAL		100. 00	100. 00	
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00	
	1 COMP	OUNDS OF	CLASSIFICATION 4	58. OB	100.00	100.00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	00	. 00	
,	1 COMP	DUND COMF	POSITE	58. 08	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT INFORMATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-001-02 (REFER ALSO TO SUMMARY INDEX)

POINT SOURCE EVAPORATION, SURFACE COATING PAINT SOLVENT, ETHYL ACETATE DATA CONFIDENCE LEVEL. II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0220

ND.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43433	ETHYL ACETATE TOTAL	88. 10	100. 00 100. 00	100. 00 100. 00	5 MISCELLANEOUS
		_				
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 5	88. 10	100.00	100. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	OUND COMPOSITE	88. 10	100.00	100. 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-001-03 (REFER ALSO TO SUMMARY INDEX)

DECEMBER 14, 1978 TABLE 4-02-001D

POINT SOURCE EVAPORATION, SURFACE COATING PAINT SOLVENT, METHYL ETHYL KETONE DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0221

.INE NO.	SARDAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43552	METHYL.	ETHYL KETONE	72. 10	100.00	100. 00	4	CARBONYL	
			TOTAL		100. 00	100. 00			
	O COMP	DUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	DUNDS DI	CLASSIFICATION 3	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 4	72. 10	100.00	100. 00			
	O COMP	DUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	DUND COM	POSITE	72. 10	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-001-04 (REFER ALSO TO SUMMARY INDEX)

POINT SOURCE EVAPORATION, SURFACE COATING

VARNISH/SHELLAC, GENERAL DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0127

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATI ON
1	43551	ACETONE	58. 08	38. 70	45. 68	4	CARBONYL
2	43552	METHYL ETHYL KETONE	72. 10	41.60	39. 57	4	CARBONYL
3	43560	METHYL ISOBUTYL KETONE	100.16	16. 70	11. 45	4	CARBONYL
4	43367	GLYCOL ETHER	62. 07	3. 00	3. 29	5	MISCELLANEOUS
		TOTAL		100. 00	99. 99		
	о спме	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 2	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	68. 80	97. 00	96, 70		
	1 COMP	OUNDS OF CLASSIFICATION 5	62. 07	3, 00	3, 29		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	4 COMP	OUND COMPOSITE	68. 5 <u>8</u>	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA

ENGINEERING EVALUATION OF GUESTIONNAIRE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 3)

POINT SOURCE EVAPORATION, SURFACE COATING VARNISH/SHELLAC, VARNISH RESIN DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0132

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43551	ACETONE	58, 08	40. 00	46. 12	4	CARBONYL
2	43310	METHYL CELLOSOLVE	76 . 11	30. 00	26. 37	5	MISCELLANEOUS
3	43450	DIMETHYLFORMAMIDE	73. 09	30. 00	27. 51	5	MISCELLANEOUS
		TOTAL		100.00	100. 00		
	О СОМР	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 4	58, 08	40.00	46. 12		
	2 COMP	OUNDS OF CLASSIFICATION 5	74, 57	60.00	53. 88		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
		OUND COMPOSITE	66. 96	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: PRINTED CIRCUIT BOARD MANUFACTURER

DECEMBER 14, 1978

TABLE 4-02-003C

POINT SOURCE EVAPORATION, SURFACE COATING VARNISH/SHELLAC, LXH-221 AIR CONVERTING VARNISH DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0278

.INE NO.	SARDAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
MU.	CODE	MARIE	MCIGHI	MC1641	AOCOME	CL	HOSTFICATION
1	43233	N-OCTANE	114, 23	1. BO	1. 48	1	PARAFFIN
2	45102	ISOMERS OF XYLENE	106. 16	2. 30	2. 03	3	AROMATIC
3	45202	TOLVENE	92. 13	35. 50	35. 58	3	AROMATIC
4	43305	N-BUTYL ALCOHOL	74, 12	21.00	26. 16	5	MISCELLANEOUS
5	43435	N-BUTYL ACETATE	116. 16	3. 20	2. 59	5	MISCELLANEOUS
6	43444	ISOPROPYL ACETATE	104.00	36. 20	32. 16	5	MISCELLANEOUS
		TOTAL		100.00	100. 00		
	1 COMP	DUNDS OF CLASSIFICATION 1	114, 23	1.80	1. 48		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	2 COMP	OUNDS OF CLASSIFICATION 3	9 2. 89	37. BO	37. 61		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	3 СО М Р	OUNDS OF CLASSIFICATION 5	91.68	60. 40	60. 91		
	O COMP	CUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
		OUND COMPOSITE	92.47	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: WOOD FURNITURE MANUFACTURER

DECEMBER 14, 1978 TABLE 4-02-003D

POINT SOURCE EVAPORATION, SURFACE COATING, COMPOSITE VARNISH/SHELLAC, A. BROWN CO., BROLITE, MIL-V-173B, TT-V-109B DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0133

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL	
NO.	CODE	NAME	MEIGHT	WEIGHT	VOLUME	CLASSIFICATION	
1	45202	TOLUENE	9 2. 13	20. 00	23. 33	3 AROMATIC	
2	43433	ETHYL ACETATE	88. 10	6. 30	7. 74	5 MISCELLANEOUS	
3	43435	N-BUTYL ACETATE	116. 16	67. 60	62. 58	5 MISCELLANEOUS	
4	43444	ISOPROPYL ACETATE	1 04 . 0 0	6. 10	6. 34	5 MISCELLANEOUS	
		TOTAL		100. 00	99 . 99		
	A COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	92.13	20. 00	23. 33		
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	112.32	80.00	76. 66		
		OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	4 COMP	DUND COMPOSITE	107. 61	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: AIRCRAFT MANUFACTURER

DECEMBER 14, 1978 TABLE 4-02-003E

POINT SOURCE EVAPORATION, SURFACE COATING VARNISH/SHELLAC SOLVENT, XYLENE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0223

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	45102	ISOMERS	OF XYLENE TOTAL	106. 16	100. 00 100. 00	100. 00 100. 00	3	AROMATIC	
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 3	106. 16	100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	OUND COM	POSITE	106. 16	100.00	100. 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGMENT

C. APPLICABLE SCC CATEGORIES: 4-02-003-05, 4-02-009-24

4.02-13

26200-795

DECEMBER 14, 1978 TABLE 4-02-004A

POINT SOURCE EVAPORATION, SURFACE COATING LACQUER, PAPERBOARD PRODUCTS AND CONTAINERS DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0149

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43118	MINERAL SPIRITS	114.00	21. 40	17. 03	1	PARAFFIN
2	45202	TOLUENE	92. 13	9. 40	9. 24	3	AROMATIC
3	43552	METHYL ETHYL KETONE	72. 10	14. 20	17. 84	4	CARBONYL
4	43311	CELLOSOLVE	90. 12	26. 60	26. 72	5	MISCELLANEOUS
5	43433	ETHYL ACETATE	88. 10	28. 40	29 . 17	5	MISCELLANEOUS
		TOTAL		100.00	100.00		
	1 COMP	OUNDS OF CLASSIFICATION 1	114. 00	21. 40	17. 03		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	92.13	9. 40	9. 24		
		OUNDS OF CLASSIFICATION 4	72. 10	14, 20	17.84		
		DUNDS OF CLASSIFICATION 5	89. 07	55. 00	55. 89		
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
-	5 COMP	DUND COMPOSITE	90. 57	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

- B. REFERENCES: PRESSURE SENSITIVE LABEL MANUFACTURER
- C. APPLICABLE SCC CATEGORIES: 4-02-004-01, -99

DECEMBER 14, 1978 TABLE 4-02-004B

POINT SOURCE EVAPORATION, SURFACE COATING LACQUER, METAL FURNITURE DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0148

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	25. 00	17. 82	1	PARAFFIN
2	45102	ISOMERS OF XYLENE	106. 16	7. 00	5. 37	3	AROMATIC
3	45202	TOLUENE	92 . 13	10.00	8. 87	3	AROMATIC
4	43264	CYCLOHEXANONE	98 . 15	8. 00	6. 67	4	CARBONYL
5	43551	ACETONE	58. 08	13.00	18. 23	4	CARBONYL
6	43552	METHYL ETHYL KETONE	72. 10	18.00	20. 34	4	CARBONYL
7	43304	ISOPROPYL ALCOHOL	60. 09	15.00	20. 34	5	MISCELLANEOUS
8	43445	METHYL AMYL ACETATE	140.00	4. 00	2. 36	5	MISCELLANEOUS
		TOTAL		100. 00	100.00		
	1 COMP	OUNDS OF CLASSIFICATION 1	114. 00	25. 00	17. 82		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	97. 4 2	17. 00	14. 24		
		OUNDS OF CLASSIFICATION 4	70. 29	39. 00	45. 24		
		OUNDS OF CLASSIFICATION 5	68. 40	19.00	22. 70		
	•	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	8 COMP	OUND COMPOSITE	81. 51	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: METAL FURNITURE MANUFACTURER

POINT SOURCE EVAPORATION, SURFACE COATING LACGUER, LXB-472-E SEMIGLOSS TOP COAT

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0147

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	MEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	6. 40	4. 99	1	PARAFFIN
2	43233	N-OCTANE	114. 23	4. 60	3. 57	1	PARAFFIN
3	43248	CYCLOHEXANE	84. 16	6. 40	6. 77	1	PARAFFIN
4	45202	TOLUENE	92 . 13	13. 90	13. 46	3	AROMATIC
5	43552	METHYL ETHYL KETONE	72. 10	16. 60	20. 50	4	CARBONYL
6	43304	ISOPROPYL ALCOHOL	60. 09	11. 70	17. 38	5	MISCELLANEOUS
7	43305	N-BUTYL ALCOHOL	74. 12	4. 00	4. 81	5	MISCELLANEOUS
8	4330B	BUTYL CELLOSOLVE	102. 00	5. 60	4. 90	5	MISCELLANEOUS
9	43435	N-BUTYL ACETATE	116. 16	30. 80	23. 62	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	3 COMP	OUNDS OF CLASSIFICATION 1	100.88	17. 40	15. 33		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 3	92. 13	13. 90	13. 46		
	1 COMP	OUNDS OF CLASSIFICATION 4	72. 10	16. 60	20. 50		
	4 COMP	OUNDS OF CLASSIFICATION 5	91. 59	52. 10	50. 71		
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
•	9 COMP	OUND COMPOSITE	89.09	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: WOOD FURNITURE MANUFACTURER

DECEMBER 14, 1978 TABLE 4-02-004D

POINT SOURCE EVAPORATION, SURFACE COATING LACQUER, AIRCRAFT COATING

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0155

LINE	SAROAD		CHEMICAL.	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE		NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	43118	MINERAL	SPIRITS	114. 00	80.00	80. 05	1	PARAFFIN	
2	45101	NAPHTHA		114.00	20. 00	19. 95	3	AROMATIC	
			TOTAL		100. 00	100. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 1	114, 00	80.00	80. 05			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 3	114.00	20.00	19. 95			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 0 0			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 0 0	. 00	. 00			
	2 COMP	OUND COM	POSITE	114.00	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: AIRCRAFT MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 4-02-004-01, -99

4.02-17

POINT SOURCE EVAPORATION, SURFACE COATING LACGUER, AIRCRAFT PARTS

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0146

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	10. 70	7. 76	1 PARAFFIN
2	45101	NAPHTHA	114.00	1.00	. 74	3 AROMATIC
3	45102	ISOMERS OF XYLENE	106. 16	4. 80	3. 71	3 AROMATIC
4	45202	TOLUENE	92. 13	3. 80	3. 38	3 AROMATIC
5	43551	ACETONE	58. 08	11. 20	15. 92	4 CARBONYL
6	43552	METHYL ETHYL KETONE	72. 10	12. 00	13. 70	4 CARBONYL
7	43560	METHYL ISOBUTYL KETONE	100. 16	4. 60	3. 80	4 CARBONYL
8	43301	METHYL ALCOHOL	32. 04	1. 60	4. 13	5 MISCELLANEOUS
9	43302	ETHYL ALCOHOL	46. 07	1. 20	2, 15	5 MISCELLANEOUS
10	43304	ISOPROPYL ALCOHOL	6 0. 0 9	4. 50	6. 19	5 MISCELLANEOUS
l 1	43305	N-BUTYL ALCOHOL	74. 12	6. 50	7. 26	5 MISCELLANEOUS
12	43367	GLYCOL ETHER	62. 07	5. 40	7. 18	5 MISCELLANEOUS
l3	43433	ETHYL ACETATE	88. 10	2. 60	2. 48	5 MISCELLANEOUS
4	43435	N-BUTYL ACETATE	116. 16	27. 50	19. 55	5 MISCELLANEOUS
15	43444	ISOPROPYL ACETATE	104. 00	2. 60	2. 06	5 MISCELLANEOUS
		TOTAL		100. 00	100. 01	
	1 COMP	DUNDS OF CLASSIFICATION 1	114. 00	10. 70	7. 76	
	O COMPI	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	3 COMP	DUNDS OF CLASSIFICATION 3	100. 84	9. 60	7. 83	
	3 COMP	DUNDS OF CLASSIFICATION 4	68. 61	27. 80	33. 42	
	8 COMP	DUNDS OF CLASSIFICATION 5	84. 13	51. 90	51.00	
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
		DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
-	15 COMP	DUND COMPOSITE	82. 57	100.00	100. 01	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

C. APPLICABLE SCC CATEGORIES: 4-02-004-01, 4-02-006-99

POINT SOURCE EVAPORATION, SURFACE COATING

LACQUER, PLASTIC COATING DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0150

.INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	45202	TOLUENE	92. 13	20, 40	16. 31	3	AROMATIC
2	43551	ACETONE	58. 08	18. 70	23. 76	4	CARBONYL
3	43305	N-BUTYL ALCOHOL	74. 12	32. 40	32. 25	5	MISCELLANEOUS
4	43310	METHYL CELLOSOLVE	76. 11	28. 50	27. 68	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	92. 13	20. 40	16. 31		
	1 COMP	OUNDS OF CLASSIFICATION 4	58. 08	18. 70	23. 76		
	2 COMP	OUNDS OF CLASSIFICATION 5	75. 04	60. 90	59, 93		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	4 COMP	OUND COMPOSITE	73. 80	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: PLASTIC MOLDING AND METALIZING MANUFACTURER

4.02-20

DECEMBER 14, 1978 TABLE 4-02-005A

POINT SOURCE EVAPORATION, SURFACE COATING

ENAMEL, GENERAL COMPOSITE DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0156

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEICHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	67. 60	57. 57	1	PARAFFIN
2	45102	ISOMERS OF XYLENE	106. 16	4. 00	3. 69	3	ARDMATIC
3	45202	TOLUENE	92 . 13	3. 20	3. 40	3	AROMATIC
4	43551	ACETONE	58. 08	6. 30	10. 58	4	CARBONYL
5	43552	METHYL ETHYL KETONE	72. 10	6. 80	7 . 13	4	CARBONYL
6	43560	METHYL ISOBUTYL KETONE	100. 16	2. 70	2. 62	4	CARBONYL
7	43301	METHYL ALCOHOL	32. 04	. 70	2. 14	5	MISCELLANEOUS
8	43302	ETHYL ALCOHOL	46. 07	. 50	1. 07	5	MISCELLANEOUS
9	43304	ISOPROPYL ALCOHOL	60. 0 9	1. 70	3. 11	5	MISCELLANEOUS
10	43305	N-BUTYL ALCOHOL	74. 12	2. 80	3. 69	5	MISCELLANEOUS
11	43433	ETHYL ACETATE	88. 10	. 30	. 29	5	MISCELLANEOUS
12	43435	N-BUTYL ACETATE	116. 16	2. 90	2. 43	5	MISCELLANEOUS
13	43444	ISOPROPYL ACETATE	104.00	. 30	. 29	5	MISCELLANEOUS
		TOTAL		100. 00	100. 01		
	1 COMP	DUNDS OF CLASSIFICATION 1	114. 00	67. 60	57. 57		
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	99. 43	7. 20	7. 09		
		DUNDS OF CLASSIFICATION 4	68. 75	15. BO	22, 33		
		DUNDS OF CLASSIFICATION 5	70. 37	9. 40	13. 02		
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
		Constant de Constant de la Constant					
,	13 COMP	DUND COMPOSITE	97. 18	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

- B. REFERENCES: LITERATURE TEST DATA (REF. 3)
- C. APPLICABLE SCC CATEGORIES: 4-02-005-01,-99

DECEMBER 14, 1978 TABLE 4-02-005B

POINT SOURCE EVAPORATION, SURFACE COATING ENAMEL, POLYESTER, MODIFIED ACRYLIC

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0159

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	45101	NAPHTHA	114. 00	12. 40	10. 06	3	ARUMATIC
2	45202	TOLUENE	92. 13	4. 80	4. 80	3	AROMATIC
3	43552	METHYL ETHYL KETONE	72. 10	32. 20	41. 24	4	CARBONYL
4	43308	BUTYL CELLOSOLVE	102.00	34. 30	31, 00	5	MISCELLANEOUS
5	43446	ISOBUTYL ACETATE	116. 16	16. 30	12. 92	5	MISCELLANEOUS
		TOTAL		100. 00	100. 02		
	0 0040	MINDO OF 01 ASSETS ATTOM 4	20	50	22		
		OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	106. 94	17. 20	14. 86		
	1 COMP	OUNDS OF CLASSIFICATION 4	72. 10	32. 20	41. 24		
	2 COMP	OUNDS OF CLASSIFICATION 5	106. 17	50. 60	43. 92		
	O COMP	CUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	5 COMP	OUND COMPOSITE	92. 23	100.00	100. 02		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF INDUSTRY QUESTIONNAIRE DATA

B. REFERENCES: SHEETMETAL FABRICATOR

POINT SOURCE EVAPORATION, SURFACE COATING ENAMEL, COMPOSITE FOR WOOD FURNITURE DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0157

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	13. 20	10. 98	1	PARAFFIN
2	43233	N-OCTANE	114. 23	5. 30	4. 36	1	PARAFFIN
3	43248	CYCLOHEXANE	84. 16	13. 20	14. 87	1	PARAFFIN
4	45202	TOLUENE	9 2. 13	12. 30	12. 69	Э	AROMATIC
5	43552	METHYL ETHYL KETONE	72. 10	24. 70	32. 48	4	CARBONYL
6	43435	N-BUTYL ACETATE	116. 16	12. 40	10. 13	5	MISCELLANEOUS
7	43452	2-ETHOXYETHYL ACETATE	132. 00	5. 00	3. 60	5	MISCELLANEOUS
8	43823	DICHLORODIFLUOROMETHANE	120. 91	13. 90	10. 89	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	3 COMP	OUNDS OF CLASSIFICATION 1	99. 35	31. 70	30, 21		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 3	92. 13	12. 30	12. 69		
	1 COMP	OUNDS OF CLASSIFICATION 4	72. 10	24. 70	32. 48		
	3 COMP	DUNDS OF CLASSIFICATION 5	120. 58	31. 30	24. 62		
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	8 COMP	OUND COMPOSITE	94. 81	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF GUESTIONNAIRE DATA

B. REFERENCES: WOOD FURNITURE MANUFACTURER

POINT SOURCE EVAPORATION, SURFACE COATING ENAMEL, AIRCRAFT INDUSTRY

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0164

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43433	ETHYL ACETATE	88. 10	7. 90	10. 11	5	MISCELLANEOUS
2	43435	N-BUTYL ACETATE	116. 16	84. 50	B1. 69	5	MISCELLANEOUS
3	43444	ISOPROPYL ACETATE	104. 00	7. 60	8. 20	5	MISCELLANEOUS
		TOTAL		100. 00	100, 00		
	O COMP	OUNDS OF CLASSIFICATION 1	00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	00	. 00	. 00		
	3 COMP	OUNDS OF CLASSIFICATION 5	112. 33	100.00	100.00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	3 COMP	OUND COMPOSITE	112. 33	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: AIRCRAFT INDUSTRY

DECEMBER 14, 1978 TABLE 4-02-005E

POINT SOURCE EVAPORATION, SURFACE COATING ENAMEL, CELLOSOLVE ACETATE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0222

NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43452	2-ETHOXYETHYL ACETATE TOTAL		132. 00	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF	CLASSIFICATION 1	L . 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 2	2 . 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	00	. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 5	5 132. 0 0	100.00	100. 00		
	O COMP	OUNDS OF	CLASSIFICATION &	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 7	7 . 00	. 00	. 00		
	1 COMP	DUND COMF	POSITE	132.00	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-005-02, 4-02-009-07

DECEMBER 14, 1978 TABLE 4-02-006A

POINT SOURCE EVAPORATION, SURFACE COATING

PRIMER, GENERAL

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0134

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL		
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION		
1	43118	MINERAL SPIRITS	114.00	41.00	31. 28	1 PARAFFIN		
2	45101	NAPHTHA	114.00	1.00	. 78	3 AROMATIC		
3	45102	ISOMERS OF XYLENE	106. 16	4. 90	4. 00	3 AROMATIC		
4	45202	TOLUENE	92 . 13	3. 9 0	3. 65	3 AROMATIC		
5	43551	ACETONE	58. 08	7. 10	10. 60	4 CARBONYL		
6	43552	METHYL ETHYL KETONE	72. 10	7. 70	9. 3 0	4 CARBONYL		
7	43560	METHYL ISOBUTYL KETONE	100. 16	3. 10	2. 69	4 CARBONYL		
8	43301	METHYL ALCOHOL	32. 04	1.00	2. 69	5 MISCELLANEOUS		
9	43302	ETHYL ALCOHOL	46. 07	80	1.48	5 MISCELLANEOUS		
10	43304	ISOPROPYL ALCOHOL	60. 09	2. 80	4. 08	5 MISCELLANEOUS		
11	43305	N-BUTYL ALCOHOL	74. 12	4. 00	4. 69	5 MISCELLANEOUS		
12	43367	CLYCOL ETHER	62. 07	11. 50	16. 07	5 MISCELLANEOUS		
13	43433	ETHYL ACETATE	88. 10	. 90	. 87	5 MISCELLANEOUS		
14	43435	N-BUTYL ACETATE	116. 16	9. 50	7. 12	5 MISCELLANEOUS		
15	43444	ISOPROPYL ACETATE	104.00	. 80	. 70	5 MISCELLANEOUS		
		TOTAL		100. 00	100. 00			
	1 COMP	OUNDS OF CLASSIFICATION 1	114. 00	41. 00	31. 28			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	3 COMP	OUNDS OF CLASSIFICATION 3	100. 81	9. 80	8. 43			
	3 COMP	DUNDS OF CLASSIFICATION 4	68. 86	17. 90	22. 59			
	8 COMP	DUNDS OF CLASSIFICATION 5	72 . 18	31. 30	37. 70			
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
_	15 COMP	OUND COMPOSITE	86, 93	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA

POINT SOURCE EVAPORATION, SURFACE COATING PRIMER, PAPERBOARD PRODUCTS AND CONTAINERS DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0137

.INE NO.	SARDAD	CHEMICAL NAME	MOLECULAR WEICHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
140.	OUDC	TATA IC	WC 1 0.11	WEIGHT	VOLUNE	UL.	10011 1011 1011
1	43231	N-HEXANE	86. 17	62. 70	57. 37		PARAFFIN
2	45202	TOLUENE	92. 13	5. 20	4. 41	3	ARDMATIC
3	43551	ACETONE	58. 08	8. 40	11. 43	4	CARBONYL
4	43119	LACTOL SPIRITS	114.00	8.00	5. 52	5	MISCELLANEOUS
5	43301	METHYL ALCOHOL	32. 04	4. 70	11. 58	5	MISCELLANEOUS
6	43311	CELLOSOLVE	90. 12	5. 60	4. 89	5	MISCELLANEOUS
7	43433	ETHYL ACETATE	88. 10	5. 40	4. 81	5	MISCELLANEOUS
		TOTAL		100. 00	100. 01		
	1 COMP	OUNDS OF CLASSIFICATION 1	86. 17	62. 70	57. 37		
	O COMP	OUNDS OF CLASSIFICATION 2	. 0 0	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 3	92. 13	5. 20	4. 41		
	1 COMP	OUNDS OF CLASSIFICATION 4	58. OB	8. 40	11, 43		
	4 COMP	OUNDS OF CLASSIFICATION 5	69. 58	23. 70	26. 80		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	7 COMP	OUND COMPOSITE	78. 78	100. 00	100.01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: PRESSURE SENSITIVE LABEL MANUFACTURER

POINT SOURCE EVAPORATION, SURFACE COATING PRIMER, METAL FURNITURE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0136

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	9. 70	7. 31	1	PARAFFIN
2	43233	N-DCTANE	114. 23	. 80	. 60	1	PARAFFIN
3	43248	CYCLOHEXANE	B4. 16	9. 70	9. 89	1	PARAFFIN
4	45202	TOLUENE	9 2. 13	14. 80	13. 84	3	AROMATIC
5	43552	METHYL ETHYL KETONE	72. 10	23. 40	27. 94	4	CARBONYL
6	43304	ISOPROPYL ALCOHOL	60. 09	9. 20	13. 16	5	MISCELLANEOUS
7	43305	N-BUTYL ALCOHOL	74. 12	4. 30	4. 99	5	MISCELLANEOUS
8	43435	N-BUTYL ACETATE	116. 16	11. 40	B. 43	5	MISCELLANEOUS
9	43444	ISOPROPYL ACETATE	104. 00	16. 70	13. B4	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	3 COMP	OUNDS OF CLASSIFICATION 1	97. 43	20. 20	17. 80		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 3	92 . 13	14. 80	13. 84		
	1 COMP	OUNDS OF CLASSIFICATION 4	72. 10	23. 40	27. 94		
	4 COMP	OUNDS OF CLASSIFICATION 5	BB. 55	41.60	40. 42		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	9 COMP	OUND COMPOSITE	86. 03	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF GUESTIONNAIRE DATA

B. REFERENCES: METAL FURNITURE MANUFACTURER

DECEMBER 14, 1978

TABLE 4-02-006D

POINT SOURCE EVAPORATION, SURFACE COATING PRIMER, RED OXIDE SHOP COAT, KOPPER P-470-A-66 DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0331

INE	SAROAD		CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE		NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	43118	MINERAL	SPIRITS	114. 00	93. 20	92. 74	1	PARAFFIN	
2	45102	ISOMERS	OF XYLENE TOTAL	106. 16	6. 80 100. 00	7. 26 100. 00	3	AROMATIC	
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	93. 20	9 2. 7 4			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 0 0			
	1 COMP	OUNDS OF	CLASSIFICATION 3	106. 16	6. 80	7. 26			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 0 0			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	2 COMPOUND COMPOSITE			113.43	100.00	100, 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: SHEETMETAL FABRICATOR

DECEMBER 14, 1978 TABLE 4-02-006E

POINT SOURCE EVAPORATION, SURFACE COATING PRIMER, WATER BASED AUTOMOTIVE PAINT SPRAY BOOTH

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0280

LINE	SARDAD		CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE		NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATI ON
1	45202	TOLUENE	,	92. 13	8. 70	6. 00	3	ARDMATIC
2	43551	ACETONE		58. 08	68. 9 0	75. 69	4	CARBONYL
3	45201	BENZENE		78. 11	22. 40	18. 32	7	NON-REACTIVE
			TOTAL		100. 00	100. 01		
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 3	92 , 13	8. 70	6. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 4	58. 08	68. 90	75. 69		
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 7	78. 11	22. 40	18. 32		
	3 COMPOUND COMPOSITE			63. 79	100. 00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH AT STACK MIDPOINT

B. REFERENCES: KVB TEST DATA (REF. 3)

DECEMBER 14, 1978 TABLE 4-02-006F

POINT SOURCE EVAPORATION, SURFACE COATING PRIMER, BLACK KOPPERS A-1131-66
DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0281

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION	
1	43118	MINERAL	SPIRITS TOTAL	114. 00	100. 00 100. 00	100. 00 100. 00	1 PARAFFIN	•
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	100. 00	100. 00		
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COM	POSITE	114.00	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: SHEETMETAL FABRICATOR

4.02-31

26200-795

DECEMBER 14, 1978 TABLE 4-02-006G

POINT SOURCE EVAPORATION, SURFACE COATING

PRIMER, NAPHTHA

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0282

LINE NO.	SARDAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	45101	NAPHTHA	TOTAL	114. 00	100. 00 100. 00	100. 00 100. 00	3 AROMATIC
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	00	. 00	. 00	
	1 COMP	OUNDS OF	CLASSIFICATION 3	114.00	100.00	100.00	
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	00	. 00	
	1 COMP	OUND COMP	POSITE	114. 00	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-006-02 (REFER ALSO TO SUMMARY INDEX)

POINT SOURCE EVAPORATION, SURFACE COATING

PRIMER, MINERAL SPIRITS DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0283

LINE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43118	MINERAL	SPIRITS	114. 00	100.00	100.00	1 PARAFFIN
			TOTAL		100. 00	100. 00	
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	100. 00	100. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 🗘	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	DUND COM	POSITE	114.00	100.00	100.00	

- NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION
 - B. REFERENCES: ENGINEERING JUDGEMENT
 - C. APPLICABLE SCC CATEGORIES: 4-02-006-04 (REFER ALSO TO SUMMARY INDEX)

DECEMBER 14, 1978 TABLE 4-02-006J

POINT SOURCE EVAPORATION, SURFACE COATING

PRIMER, MINERAL SPIRITS, SPECIATION

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0225

LINE NO.	SARDAD		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME		CHEMICAL ASSIFICATION	
140.	CODE		14HIIL	MEIGH	WEIGHT	VOLUNE	VL	HOST TON TON	
1	43107	ISOMERS C	OF OCTANE	114. 23	. 80	. 97	1	PARAFFIN	
2	43108	ISOMERS C	OF NONANE	128. 25	27. 30	29. 42	1	PARAFFIN	
3	43109	ISOMERS C	OF DECANE	142. 28	69. 30	67. 27	1	PARAFFIN	
4	43110	ISOMERS C	OF UNDECANE	156 . 30	2. 60	2. 35	1	PARAFFIN	
			TOTAL		100. 00	100. 01			
	A COMP	OTHUNG UE (CLASSIFICATION 1	138, 21	100. 00	100. 01			
			CLASSIFICATION 2	. 00	. 00	. 00			
			CLASSIFICATION 3	. 00	. 00	. 00			
			CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF C	LASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF C	CLASSIFICATION 6	00	. 00	00			
	O COMP	OUNDS OF C	CLASSIFICATION 7	. 00	. 00	. 00			
	4 COMP	OUND COMPO	SITE	138. 21	100.00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF CHEMICAL FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-006-04

D. See Appendix I-F

26200-795

DECEMBER 14, 1978 TABLE 4-02-006K

POINT SOURCE EVAPORATION, SURFACE COATING

PRIMER, SHELL M-75

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0284

LINE	SAROAD		CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
ND.	CODE		NAME	MEIGHT	MEICHT	VOLUME	CL	ASSIFICATION	
1	43118	MINERAL	SPIRITS	114.00	91. 70	90. 74	1	PARAFFIN	
2	45101	NAPHTHA		114.00	3. 90	3. 84	3	AROMATIC	
3	45202	TOLUENE		92. 13	4. 40	5. 42	3	AROMATIC	
			TOTAL		100. 00	100. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	91. 70	90, 74			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	2 COMP	OUNDS OF	CLASSIFICATION 3	101. 20	8. 30	9. 26			
	O COMP	DUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	3 COMP	OUND COM	POSITE	112. 81	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: SHEETMETAL FABRICATOR

DECEMBER 14, 1978 TABLE 4-02-007A

POINT SOURCE EVAPORATION, SURFACE COATING

ADHESIVES, LABEL

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0141

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	45102	ISOMERS OF XYLENE	106. 16	22. 00	17. 69	3	AROMATIC	
2	45202	TOLVENE	9 2. 13	54. 00	50. 09	3	AROMATIC	
3	43551	ACETONE	58. 08	13. 00	19. 15	4	CARBONYL	
4	43552	METHYL ETHYL KETONE	<i>7</i> 2. 10	11.00	13.08	4	CARBONYL	
		TOTAL		100. 00	100. 01			
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	2 COMP	OUNDS OF CLASSIFICATION 3	95. 79	76. 00	67. 78			
	2 COMP	OUNDS OF CLASSIFICATION 4	63. 77	24. 00	32. 23			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O ÇOMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	4 COMP	OUND COMPOSITE	85. 47	100. 00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: PRESSURE SENSITIVE LABEL MANUFACTURER

DECEMBER 14, 1978

TABLE 4-02-007B

POINT SOURCE EVAPORATION, SURFACE COATING ADHESIVE, METAL FURNITURE

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0142

LINE NO.	SARDAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43118	MINERAL	SPIRITS	114.00	50, 00	37. 43	1	PARAFFIN	
2	45202	TOLUENE	<u> </u>	92. 13	20.00	18. 50	3	ARDMATIC	
3	43551	ACETONE		58. 08	30. 00	44. OB	4	CARBONYL	
			TOTAL		100. 00	100. 01			
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	50. 00	37. 43			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 3	92. 13	20. 00	18, 50			
	1 COMP	OUNDS OF	CLASSIFICATION 4	58. 08	30. 00	44. 08			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	3 COMP	DUND COMP	POSITE	85. 31	100.00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF GUESTIONNAIRE DATA

B. REFERENCES: CUSHION SPRING MANUFACTURER FOR FURNITURE AND BEDDING

TABLE 4-02-007C

POINT SOURCE EVAPORATION, SURFACE COATING ADHESIVE, AUTOMOTIVE VINYL TOP SPRAY BOOTH DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0285

LINE NO.	SARDAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT, WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43231	N-HE XANE	86. 17	14. 50	13. 40	1	PARAFFIN
2	45202	TOLVENE	92. 13	14.30	12. 36	3	AROMATIC
3	43551	ACETONE	58.08	12. 50	17. 15	4	CARBONYL
4	43433	ETHYL ACETATE	88. 10	24. 30	22. 01	5	MISCELLANEOUS
5	45201	BENZENE	78. 11	34. 40	35. 09	7	NON-REACTIVE
		TOTAL		100.00	100, 01		
	1 COMP	OUNDS OF CLASSIFICATION 1	86. 17	14. 50	13. 40		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 3	92. 13	14, 30	12, 36		
	1 COMP	OUNDS OF CLASSIFICATION 4	58, 08	12, 50	17, 15		
	1 COMP	OUNDS OF CLASSIFICATION 5	88. 10	24. 30	22, 01		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 7	78. 11	34. 40	35. 09		
	5 COMP	OUND COMPOSITE	79.69	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH AT STACK MIDPOINT

B. REFERENCES: KVB TEST DATA (REF. 3)

TABLE 4-02-007D

POINT SOURCE EVAPORATION, SURFACE COATING ADHESIVES, FOSTER BOND SEAL NO. 107 DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0145

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43817	PERCHLOROETHYLENE TOTAL	165, 83	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 0 0	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 5	165. 83	100.00	100. 00		
	9 COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	DUND COMPOSITE	165. 83	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

- B. REFERENCES: KVB TEST DATA (REF. 3)
- C. APPLICABLE SCC CATEGORIES: 4-02-007-01,-99

DECEMBER 14, 1978 TABLE 4-02-007E

POINT SOURCE EVAPORATION, SURFACE COATING

ADHESIVE, BENZENE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0287

NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	45201	BENZENE		78. 11	100.00	100. 00	7 NON-REACTIVE
			TOTAL		100. 00	100.00	
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	00	. 00	
	O COMF	OUNDS OF	CLASSIFICATION 3	. 00	00	. 00	
	O COMF	OUNDS OF	CLASSIFICATION 4	. 00	00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	00	. 00	
	O COMF	OUNDS OF	CLASSIFICATION 6	00	00	. 00	
	1 COMF	OUNDS OF	CLASSIFICATION 7	78. 11	100.00	100. 00	
	1 COMF	OUND COME	POSITE	78. 11	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

DECEMBER 14, 1978 TABLE 4-02-999A

POINT SOURCE EVAPORATION, SURFACE COATING

CITRUS COATING WAX, BROGDEX 502 DATA CONFIDENCE LEVEL: I

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0293

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEICHT	HEICHT	VOLUME	CL	ASSIFICATION	
1	43231	N-HEXANE	86. 17	34. 00	34. 83	1	PARAFFIN	_
2	43248	CYCLOHEXANE	84. 16	43. 10	45. 15	1	PARAFFIN	
3	45102	ISOMERS OF XYLENE	106. 16	6. 60	5. 47	3	ARDMATIC	
4	45202	TOLUENE	92 . 13	7. 60	7. 32	3	ARDMATIC	
5	45203	ETHYLBENZENE	106. 16	8. 70	7. 23	3	ARDMATIC	
		TOTAL		100.00	100.00			
	2 COMP	OUNDS OF CLASSIFICATION 1	85. 04	77. 10	79. 9 8			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	3 COMP	OUNDS OF CLASSIFICATION 3	101.03	22. 90	20. 02			
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	5 COMP	DUND COMPOSITE	88. 24	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: CITRUS COATING WAX MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 4-02-999-99

4.02-40

DECEMBER 14, 1978 TABLE 4-02-9998

POINT SOURCE EVAPORATION, SURFACE COATING CITRUS COATING WAX, FLAVORSEAL 320-0820 DATA CONFIDENCE LEVEL: I

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0294

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	-	CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	45102	ISOMERS OF XYLENE	106. 16	41.60	42. 38	3	AROMATIC	
2	45104	ISOMERS OF ETHYLTOLUENE	120. 19	12. 20	11.03	3	AROMATIC	
3	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	20. 10	18. 05	3	AROMATIC	
4	45108	ISOMERS OF PROPYLBENZENE	120. 19	2. 10	1. 95	3	AROMATIC	
5	45202	TOLUENE	92. 13	13. 70	16. 11	3	AROMATIC	
6	45203	ETHYLBENZENE	106. 16	10. 30	10. 49	3	AROMATIC	
		TOTAL		100. 00	100. 01			
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 2	00	. 00	. 00			
	6 COMP	OUNDS OF CLASSIFICATION 3	108. 25	100.00	100. 01			
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	00	. 00			
	6 COMP	OUND COMPOSITE	108. 25	100. 00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: CITRUS COATING WAX MANUFACTURER

DECEMBER 14, 1978 TABLE 4-02-9990

POINT SOURCE EVAPORATION, SURFACE COATING CITRUS COATING WAX, FLAVORSEAL 115-1800 DATA CONFIDENCE LEVEL: I

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0295

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	45102	ISOMERS OF XYLENE	106. 16	32. 63	31. 07	3	AROMATIC
2	45104	ISOMERS OF ETHYLTOLUENE	120. 1 9	7. 33	6. 17	3	AROMATIC
3	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	8. 92	7. 49	3	ARDMATIC
4	45108	ISOMERS OF PROPYLBENZENE	120. 19	1. 91	1. 62	3	AROMATIC
5	45202	TOLUENE	92. 13	39. 07	42. 91	3	AROMATIC
6	45203	ETHYLBENZENE	106. 16	7. 03	6. 68	3	AROMATIC
7	45201	BENZENE	78. 11	3. 11	4. 05	7	NON-REACTIVE
		TOTAL		100. 00	99. 99		
	O COMP	DUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	6 COMP	DUNDS OF CLASSIFICATION 3	102. 12	96. 89	95. 94		
	O COMP	DUNDS OF CLASSIFICATION 4	. 0 0	. 00	. 00		
	O COMPI	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	O COMPI	DUNDS OF CLASSIFICATION 6	. 0 0	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION 7	78. 11	3. 11	4. 05		
	7 COMP	OUND COMPOSITE	101. 15	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: CITRUS COATING WAX MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 4-02-999-99

4.02-42

REFERENCES

- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 2. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

4.02-43 26200-795

4-02 POINT SOURCE EVAPORATION, SURFACE COATING

4-02-008 COATING OVENS

Process Description 1

Coating ovens involve the process of drying and the process of baking, curing, or polymerizing coatings. In both processes, heat is used to remove residual solvents, but in baking, curing or polymerizing, the heat also serves to produce desired chemical changes in the coating.

Bake ovens are designed for processing on either an intermittent batch basis, or on a continuous web or conveyor-fed basis. Common methods of oven heating include:

- electric resistance heatersinfrared heating
- . steam indirect method, and
- . waste heat discharged from other equipment

For further process information, consult AP-40 (Ref. 1).

Emissions

The air pollutants emitted from a surface coating oven are generally composed of organic vapors and aerosols. Particulates and the products of incomplete combustion can pose additional air pollution problems when gas or oil-fired heating systems are used.

Organic solvent vapors are emitted during the evaporation of the organic thinners and dilutents contained in the surface coating. The composition of the organic solvent vapors emitted will differ based on the composition of the solvents used in the coating material.

Estimating the emission rate of VOC's from an oven involves consideration of:

1. The quantity and composition of coating material used,

- 2. method of application
- 3. factors affecting solvent evaporation prior to oven treatment (ambient temperature, pressure and humidity, air movement, surface characteristics of the coating, solvent volatility, time), and
- 4. reduction by combustion in the oven heating system.

An illustrative example is presented in AP-40 as an aid in estimating emissions (Ref. 1).

Profiles 4-02-008A through C present the VOC emissions estimated to be eminating from various coating oven operations (Ref. 2).

Controls

Effluent streams from solvent based surface coating ovens can best be controlled through the use of afterburner equipment.

The choice between direct flame and catalytic incineration methods must be based on economic factors and on the requirements of local air pollution control agencies.

Process Modification

Appropriate substitutes for organic solvent borne coatings exist in the increasingly popular water-borne coatings. Generally known as water-based paints or latexes, they do have lower organic solvent contents than the high solvent based coatings.

Profile Basis²

profiles 4-02-008A through F are based on field tests conducted at various coating oven operations. VOC samples were collected by means of gas collecting bottles and NIOSH charcoal sample tubes. Details of each test can be found in Reference 2.

Data Qualification

Details of the methods used to estimate VOC emission rates are presented in AP-40 (Ref. 1).

Profiles 4-02-008A through F may be used to characterize the VOC emissions for the indicated surface coating oven operations as described in each profile.

POINT SOURCE EVAPORATION, SURFACE COATING COATING OVEN, METAL PARTS, GENERAL DATA CONFIDENCE LEVEL. III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0092

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43204	PROPANE	44. 09	3. 10	1. 44	1	PARAFFIN	
2	43212	N-BUTANE	58, 12	23. 40	8. 30	1	PARAFFIN	
3	43203	ETHYLENE	28. 05	2.00	1.46	2	OLEFIN	
4	43201	METHANE	16. 04	66. 50	85 . 38	6	METHANE	
5	43202	ETHANE	30. 07	5. 0 0	3. 42	7	NON-REACTIVE	
		TGTAL		100. 00	100. 00			
	2 COMP	OUNDS OF CLASSIFICATION 1	56, 05	26. 50	9 . 7 4			
	1 COMP	DUNDS OF CLASSIFICATION 2	28. 05	2. 00	1.46			
	O COMP	OUNDS OF CLASSIFICATION 3	00	00	. 00			
		OUNDS OF CLASSIFICATION 4	00	00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	00	. 00	. 00			
	1 COMP	OUNDS OF CLASSIFICATION 6	16.04	66, 50	85. 38			
	_	OUNDS OF CLASSIFICATION 7	30. 07	5. 00	3. 42			
	5 COMP	OUND COMPOSITE	20. 59	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF KVB TEST DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

DECEMBER 14, 1978 TABLE 4-008B

POINT SOURCE EVAPORATION, SURFACE COATING COATING OVEN, WATER BASED AUTOMOTIVE PRIMER, NATURAL GAS FIRED DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE
PROCESS MODIFICATION: NONE
KVB PROFILE KEY 0279

INE NO.	SARDAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43203	ETHYLENE	28. 05	2. 40	2. 47	2	OLEFIN
2	45202	TOLUENE	92. 13	10. 60	3. 31	3	AROMATIC
3	43551	ACETONE	58. 08	30. 20	14. 96	4	CARBONYL
4	43201	METHANE	16. 04	39. 90	71. 60	6	METHANE
5	43202	ETHANE	30. 07	2. 40	2. 30	7	NON-REACTIVE
6	45201	BENZENE	78. 11	14. 50	5. 35	7	NON-REACTIVE
		TOTAL		100. 00	99. 99		
	O COMP	OUNDS OF CLASSIFICATION	1 .00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION		2. 40	2. 47		
	1 COMP	DUNDS OF CLASSIFICATION	3 92.13	10. 60	3. 31		
	1 COMP	DUNDS OF CLASSIFICATION	4 58.08	30. 20	14. 96		
	O COMP	DUNDS OF CLASSIFICATION	5 .00	. 00	. 00		
	1 COMP	DUNDS OF CLASSIFICATION	6 16.04	39. 90	71.60		
	2 COMP	DUNDS OF CLASSIFICATION	7 63. 67	16. 90	7. 65		
	6 COMP	DUND COMPOSITE	28. 79	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH AT STACK MIDPOINT

B. REFERENCES: KVB TEST DATA (REF. 2)

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0056

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATI ON	
1	43106	ISOMERS OF HEPTANE	100. 20	. 70	. 81	1	PARAFFIN	
2	43107	ISOMERS OF OCTANE	114, 23	. 60	. 58	1	PARAFFIN	
3	43108	ISOMERS OF NONANE	128. 25	10. 10	9. 14	1	PARAFFIN	
4	43115	C-7 CYCLOPARAFFINS	98 . 19	1.40	1. 62	1	PARAFFIN	
5	43116	C-B CYCLOPARAFFINS	112. 23	25. 60	26. 39	1	PARAFFIN	
6	43117	C-9 CYCLOPARAFFINS	126. 26	5. 00	4. 63	1	PARAFFIN	
7	43231	N-HEXANE	86. 17	. 20	. 23	1	PARAFFIN	
8	43232	N-HEPTANE	100. 20	1.40	1.62	1	PARAFFIN	
9	43233	N-OCTANE	114, 23	46. 50	47. 11	1	PARAFFIN	
10	43235	N-NONANE	128. 25	7. 30	6. 60	1	PARAFFIN	
11	45102	ISOMERS OF XYLENE	106. 16	1.00	1. 04	3	AROMATIC	
12	45202	TOLUENE	9 2. 13	. 20	. 23	3	AROMATIC	
		TOTAL		100. 00	100. 00			
	10 COMP	OUNDS OF CLASSIFICATION 1	115. 82	98. 80	98. 73			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	2 COMP	OUNDS OF CLASSIFICATION 3	103. 62	1. 20	1. 27			
	O COMP	QUNDS OF CLASSIFICATION 4	. 00	00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	12 COMP	OUND COMPOSITE	115. 67	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 2)

DECEMBER 14, 1978 TABLE 4-02-008D

POINT SOURCE EVAPORATION, SURFACE COATING COATING OVEN, ADHESIVE, AUTOMOBILE VINYL TOP, NATURAL GAS FIRED DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0286

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44. 09	1. 60	2. 65	1	PARAFFIN
2	43214	ISOBUTANE	58. 12	11. 10	14. 03	1	PARAFFIN
3	45202	TOLUENE	92 . 13	2. 70	2. 13	3	AROMATIC
4	43551	ACETONE	58. 08	25. 40	32. 11	4	CARBONYL
5	43433	ETHYL ACETATE	88. 10	12. 40	10. 36	5	MISCELLANEOUS
6	43817	PERCHLOROETHYLENE	165. 83	41. 90	18. 59	5	MISCELLANEOUS
7	43201	METHANE	16. 04	3. BO	17. 41	6	METHANE
8	43202	ETHANE	30. 07	1. 10	2. 7 2	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP	OUNDS OF CLASSIFICATION 1	55. 89	12. 70	16. 68		
		OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	92. 13	2. 70	2. 13		
		OUNDS OF CLASSIFICATION 4	58. 08	25. 40	32. 11		
		OUNDS OF CLASSIFICATION 5	138. 01	54. 30	28. 95		
		OUNDS OF CLASSIFICATION 6	16. 04	3. 80	17. 41		
		OUNDS OF CLASSIFICATION 7	30. 07	1. 10	2. 72		
	B COMP	OUND COMPOSITE	73. 50	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH AT STACK MIDPOINT

B. REFERENCES: KVB TEST DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 4-02-008-99

POINT SOURCE EVAPORATION, SURFACE COATING COATING OVEN, ENAMEL GENERAL, GAS FIRED DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0162

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	25. 80	14. 13	1 PARAFFIN
2	43204	PROPANE	44. 09	. 30	. 44	1 PARAFFIN
3	43212	N-BUTANE	58. 12	. 10	. 13	1 PARAFFIN
4	43220	N-PENTANE	72. 15	. 60	. 50	1 PARAFFIN
5	43205	PROPYLENE	42.08	2. 40	3. 56	2 OLEFIN
6	43213	BUTENE	56. 10	. 90	1.00	2 OLEFIN
7	43224	1-PENTENE	70. 1 3	2. 40	2. 13	2 OLEFIN
8	45202	TOLUENE	92 . 13	2. 10	1. 44	3 AROMATIC
9	43502	FORMALDEHYDE	30. 03	3. 30	6. 88	4 CARBONYL
10	43551	ACETONE	58 . 08	. 60	. 63	4 CARBONYL
11	43206	ACETYLENE	26. 04	. 30	. 75	5 MISCELLANEOUS
12	43304	ISOPROPYL ALCOHOL	60. 09	7. 00	7. 31	5 MISCELLANEOUS
13	43432	METHYL ACETATE	74. 08	3. 10	2. 63	5 MISCELLANEOUS
14	43435	N-BUTYL ACETATE	116. 16	29. 60	15. 94	5 MISCELLANEOUS
15	43444	ISOPROPYL ACETATE	104.00	12. 50	7. 50	5 MISCELLANEOUS
16	43201	METHANE	16. 04	9. 00	35. 06	6 METHANE
		TOTAL		100. 00	100. 03	
	4 СЛМР	OUNDS OF CLASSIFICATION 1	110. 12	26. 80	15. 20	
		OUNDS OF CLASSIFICATION 2	53. 11	5. 70	6. 69	
		QUNDS OF CLASSIFICATION 3	92. 13	2. 10	1. 44	
		OUNDS OF CLASSIFICATION 4	32. 38	3. 90	7. 51	
		OUNDS OF CLASSIFICATION 5	96. 26	52. 50	34, 13	
		DUNDS OF CLASSIFICATION 6	16. 04	9. 00	35. 06	
		DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
-	16 COMP	DUND COMPOSITE	62. 51	100.00	100. 03	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: APPLIANCE MANUFACTURER

C. APPLICABLE SCC CATEGORIES: 4-02-008-99

POINT SOURCE EVAPORATION, SURFACE COATING DRYING OVEN, LACQUER, AUTOMOTIVE, GAS FIRED DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE KVB PROFILE KEY 0154

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43122	ISOMERS OF PENTANE	72. 15	3. 90	4. 13	1	PARAFFIN	
2	43212	N-BUTANE	58. 12	13. 80	18. 11	1	PARAFFIN	
3	43214	ISOBUTANE	58. 12	10. 60	13. 90	1	PARAFFIN	
4	43231	N-HEXANE	86. 17	6. 20	5. 50	1	PARAFFIN	
5	43232	N-HEPTANE	100. 20	12. 80	9. 7 8	1	PARAFFIN	
6	45202	TOLUENE	92 . 13	31.80	26. 36	3	AROMATIC	
7	43551	ACETONE	58. 08	5. 30	6. 95	4	CARBONYL	
8	45201	BENZENE	78. 11	15. 60	15. 28	7	NON-REACTIVE	
		TOTAL		100. 00	100. 01			
	5 COMP	DUNDS OF CLASSIFICATION 1	70, 25	47. 30	51. 42			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 3	92. 13	31.80	26. 36			
		OUNDS OF CLASSIFICATION 4	58. 08	5. 30	6. 95			
		OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 7	78. 11	15. 60	15. 28			
•	8 COMP	OUND COMPOSITE	76. 37	100. 00	100. 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 2)

REFERENCES

- 1. Danielson, J. A., (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

4.02-53 26200-795

4-02 POINT SOURCE EVAPORATION, SURFACE COATING

4-02-009 SOLVENTS

Process Description 1

As described in AP-40, organic solvents are some of the most common and widely used products of our society. They are involved in our daily lives in such activities as making and cleaning the clothes we wear, making and coating the vehicles we drive, packaging the foods we eat, printing the materials we read, and finishing the furniture we use.

Emissions

After an organic solvent has served its purpose, its continued presence in the product is usually undesired and it must be removed. In so doing, it may be recovered for reuse and recycling. Too often, however, the solvent is wasted to the atmosphere by natural or forced evaporation. When architectural coatings are applied with solvents, the solvents must evaporate into the atmosphere so that the coating can form a film or barrier. When industrial coatings are applied with solvents, the solvents are discharged into the atmosphere by forced evaporation in ovens. When clothes are cleaned with solvents, the solvents must be removed, usually by heat, before the clothes can be worn again.

These organic emissions may represent a substantial portion of all organic vapors present in a community's atmosphere. A rule of thumb which has been reasonably close for Los Angeles County indicates that about 1/6 pound of solvent is emitted each day for each person.

Profiles 4-02-009A through K characterize the VOC species being emitted from various solvents.

Controls 1

Controls for volatile organic vapors from point sources generally take the form of thermal or catalytic incinerators, or activated carbon adsorbers.

Where large volumes of air are involved, neither control method may prove feasible. In such instances, it has proven more ecomonical to reformulate the solvent systems to the extent of making them nonphotochemically reactive and thereby removing the limitation on the quantity of organic material which may be emitted. This is especially true in the area of architectural surface coatings - see Section 9-35-103.

Various problems such as cost considerations, relative solvency, evaporation rates, compatibilities, and partial solvation of undercoats to name a few, however, are encountered in this approach.

The degree of control is generally mandated by local air pollution regulations.

Profile Basis²

Profiles 4-02-009A through K are based on an engineering evaluation of the various solvents used in formulating surface coatings or in the application of these coatings for finishing purposes.

Data Qualification

Profiles 4-02-009A through K may be used to characterize the volatile organic vapors eminating from these solvents as indicated on the different profiles.

DECEMBER 14, 1978 TABLE 4-02-009A

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, GENERAL

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0096

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
ND.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLA	SSIFICATION
1	43107	ISOMERS OF OCTANE	114. 23	40	. 34	1	PARAFFIN
2	43108	ISOMERS OF NONANE	128. 25	10. 90	7. 18	1	PARAFFIN
3	43109	ISOMERS OF DECANE	142. 28	27 . 80	16, 47	1	PARAFFIN
4	43110	ISOMERS OF UNDECANE	156. 30	1.00	. 51	1	PARAFFIN
5	45102	ISOMERS OF XYLENE	106. 16	4. 00	3, 21	3	AROMATIC
6	45202	TOLUENE	9 2. 13	4.00	3. 63	3	AROMATIC
7	43551	ACETONE	58. OB	10.00	14. 53	4	CARBONYL
8	43552	METHYL ETHYL KETONE	72. 10	10.00	11. 74	4	CARBONYL
9	43560	METHYL ISOBUTYL KETONE	100. 16	5. 00	4. 22	4	CARBONYL
10	43301	METHYL ALCOHOL	32. 04	5. 60	14. 7B	5	MISCELLANEOUS
11	43302	ETHYL ALCOHOL	46. 07	5. 60	10. 30	5	MISCELLANEOUS
12	43304	ISOPROPYL ALCOHOL	60. 09	5. 70	8. 02	5	MISCELLANEOUS
13	43817	PERCHLOROETHYLENE	16 5 . 8 3	10.00	5. 07	5	MISCELLANEOUS
		TOTAL		100. 00	100.00		
	4 COMP	OUNDS OF CLASSIFICATION 1	138. 07	40. 10	24. 50		
		OUNDS OF CLASSIFICATION 2	. 00	00	00		
	2 COMP	DUNDS OF CLASSIFICATION 3	98. 71	8.00	6. 84		
	3 CDMP	DUNDS OF CLASSIFICATION 4	69. 30	25. 00	30. 49		
		DUNDS OF CLASSIFICATION 5	59. 49	26. 90	38. 17		
		DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 7	. 00	00	. 00		
	13 COMP	OUND COMPOSITE	84. 42	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3)

DECEMBER 14, 1978 TABLE 4-02-009B

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, BUTYL ACETATE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0288

.INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43435	N-BUTYL	ACETATE TOTAL	116. 16	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS	
	O COMP	DUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. QO	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 5	116. 16	100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	OUND COM	POSITE	116. 16	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-009-03

4.02-57

TABLE 4-02-009C

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, BUTYL ALCOHOL

DATA CONFIDENCE LEVEL: 11

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0289

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43305	N-BUTYL	ALCOHOL TOTAL	74. 12	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 4	. 0 0	00	. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 5	74. 12	100.00	100.00		
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COM	POSITE	74. 12	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, CELLOSOLVE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0290

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43311	CELLOSO	_ ·	90. 12	100.00	100. 00	5	MISCELLANEOUS
			TOTAL		100. 00	100. 00		
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF	CLASSIFICATION 5	90. 12	100.00	100. 00		
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COM	POSITE	90. 12	100.00	100, 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-009-06, 4-05-003-03

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, DIMETHYLFORMAMIDE DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0292

LINE ND.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43450	DIMETHYLFORMAMIDE	73. 09	100.00	100.00	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	73. 09	100.00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COMPOSITE	73. 09	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-009-08, 4-05-005-02

TABLE 4-02-009F

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, ETHYL ALCOHOL

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0226

.INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43302	ETHYL ALCOHOL TOTAL	46. 07	100. 00 100. 00	100. 00 100. 00	5 MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 5	46. 07	100.00	100.00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	COUND COMPOSITE	46, 07	100.00	100, 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-02-009-10, 4-05-003-04, 4-05-005-04

DECEMBER 14, 1978 TABLE 4-02-0090

POINT SOURCE EVAPORATION, SURFACE COATING SOLVENT, ISOPROPYL ALCOHOL DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0227

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43304	ISOPROPYL ALCOHOL TOTAL	60. 09	100. 00 100. 00	100. 00 100. 00	5 MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 5	60. 09	100.00	100. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 0 0	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	00	. 00	
	1 COMP	DUND COMPOSITE	60. 09	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

POINT SOURCE EVAPORATION, SURFACE COATING SOLVENT, ISOPROPYL ACETATE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0228

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43444	ISOPROPYL ACETATE TOTAL	104. 00	100. 00 100. 00	100. 00 100. 00	5 MISCELLANEOUS
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 5	104.00	100.00	100.00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMPOUND COMPOSITE		104.00	100.00	100. 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

DECEMBER 14, 1978 TABLE 4-02-009J

POINT SOURCE EVAPORATION, SURFACE COATING

SOLVENT, LACTOL SPIRITS DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0229

NO.	SAROAD CODE	CHEMICAL NAME		MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION	
1	43119	LACTOL	SPIRITS TOTAL	114. 00	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS
	O COMP	OUNDS OF	F CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF	F CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 0 -2	. 00		
	1 COMP	DUNDS OF	CLASSIFICATION 5	114.00	100.00	100.00		
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMPOUND COMPOSITE		114. 00	100.00	100.00			

NOTES: A, METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

DECEMBER 14, 1978 TABLE 4-02-009K

POINT SOURCE EVAPORATION, SURFACE COATING SOLVENT, METHYL ALCOHOL DATA CONFIDENCE LEVEL. II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0291

INE NO.	SAROAD CODE	CHEMICAL Name	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43301	METHYL ALCOHOL	32. 04	100.00	100.00	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	32. 04	100.00	100, 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COMPOSITE	32. 04	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

REFERENCES

- Danielson, J. A., (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 3. Lunche, R. G., et al., "Products Emitting Organic Vapors in Los Angeles County," Chemical Engineering Progress, Vol. 53, No. 8, August 1957.

4-03 POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

4-03-001 FIXED ROOF TANKS--GASOLINE

--CRUDE OIL

--MISC. PETROLEUM PRODUCTS

--LIQUIFIED PETROLEUM GAS

Process Description 1

Basically, the petroleum industry consists of three operations:

- 1. Petroleum production and transportation
- 2. Petroleum refining
- Transportation and marketing of finished petroleum products

All three operations require some basic storage for petroleum liquids, and these storage tanks, for both the crude and finished products, can be significant sources of evaporative emissions.

Four basic tank designs are used for petroleum storage vessels: fixed roof, floating roof, variable vapor space, and pressurized tanks. For more detailed descriptions of these tank designs, consult References 1 and 2.

Emissions and Controls 1,4

There are six sources of emissions from petroleum liquids in storage: fixed roof breathing losses, fixed roof working losses, floating roof storage losses, floating roof withdrawal losses, variable vapor space filling losses, and pressure tank losses (Ref. 4).

Fixed Roof Tanks--Fixed roof breathing losses consist of vapor expelled from a tank because of the thermal expansion of existing vapors, vapor expansion caused by barometric pressure changes, and/or an increase in the amount of vapor due to added vaporization in the absence of a liquid-level change.

Fixed roof working losses consist of vapor expelled from a tank as a result of filling and emptying operations. Filling loss is the result of vapor displacement by the input of liquid. Emptying loss is the expulsion of vapors subsequent to product withdrawal, and is attributable to vapor growth as the newly inhaled air is saturated with hydrocarbons.

The method most commonly used to control emissions from fixed roof tanks is a vapor recovery system that collects emissions from the storage vessels and converts them to liquid product. Often a natural gas blanket is introduced to prevent excessive tank breathing. The expulsion of vapors through a gauge hatch will often consist of the VOC's from the gas blanket. To recover vapor, one or a combination of four methods may be used: vapor/liquid absorption, vapor compression, vapor cooling, and vapor/solid adsorption. Overall control efficiencies of vapor recovery systems vary from 90% to 95% depending on the method used, the design of the unit, the composition of vapors recovered, and the mechanical condition of the system.

Emissions from fixed roof tanks can also be controlled by the addition of an internal floating cover or covered floating roof to the existing fixed roof tank. API reports that this can result in an average loss reduction of 90% of the total evaporation loss sustained from a fixed roof tank.

Evaporative emissions can be minimized by reducing tank heat input with water sprays, mechanical cooling, underground storage, tank insulation, and optimum scheduling of tank turnovers.

Floating Roof Tanks--Floating roof standing storage losses result from causes other than breathing or changes in liquid level. The largest potential source of this loss is due to an improper fit of the seal and the roof.

Floating roof withdrawal losses result from evaporation of stock which wets the tank wall as the roof descends during emptying operations. This loss is small in comparison to other types of losses.

Evaporative emissions from floating roof tanks can be minimized by reducing tank heat input and by maximizing maintenance on the tank scals.

Variable Vapor Space Tanks--Variable vapor space filling losses result when vapor is displaced by the liquid input during filling operations. Since the variable vapor space tank has an expandable vapor storage capability, this loss is not as large as the filling loss associated with fixed roof tanks. Loss of vapor occurs only when the vapor storage capacity of the tank is exceeded.

Pressure tank losses occur when the pressure inside the tank exceeds the design pressure of the tank, which results in relief vent opening. This happens only when the tank is filled improperly, or when abnormal vapor expansion occurs. These are not regularly occurring events, and pressure tanks are not a significant source of loss under normal operating conditions.

The total amount of evaporation loss from storage tanks depends upon the rate of loss and the period of time involved. Factors affecting the rate of loss include:

- 1. True vapor pressure of the liquid stored
- 2. Temperature changes in the tank
- 3. Height of the vapor space (tank outage)
- 4. Tank diameter
- 5. Schedule of tank filling and emptying
- 6. Mechanical condition of tank and seals
- 7. Type of tank and type of paint applied to outer surface

Evaporative emissions from variable vapor space tanks are negligible and can be minimized by optimum scheduling of tank turnovers and by reducing tank heat input. Vapor recovery systems can be used with variable vapor space systems to collect and recover filling losses.

Vapor recovery systems capture hydrocarbon vapors displaced during filling operations and recover the hydrocarbon vapors by the use of refrigeration, absorption, adsorption, and/or compression. Control efficiencies range from 90% to 98%, depending on the nature of the vapors and the recovery equipment used.

Pressure Tanks—Pressure tanks incur vapor losses when excessive internal pressures result in relief valve venting. In some pressure tanks vapor venting is a design characteristic, and the vented vapors must be routed to a vapor recovery system. However, for most pressure tanks vapor venting is not a normal occurrence, and the tanks can be considered closed systems. Fugitive losses are also associated with pressure tanks and their equipment, but with proper system maintenance they are insignificant. Correlations do not exist for estimating vapor losses from pressure tanks.

Emission Factors

AP-42 may be used to calculate emissions from petroleum storage tanks. Many studies have been undertaken recently to better understand tank emissions. Revision of current emission factors is being considered by the EPA.

Profile Basis

Approximately 45% of the crude oil refined in the California South Coast Air Basin is produced locally. The balance comes from several sources including domestic and foreign suppliers. There exist significant differences in VOC speciation between the crude oils stored in production and refining in the Basin and that involved in marketing operations. The following discussion is therefore divided into the three general areas of production, marketing, and refinery operations.

Production Operations - Results from the current inventory (Ref. 3) show that the organic compound emissions from crude oil storage for petroleum production operations account for 35 tons per day or approximately 5% of all manmade sources. The emission profiles for the vapors released from these sources were developed using data from the field tests conducted on the program (Ref. 3) and the results of a recently completed study on fixed roof tank emissions sponsored by the Western Oil and Gas Association (WOGA) (Ref. 6).

A summary of the data used to determine this profile is included in Table 4-03D. Since data from Reference 6 did not identify "normal" and "isomer" compounds, the organic compounds for these tests have been identified as "normal". The layout of this table is similar to the following tables with the identification of the crude oils across the top and the organic compounds listed vertically. The numbers in each column represent the weight percent of that specie in the associated crude oil vapors.

As seen in Table 4-03D, there was a wide variation in the organic constituents of the crude oil vapors within the study area. The composite emission profile was developed using the weighting factor listed with each crude oil which represents roughly the fraction of the total crude oil produced by refineries within the Basin according to the California Division of Oil and Gas (Ref. 10).

Table 4-03-001C is a detailed profile specie table which is a composite of crude oil storage for petroleum production operations. This composite profile was developed from Table 4-03D and as discussed above.

Refining Operations - A similar procedure was used to develop an emission profile for crude oil storage associated with refining operations. A summary of the data used to calculate a composite profile is included in Table 4-03E. The crude oil stored in refineries showed a much larger variation in vapor compositions reflecting the diverse origins in the crude oil.

Data in Table 4-03E represents vapors collected from both fixed and floating roof tanks. However, insufficient data were available to allow a differentiation between vapors emitted from each tank type. Regulations by the California South Coast Air Quality Management District (SCAQMD) require storage of petroleum liquids with a vapor pressure greater than 1.5 psia in floating roof tanks. Therefore, it would seem reasonable to assume that differences between the vapor composition for crude oils above and below this limit would exist.

Table 4-03-001D is a detailed profile specie table that is a composite of crude oil storage for petroleum refining operations. This profile was developed from Table 4-03E and as discussed above.

Crude Oil Storage and Transfer Operations - Table 4-03F presents the data employed to determine a composite emission profile for crude oil storage and transfer operations primarily at marine terminals. Because significant variations in the crude oil vapors were observed between the various samples analyzed by KVB (Ref. 3), these test data were not used for this profile. The data shown were taken from the WOGA Fixed Roof Tank Study (Ref. 6).

Table 4-06-002 is a detailed profile specie table that is a composite of crude oil storage and transfer operations at marine terminals. The remainder of the profiles following this section are self-explanatory and may be used as indicated in the profile titles or by their applicable SCC categories.

4.03-5

TABLE 4-03D. CRUDE OIL STORAGE, PETROLEUM PRODUCTION OPERATIONS

					Production	Field					
	† Huntington Beach	† Saticoy	§ Rosecrans	§ Wilmington	§ Seal Beach	§ Ventura	§ Santa Fe Springs	§ Brea Olinda	§ South Mountain	§ Ojai	Composite
Fraction of Production	0.13	0.01	0.10	0.45	v.05	0.12	0.02	0.08	0.05	0.01	1.0
Organic Compound (wt. %)											
Methane	3.0	7.1	1.0	10.8	2.0	1.2	5.3	1.4	7.9	0.2	6.2
Ethane	6.7	17.6	1.9	6.7	1.9	2.6	15.4	2.0	9.2	2.1	5.6
Propane	20.2	30.8	12.6	18.1	12.5	19.6	19.7	9.5	25.5	8.4	17.6
l-Butane	19.9	19.5	24.7	27.4	25.6	34.5	26.2	30.7	26.5	30.9	27.1
-Butane	11.0	9.9									1.5
I-Pentane	8.0	5.1	19.3	13.1	30.7	14.7	13.0	23.5	10.5	25.6	14.6
I-Pentane	11.1	5.8									1.5
łexane	5.0	2.1	12.9	6.8	12.3	9.4	5.8	11.3	5.8	11.3	7.9
-Hexane	0.8										
leptane			17.2	9.3	9.2	10.7	6.2	13.7	7.5	11.8	9.2
Octane*			10.4	7.8	5.8	7.3	8.4	7.9	7.1	9.7	6.9
-7 Cycloparaffins	9.9	0.6									1.3
-8 Cycloparaffins	3.5										0.5
Benzene	1.0	0.1									0.1
Cyclohexane											
B-Methyl Pentane		1.4									_==
	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100.0

^{*} C +

[†] Tests conducted during the current project

[§] Ref. 6

TABLE 4-03E. CRUDE OIL STORAGE, PETROLEUM REFINING OPERATIONS

Organic Compound	1		Crude Sources			
(wt. %)	Foreign*	Domestic*	Domestic*	Domestic*	Unknown_	Composite
Methane	5.3	9.0	0.1	2.7	26.8	8.8
Ethane	4.5	6.0	2.1		0.9	2.7
Propane	17.1	26.0	20.6	7.3	9.7	16.1
N-Butane	18.6	26.0	22.7	15.4	21.7	20.8
I-Butane	10.7	11.0	15.5	9.5		9.3
N-Pentane	15.0	8.0	8.0		19.4	10.1
I-Pentane	13.4	8.0	12.3	22.2		11.2
Hexane	7.3	3.0	0.2		12.8	4.7
I-Hexane	5.2	2.0	1.0	17.1		5.1
Heptane	0.9		0.2		8.7	2.0
I-Heptane	2.0	1.0	8.1	14.0		5.0
I-Octane			1.9			0.4
Benzene			4.3	7.9		2.4
Toluene			3.0	3.9		1.4
	100.0	100.0	100.0	100.0	100.0	100.0

^{*} Tests conducted for the program

[†] Ref. 6

TABLE 4-03F. CRUDE OIL STORAGE, PETROLEUM MARKETING OPERATIONS

Organic Compound		Crude S	ource	
(wt. %)	Foreign	Foreign	Domestic	Composite
Methane	6.0	0.4	1.5	2.6
Ethane	8.1	1.1	2.0	3.7
Propane	15.0	19.3	13.0	15.8
Butane	31.1	33.7	25.7	30.2
Pentane	12.8	22.7	20.1	18.5
Hexane	7.4	9.7	9.3	8.8
Heptane	8.8	8.5	15.0	10.8
Octane +	10.8	4.6	13.4	9.6
	100.0	100.0	100.0	100.0

Ref. 6

Data Qualifications

The following profiles may be used to characterize the organic emissions from typical petroleum product storage. Because of the variety of crude oil or petroleum product specifications, some VOC species will differ from region to region.

4.03-9 26200-795

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE FIXED ROOF, GASOLINE COMPOSITE, WORKING AND BREATHING LOSSES COMP DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE PROCESS MODIFICATION: NONE KVB PROFILE KEY 0098

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43204	PROPANE	44. 09	. 10	. 19	1 PARAFFIN
2	43212	N-BUTANE	58. 12	2. 20	3. 5 9	1 PARAFFIN
3	43214	ISOBUTANE	58. 12	. 50	. 85	1 PARAFFIN
4	43220	N-PENTANE	7 2. 1 5	12. 20	15. 96	1 PARAFFIN
5	43231	N-HEXANE	86. 17	16. 30	17. 85	1 PARAFFIN
6	43232	N-HEPTANE	100. 20	9. 30	8. 78	1 PARAFFIN
7	43233	N-OCTANE	114. 23	10. 10	8. 31	1 PARAFFIN
8	43235	N-NONANE	128. 25	1.80	1. 32	1 PARAFFIN
9	43238	N-DECANE	142. 28	2. 60	1. 70	1 PARAFFIN
10	43213	BUTENE	56. 10	40	. 66	2 OLEFIN
11	43224	1-PENTENE	70. 13	2. 50	3. 40	2 OLEFIN
12	45102	ISOMERS OF XYLENE	106. 16	13. 50	11. 99	3 AROMATIC
13	45104	ISOMERS OF ETHYLTOLUENE	120. 19	6. 50	5. 10	3 AROMATIC
14	45105	ISOMERS OF BUTYLBENZENE	134, 21	4. 40	3. 12	3 AROMATIC
15	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	5. 50	4. 34	3 AROMATIC
16	45202	TOLUENE	92. 13	9. 70	9. 92	3 AROMATIC
17	45201	BENZENE	78. 11	2. 40	2. 93	7 NON-REACTIVE
		TOTAL		100. 00	100. 01	
	9 COMP	OUNDS OF CLASSIFICATION 1	80. 75	55. 10	58. 55	
	2 COMP	DUNDS OF CLASSIFICATION 2	67. 85	2. 90	4.06	
	5 COMP	DUNDS OF CLASSIFICATION 3	108. 50	39. 60	34. 47	
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	00	. 00	
	1 COMP	DUNDS OF CLASSIFICATION 7	78. 11	2. 40	2. 93	
	17 COMP	DUND COMPOSITE	94. 40	100.00	100. 01	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-03-001-01,-03 (REFER ALSO TO SUMMARY INDEX)

DECEMBER 14, 1978 TABLE 4-03-001B

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE FIXED ROOF, GASOLINE BREATHING, FCC UNIT AND REFORMER BLEND DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0187

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	5. 20	4. 12	1	PARAFFIN	
2	43106	ISOMERS OF HEPTANE	100. 20	18. 40	12. 63	1	PARAFFIN	
3	43122	ISOMERS OF PENTANE	72. 15	3. 20	3. 02	1	PARAFFIN	
4	43212	N-BUTANE	58. 12	58. 30	68. 84	1	PARAFFIN	
5	43214	ISOBUTANE	58. 12	. 30	. 34	1	PARAFFIN	
6	43220	N-PENTANE	72, 15	1. 50	1. 44	1	PARAFF IN	
7	43231	N-HEXANE	86. 17	4. 60	3. 64	1	PARAFFIN	
8	43232	N-HEPTANE	100, 20	2. 70	1. 85	1	PARAFFIN	
9	45102	ISOMERS OF XYLENE	106, 16	1. 90	1. 24	3	AROMATIC	
10	45202	TOLUENE	92 , 13	3. 90	2. 88	3	AROMATIC	
		TOTAL		100. 00	100. 00			
	в сомр	OUNDS OF CLASSIFICATION 1	67. 40	94 . 20	95. BB			
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00			
	2 COMP	OUNDS OF CLASSIFICATION 3	96, 35	5. BO	4. 12			
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
	10 COMP	OUND COMPOSITE	68. 59	100. 00	100. 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE TAKEN FROM ROOF HATCH

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-03-001-01 (REFER ALSO TO SUMMARY INDEX)

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE, PRODUCTION

FIXED ROOF, COMPOSITE PROFILE FOR CRUDE OIL

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0296

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43115	C-7 CYCLOPARAFFINS	98. 19	1. 30	. 67	1	PARAFFIN
2	43116	C-8 CYCLOPARAFFINS	112. 23	. 50	. 26	1	PARAFFIN
3	43122	ISOMERS OF PENTANE	72. 15	1. 50	1.08	1	PARAFFIN
4	43204	PROPANE	44. 09	17. 60	20. 46	1	PARAFFIN
5	43212	N-BUTANE	58. 12	27. 10	23. 90	1	PARAFFIN
6	43214	ISOBUTANE	58. 12	1. 50	1. 33	1	PARAFFIN
7	43220	N-PENTANE	72. 15	14.60	10. 36	1	PARAFFIN
5 3	43231	N-HEXANE	86. 17	7. 90	4. 72	1	PARAFFIN
9	43232	N-HEPTANE	100. 20	9. 20	4. 72	1	PARAFFIN
10	43233	N-OCTANE	114. 23	6. 90	3.08	1	PARAFFIN
11	43201	METHANE	16.04	6. 20	19. 85	6	METHANE
12	43202	ETHANE	30. 07	5. 60	9. 54	7	NON-REACTIVE
13	45201	BENZENE	78. 11	. 10	. 05	7	NON-REACTIVE
		TOTAL		100. 00	100. 02		
	10 COMP	DUNDS OF CLASSIFICATION 1	64. 05	88. 10	70. 58		
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 6	16. 04	6. 20	19. 85		
	2 COMP	DUNDS OF CLASSIFICATION 7	30. 32	5. 70	9. 59		
•	13 COMP	DUND COMPOSITE	51. 28	100.00	100. 02		

- NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE
 - B. REFERENCES: KVB TEST DATA (REF. 3), LITERATURE TEST DATA (REF. 6)
 - C. APPLICABLE SCC CATEGORIES: 4~03-001-02,-04

DECEMBER 14, 1978 TABLE 4-03-001D

PDINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE, REFINERY FIXED ROOF, COMPOSITE PROFILE FOR CRUDE OIL

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0297

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	Bado	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATI ON
1	43105	ISOMERS OF HEXANE	86. 17	5. 10	2.88	1	PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	5. 00	2. 44	1	PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	40	. 20	1	PARAFFIN
4	43122	ISOMERS OF PENTANE	72. 15	11.20	7. 56	1	PARAFFIN
5	43204	PROPANE	44. 09	16. 10	17. 80	1	PARAFFIN
6	43212	N-BUTANE	58. 12	20. 80	17. 45	1	PARAFFIN
7	43214	ISOBUTANE	58. 12	9. 30	7. 80	1	PARAFFIN
8	43220	N-PENTANE	72. 15	10.10	6. 83	1	PARAFFIN
9	43231	N-HEXANE	86.17	4. 70	2. 68	1	PARAFFIN
10	43232	N-HEPTANE	100. 20	2. 00	. 98	1	PARAFFIN
11	45202	TOLUENE	92. 13	1.40	73	3	AROMATIC
12	43201	METHANE	16. 04	8. 80	26. 77	6	METHANE
13	43202	ETHANE	30. 07	2. 70	4. 39	7	NON-REACTIVE
14	45201	BENZENE	78. 11	2. 40	1. 51	7	NON-REACTIVE
		TOTAL		100. 00	100. 02		
	10 COMP	OUNDS OF CLASSIFICATION 1	62 . 07	84. 70	66. 62		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 3	92. 13	1.40	. 73		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	8. 80	26. 77		
	2 COMP	OUNDS OF CLASSIFICATION 7	42. 37	5. 10	5. 90		
•	14 COMP	OUND COMPOSITE	48. 81	100.00	100. 02		

- NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLES
 - B. REFERENCES: KVB TEST DATA (REF. 3), LITERATURE TEST DATA (REF. 6)
 - C. APPLICABLE SCC CATEGORIES: 4-03-001-02, -04, 4-03-002-03, -04

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

FIXED ROOF, COMMERCIAL JET FUEL (JET-A)

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0100

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	43232	N-HEPTANE	100. 20	. 10	. 17	1	PARAFFIN	
2	43233	N-OCTANE	114. 23	. 50	. 66	1	PARAFFIN	
3	43235	N-NONANE	128. 25	4. 70	6. 11	1	PARAFFIN	
4	43238	N-DECANE	142. 28	19. 60	22. 77	1	PARAFFIN	
5	43241	N-UNDECANE	156. 30	20. 30	21. 45	1	PARAFFIN	
6	43255	N-DODECANE	170. 33	18. 20	17. 66	1	PARAFFIN	
7	43258	N-TR I DECANE	184. 36	17. 70	15. 84	1	PARAFFIN	
8	43259	N-TETRADECANE	198. 38	11. 70	9. 74	1	PARAFFIN	
9	43260	N-PENTADECANE	212. 41	7. 20	5. 61	1	PARAFFIN	
		TOTAL		100. 00	100. 01			
	9 COMP	QUINDS OF CLASSIFICATION 1	165. 19	100. 00	100. 01			
	O COMP	OUNDS OF CLASSIFICATION 2	. 0 0	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00			
		OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00			
•	9 COMP	OUND COMPOSITE	165. 19	100.00	100, 01			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 11)

C. APPLICABLE SCC CATEGORIES: 4-03-001-05, -50, 4-03-002-05, 4-03-003-03

DECEMBER 14, 1978 TABLE 4-03-001F

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

FIXED ROOF, BENZENE

DATA CONFIDENCE LEVEL: 11

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0298

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	45201	BENZENE		78. 11	100. 00	100. 00	7 NON-REACTIVE
			TOTAL		100. 00	100. 00	
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	QUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00	
	1 COMP	OUNDS OF	CLASSIFICATION 7	78. 11	100.00	100. 00	
	1 COMP	OUND COMP	POSITE	78, 11	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-08, -53, 4-03-002-08, 4-03-003-06

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE DECEMBER 14, 1978 TABLE 4-03-001G

FIXED ROOF, CYCLOHEXANE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0299

INE NO.	SARUAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43248	CYCLOHEXANE	84.16	100.00	100.00	1 PARAFFIN
		TOTAL		100. 00	100. 00	
	1 COMP	OUNDS OF CLASSIFICATION 1	84. 16	100. 00	100. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	DUND COMPOSITE	84. 16	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-09, -54, 4-03-002-09, 4-03-003-07

DECEMBER 14, 1978 TABLE 4-03-001H

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE FIXED ROOF, CYCLOPENTANE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0300

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION	
1	43242	CYCLOPENTANE	70. 14	100.00	100.00	1 PARAFFIN	
		TOTAL		100. 00	100. 00		
	1 COMP	OUNDS OF CLASSIFICATION 1	70. 14	100. 00	100. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	1 COMP	OUND COMPOSITE	70. 14	100.00	100, 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-10, -55, 4-03-002-10, 4-03-003-08

4.03-18

26200-795

DECEMBER 14, 1978 TABLE 4-03-001J POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

FIXED ROOF, HEPTANE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0301

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43232	N-HEPTA	NE	100. 20	100.00	100.00	1	PARAFFIN	
			TOTAL		100. 00	100. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 1	100. 20	100.00	100.00			
	O COMP	DUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	OUND COM	POSITE	100. 20	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-11,-56, 4-03-002-11, 4-03-003-09

DECEMBER 14, 1978 TABLE 4-03-001K POINT SOURCE EVAPORATION, PETROLEUM PRODUCTS STORAGE

FIXED ROOF, HEXANE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KV9 PROFILE KEY 0230

.INE NO.	SARGAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	43231	N-HEXANE		86. 17	100. 00	100.00	1	PARAFFIN	
			TOTAL		100. 00	100. 00			
	1 COMP	DUNDS OF	CLASSIFICATION 1	86. 17	100. 00	100. 00			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	DUNDS OF	CLASSIFICATION 5	. 0 0	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	OUND COMP	OSITE	B6. 17	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-12,-57 4-03-002-12, 4-03-003-10

DECEMBER 14, 1978 TABLE 4-Q3-001L POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE FIXED ROOF, ISO-OCTANE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0302

ND.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43107	ISOMERS	OF OCTANE TOTAL	114. 23	100.00	100. 00 100. 00	1 PARAFFIN
					-00	1-0. 00	
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 23	100. 00	100.00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	DUND COMP	POSITE	114. 23	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-13, -58, 4-03-002-13, 4-03-003-11

DECEMBER 14, 1978 TABLE 4-03-001M POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

FIXED ROOF, ISOPENTANE
DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0231

LINE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43122	ISOMERS	OF PENTANE TOTAL	72. 15	100. 00 100. 00	100. 00 100. 00	1 PARAFFIN
	1 COMP	OUNDS OF	CLASSIFICATION 1	72. 15	100. 00	100. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 6	00	00	. 00	
	O COMP	DUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	OUND COM	POSITE	72. 15	100. 00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-14,-59 4-03-002-14, 4-03-003-12

DECEMBER 14, 1978

TABLE 4-03-001N

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

FIXED ROOF, PENTANE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0303

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43220	N-PENTANE	72. 15	100.00	100.00	1 PARAFFIN
		TOTAL		100. 00	100. 00	
	1 COMP	OUNDS OF CLASSIFICATION 1	72. 15	100. 00	100. 00	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	DUND COMPOSITE	72. 15	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-15, -60, 4-03-002-15, 4-03-003-13

DECEMBER 14, 1978 TABLE 4-03-001P POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE

FIXED ROOF, TOLUENE

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0185

INE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION	
1	45202	TOLUENE		92. 13	100. 00	100.00	3	AROMATIC	
			TOTAL		100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION 1	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 3	92 . 13	100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00			
	1 COMP	OUND COME	POSITE	92. 13	100.00	100, 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF STORED PRODUCT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-03-001-16, -61, 4-03-002-16, 4-03-003-14

DECEMBER 14, 1978 TABLE 4-03-0010

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE FIXED ROOF, COMPOSITE PROFILE FOR CRUDE OIL AND WASTE WATER DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0188

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION	
1	43105	ISOMERS OF HEXANE	86. 17	1. 30	. 56	1 PARAFFIN	_
2	43112	ISOMERS OF DODECANE	170. 33	30	. 08	1 PARAFFIN	
3	43115	C-7 CYCLOPARAFFINS	98. 19	4. 50	1. 73	1 PARAFFIN	
4	43116	C-B CYCLOPARAFFINS	112. 23	1. 30	. 45	1 PARAFFIN	
5	43122	ISOMERS OF PENTANE	72. 15	7. 10	3. 68	1 PARAFFIN	
6	43204	PROPANE	44. 09	21.30	18. 12	1 PARAFFIN	
7	43212	N-BUTANE	58. 12	16. 60	10. 73	1 PARAFFIN	
8	43214	ISOBUTANE	58. 12	8. 30	5. 37	1 PARAFFIN	
9	43220	N-PENTANE	72. 15	5. 80	3. 00	1 PARAFFIN	
10	43231	N-HEXANE	86. 17	4. 20	1.84	1 PARAFFIN	
11	43201	METHANE	16. 04	16. 80	39. 29	6 METHANE	
12	43202	ETHANE	30. 07	11. 90	14, 86	7 NON-REACTIVE	
13	45201	BENZENE	78. 11	. 60	. 30	7 NON-REACTIVE	
		TOTAL		100. 00	100. 01		
	10 COMP	DUNDS OF CLASSIFICATION 1	58, 33	70. 70	45. 56		
		DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
		DUNDS OF CLASSIFICATION 6	16. 04	16. 80	39. 2 9		
		DUNDS OF CLASSIFICATION 7	31. 02	12. 50	15. 16		
	13 COMP	DUND COMPOSITE	37. 58	100. 00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLES TAKEN WITHIN LIQUID-AIR INTERFA

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-03-001-98

26200-795

DECEMBER 14, 1978 TABLE 4-03-003

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE VARIABLE VAPOR SPACE, LIQUIFIED PETROLEUM GAS DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0232

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43204	PROPANE	44. 09	90. 60	88. 77	1	PARAFFIN
2	43214	ISOBUTANE.	58. 12	. 20	. 13	1	PARAFFIN
3	43205	PROPYLENE	42. 08	5. 10	5. 23	2	OLEFIN
4	43202	ETHANE	30. 07	4. 10	5. 87	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP	OUNDS OF CLASSIFICATION 1	44. 11	90. 80	88. 90		
		OUNDS OF CLASSIFICATION 2	42. OB	5. 10	5. 23		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	4. 10	5. 87		
	4 COMP	OUND COMPOSITE	43. 18	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-03-003-99

POINT SOURCE EVAPORATION, PETROLEUM PRODUCT STORAGE FIXED ROOF, COMPOSITE FOR CRUDE OIL, MARINE TERMINAL DATA CONFIDENCE LEVEL: 111

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0305

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43204	PROPANE	44. 09	15. 80	20. 90	1 PARAFFIN
2	43212	N-BUTANE	58 . 12	30. 20	30. 36	1 PARAFFIN
3	43220	N-PENTANE	72. 15	18. 50	14. 94	1 PARAFFIN
4	43231	N-HEXANE	86. 17	8. 80	5. 95	1 PARAFFIN
5	43232	N-HEPTANE	100. 20	10.80	6. 30	1 PARAFFIN
6	43233	N-OCTANE	114. 23	9. 60	4. 90	1 PARAFFIN
7	43201	METHANE	16. 04	2. 60	9. 46	6 METHANE
8	43202	ETHANE	30. 07	3. 70	7. 18	7 NON-REACTIVE
		TOTAL		100. 00	99. 99	
	6 COMP	OUNDS OF CLASSIFICATION 1	65. 60	93. 70	63, 35	
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00	
		OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	_	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
		DUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	-	DUNDS OF CLASSIFICATION 6	16. 04	2. 60	9, 46	
		DUNDS OF CLASSIFICATION 7	30. 07	3. 70	7. 18	
•	В СОМР	DUND COMPOSITE	58. 36	100.00	99, 99	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING JUDGEMENT OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 6)

C. APPLICABLE SCC CATEGORIES: 4-06-002-02,-27

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Danielson, J. A. (Ed), "Air Pollution Engineering Manual," Environmental Protection Agency, Reserach Triangle Park, NC, AP-40, May 1973.
- 3. Taback, H. J. et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. "Evaporation Loss in the Petroleum Industry Causes and Control," American Petroleum Institute, Evaporation and Loss Committee, API Bulletin 2513, Washington, DC, 1959.
- 5. "Evaluation of Hydrocarbon Emissions from Floating Roof Petroleum Tanks," Engineering-Science, Inc., sponsored by the Western Oil and Gas Association, January 1977.
- 6. "Hydrocarbon Emissions from Fixed-Roof Petroleum Tanks," Engineering-Science, Inc., sponsored by the Western Oil and Gas Association, July 1977.
- 7. Adrian, R., "Emissions from Fixed Roof Tanks Santa Barbara County," California Air Resources Board, November 1978.
- 8. Burklin, C. E., et al., "Revision of Evaporative Hydrocarbon Emission Factors," report for EPA-450/3-76-039, 1976.
- 9. Jonker, P. E., et al., "Control Floating Roof Tank Emissions," Hydrocarbon Processing, Gulf Publishing Co., May 1977.
- 10. "61st Annual Report of the State Oil and Gas Supervisor," California Division of Oil and Gas, Report No. PRO6, 1975.
- 11. Mayrsohn, H. and Crabtree, J., Source Reconciliation of Atmospheric Hydrocarbons in the South Coast Air Basin, 1975, California Air Resources Board, December 1976.

4-05 POINT SOURCE EVAPORATION, PRINTING PRESS GRAPHIC ART PROCESSES Introduction 1,2

There are four major classes of graphic art printing processes:

- 1. letterpress,
- 2. lithographic,
- 3. flexographic, and
- 4. rotogravure.

The first two processes use oil or paste inks, and the last two use solvent inks. These inks vary in physical appearance, composition, method of application, and drying mechanism. Although flexographic and rotogravure inks have many elements in common with paste inks, they differ because of their low viscosity and dry by evaporation of highly volatile solvents.

There are three general processes in the manufacture of printing inks: (1) cooking the vehicle and adding dyes, (2) grinding of a pigment into the vehicle using a roller mill, and (3) replacing water in the wet pigment pulp by an ink vehicle (commonly known as the flushing process). The ink "varnish" or vehicle is generally cooked in large kettles at 200 ° to 600 °F (93 ° to 315 °C) for an average of eight to twelve hours in much the same way that regular varnish is made. Mixing of the pigment and vehicle is done in dough mixers or in large agitated tanks. Grinding is most often carried out in three-roller or five-roller horizontal or vertical mills.

Since the information and details of each of the different printing processes vary considerably, it was decided that each process should be treated as a separate section in this report.

4.05 POINT SOURCE EVAPORATION, PRINTING PRESS

4-05-002 LETTERPRESS PROCESS

Process Description 1

Letterpress printing is the oldest and most basic form of printing and still predominates in periodical and newspaper publishing. In letterpress printing, ink is transferred to the paper from the image surface. This surface is raised relative to the nonprinting surface of the plate. Consult Reference 2 for a more detailed explanation of this graphic art process.

Emissions

The major sources of hydrocarbon emissions from letterpress printing occur from the:

- . hot air dryer
- . press unit, and
- . chill rolls

In letterpress printing operations, the ink is the major source of hydrocarbons. Hydrocarbon emissions rate data and process information can be obtained from Reference 1. Hydrocarbon species data on emissions from these sources is presented in

Controls

Control of the hydrocarbon emissions from letterpress operations in general are categorized according to the following:

- . Process modification--microwave drying
 --infrared drying
 --electron beam drying
 --ultraviolet drying
- . Ink modification--aqueous inks
 --solventless inks
- . Adsorption--activated carbon

For a more thorough explanation of each control method, consult References 1, 3 and 4.

Profile Basis³

The basis for profiles 4-05-002A and B was test data obtained from a typical letterpress printing operation employing paste type inks and a hot air dryer. Profile 4-05-002A included VOC's from incomplete combustion from the direct fired dryer. A sampling train consisting of a glass gas collection bottle and NIOSH type charcoal tubes was employed. Samples were taken at points within the exhaust duct work located upstream and downstream of a thermal afterburner.

Data Qualifications

The following profiles may be used to characterize the organic emissions from a typical lettpress printing operation employing a hot air dryer and a thermal gas fired afterburner.

POINT SOURCE EVAPORATION, PRINTING PRESS LETTERPRESS, INKING AND DRYING (DIRECT-FIRED DRIER)

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0334

.INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43204	PROPANE	44. 09	3. 00	1. 30	1	PARAFFIN
2	43212	N-BUTANE	58. 12	3. 00	1.00	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	1. 00	. 33	1	PARAFFIN
4	43203	ETHYLENE	28. 05	13. 00	8. 88	2	OLEFIN
5	43205	PROPYLENE	42. 08	3. 00	1. 36	2	OLEFIN
6	43213	BUTENE	56. 10	1.00	. 34	2	OLEFIN
7	43201	METHANE	16. 04	69. 00	82. 33	6	METHANE
8	43202	ETHANE	30. 07	7. 00	4. 46	7	NON-REACTIVE
		TOTAL		100. 00	100.00		
	3 СОМР	OUNDS OF CLASSIFICATION 1	51. 19	7. 00	2. 63		
	3 COMP	OUNDS OF CLASSIFICATION 2	30, 75	17. 00	10. 58		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	6 9. 00	82. 33		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	7. 00	4. 46		
	8 COMP	OUND COMPOSITE	19. 15	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-05--002-99

DECEMBER 14, 1978 TABLE 4-05-002

POINT SOURCE EVAPORATION, PRINTING PRESS

LETTERPRESS, INKING PROCESS DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: CATALYTIC AFTERBURNER

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0166

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44. 09	12. 00	5. 42	1	PARAFFIN
2	43212	N-BUTANE	58. 12	1.00	. 34	1	PARAFFIN
3	43201	METHANE	16. 04	63. 00	78. 33	6	METHANE
4	43202	ETHANE	30. 07	24.00	15. 91	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP	OUNDS OF CLASSIFICATION 1	44. 92	13. 00	5. 76		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	63. 00	78 . 33		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	24. 00	15. 91		
	4 COMP	OUND COMPOSITE	19. 94	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-05-002-99

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 5. Sonnichsen, T. W., KVB Engineer.

4-05 POINT SOURCE EVAPORATION, PRINTING PRESS

4-05-003 FLEXOGRAPHIC PROCESS

Process Description 2

Flexographic printing is similar to letterpress, where the image area is raised above the surface of the plate. Ink is transferred directly to the image area of the plate and directly from the plate to the paper or substrate. Whenever the plate is made of rubber and alcohol based inks are used, the process is flexography. The process is always web fed and is used for medium or long runs on a variety of substrates, including heavy paper, fiberboard, metal, and plastic foil.

Flexographic processes differ among themselves mainly in the type of ink used. Most flexographic inks are fluid in consistency and contain about 55% organic solvent. The solvent may be alcohol or alcohol mixed with aliphatic hydrocarbons or esters.

Flexography printing uses two similar but different processes. The composition of the ink and the inclusion of drying are the main areas where the processes differ. The two types of flexographic printing are:

- 1. flexographic, publication and
- 2. flexographic, newspaper.

For further process information, consult Reference 2.

Emissions 1

The major sources of hydrocarbon emissions from flexographic printing occur from the:

- . hot air dryer
- . press unit, and
- . chill rolls

In a typical flexographic operation, the ink is the major source of hydrocarbons. Emission rates and information can be found in Reference 1.

Organic species data on emission from these sources using alcohol based inks is presented in profiles 4-05-003A and B.

Controls 1

Control of hydrocarbon emission from flexographic web fed offset printing are categorized according to the following:

- process modification microwave drying infrared drying electron beam drying ultraviolet drying
- ink modification aqueous inks solventless inks
- . incineration, exhaust gas thermal catalytic combination, and
- . adsorption activated carbon

For a thorough explanation of each control method, consult References 1, 3 and 4.

Profile Basis 3,5

Profiles 4-05-003A and B were developed using engineering evaluation and judgement of data obtained from industry questionnaires and pertinent, literature, respectively.

Data Qualifications

The following profiles may be used to characterize the organic emissions from a typical flexographic printing operation using alcohol based inks.

POINT SOURCE EVAPORATION, PRINTING PRESS FLEXOGRAPHIC, COMPOSITE OF ALCOHOL BASED SOLVENT DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

VB PROFILE KEY 0172

LINE NO.	SAROAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION	
,,,,,			***************************************	WE101	VOLU.12		
1	43302	ETHYL ALCOHOL	46. 07	47. 60	57. 10	5 MISCELLANEOUS	
2	43303	N-PROPYL ALCOHOL	60. 09	24. 30	22. 33	5 MISCELLANEOUS	
3	43304	ISOPROPYL ALCOHOL	60. 09	10. 70	9. 84	5 MISCELLANEOUS	
4	43351	ETHYL ETHER	74. 12	. 9 0	. 66	5 MISCELLANEOUS	
5	43433	ETHYL ACETATE	88. 10	15. 10	9. 45	5 MISCELLANEOUS	
6	43452	2-ETHOXYETHYL ACETATE	132. 00	1. 40	. 61	5 MISCELLANEOUS	
		TOTAL		100.00	99. 99		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 0 0	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	6 COMP	OUNDS OF CLASSIFICATION 5	55 . 26	100.00	99. 99		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
				100 00			
	6 CUMP	OUND COMPOSITE	55. 26	100.00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

B. REFERENCES: PLASTIC COATED PAPER MILK CARTON MANUFACTURERS

C. APPLICABLE SCC CATEGORIES: 4-05-003-01, -99

4.05-10

26200-795

DECEMBER 14, 1978 TABLE 4-05-003B

POINT SOURCE EVAPORATION, PRINTING PRESS FLEXOGRAPHIC, N-PROPYL ALCOHOL DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0304

NO.	SAROAD CODE		CHEMICAL NAME		MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CLA	CHEMICAL ASSIFICATION	
1	43303	N-PROPYI	- ALCOHOL TOTAL		60. 09	100. 00 100. 00	100. 00 100. 00	5	MISCELLANEOUS	
			TOTAL			100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION	1	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	2	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	3	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	4	. 00	. 00	. 00			
	1 COMP	OUNDS OF	CLASSIFICATION	5	60. 09	100.00	100.00			
	O COMP	OUNDS OF	CLASSIFICATION	6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION	7	. 00	. 00	. 00			
	1 COMP	OUND COM	POSITE	-	60. 09	100.00	100. 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA INSPECTION OF SOLVENT FORMULATION

B. REFERENCES: ENGINEERING JUDGEMENT

C. APPLICABLE SCC CATEGORIES: 4-05-003-06, 4-05-005-09

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- 2. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 5. Sonnichsen, T. W., KVB Engineer.

4-05 POINT SOURCE EVAPORATION, PRINTING PRESS

4-05-004 LITHOGRAPHIC PROCESS

Process Description

Lithography printing is characterized by having the image area on the same plane as the non-image area. The image area chemically attracts the ink while the non-image area chemically repels ink. The printing image is applied to a cylinder which transfers the inked image directly to the substrate. This process is direct lithography. A second process called offset lithography involves the applying of a printed image to a cylinder where the inked image is transferred to a rubber blanket cylinder which in the same revolution prints the wet inked image onto the substrate. When a web or continuous roll of paper is employed with the offset process, it is called web-offset printing.

For further process information, consult Reference 1.

Emissions 1

The major areas of hydrocarbon emissions from web-offset printing occur from the:

- press
- . dryer
- . chill rolls, and
- . ink fountains

The ink and the coating on the paper are the major sources of hydrocarbons in web-offset printing operations. Reference 1 should be consulted for further information on emission rates and process information.

Profiles 4-05-004A and B present the hydrocarbon species emissions from these sources.

Controls

Control of hydrocarbon emissions from web-offset printing are cateqorized according to the following:

- process modification microwave infrared drying electron beam drying ultraviolet drying
- . ink modification aqueous inks solventless inks
- . combustion, exhaust gas thermal catalytic combination
- . adsorption activated carbon

For a thorough explanation of each control method, consult References 1, 3 and 4.

Profile Basis 3

The data presented in profiles 3-05-004A and B is based on tests conducted at a typical lithographic printing operation. A hot air dryer was employed in profile 4-05-004A and the VOC species reflect the incomplete combustion. Samples of the exhaust gas were taken from points within the exhaust duct work located upstream and downstream of a thermal afterburner. Gas samples were collected in glass collecting bottles and NIOSH type charcoal tubes.

Data Qualification

The following profiles may be used to characterize the organic emissions from a typical lithographic printing operation employing a hot air dryer and a thermal gas-fired afterburner.

POINT SOURCE EVAPORATION, PRINTING PRESS
LITHOGRAPHY, INKING AND DRYING (DIRECT-FIRED DRIER)

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0333

LINE NO.	SAROAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	Ci	CHEMICAL ASSIFICATION
NU.	CODE		MCIGIII	WC19III	VOLONIE	CL	ASSIT TON TON
1	43204	PROPANE	44. 09	3. 30	2. 20	1	PARAFFIN
2	43212	N-BUTANE	58. 12	11.50	5. 80	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	1.80	. 9 1	1	PARAFFIN
4	43304	ISOPROPYL ALCOHOL	60. 09	. 90	. 44	5	MISCELLANEOUS
5	43802	DICHLOROMETHANE	84. 94	34 . 90	12. 04	5	MISCELLANEOUS
6	43201	METHANE	16.04	37. 70	68. 85	6	METHANE
7	43202	ETHANE	30. 07	10.00	9 . 76	7	NON-REACTIVE
		TOTAL		100. 10	100.00		
	3 СОМР	OUNDS OF CLASSIFICATION 1	54. 6 6	16. 60	8. 91		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	00	. 0 0		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	2 COMP	OUNDS OF CLASSIFICATION 5	84.06	35. 80	12. 48		
		OUNDS OF CLASSIFICATION 6	16.04	37. 70	68. 8 5		
		OUNDS OF CLASSIFICATION 7	30. 07	10. 00	9 . 76		
	7 COMP	DUND COMPOSITE	29. 34	100. 10	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-05-004-99

DECEMBER 14, 1978 TABLE 4-05-004B

POINT SOURCE EVAPORATION, PRINTING PRESS LITHOGRAPHY, INKING AND DRYING

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: THERMAL AFTERBURNER

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0332

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATI ON
1	43203	ETHYLE NE	28. 05	2. 00	2. 31	2	OLEFIN
2	43205	PROPYLENE	42. 08	33. 60	25. 99	2	OLEFIN
3	43213	BUTENE	56. 10	13. 60	7. 87	2	OLEFIN
4	43224	1-PENTENE	70. 13	11. 90	5. 53	2	OLEFIN
5	43502	FORMALDEHYDE	30. 03	21. 80	23. 62	4	CARBONYL
6	43201	METHANE	16. 04	17. 10	34. 68	6	METHANE
		TOTAL		100. 00	100. 00		
	O COME	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 2	47. 67	61. 10	41. 70		
		OUNDS OF CLASSIFICATION 3	00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	30. 03	21. 80	23. 62		
		OUNDS OF CLASSIFICATION 5	. 00	00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16, 04	17. 10	34. 68		
		POUNDS OF CLASSIFICATION 7	. 00	00	. 00		
	A COMP	OUND COMPOSITE	32, 53	100. 00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF SAMPLING TRAIN CATCH

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-05-004-99

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- "Compilation of Pollutant Emission Factors," Environmental Protection Agency. Research Triangle Park, NC, AP-42, August 1977.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol I and II, KVB, Inc., Tustin, CA, June 1978.
- Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973.
- 5. Sonnichsen, T. W., KVB Engineer.

POINT SOURCE EVAPORATION, PRINTING PRESS

4-05-005 GRAPHIC ARTS--ROTOGRAVURE

Process Description

4-05

Gravure printing is a type of printing where the image area is recessed relative to the surface of the image carrier. Ink is picked up in the engraved area, and excess ink is scraped off the nonimage area with a "doctor blade."

Ink is transerred directly from the image carrier to the paper or film. Gravure may be sheet-fed or roll-fed. Sheet-fed gravure uses either a flat plate for an image carrier, or a curved plate which is attached to a cylinder. In roll-fed gravure, or rotogravure, the image is engraved in the cylinder itself. Rotogravure may be used for coated or uncoated paper, film, foil, and many combinations thereof.

The ink used in high speed gravure printing contains a relatively large amount of low-boiling solvent and has a low viscosity. The rotogravure inks contain approximately 65% highly volatile, aromatic solvent which is not subject to decomposition in the drying process. Control of solvent vapors around the ink fountain is desirable to avoid the danger of explosion. For most commercial operations, the solvent concentration in the exhaust gases ranges between 25% and 40% of the lower explosive limit.

For further process information consult Reference 1.

Emissions 1

The major areas of hydrocarbon emissions from rotogravure printing occur from the:

- . hot air dryer
- . press unit
- . chill rollers, and
- . ink fountain

In gravure printing operations the ink is the major source of hydrocarbon emissions. Emission rates and information can be found in Reference 1. Hydrocarbon species data on emissions from these sources are presented in profiles 4-05-005A, B, C and D.

Controls

Control of hydrocarbon emissions from gravure printing are categorized according to the following:

- . adsorption activated carbon

For a thorough explanation of each control method, consult References 1, 3 and 4.

Profile Basis 3,5

Profiles 4-05-005A, B, C and D were developed from data obtained from industry questionnaires and literature. An engineering evaluation of the data was performed.

Data Qualification

The following profiles may be used to characterize the organic emissions from a typical rotogravure printing process.

POINT SOURCE EVAPORATION, PRINTING PRESS GRAVURE, PAPERBOARD PRINTING

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0181

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEICHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43118	MINERAL SPIRITS	114. 00	5. 40	3. 80	1	PARAFFIN
2	43231	N-HEXANE	86. 17	6. 20	5. 82	1	PARAFFIN
3	43232	N-HEPTANE	100. 20	6. 60	5. 33	1	PARAFFIN
4	45202	TOLUENE	92 . 13	12. 00	10. 50	3	AROMATIC
5	43551	ACETONE	58. 08	. 80	1. 13	4	CARBONYL
6	43552	METHYL ETHYL KETONE	72. 10	7. 80	8. 72	4	CARBONYL
7	43119	LACTOL SPIRITS	114.00	1.00	. 73	5	MISCELLANEOUS
8	43302	ETHYL ALCOHOL	46. 07	8. 60	15. 11	5	MISCELLANEOUS
9	43303	N-PROPYL ALCOHOL	60. 09	. 30	. 40	5	MISCELLANEOUS
10	43304	ISOPROPYL ALCOHOL	60. 09	10. 80	14. 54	5	MISCELLANEOUS
11	43433	ETHYL ACETATE	88. 10	18. 20	16. 72	5	MISCELLANEOUS
12	43434	PROPYL ACETATE	102. 13	1. 30	1.05	5	MISCELLANEOUS
13	43444	ISOPROPYL ACETATE	104.00	20. 10	15. 59	5	MISCELLANEOUS
14	43452	2-ETHOXYETHYL ACETATE	132. 00	. 90	. 57	5	MISCELLANEOUS
		TOTAL.		100. 00	100. 01		
	з сомР	OUNDS OF CLASSIFICATION 1	98 . 25	18. 20	14. 95		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 3	9 2. 13	12.00	10. 50		
	2 COMP	OUNDS OF CLASSIFICATION 4	70. 49	8. 60	9. 85		
	8 COMP	OUNDS OF CLASSIFICATION 5	76. 56	61. 20	64. 71		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	14 COMP	OUND COMPOSITE	80. 84	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF QUESTIONNAIRE DATA

4.05-19

B. REFERENCES: REPRESENTATIVE PRINTING COMPANY

C. APPLICABLE SCC CATEGORIES: 4-05-005-01

TABLE 4-05-005B

POINT SOURCE EVAPORATION, PRINTING PRESS GRAVURE, PERIODICALS, PRINTING SOLVENT DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0183

LINE	SAROAD		CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL	
NO.	CODE		NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION	
1	43118	MINERAL	SPIRITS	114. 00	83. 00	80. 26	1	PARAFFIN	
2	45102	ISDMERS	OF XYLENE	106. 16	4. 00	4. 19	3	AROMATIC	
3	45202	TOLUENE		9 2. 13	13. 00	15. 55	3	ARDMATIC	
			TOTAL		100. 00	100.00			
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	83. 00	BO. 26			
	_		CLASSIFICATION 2	. 00	. 00	. 00			
			CLASSIFICATION 3	95. 11	17. 00	19. 74			
			CLASSIFICATION 4	. 00	. 00	. 00			
			CLASSIFICATION 5	. 00	. 00	. 00			
			CLASSIFICATION 6	. 00	. 00	. 00			
			CLASSIFICATION 7	. 00	. 00	. 00			
	3 COMF	OUND COM	POSITE	110, 27	100.00	100.00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

C. APPLICABLE SCC CATEGORIES: 4-05-005-01, 4-05-005-99

POINT SOURCE EVAPORATION, PRINTING PRESS GRAVURE, COMMERCIAL PRINTING SOLVENT

DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0184

LINE NO.	SAROAD CODE		CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATI ON	
1	43118	MINERAL	SPIRITS	114.00	83. 00	BO. 26	1	PARAFFIN	
2	45102	ISOMERS	OF XYLENE	106. 16	4. 00	4. 19	3	AROMATIC	
3	45202	TOLUENE		92. 13	13.00	15. 55	3	AROMATIC	
			TOTAL		100.00	100. 00			
	1 COMP	OUNDS OF	CLASSIFICATION 1	114. 00	83. 00	BO. 26			
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00			
	2 COMP	OUNDS OF	CLASSIFICATION 3	9 5. 11	17. 00	19. 74			
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 6	. 00	. 00	. 00			
	O COMP	OUNDS OF	CLASSIFICATION 7	00	. 00	. 00			
	3 COMP	OUND COM	POSITE	110. 27	100. 00	100. 00			

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

C. APPLICABLE SCC CATEGORIES: 4-05-005-01,-99

POINT SOURCE EVAPORATION, PRINTING PRESS

GRAVURE, GENERAL SOLVENT DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0182

INE NO.	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CL	CHEMICAL ASSIFICATION
MU.	CODE	NAME	WEIGHT	WEIGHT	VULUME	CL	MSS1F1CA110N
1	43118	MINERAL SPIRITS	114.00	51. 00	40. 20	1	PARAFFIN
3	43248	CYCLDHEXANE	84. 16	10.00	10. 70	1	PARAFFIN
3	45106	ISOMERS OF DIETHYLBENZENE	134. 21	7. 00	4. 68	3	AROMATIC
4	45203	ETHYLBENZENE	106. 16	6. 00	5. 13	3	AROMATIC
5	43301	METHYL ALCOHOL	32. 04	6. 00	16. 82	5	MISCELLANEOUS
6	43304	ISOPROPYL ALCOHOL	60. 09	6. 00	8. 99	5	MISCELLANEOUS
7	43305	N-BUTYL ALCOHOL	74. 12	6. 00	7. 28	5	MISCELLANEOUS
8	43435	N-BUTYL ACETATE	116. 16	8. 00	6. 21	5	MISCELLANEOUS
		TOTAL		100.00	100. 01		
	2 COMP	OUNDS OF CLASSIFICATION 1	107. 73	61. 00	50. 90		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	2 COMP	OUNDS OF CLASSIFICATION 3	119. 54	13. 00	9. 81		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	4 COMP	OUNDS OF CLASSIFICATION 5	59. 54	26. 00	39. 30		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	8 COMP	OUND COMPOSITE	8 9. 95	100. 00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 4-05-005-01,-99

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rates and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- 2. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol I and II, KVB, Inc., Tustin, CA, June 1978.
- 4. Danielson, J. A. (ed.), "Air Pollution Engineering Manual," Environmental Protection Agency, Research Triangle Park, NC, AP-40, May 1973
- 5. Sonnichsen, T. W., KVB Engineer.

5-01 SOLID WASTE

5-01 SOLID WASTE, GOVERNMENT

5-01-002 OPEN BURNING DUMP--LANDSCAPE/PRUNING
--AGRICULTURAL

Process Description 1,2

Disposal of agricultural/landscape wastes by open burning is imperative because refuse piles retain horticultural diseases and agricultural pests.

Open burning is performed in many areas as a practical means of clearing the land of these wastes. Open burning is done in open drums or baskets, and in large-scale open dumps, piles or pits.

Agricultural waste burning occurs in two basic patterns, head fires and back fires. Head fires are started at the upwind side of a field and allowed to progress in the direction of the wind whereas back fires are started at the downwind edge and forced to progress in a direction opposing the wind.

Emissions^{1,2}

Ground level open burning is affected by many variables including wind, ambient temperature, agricultural waste composition and moisture content, and compactness of the waste pile.

The relatively low temperatures associated with open burning causes emissions of large quantities of unburned particulates, carbon monoxide and hydrocarbons, while suppressing the emissions of oxides of nitrogen.

Emissions from agricultural/landscape waste burnings are dependent mainly on moisture content and in the case of field crops, whether the refuse is burned in a head-fire or back-fire mode. Other variables such as fuel loading (how much refuse material is burned per unit of land area?) and how the refuse is arranged (e.g., piles, rows or spread out).

Emission factors for open burning (pile form) are presented in the following table. For further information on the variables affecting emissions consult Reference 3.

		Hydrocarbon,	lbs/ton of	Fuel Burne	i	····
<5	5-10	10-15	15-20	>20	Average Emissions	Average Fuel Moist (lbs/ton
Field and v	weeds				17.3	@ 12.3
tule	sorghum mixed weeds cotton	rice ditch bank weeds	barley corn hay wheat	alfalfa asparagus bean oats peas safflower		
Orchard and	d vines				11.1	@ 40.3
apple boysen- berry nectarine prune	almond apricot date fig peach pear walnut	cherry	olive	avocado	7.5	@ 23.7

Source: Reference 3

Controls 3

Landscape/agricultural open burning is an uncontrolled pollution problem from the equipment application point of view. Controls in this instance generally take the form of:

- Burning on days of favorable weather (e.g., low winds, no rain)
- . Drying of debris to a low moisture content
 - -- Orchard and vines, 35-25% moisture
 - -- Field and weeds, ∿12% moisture
- . Cold start type of ignition instead of roll-on type of ignition (especially for orchard crops produces lower hydrocarbon emissions).

^{*} The presented factor ranges are based on the assumption that burning would be conducted under those conditions where lower emissions could be expected (pile form).

For further information on this subject, the reader is advised to consult the references cited at the end of this section.

Profile Basis

Numerous articles on the various aspects of agricultural/landscape waste burning were researched prior to constructing the composite profile of hydrocarbon specie emissions (Profile 5-01-002).

Data Qualification

Profile 5-01-002 is a composite for agricultural/landscape/pruning open burning and may be used to characterize the VOC emissions for government and agricultural operations involving these materials. It should be used with discretion, however, when applying it to open burning of refuse for commercial, institutional and industrial operations unless the same materials discussed above are being burned.

DECEMBER 14, 1978 TABLE 5-01-002

SOLID WASTE, COVERNMENT, OPEN BURNING DUMP

LANDSCAPE/PRUNING

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE KVB PROFILE KEY 0121

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL.
NO.	CODE	NAME	WEIGHT	WEICHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44. 09	1. 90	2. 44	1	PARAFFIN
2	43212	N-BUTANE	5 8. 12	1. 90	1.87	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	1. 90	1. 87	1	PARAFFIN
4	43220	N-PENTANE	72. 15	1. 90	1. 48	1	PARAFFIN
5	43231	N-HEXANE	86. 17	13. <i>9</i> 0	9. 14	1	PARAFFIN
6	43232	N-HEPTANE	100. 20	13. 90	7. 89	1	PARAFFIN
7	43233	N-OCTANE	11423	13. 80	ė. 87	1	PARAFFIN
8	43121	ISOMERS OF PENTENE	70. 13	11, 80	9. 5 3	2	OLEFIN
9	43203	ETHYLENE	28. 05	19. 40	39. 27	2	OLEFIN
10	43213	BUTENE	56. 10	5. 90	5. 96	2	OLEFIN
11	43224	1-PENTENE	70. 13	11.80	9. 5 3	2	OLEFIN
12	43206	ACETYLENE	26. 04	1. 90	4. 14	5	MISCELLANEOUS
		TOTAL.		100. 00	99. 99		
	7 COMP	OUNDS OF CLASSIFICATION 1	88. 55	49, 20	31. 56		
	4 COMP	OUNDS OF CLASSIFICATION 2	43. 13	48. 90	64. 29		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	26, 04	1. 90	4, 14		
		OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	12 COMP	OUND COMPOSITE	56, 76	100.00	99, 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER

C. APPLICABLE SCC CATEGORIES: 5-02-001-02, 5-02-002-02, 5-03-002-02

REFERENCES

- 1. Formica, P. N., "Controlled and Uncontrolled Emission Rate and Applicable Limitations for Eighty Processes," prepared for EPA Office of Air Quality Planning and Standards, Research Triangle Park, NC, EPA-340/1-78-004, April 1978.
- Wayne, L. G. and McQueary, M. L., "Calculation of Emission Factors for Agricultural Burning Activities," prepared for the EPA Office of Air and Waste Management, Office of Air Planning and Standards, Research Triangle Park, NC, EPA-450/3-74-017 and 018 and 450/3-75-087.
- 3. Darley, Ellis F., "Emission Factors from Burning Agricultural Wastes Collected in California," Final Report, Cal/ARB Project 4-011, January 1977.
- 4. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Station-ary Sources in The California South Coast Air Basin," Vol I and II, KVB, Inc., Tustin, CA, June 1978.

5.01-5

26200-795

5-01 SOLID WASTE, GOVERNMENT

5-01-005 INCINERATOR--BAR SCREENED SEWAGE WASTE

Process Description

Sewage waste of sufficient size is filtered out of the incoming, untreated sewage effluent by a rotating bar screen and transported by means of a conveyor to the incinerator where it is combusted to an ash residue.

Composition of this sewage waste is approximately twenty percent by weight solid and eighty percent by weight liquid. Onsite digester gas is generally used to combust the waste.

A normal cycle consists of conveying the wet waste into the incinerator during a low-firing mode for approximately fifteen minutes. Conveying then halts and a period of high firing for 45 minutes takes place. The waste is then presumed to be combusted to an ash residue where it then falls through a grating into an ash pile. The cycle then repeats itself. Unburned ash waste, if present, is again combusted along with the fresh waste during the next cycle and normally removed once a day.

Emissions²

Gaseous hydrocarbon emissions are the result of the incomplete oxidation of the sewage waste and digester gas due mainly to flame quenching, poor mixing and short residence time.

Methane accounts for $^{\circ}80\%$ by weight of the gaseous hydrocarbons being emitted. A large percent of this is due to the incomplete oxidation of the digester gas. A total hydrocarbon emissions rate of 0.25 lb HC per hr (6.3 x 10^{-4} grams/DSCF) was measured. Profile 5-01-005 presents the organic species measured.

Controls^{1,3}

Operating conditions, waste composition and basic incinerator design have a pronounced effect on emissions. The composition of the waste material incinerated will play a major role in the type and degree of control necessary.

Listed below are a number of control methods that may be used. Economics, local air pollution regulations, and incinerator design will dictate which avenue of control to pursue.

- . Drying of the waste prior to incineration
- . Improved combustion increasing the time, temperature and turbulence of the combustion reaction
- . After burner installation

Profile Basis²

The approach to establish a profile was to use the results of an actual field test (Ref. 2) conducted on the unit described under the sections titled, "Process Description" and "Emissions."

The test method consisted of extraction of the flue gas sample by means of a gas collection bottle. A standard pitot plus thermometer was used to establish a stack gas flow rate.

Data Qualification

The reported hydrocarbon gaseous emissions resulting from the incineration of the described sewage waste applies to the installation as described above. The 195 million gallons per day (MMGD) sewage treatment plant tested serves a metropolitan area characterized by single family, multiple dwelling (apartments) and light industry (electronics, pharmaceuticals, etc.). Single family and multiple dwelling unit type sewage dominates the effluent. The sewage effluent entering the treatment plant definitely characterizes the emissions of an incinerator combusting bar screen waste. Consideration must be given to the make-up of material screened and incinerated when using profile 5-01-005.

5.01-7 26200-795

SOLID WASTE, GOVERNMENT, INCINERATOR BAR SCREEN WASTE INCINERATOR DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0122

INE	SAROAD	CHEHICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
ND.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CL	ASSIFICATION
1	43203	ETHYLENE	28. 05	8. 70	5. 61	2	OLEFIN
2	43205	PROPYLENE	42. 08	. 50	. 22	2	OLEFIN
3	43201	METHANE	16.04	80. 40	90. 75	6	METHANE
4	43202	ETHANE	30. 07	2. 70	1. 63	7	NON-REACTIVE
5	45201	BENZENE	78. 11	7. 70	1. 79	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	2 COMP 0 COMP 0 COMP 0 COMP	OUNDS OF CLASSIFICATION 1 OUNDS OF CLASSIFICATION 2 OUNDS OF CLASSIFICATION 3 OUNDS OF CLASSIFICATION 4 OUNDS OF CLASSIFICATION 5 OUNDS OF CLASSIFICATION 6	. 00 28. 58 . 00 . 00 . 00	. 00 9. 20 . 00 . 00 . 00 80. 40	. 00 5. 83 . 00 . 00 . 00 . 75		
		OUNDS OF CLASSIFICATION 7	55. 21	10. 40	3.42		
	5 COMP	OUND COMPOSITE	18. 11	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 5-01-005-99, 5-02-005-99, 5-03-005-99

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- Considine, D. M., (ed), "Chemical and Process Technology Encyclopedia," McGraw-Hill Book Co., 1974.
- 3. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

5.01-9 26200-795

AREA SOURCE EMISSIONS

- 6-01 RESIDENTIAL FUEL
- 6-06 GASOLINE FUEL
- 6-07 DIESEL FUEL
- 6-11 MEASURED VEHICLE MILES
- 6-13 MISCELLANEOUS BURNING
- 6-35 SOLVENT USE
- 6-47 GEOGENIC
- 6-49 SOLID WASTE

6-01

AREA SOURCE EMISSIONS, FUEL USE

6-01-005 RESIDENTIAL--NATURAL GAS COMBUSTION

Process Description 1

Natural gas has become one of the major fuels used for home heating in many parts of the country. The primary component of natural gas is methane, with varying amounts of ethane and smaller amounts of nitrogen, helium, and carbon dioxide.

Natural gas fired residential heating units generally use an atmospheric injection type burner. Common excess air rates range from ten to fifteen percent.

Emissions^{1,2}

Modern residential natural gas fired home heating units have been designed to the point where essentially complete combustion does take place. However, even though natural gas is considered to be a relatively clean fuel, some emissions can occur from the combustion reaction. For example, improper operating conditions, including poor mixing, insufficient air, etc., may cause large amounts of smoke, carbon monoxide, and hydrocarbons to be produced.

AP-42 presents an 8.0 lbs/ 10^6 ft³ (as CH₄) emission factor for natural gas combustion, for both domestic and commercial heating. Also, a population based emission factor of 0.16 tons per 1000 people was developed by KVB based on the 8.0 lbs/ 10^6 ft³ (as CH₄) emission factor and the reported residential natural gas consumption for the California South Coast Air Basin population (Ref. 2).

Profile 6-01-005 presents the volatile organic compounds emitted from a natural gas fired home heating unit (Ref. 2).

Controls

Modern burner equipment has generally been perfected to the point where all common fuels can be burned without causing excessive discharges of the products of incomplete combustion. This is basically accomplished through the proper combination of burner and fuel coupled with a proper operation and maintenance program.

Control of volatile organic compounds, which in natural gas fired equipment is considered to be negligible, would therefore basically amount to improving the combustion efficiency of the unit.

Profile Basis²

Profile 6-01-005 is based on a hydrocarbon grab sample taken from the exhaust vent of a 125,000 Btu/hr, atmospheric injection type burner, residential natural gas space heating unit. A glass gas collecting bottle was used to contain the flue gas sample.

Data Qualification_

Details on the development of the 8 lbs/ 10^6 ft 3 (as CH $_4$) hydrocarbon emission factor is discussed in AP-42 (Ref. 1).

Profile 6-01-005 may be used to characterize the volatile organic compounds emitted from a typical residential natural gas fired space heating unit.

DECEMBER 14, 1978

TABLE 6-01-005

AREA SOURCE EMISSIONS, RESIDENTIAL FUEL

NATURAL GAS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0195

INE NO.	SAROAD CODE		CHEMICAL NAME	MDLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43201	METHANE		16. 04	100.00	100. 00	6 METHANE
			TOTAL		100. 00	100.00	
	O COMP	OUNDS OF	CLASSIFICATION 1	00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 2	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF	CLASSIFICATION 5	. 00	. 00	. 00	
	1 COMP	OUNDS OF	CLASSIFICATION 6	16. 04	100.00	100.00	
	O COMP	OUNDS OF	CLASSIFICATION 7	. 00	. 00	. 00	
	1 COMP	OUND COME	POSITE	16. 04	100. 00	100. 00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 6-01-005-00

REFERENCES

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 2. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

6-06

AREA SOURCE EMISSIONS, GASOLINE ENGINES

6-06-021 LIGHT DUTY VEHICLES

Process Description 1

The internal combustion engine in mobile applications is a major source of air pollutant emissions (carbon monoxide, hydrocarbons, and nitrogen oxides). Passenger cars, light duty trucks, and heavy duty trucks comprise the three main categories of highway vehicles (less motorcycles).

Emissions

Emission control technology has changed both the total mass of hydrocarbons being emitted from automobiles and the detailed patterns of these emissions. Vehicle exhaust emissions are determined by the type of control system used, the type of fuel, ambient temperature and pressure, the vehicle history, and the driving patterns. Section 3 of AP-42 presents a discussion on automotive emission factors.

Emissions from automobiles also occur from evaporation of VOC's after operation (profile $m{6}$ -06-021F), and from evaporation of the gasoline during filling of the gas tank (profile $m{6}$ -06-021G).

Emission control has essentially eliminated emissions from crankcase blow-by in low mileage cars. 4 Proper maintenance of the system eliminates about 98% of the emissions. 14

Controls

The composition of automobile exhaust is significantly influenced by a control system. The amount and type of VOC's in pre-emission control automobiles differ from automobiles with controls. Considerable evidence has been presented that the type of control system will also affect hydrocarbon emissions and their pattern. The proportion of methane, having relatively low potential for the formation of oxidant in atmospheric photochemical reactions, is greater in the exhaust of catalytic converter-equipped automobiles (profile 6-06-021B) than in the exhaust of thermal converter or lean-burn operating cars (profile 6-06-021C) or pre-emission control cars (profile 6-06-021D).

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The catalyst systems also appear to preferentially remove the more reactive hydrocarbon species. In general, vehicles employing oxidation catalysts for exhaust hydrocarbon control have a lower percentage of unsaturated hydrocarbons than do vehicles without exhaust hydrocarbon control.

Profile Basis

Profile 6-06-021A is a composite of vehicle exhaust emission profiles for controlled and uncontrolled vehicles. The composite was based upon the percentage of unleaded gasoline sold in California as representing the usage of catalyst controlled vehicles (profile 6-06-021B) and the percentage of leaded gasoline, regular and premium (profile 6-06-021D) as representing the usage of uncontrolled vehicles.

Profiles 6-06-021B and C were based on the data found in Reference 3. The species pattern was averaged for 1972-1977 vehicles with catalyst control (17 autos) and with lean-burn control (5 autos) burning a 26% aromatic unleaded fuel. The data which listed 56 different hydrocarbon compounds was reorganized into 20 classifications and adjusted for aldehydes.

Profiles **6**-06-021D and E for exhaust hydrocarbons from uncontrolled vehicles were based on data found in Reference 10. The profiles for ten 1970 73 automobiles without controls were separated into the type of fuel tested i.e. a 22% aromatic, leaded fuel corresponding to a premium grade and a 40% aromatic, unleaded fuel. The species patterns were averaged by weight percent into nine major classifications, adjusted to include aldehydes, and expanded into the 21 hydrocarbon species as shown.

Profile 6-06-021F for evaporative emissions is a composite of VOC's emitted after the vehicle has stopped operation, e.g. vapors from the carburetor. Profile 6-06-021F is a composite profile of gasoline vapors lost during the filling of the gas tank.

Data Qualification

Variations in gasoline engine exhaust VOC species occur from vehicle to vehicle depending upon the type of fuel used, the ambient pressure and

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temperature, fuel additives, the engine type, the driving pattern, and most importantly, exhaust controls. Research is continuously refining the VOC emission profiles. The profiles should be used in a manner which best represents the driving patterns found in the user's geographical region.

DATA CONFIDENCE LEVEL: III

DATA CONFIDENCE LEVEL. III

CONTROL DEVICE: COMPOSITE OF CONTROLLED AND UNCONTROLLED VEHICLES

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0329

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43212	N-BUTANE	58. 12	2. 90	2. 19	1 PARAFFIN
2	43220	N-PENTANE	72. 15	6. 90	4. 20	1 PARAFFIN
3	43231	N-HEXANE	86. 17	2. 50	1. 27	1 PARAFFIN
4	43232	N-HEPTANE	100. 20	3.00	1. 31	1 PARAFFIN
5	43233	N-OCTANE	114. 23	12. 30	4. 72	1 PARAFFIN
6	43235	N-NONANE	128. 25	1.00	. 35	1 PARAFFIN
7	43203	ETHYLENE	28. 05	11.40	17. 75	2 OLEFIN
8	43205	PROPYLENE	42.08	7. 30	7. 61	2 OLEFIN
9	43213	BUTENE	56. 10	4. 40	3. 41	2 OLEFIN
0	43224	1-PENTENE	70. 13	2. 80	1.75	2 OLEFIN
1	43245	1-HEXENE	84.16	2. 10	1.09	2 OLEFIN
2	45102	ISOMERS OF XYLENE	106. 16	4. 00	1.66	3 AROMATIC
.3	45105	ISOMERS OF BUTYLBENZENE	134. 21	10	. 04	3 AROMATIC
4	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	2. 10	. 79	3 AROMATIC
5	45202	TOLUENE	92. 13	10. 90	5. 16	3 AROMATIC
6	45203	ETHYLBENZENE	106. 16	. 80	35	3 AROMATIC
7	43502	FORMALDEHYDE	30. 03	3. 90	5. 68	4 CARBONYL
8	43206	ACETYLENE	26. 04	8. 70	14. 60	5 MISCELLANEOUS
9	43201	METHANE	16. 04	8. 40	22, 91	6 METHANE
0	43202	ETHANE	30. 07	70	1.01	7 NON-REACTIVE
1	45201	BENZENE	78. 11	3. 80	2. 14	7 NON-REACTIVE
		TOTAL		100. 00	99. 99	
	6 COMPO	DUNDS OF CLASSIFICATION 1	B9. 39	28. 60	14. 04	
	5 COMPO	OUNDS OF CLASSIFICATION 2	38, 72	28. 00	31. 61	
		OUNDS OF CLASSIFICATION 3	98. 64	17. 90	R. 00	
		OUNDS OF CLASSIFICATION 4	30. 03	3. 90	5. 68	
		DUNDS OF CLASSIFICATION 5	26. 04	8. 70	14. 60	
		OUNDS OF CLASSIFICATION 6	16.04	B. 40	22. 91	
		JUNDS OF CLASSIFICATION 7	62. 71	4. 50	3. 15	
-	21 COMPO	OUND COMPOSITE	43. 84	100. 00	99. 99	

26200-795

6.06-4

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: PROFILES FOR CONTROLLED AND UNCONTROLLED VEHICLES

C. APPLICABLE SCC CATEGORIES: 6-06-021-00, (REFER ALSO TO SUMMARY INDEX)

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: CATALYST PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0326

INE	SARGAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEICHT	WEIGHT	VOLUME	CLA	SSIFICATION
1	43212	N-BUTANE	58. 12	5. 50	3. 91	1	PARAFFIN
2	43220	N-PENTANE	72. 15	14. 00	7. 98	1	PARAFFIN
3	43231	N-HEXANE	86 . 17	1. 60	. 78	1	PARAFFIN
4	43232	N-HEPTAME	100. 20	1. 90	. 78	1	PARAFFIN
5	43233	N-OCTANE	114. 23	10. 30	3. 70	1	PARAFFIN
6	43235	N-NONANE	128. 25	. 60	. 21	1	PARAFFIN
7	43203	ETHYLENE	28 . 0 5	9. 60	14. 07	2	OLEFIN
8	43205	PROPYLENE	42. 08	4. 90	4. 77	2	OLEFIN
9	43213	BUTENE	56. 10	2. 80	2. 06	2	OLEFIN
10	43224	1-PENTENE	70. 13	3. 30	1. 93	2	OLEFIN
£ 1	43245	1-HEXENE	84. 16	3. 50	1. 73	2	OLEFIN
12	45162	ISOMERS OF XYLENE	106. 16	3. 30	1. 28	3	AROMATIC
13	45105	ISOMERS OF BUTYLBENZENE	134. 21	. 10	. 04	3	ARUMATIC
14	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	1. BO	. 62	3	AROMATIC
5	45202	TOLUENE	92 . 13	8. 30	3. 70	3	AROMATIC
4	45203	ETHYLBENZENE	106. 16	. 90	. 37	3	AROMATIC
7	43502	FURMALDEHYDE	30. 03	4. 50	6. 17	4	CARBONYL
8	43206	ACETYLENE	26. 04	4. 70	7. 45	5	MISCELLANEOUS
9	43201	METHANE	16.04	- 13 . 20	33. 97	6	METHANE
:0	43202	ETHANE	30. 07	2. 20	3. 00	7	NON-REACTIVE
1	45201	BENZENE	78. 11	3. 00	1. 56	7	NON-REACTIVE
		TOTAL		100. 00	99. 9 8		
	6 COMP	DUMDS OF CLASSIFICATION 1	80. 53	33. 90	17. 36		
	5 COMPOUNDS OF CLASSIFICATION 2 5 COMPOUNDS OF CLASSIFICATION 3		40. 39	24. 10	24. 56		
			99. 16	14. 40	6 . 01		
	1 COMP	DUNDS OF CLASSIFICATION 4	30. 03	4. 50	6. 17		
	1 COMP	OUNDS OF CLASSIFICATION 5	26. 04	4. 70	7. 45		
	1 COMPOUNDS OF CLASSIFICATION 6		16. 04	13. 20	33. 87		
	5 COMPO	DUNDS OF CLASSIFICATION 7	46. 50	5. 20	4. 56		
-	21 COMPO	UND COMPOSITE	41. 21	100.00	99. 98		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3 AND 9)
C. APPLICABLE SCC CATEGORIES: 6-06-021-00, 6-06-022-00, 6-06-023-00

AREA SOURCE EMISSIONS, GASOLINE FUEL, UNLEADED 26% AROMATIC LIGHT, HEAVY AND OFF HIGHWAY VEHICLES

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: LEAN-BURN PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0327

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEIGHT	VOLUME	CLA	ASSIFICATION
1	43212	N-BUTANE	58. 12	3. 80	3. 07	1	PARAFFIN
2	43220	N-PENTANE	72. 15	9. 00	5. 90	1	PARAFFIN
3	43231	N-HEXANE	86. 17	1.80	. 99	1	PARAFFIN
4	43232	N-HEPTANE	100. 20	2. 20	1.04	1	PARAFFIN
5	43233	N-OCTANE	114. 23	8. 80	3. 63	1	PARAFFIN
6	43235	N-NONANE	128. 25	70	. 28	1	PARAFFIN
7	43203	ETHYLENE	28. 05	15. 50	26. 10	2	OLEFIN
8	43205	PROPYLENE	42.08	7. 80	8. 73	2	OLEFIN
9	43213	BUTENE	56. 10	4. 70	3. 9 6	2	OLEFIN
10	43224	1-PENTENE	70. 13	2. 80	1. 89	2	OLEFIN
11	43245	1-HEXENE	84. 16	3. 30	1.84	2	OLEFIN
12	45102	ISOMERS OF XYLENE	106. 16	4. 70	2. 08	3	AROMATIC
13	45105	ISOMERS OF BUTYLBENZENE	134. 21	10	. 05	3	ARDMATIC
14	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	2. 50	. 99	3	AROMATIC
15	45202	TOLUENE	92. 13	12.60	6. 47	3	AROMATIC
16	45203	ETHYLBENZENE	106. 16	. 90	. 42	3	AROMATIC
17	43502	FORMALDEHYDE	30. 03	4. 50	7.08	4	CARBONYL
18	43206	ACETYLENE	26. 04	6. 80	12, 32	5	MISCELLANEOUS
19	43201	METHANE	16. 04	3. 40	10. 00	6	METHANE
20	43202	ETHANE	30. 07	. 70	1.09	7	NON-REACTIVE
21	45201	BENZENE	78. 11	3. 40	2.08	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	6 COMP	DUNDS OF CLASSIFICATION 1	83. 45	26. 30	14. 91		
	5 COMP	DUNDS OF CLASSIFICATION 2	37. 84	34. 10	42. 52		
	5 COMP	DUNDS OF CLASSIFICATION 3	98. 62	20.80	10. 01		
	1 COMP	DUNDS OF CLASSIFICATION 4	30. 03	4. 50	7.08		
	1 COMPO	DUNDS OF CLASSIFICATION 5	26. 04	6. 80	12. 32		
	1 COMPO	OUNDS OF CLASSIFICATION 6	16. 04	3. 40	10.00		
	2 COMP	JUNDS OF CLASSIFICATION 7	61. 59	4. 10	3. 17		
-	21 COMP	DUND COMPOSITE	47. 29	100.00	100, 01		

26200-795

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3 AND 9)
C. APPLICABLE SCC CATEGORIES: 6-06-021-00, 6-06-022-00, 6-06-023-00

LIGHT, HEAVY AND OFF HIGHWAY VEHICLES

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0328

.INE NO.	SAROAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
110.	CODE	NAIL	WEIGHT	WEIGHT	VOLOTIL	ULI	10H110H
1	43212	N-BUTANE	58. 12	2. 50	1. 90	1	PARAFFIN
2	43220	N-PENTANE	72. 15	5. 80	3. 53	1	PARAFFIN
3	43231	N-HEXANE	86. 17	2. 60	1. 32	1	PARAFFIN
4	43232	N-HEPTANE	100. 20	3. 20	1.41	1	PARAFFIN
5	43233	N-OCTANE	114. 23	12. 60	4. 85	1	PARAFFIN
6	43235	N-NONANE	128. 25	1. 00	. 35	1	PARAFFIN
7	43203	ETHYLENE	28. 05	11. 70	18. 40	2	OLEFIN
8	43205	PROPYLENE	42 . 0 8	7. 70	8. 08	2	OLEFIN
9	43213	BUTENE	56. 10	4. 60	3. 62	2	OLEFIN
10	43224	1-PENTENE	70. 13	2. 70	1. 72	2	OLEFIN
11	43245	1-HEXENE	84. 16	1.80	. 9 3	2	OLEFIN
12	45102	ISOMERS OF XYLENE	106. 16	4. 20	1. 77	3	AROMATIC
13	45105	ISOMERS OF BUTYLBENZENE	134. 21	. 10	. 04	3	AROMATIC
1 辱	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	2. 20	. 7 9	3	AROMATIC
15	45202	TOLUENE	92. 13	11. 30	5. 43	3	AROMATIC
16	45203	ETHYLBENZENE	106. 16	. 80	. 35	3	ARDMATIC
17	43502	FORMALDEHYDE	3 0. 0 3	3. 80	5. 60	4	CARBONYL
18	43206	ACETYLENE	26. 04	9 . 30	15. 75	5	MISCELLANEOUS
19	43201	METHANE	16. 04	7. 70	21. 18	6	METHANE
20	43202	ETHANE	30. 07	. 50	. 75	7	NON-REACTIVE
21	45201	BENZENE	78. 11	3. 7 0	2. 21	7	NON-REACTIVE
		TOTAL		100. 00	99 . 98		
	6 COMP	OUNDS OF CLASSIFICATION 1	91. 25	27. 70	13. 36		
	5 COMP	OUNDS OF CLASSIFICATION 2	38. 42	28. 50	32. 75		
	5 COMP	DUNDS OF CLASSIFICATION 3	98. 53	18. 60	8. 38		
	1 COMP	DUNDS OF CLASSIFICATION 4	30. 03	3. 80	5. 60		
	1 COMP	DUNDS OF CLASSIFICATION 5	26. 04	9. 30	15. 75		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	7. 70	21. 18		
	2 COMP	DUNDS OF CLASSIFICATION 7	65. 94	4. 40	2. 96		
-	21 COMP	OUND COMPOSITE	44. 17	100.00	99. 98		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3, 9 AND 10)
C. APPLICABLE SCC CATEGORIES: 6-06-021-00, 6-06-022-00, 6-06-023-00

AREA SOURCE EMISSIONS, GASOLINE FUEL, UNLEADED 40% AROMATIC LIGHT, HEAVY AND OFF HIGHWAY VEHICLES

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0335

INE	SARDAD	CHEMICAL.	MOLECULAR	PERCENT	PERCENT	СН	EMICAL
NO.	CODE	NAME	MEICHT	WEIGHT	VOLUME	CLASS	IFICATION
1	43212	N-BUTANE	58. 12	2. 20	1. 87	1 P	ARAFFIN
2	43220	N-PENTANE	<i>7</i> 2. 15	5. 30	3. 65	1 P	ARAFFIN
3	43231	N-HEXANE	86.17	1.80	1.03	1 P	ARAFFIN
4	43232	N-HEPTANE	100. 20	2. 30	1. 13	1 P	ARAFFIN
5	43233	N-OCTANE	114. 23	8. 90	3. 84		ARAFFIN
6	43235	N-NONANE	128, 25	. 70	. 30		ARAFFIN
7	43203	ETHYLENE	28. 05	9. 20	16. 16		LEFIN
8	43205	PROPYLENE	42. 08	5. 10	5. 96		LEFIN
9	43213	BUTENE	56. 10	3. 00	2. 66		LEFIN
0	43224	1-PENTENE	70. 13	1. 80	1. 28	_	LEFIN
. 1	43245	1-HEXENE	84. 16	1. 60	. 94		LEFIN
.2	45102	ISOMERS OF XYLENE	106. 16	7. 80	3. 65	3 A	ROMATIC
.3	45105	ISOMERS OF BUTYLBENZENE	134. 21	. 20	. 10	3 4	ROMATIC
.4	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	4. 20	1. 72	3 A	ROMATIC
5	45202	TOLUENE	92. 13	21. 10	11.28	3 A	ROMATIC
6	45203	ETHYLBENZENE	106. 16	1. 50	. 69	3 4	ROMATIC
7	43502	FORMALDEHYDE	30. 03	4. 50	7. 39	4 (ARBONYL
8	43206	ACETYLENE	26 . 04	7. 90	14. 93	5 M	ISCELLANEOUS
9	43201	METHANE	16.04	5. 80	17. 83	6 1	1ETHANE
20	43202	ETHANE	30. 07	. 40	. 64	7 N	ION-REACTIVE
1	45201	BENZENE	78. 11	4. 70	2. 96	7 1	ION-REACTIVE
		TOTAL		100. 00	100. 01		
	6 COMP	OUNDS OF CLASSIFICATION 1	88. 93	21. 20	11. 82		
	5 COMP	DUNDS OF CLASSIFICATION 2	37. 86	20.70	27. 00		
	5 COMP	OUNDS OF CLASSIFICATION 3	98. 63	34. 80	17.44		
	1 COMP	DUNDS OF CLASSIFICATION 4	30. 03	4. 50	7. 39		
	1 COMP	DUNDS OF CLASSIFICATION 5	26. 04	7. 9 0	14. 93		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	5. 80	17. 83		
	2 COMP	DUNDS OF CLASSIFICATION 7	69 . 5 7	5. 10	3. 60		
	21 COMP	DUND COMPOSITE	49, 40	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 3, 9 AND 10)
C. APPLICABLE SCC CATEGORIES: 6-06-021-00, 6-06-022-00, 6-06-023-00

DECEMBER 14, 1978 TABLE €-06-021F

AREA SOURCE EMISSIONS, EVAPORATION LIGHT, HEAVY AND OFF HIGHWAY VEHICLES DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0336

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	MEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION
1	43204	PROPANE	44. 09	1.80	2. 87	1	PARAFFIN
2	43212	N-BUTANE	58 . 12	19. 70	23. 71	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	6. 60	7. 9 7	1	PARAFFIN
4	43220	N-PENTANE	72. 15	39. 40	38. 18	1	PARAFFIN
5	43231	N-HEXANE	86. 17	13. 50	10. 98	1	PARAFFIN
6	43232	N-HEPTANE	100. 20	2. 80	1. 96	1	PARAFFIN
7	43233	N-OCTANE	114. 23	2. 00	1. 26	1	PARAFFIN
8	43235	N-NONANE	128. 25	. 20	. 14	1	PARAFFIN
9	43213	BUTENE	56. 10	1. 80	2. 24	2	OLEFIN
10	43224	1-PENTENE	70. 13	6. 30	6. 29	2	OLEFIN
11	45102	ISOMERS OF XYLENE	106. 16	1. 60	1.05	3	AROMATIC
12	45104	ISOMERS OF ETHYLTOLUENE	120. 19	. 50	. 28	3	AROMATIC
13	45105	ISOMERS OF BUTYLBENZENE	134. 21	. 10	. 07	3	AROMATIC
14	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	. 40	. 21	3	AROMATIC
15	45202	TOLUENE	9 2. 13	2. 10	1. 61	3	AROMATIC
16	43202	ETHANE	30. 07	. 10	. 21	7	NON-REACTIVE
17	45201	BENZENE	78. 11	1. 10	. 98	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	8 COMP	OUNDS OF CLASSIFICATION 1	69. 22	86. 00	87. 07		
	2 COMP	OUNDS OF CLASSIFICATION 2	66, 45	8. 10	8. 53		
	5 COMP	OUNDS OF CLASSIFICATION 3	101.89	4. 70	3. 22		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
	2 COMP	DUNDS OF CLASSIFICATION 7	69. 63	1. 20	1. 19		
-	17 COMP	DUND COMPOSITE	70. 04	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 7)

C. APPLICABLE SCC CATEGORIES: 6-06-021-00, 6-06-022-00, 6-06-023-00

AREA SOURCE EMISSIONS, GASOLINE VAPOR, COMPOSITE

LIGHT, HEAVY AND DFF HIGHWAY VEHICLES

DATA CONFIDENCE LEVEL: II

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0337

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION
1	43204	PROPANE	44. 09	. 10	. 19	1 PARAFFIN
2	43212	N-BUTANE	58. 12	2. 20	3. 5 9	1 PARAFFIN
3	43214	ISOBUTANE	58. 12	. 50	. 85	1 PARAFFIN
4	43220	N-PENTANE	72. 15	12. 20	15. 96	1 PARAFFIN
5	43231	N-HEXANE	86. 17	16. 30	17. 85	1 PARAFFIN
6	43232	N-HEPTANE	100. 20	9. 30	8. 78	1 PARAFFIN
7	43233	N-OCTANE	114, 23	10. 10	8. 31	1 PARAFFIN
8	43235	N-NONANE	128. 25	1.80	1. 32	1 PARAFFIN
7	43238	N-DECANE	142, 28	2. 60	1. 70	1 PARAFFIN
10	43213	BUTENE	56, 10	. 40	. 66	2 OLEFIN
11	43224	1-PENTENE	70, 13	2. 50	3. 40	2 OLEFIN
12	45102	ISOMERS OF XYLENE	106, 16	13. 50	11. 99	3 AROMATIC
13	45104	ISOMERS OF ETHYLTOLUENE	120, 19	6. 50	5. 10	3 ARDMATIC
14	45105	ISOMERS OF BUTYLBENZENE	134, 21	4. 40	3. 12	3 AROMATIC
15	45107	ISOMERS OF TRIMETHYLBENZENE	120, 19	5. 50	4, 34	3 AROMATIC
16	45202	TOLUENE	92, 13	9. 70	9. 92	3 ARDMATIC
17	45201	BENZENE	78, 11	2. 40	2, 93	7 NON-REACTIVE
		TOTAL		100.00	100. 01	
	9 COMPO	DUNDS OF CLASSIFICATION 1	88. 75	55. 10	58, 55	
		OUNDS OF CLASSIFICATION 2	67. 85	2. 90	4. 06	
		OUNDS OF CLASSIFICATION 3	108, 50	39. 60	34, 47	
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
		OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
		OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
		OUNDS OF CLASSIFICATION 7	78. 11	2. 40	2. 93	
-	17 COMPC	OUND COMPOSITE	94. 40	100. 00	100. 01	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 7)

C. APPLICABLE SCC CATEGORIES: 6-06-021-00, 6-06-022-00, 6-06-023-00

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC AP-42, August 1977.
- 2. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I & II, KVB, Inc., Tustin, CA, June 1978.
- 3. Black, F. and High, L., "Automotive Hydrocarbon Emission Patterns and the Measurements of Non-methane Hydrocarbon Emission Rates," Society of Automotive Engineers, Paper No. 770144, February 28-March 4, 1977.
 - 4. Black, F., "The Impact of Emission Control Technology on Passenger Car Hydrocarbon Emission Rates and Patterns," International Conference on Photochemical Oxidant Pollution and its Control Proceeding: Volume II, Environmental Protection Agency, EPA-600/3-77-001b, January 1977.
 - 5. Black, F. M. and Bradow, R. L., "Patterns of Hydrocarbon Emissions from 1975 Production Cars," Environmental Protection Agency.
 - 6. Black, F., Personal Communication to EPA, Mobile Source Group, December 1978, "Unknowns" in profile split 80% paraffin and 20% olefin.
 - 7. Mayrsohn, H. and Crabtree, J., Source Reconciliation of Atmospheric Hydrocarbons, California Air Resources Board, March 1975.
 - 8. Mayrsohn, H. and Crabtree, J., Source Reconciliation of Atmospheric Hdyrocarbons, California Air Resources Board, July 1975.
 - 9. U.S. Bureau of Mines, "Aldehydes and Reactive Organic Emissions from Motor Vehicles, Part I Advanced Automotive Control Systems Vehicles," EPA-IAG-0188(D), March 1973.
- 10. U.S. Bureau of Mines, "Aldehyde and Reactive Organic Emissions from Motor Vehicles, Part II Characterization of Emissions from 1970 through 1973 Model Vehicles, MSPCP-IAG-001, March 1973.
- 11. Powers, T. R., "Effect of Hydrocarbon Composition on Oxidant-Hydrocarbon Relationships, Phase I. Exhaust Blends from Non-Catalytic and Catalyst Equipped Vehicles," EPA 600/3-77-109a. September 1977.
- 12. Ashby, H. A., et al., "Vehicle Emissions Summer to Winter," Society of Automotive Engineers, Paper No. 741053, October 21-25, 1974.

- 13. Koehl, N. J., "Mathematical Models for Prediction of Fuel Tank and Carburetor Evaporation Losses," <u>Society of Automotive Engineers</u>, Paper No. 690506, May 19-23, 1969.
- 14. Hurn, R. W., Cox, F. W. and Allsup, J. R., "Effect of Gasoline Additives on Gaseous Emissions, Part II," Environmental Protection Agency, EPA 600/2-76-026, February 1976.
- 15. Automotive Environmental Systems, Inc., "Accelerated Decay of Non-Fuel Evaporative Emissions," Task No. 1, Environmental Protection Agency, EPA-460/3-76-026, August 1976.
- 16. Crabtree, J., Personal communication to CARB, December 1978, Gasoline sales in 1974, split 13% unleaded and 87% leaded.

6-07 AREA SOURCE EMISSIONS, DIESEL ENGINES

6-07-021 LIGHT DUTY VEHICLES

Process Description

In comparison with the conventional, "uncontrolled," gasoline-powered, spark ignited, automotive engine, the uncontrolled diesel automotive engine is a low pollution power plant. On a grams per mile basis, uncontrolled diesel engines emit less carbon monoxide, hydrocarbons, and somewhat less nitrogen oxides than comparable uncontrolled gasoline engines. A small number of light duty, diesel-powered vehicles are in use in the United States. In contrast, a relatively large number of heavy-duty diesel engines used in trucks and buses are in current use throughout the United States. Diesel engines in any application demonstrate operating principles that are significantly different from those of the gasoline engine.

Emissions

VOC emission species from diesel engines are generally characterized by the diesel fuel species. The $\rm C_1^{-C}_{10}$ hydrocarbons result almost entirely from the combustion process, the cracking of higher molecular weight materials. The $\rm C_{10}^{-C}_{40}$ hydrocarbons result from uncombusted fuel, $\rm C_{10}^{-C}_{25}$, and lubricants, $\rm C_{15}^{-C}_{40}$. The diesel engine produces far more aldehydes than does the gasoline engine. $\rm ^3$

The vapor pressure of diesel fuels under ambient conditions is so low that evaporative emissions can be ignored. 2

The quantity and scope of diesel exhaust test data is rather limited when compared with that of gasoline exhaust studies, although presentation of current test results is scheduled for early 1979 (Ref. 10). An explanation of light and heavy duty diesel exhaust emission factors and the variables affecting them are presented in AP-42 (Ref. 1).

Controls 4

Most of the current diesel exhaust emissions studies are concerned with emission controls through either engine design or the use of fuel additives. Catalytic reactors appear to also be a viable control option.

Profile Basis

The development of a composite light and heavy duty diesel powered vehicle exhaust VOC specie emission profile was based on an engineering evaluation of pertinent literature (Ref. 2-8). Profile 6-07-021 was based on data taken from Ref. 7, the weight % average of three diesel engines and expanded to the 18 hydrocarbon classes utilized. The test data was adjusted for aldehydes based upon Ref. 8 data. Because diesel VOC emissions closely relate to the uncombusted fuel burned, it was assumed that C_4-C_{40} followed the composition of the fuel used, i.e. 66.2% paraffin, 32.5% aromatic and 1.3% olefin. Paraffins $C_{10}-C_{40}$ were represented by specie n-pentadecane, and aromatics C_{10} and above were represented by naphtha.

Data Qualification

Variations in diesel engine exhaust VOC specie emission may occur from vehicle to vehicle depending on such variables as engine size and type, duty cycle, fuel (aromatic content), and age.

New emissions information is being continually developed and the profile user should be informed of any significant developments.

AREA SOURCE EMISSIONS, DIESEL FUEL, 32 % AROMATIC LIGHT, HEAVY AND OFF HIGHWAY VEHICLES DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0330

LINE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT	CHEMICAL	
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CLASSIFICATION	
1	43212	N-BUTANE	58. 12	3. 20	3. 07	1 PARAFFIN	
2	43220	N-PENTANE	72. 15	1. BO	1.40	1 PARAFFIN	
3	43231	N-HEXANE	B6. 17	1. 40	. 87	1 PARAFFIN	
4	43232	N-HEPTANE	100. 20	. 40	. 22	1 PARAFFIN	
5	43233	N-OCTANE	114. 23	. 40	. 2 2	1 PARAFFIN	
6	43235	N-NONANE	128. 25	. 40	. 17	1 PARAFFIN	
7	43260	N-PENTADECANE	212. 41	35. 10	9. 21	1 PARAFFIN	
8	43203	ETHYLENE	28. 05	10. 70	21. 33	2 OLEFIN	
9	43205	PROPYLENE	42. 08	4. 00	5. 30	2 OLEFIN	
10	45101	NAPHTHA	114.00	17. 60	B. 60	3 AROMATIC	
i 1	45102	ISOMERS OF XYLENE	106. 16	. 30	. 17	3 AROMATIC	
12	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	. 20	. 11	3 AROMATIC	
13	45202	TOLUENE	92 . 13	1. BO	1. 12	3 AROMATIC	
l 4	43502	FORMALDEHYDE	30. 03	12. 20	22. 67	4 CARBONYL	
15	43206	ACETYLENE	26. 04	3. 80	B. 15	5 MISCELLANEOUS	
16	43201	METHANE	16. 04	4. 40	15. 30	6 METHANE	
17	43202	ETHANE	30. 07	. 40	. 73	7 NON-REACTIVE	
18	45201	BENZENE	78. 11	1. 90	1. 34	7 NON-REACTIVE	
		TOTAL		100. 00	100. 00		
	7 COMP	OUNDS OF CLASSIFICATION 1	156. 88	42. 70	15. 18		
	2 COMP	OUNDS OF CLASSIFICATION 2	30. 84	14. 70	26. 63		
	4 COMP	OUNDS OF CLASSIFICATION 3	111. 49	19. 90	10.00		
	1 COMP	OUNDS OF CLASSIFICATION 4	30. 03	12. 20	22. 67		
	1 COMP	OUNDS OF CLASSIFICATION 5	26. 04	3. 80	8. 15		
	1 COMP	DUNDS OF CLASSIFICATION 6	16. 04	4. 40	15. 30		
	2 COMP	OUNDS OF CLASSIFICATION 7	61. 17	2. 30	2. 07		
-	18 COMP	DUND COMPOSITE	55. 83	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 2 AND 3)
C. APPLICABLE SCC CATEGORIES: 6-07-021-00, 6-07-022-00

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 2. Black, F. and High, L., "Diesel Hydrocarbon Emission Particulate and Gas Phase," Symposium of Diesel Particulate Emission Measurement and Characterization, May 1978.
- 3. Springer, K. J. and Baines, T. M., "Emissions from Diesel Versions of Production Passenger Cars," Society of Automotive Engineers, Paper No. 770818, September 26-30, 1977.
- 4. Cavagnaro, Diane M., "Diesel Exhaust Emission Control for Motor Vehicles," National Technical Information Service, Springfield, VA.
- 5. Springer, K., "Investigation of Diesel Powered Vehicle Emissions VII," EPA 460/3-76-034, February 1977.
- 6. Taback, H. J. and Sonnichsen, T. W., "Control of Hydrocarbon Emisions from Stationary Sources in the California South Coast Air Basin--Final Report, Vol. I & II, KVB, Inc., Tustin, CA, June 1978.
- 7. Braddock, J. M. and Bradow, R. L., "Emission Patterns of Diesel-Powered Passenger Cars," Society of Automotive Engineers, Paper No. 750682, June 3-5, 1975.
- 8. Braddock, J. N. and Gabele, P. A., "Emission Patterns of Diesel-Powered Passenger Cars--Part II," Society of Automotive Engineers, Paper No. 770168, Feb. 28 March 4, 1977.
- 9. Hare, C. T., "Methodology for Determining Fuel Effects on Diesel Particulate Emissions," Environmental Protection Agency, EPA-650/2-75-056, March 1975.
- 10. Black, F., Personal communication to EPA, Mobile Source Group, December 1978.

6-11 AREA SOURCE EMISSIONS, MEASURED VEHICLE MILES 6-11-061 GASOLINE/DIESEL

Profile Basis

Profile 6-11-061 is a composite of vehicle emissions taken during the morning rush hour beneath a busy Los Angeles tunnel. The exhaust profile was derived from a population of about 2000 vehicles (automobiles, trucks, and buses) accelerating, cruising, and decelerating in the 45-55 mph speed range. The data was adjusted for aldehydes and methane.

This profile was designed to assist individuals interested in compiling approximate highway vehicle VOC specie emissions data based on total vehicle miles traveled. Total hydrocarbon emission rates (g/mi) can be calculated based on information contained in Section 3 of AP-42.

AREA SOURCE EMISSIONS, MEASURED VEHICLE MILES COMPOSITE OF GASOLINE AND DIESEL FUELS, EXHAUST EMISSIONS DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: COMPOSITE OF CONTROLLED AND UNCONTROLLED VEHICLES

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0325

INE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	MEIGHT	WEIGHT	VOLUME	CLA	ASSIFICATION
1	43212	N-BUTANE	58. 12	3. 00	2. 47	1	PARAFFIN
2	43220	N-PENTANE	72. 15	7. 50	4. 93	1	PARAFFIN
3	43231	N-HEXANE	86. 17	6. 90	3. 79	1	PARAFFIN
4	43232	N-HEPTANE	100. 20	4. 00	1. 90	1	PAR A FFIN
5	43233	N-OCTANE	114. 23	4. 70	1. 94	1	PARAFFIN
6	43235	N-NONANE	128. 25	1.40	. 52	1	PARAFFIN
7	43238	N-DECANE	142. 28	2. 40	. 81	1	PARAFFIN
8	43203	ETHYLENE	28. 05	6. 40	10. 81	2	OLEFIN
9	43205	PROPYLENE	42. 08	2. 90	3. 27	2	OLEFIN
10	43213	BUTENE	56. 10	3. 10	2. 61	2	OLEFIN
11	43224	1-PENTENE	70. 13	2. 60	1. 75	2	OLEFIN
12	45102	ISOMERS OF XYLENE	106. 16	10. 60	4. 74	3	AROMATIC
13	45104	ISOMERS OF ETHYLTOLUENE	120. 19	5. 60	2. 23	3	AROMATIC
14	45105	ISOMERS OF BUTYLBENZENE	134. 21	2. 40	. 85	3	AROMATIC
15	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	5. 40	2. 13	3	AROMATIC
16	45202	TOLUENE	9 2. 13	7. 40	3. 79	3	AROMATIC
17	43502	FORMALDEHYDE	30. 03	4. 50	7. 11	4	CARBONYL
18	43206	ACETYLENE	26. 04	5. 20	9. 48	5	MISCELLANEOUS
19	43201	METHANE	16. 04	11.00	32. 53	6	METHANE
20	43202	ETHANE	30. 07	. 50	. 81	7	NON-REACTIVE
21	45201	BENZENE	78. 11	2. 50	1. 52	7	NON-REACTIVE
		TOTAL		100. 00	99. 99		
	7 COMP	DUNDS OF CLASSIFICATION 1	86. 79	29. 90	16. 36		
	4 COMP	DUNDS OF CLASSIFICATION 2	38. 50	15. 00	18. 44		
		DUNDS OF CLASSIFICATION 3	108. 48	31.40	13. 74		
		OUNDS OF CLASSIFICATION 4	30. 03	4. 50	7. 11		
	1 COMPO	OUNDS OF CLASSIFICATION 5	26. 04	5. 20	9. 48		
	1 COMPO	DUNDS OF CLASSIFICATION 6	16. 04	11.00	32. 53		
	2 COMPO	OUNDS OF CLASSIFICATION 7	61. 41	3. 00	2. 33		
-	21 COMP	OUND COMPOSITE	47. 46	100. 00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF GASOLINE AND DIESEL PROFILES

B. REFERENCES: LITERATURE DATA (REF. 1)

C. APPLICABLE SCC CATEGORIES: 6-11-061-00, 6-11-062-00, 6-11-063-00, 6-11-064-00

- 1. Mayrsohn, H. and Crabtree, J., Source Reconciliation of Atmospheric Hydrocarbons, California Air Resources Board, March 1975.
- 2. Mayrsohn, H. and Crabtree, J., Source Reconciliation of Atmospheric Hydrocarbons in the South Coast Air Basin, 1975, California Air Resources Board, December 1976.

6.11-3 26200-795

6-13-081 FOREST FIRES

Process Description 1

A forest "wildfire" is a large-scale natural combustion process that consumes various ages, sizes, and types of botanical specimens growing out-doors in a defined geographical area. Consequently, wildfires are potential sources of large amounts of air pollutants that should be considered when trying to relate emissions to air quality.

The size and intensity (or even the occurrence) of a wildfire is directly dependent on such variables as the local meteorological conditions, the species of trees, and their moisture content, and the weight of consumable fuel per acre (fuel loading). Once a fire begins, the dry combustible material (usually small undergrowth and forest floor litter) is consumed first, and if the energy release is large and of sufficient duration, the drying of green, live material occurs with subsequent burning of this material as well as the larger dry material. Under proper environmental and fuel conditions, this process may initiate a chain reaction that results in a widespread conflagration.

Emissions^{2,3}

Forest fire emissions are a complex mixture of solids, liquids, and gases. Carbon dioxide and water vapor constitute over 90% of the total mass emitted. Emission factors have been reported in the range from 10 to 40 lb/ton fuel burned depending on fuel and fire type and fuel loading. Methane, ethylene, and acetylene are the predominant species in the group, comprising as much as 50% of the total hydrocarbon fraction when determined by flame ionization detection methods. Lesser amounts of ethane, propane, propylene, methyl and ethyl acetylene, butene and butane isomers have been found. For more detailed information on hydrocarbon emission factors to forest wildfires, consult AP42, Section 11.1, and Reference 3.

Controls 3

Of course, the most effective means of controlling wildfire emissions is to prevent the occurrence of a forest fire through various means available to the forest ranger. A frequently used technique for reducing

wildfire occurrence is "prescribed" or hazard reduction" burning. This type of managed burn involves combustion of litter and underbrush in order to prevent fuel buildup on the forest floor and thus reduce the danger of a wildfire. Although some air pollution is generated by this preventative burning, the net amount is believed to be a relatively smaller quantity than that produced under a wildfire situation. For more detailed information on the methods and benefits of prescribed burning, consult Reference 3.

Profile Basis 4

The source of information for the profile originated from a study (Ref. 4) concerned with the emissions generated from slash burning. Nine untreated fuel beds were constructed from ponderosa logging slash collected from the San Bernardino National Forest and burned under lab conditions at the U.S. Forest Service Fire Laboratory at Riverside, California. Although small in size, these fuel beds were the equivalent of a 50 ton/acre fuel loading which is similar in size and distribution to actual logging slash.

Combustion gases were filtered and collected in 250 cc glass cylinders for analysis by chromatography.

The unidentified fraction (25% vol.) is believed to be composed of organic acids, aldehydes, ketones, phenols, and heterocyclic compounds. For more detailed information, consult Reference 4.

Data Qualification

Although these tests were run under laboratory conditions, the fuel beds constructed are similar in fuel loading and geometry to actual logging slash areas. Realizing all of the variations that can and do occur in fuel type and fuel loading from one area to another, this profile is construed to be the basis from which hydrocarbon emissions from areas where ponderosa pine predominate. Further research into the effects burning other species of trees and foliage would have on the present profile would need to be made if used where this condition does exist.

TABLE *€ -*13-081

AREA SOURCE EMSISSIONS, MISCELLANEOUS BURNING

FORREST FIRES

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0307

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	MEICHT	VOLUME	CL	ASSIFICATI ON
1	43122	ISOMERS OF PENTANE	72. 15	. 15	. 08	1	PARAFFIN
2	43204	PROPANE	44. 09	. 35	. 31	1	PARAFFIN
3	43212	N-BUTANE	58. 12	. 24	. 15	1	PARAFFIN
4	43214	ISOBUTANE	58. 12	. 11	. 08	1	PARAFFIN
5	43120	ISOMERS OF BUTENE	56. 10	. 92	. 62	2	OLEFIN
6	43203	ETHYLENE	28. 05	19. 11	26. 20	2	OLEFIN
7	43205	PROPYLENE	42. 08	3. 93	3. 58	2	OLEFIN
8	43213	BUTENE	5 6. 10	. 81	. 54	2	OLEFIN
9	43218	1,3-BUTADIENE	54. 09	. 52	. 38	2	OLEFIN
10	43223	3-METHYL-1-BUTENE	70. 14	. 17	. 08	2	OLEFIN
11	43000	UNIDENTIFIED HYDROCARBONS	86. 00	44. 59	19. 9 7	5	MISCELLANEOUS
12	43206	ACETYLENE	26. 04	B. 40	12. 43	5	MISCELLANEOUS
13	43209	METHYLACETYLENE	40. 06	. 41	. 38	5	MISCELLANEOUS
14	43201	METHANE	16. 04	9. 82	23. 55	6	METHANE
15	43202	ETHANE	30. 07	10. 47	13. 39	7	NON-REACTIVE
		TOTAL		100. 00	101. 74		
	4 COMP	DUNDS OF CLASSIFICATION 1	52, 92	. 85	. 62		
		OUNDS OF CLASSIFICATION 2	31. 11	25. 46	31.40		
	_	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	62. 73	53. 40	32. 78		
		OUNDS OF CLASSIFICATION 6	16. 04	9. 82	23. 55		
		OUNDS OF CLASSIFICATION 7	30. 07	10. 47	13. 39		
•	15 COMP	DUND COMPOSITE	37. 81	100.00	101. 74		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE DATA (REF. 4)

C. APPLICABLE SCC CATEGORIES: 6-13-081-00

- 1. "Compilation of Pollutant Emission Factors," Environmental Protection Agency, Research Triangle Park, NC, AP-42, August 1977.
- 2. McMahon, C. K. and Ryan P. W., "Some Chemical and Physical Characteristics of Emissions from Forest Fires," paper presented at 69th APCA Meeting, Portland, OR, June 27 July 1, 1976.
- 3. Southern Forestry Smoke Management Guidebook, by Southern Forest Fire Laboratory personnel, Southeastern Forest Experiment Station, Asheville, NC, and Southern Forest Fire Laboratory, Macon, Georgia, December 1977.
- 4. Sandberg, D. V. et al., "Emissions from Slash Burning and the Influence of Flame Retardant Chemicals," J. Air Poll Cont Assoc, Vol. 25, No. 3, March 1975.

6-35 AREA SOURCE EMISSION, SOLVENT USE

6-35-103 ARCHITECTURAL SURFACE COATINGS

Process Description 1

Architectural coatings are paints and other coatings that are applied to stationary surfaces, structures, and their appurtenances. Architectural coatings include air-dried coatings that are applied by spray, brush, or roller to surfaces and structures such as buildings, pavements, or curbs, but do not include industrial coatings, which are generally applied by a wider variety of methods, commonly oven cured, and used on items made in factories. The major users of architectural coatings are homeowners and painting contractors.

Emissions 1,2

When architectural coatings are applied with solvents, the solvents must evaporate into the atmosphere so that the coating can form a film or barrier. The evaporation of these solvents along with the associated use of solvents for thinning and cleaning up generate organic solvent emissions that represent a substantial portion of all organic vapors present in a community's atmosphere and therefore are a significant area source of VOC emissions.

An estimate of these emissions from work performed in Southern California for 1976 is 3.5 tons/1000 people/year (Refs. 1 and 2). In Southern California the emission of solvent vapors has been controlled since 1967 (see SCAQMD Rule 442, formerly Rule 66). Use of the above emission factor for other areas may result in a low estimate of emissions. For further information consult Reference 3 and 4.

Controls²

In order to control emissions from architectural coatings the only practical means is to change coating formulations to reduce the amount of solvent in the formulation or to use solvents including water.

In California it was estimated that if waterborne architectural coatings were to be used exclusively in place of high-solvent architectural coatings, the emissions could be reduced by up to 70% (Ref. 2).

Profile Basis

The basis for the emission profile presented in Table 6-35-103 was an architectural coating study conducted by the San Diego County, California, APCD which is summarized in Reference 5. The APCD determined by questionnaire the total amount of individual solvents used in architectural coatings in San Diego County for the year 1974-75. The VOC's listed in Table 6-35-103 and their percent composition was taken directly from the results of that investigation.

Data Qualification

As discussed above this profile was based on a study of the Southern California area. Some care should be exercised in using this profile in areas of the country where solvent emission controls are not practiced.

DECEMBER 14, 1978

TABLE 6-35-103

AREA SOURCE EMISSIONS, SOLVENT USE ARCHITECTURAL SURFACE COATINGS, COMPOSITE DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0196

LINE SAROAD CHEMICAL MOLECULAR PERCENT PERCENT CHEMICAL NO. CODE NAME WEIGHT WEIGHT VOLUME CLASSIFICATION 43231 20. 70 1 N-HEXANE 86. 17 18. 36 PARAFFIN 2 43248 CYCLOHEXANE 84, 16 20.70 18. 82 PARAFFIN 1 3 45102 ISOMERS OF XYLENE 106. 16 2. 60 1. 91 3 AROMATIC 45202 TOLUENE 92. 13 5. 20 4. 28 3 AROMATIC AROMATIC 45203 ETHYLBENZENE 106, 16 4. 30 3. 14 43551 ACETONE 58. 08 3. 20 4. 21 CARBONYL 7 43552 72. 10 CARBONYL METHYL ETHYL KETONE 5. 60 5, 97 43559 METHYL N-BUTYL KETONE 100.16 . 70 . 54 CARBONYL 9 43560 METHYL ISOBUTYL KETONE 100, 16 . 60 . 46 CARBONYL 10 43301 METHYL ALCOHOL 32. 04 3. 90 9. 33 MISCELLANEOUS 43302 ETHYL ALCOHOL 46, 07 . 99 MISCELLANEOUS 11 . 60 12 43304 ISOPROPYL ALCOHOL 60.09 16, 40 20.89 MISCELLANEOUS 13 43305 N-BUTYL ALCOHOL 74, 12 1.68 5 MISCELLANEOUS 1. 60 14 43306 ISOBUTYL ALCOHOL 74. 12 . 60 . 61 MISCELLANEOUS 15 43369 PROPYLENE GLYCOL 76, 00 . 80 . 84 MISCELLANEOUS 43370 62, 07 MISCELLANEOUS 16 ETHYLENE GLYCOL . 60 . 77 17 43435 N-BUTYL ACETATE 116. 16 2. 50 1. 68 5 MISCELLANEOUS . 99 18 43446 ISOBUTYL ACETATE 116.16 1. 50 MISCELLANEOUS 19 43450 DIMETHYLFORMAMIDE 73. 09 . 50 . 54 MISCELLANEOUS 20 43451 ISOBUTYL ISOBUTYRATE 144. 21 6.10 3. 21 MISCELLANEOUS 21 43452 2-ETHOXYETHYL ACETATE 132.00 1.30 . 77 MISCELLANEOUS TOTAL 100.00 99, 99 2 COMPOUNDS OF CLASSIFICATION 1 85, 15 41.40 37. 18 O COMPOUNDS OF CLASSIFICATION 2 . 00 . 00 . 00 3 COMPOUNDS OF CLASSIFICATION 3 99. 72 12.10 9.33 4 COMPOUNDS OF CLASSIFICATION 4 69, 33 10.10 11.18 12 COMPOUNDS OF CLASSIFICATION 5 66, 08 36, 40 42, 30 O COMPOUNDS OF CLASSIFICATION 6 . 00 . 00 . 00 O COMPOUNDS OF CLASSIFICATION 7 . 00 . 00 . 00 21 COMPOUND COMPOSITE 99. 99 76, 68 100.00

26200-795

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA CALCULATIONS FROM COMPOSITE SURVEY DATA

B. REFERENCES: SAN DIEGO COUNTY APCD SOURCE DATA QUESTIONNAIRES (REF. 5)

C. APPLICABLE SCC CATEGORIES: 6-35-103-00

- 1. "Status Report on Organic Solvent Regulations," California Air Resources Board, Staff Report 76-25-4, Nov. 24, 1976.
- 2. "Consideration of Model Organic Solvent Rule Applicable to Architectural Coatings," California Air Resources Board, Staff Report 77-14-4, June 29, 1977.
- 3. Southerland, J. H., et al., "Emission inventory/Factor Workshop," Volume II, EPA Office of Air Quality Planning and Standards, September 13-15, 1977.
- 4. Danielson, J.A. (ed), "Air Pollution Engineering Manual," U.S. Environmental Protection Agency, Office of Air Quality Planning and Standards, Research Triangle Park, NC, AP-40, May 1973.
- 5. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., June 1978.

6.35-4 26200-795

6-35-702 DOMESTIC, GENERAL

Process Description

Domestic solvents are here defined as any solvent found in products used around the house, garage, or yard. The following table lists many of the common domestic products currently in use, percent by weight of solvent, and the estimated national sales.

NATIONAL DOMESTIC AND COMMERCIAL SOLVENT SALES Based on Department of Commerce Data

Product	Estimated National Sales (10 ⁶ lbs)	Solvents (wt. %)	Total Weight of Solvents (10 ⁶ lbs)
Furniture Polish	53	40	21
Floor Polish	87	40	35
Shoe Polish	3	40	1
Metal Polish	8	40	3
Shaving Soap	56	5	3
After Shave	49	20	10
Perfumes, Toiletries & Cosmetics	17	39	7
Shampoo	152	10	15
Hair Tonics	5	5	0
Hair Spray	210	59	124
Hair Rinses	23	5	1
Mouthwash	119	14	17
Creams	74	30	22
Suntan Oil	9	50	5
Hand Lotion	51	20	10
Cleaning Lotions	23	60	14
Rubbing Alcohol	153	100	153
Deodorant	148	14	21
Nail Polish	1	50	1
Nail Polish Remover	8	90	
TC	TAL		469

Emissions

The organic emissions from domestic chemical use results from the vaporization of the low boiling point solvents contained within the product. The quantity and species of these emissions depends greatly, of course, on the product used - concentration and composition of solvent in product. A recent hydrocarbon emissions study (Ref. 1) estimated the organic emissions from domestic chemical use to be 1.1 tons per 1000 people per year.

profile 6-35-702 presents a composite estimate of the organic species eminating from common domestic solvent use.

Profile Basis

The basis for the composite profile of domestic solvent use was an extensive survey and engineering evaluation of literature data obtained from major domestic product manufacturers. The population based emission factor was mainly based on sales data obtained from the Department of Commerce.

Data Qualification

The above mentioned profile and emission factor are intended to be used to characterize the organic emissions resulting from domestic solvent use. They are at best approximations and composites of many types and brands of household chemical products in current use and should be used as such. A population based emission factor was estimated to best reflect the distribution of these emissions.

AREA SOURCE EMISSIONS, SOLVENT USE

DOMESTIC SOLVENTS, GENERAL DATA CONFIDENCE LEVEL: IV

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0197

INE	SARDAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43214	ISOBUTANE	58. 12	5. 30	5. 04	1	PARAFFIN
2	45101	NAPHTHA	114. 0 0	4. 50	2. 22	3	AROMATIC
3	43502	FORMALDEHYDE	30. 03	. 60	1.11	4	CARBONYL
4	43551	ACETONE	58. 08	1. 40	1. 33	4	CARBONYL
5	43302	ETHYL ALCOHOL	46. 07	36. 90	44. 40	5	MISCELLANEOUS
6	43304	ISOPROPYL ALCOHOL	60. 09	38. 50	35. 53	5	MISCELLANEOUS
7	43367	GLYCOL ETHER	62. 07	B. 30	7. 43	5	MISCELLANEOUS
8	43369	PROPYLENE CLYCOL	76. 00	3. 20	2. 33	5	MISCELLANEOUS
9	43435	N-BUTYL ACETATE	116. 16	1. 30	. 61	5	MISCELLANEOUS
		TOTAL		100. 00	100. 00		
	1 COMP	OUNDS OF CLASSIFICATION 1	58. 12	5. 30	5. 04		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 3	114.00	4. 50	2. 22		
	2 COMP	OUNDS OF CLASSIFICATION 4	45. 32	2. 00	2. 44		
	5 COMP	OUNDS OF CLASSIFICATION 5	54, 15	88. 20	90. 30		
	O COMP	OUNDS OF CLASSIFICATION 6	00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 7	. 00	. 00	. 00		
	9 COMP	OUND COMPOSITE	55. 46	100. 00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 1)

C. APPLICABLE SCC CATEGORIES: 6-35-702-00

Reference

1. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

6-35 AREA SOURCE EMISSIONS, SOLVENT USE

6-35-705 PESTICIDES, GENERAL

Process Description 1

Pesticides consist of any chemical that is formulated to kill pests such as insects and rodents. The chemicals used are generally arsenic, maleic anhydride, or pyridine derivatives in suspension or solution in an organic solvent (Ref. 1).

Pesticides are commonly available in the form of liquids, aerosols or powders and are applied by spraying and/or dusting.

Emissions 2-5

Organic air contaminants are emitted from the vaporization of the solvents used or from the vaporization of the pesticide. An emission factor of nine tons per 100,000 people per year (Ref. 2) for domestic and commercial use was developed from information obtained from State Pesticide Use Reports (Ref. 3-5).

The estimated organic specie emissions from pesticide use is presented in profile 6-35-705.

Controls

The fugitive organic emissions resulting from the use of common pesticides can possibly be reduced by:

- 1. use of higher boiling point solvents, and
- 2. dusting instead of over spraying and aerosols when possible.

These recommendations may not be feasible or even possible in many situations due to the effective chemical formulation and corresponding application methods necessary to accomplish the desired effect of killing pests.

Profile Basis^{2,3}

Profile 6-35-705 is based on an engineering evaluation of the information contained within the State of California Pesticide Use Reports (Ref. 2 and 3).

Data Qualification

The above mentioned profile and emission factor are intended to be used to characterize the organic emissions resulting from the domestic and commercial use of common pesticides. They are at best approximations and composites of many types and brands of pesticides in current use and should be used with this in mind. A population based emission factor was estimated to best reflect typical domestic and commercial pesticide use.

DECEMBER 14, 1978 TABLE 6-35-705

AREA SOURCE EMISSIONS, SOLVENT USE PESTICIDES, DOMESTIC AND COMMERCIAL, COMPOSITE FOR CALIFORNIA DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0076

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	8. 10	8. 71	1	PARAFFIN
2	43115	C-7 CYCLOPARAFFINS	98 . 19	15. 40	14, 55	1	PARAFFIN
3	43116	C-8 CYCLOPARAFFINS	112. 23	1. 60	1. 30	1	PARAFFIN
4	43118	MINERAL SPIRITS	114.00	15.00	12. 23	1	PARAFFIN
5	43122	ISOMERS OF PENTANE	72. 15	3. 10	3. 99	1	PARAFFIN
6	43204	PROPANE	44. 09	1. 80	3. 80	1	PARAFFIN
7	43212	N-BUTANE	58. 12	4. 40	7. 04	1	PARAFFIN
8	43214	ISOBUTANE	58. 12	1. 40	2. 22	1	PARAFFIN
9	43220	N-PENTANE	72. 15	3. 20	4. 08	1	PARAFFIN
10	43231	N-HEXANE	86. 17	3. 70	3. 99	1	PARAFFIN
11	45102	ISOMERS OF XYLENE	106. 16	15.00	13. 07	3	ARCHATIC
12	45202	TOLUENE	92. 13	5. 00	5. 00	3	AROMATIC
13	43819	METHYLENE BROMIDE	173. 85	10. 00	5. 38	5	MISCELLANEOUS
14	45201	BENZENE	7B. 11	12. 30	14. 64	7	NON-REACTIVE
		TOTAL		100. 00	100. 00		
	10 COMP	OUNDS OF CLASSIFICATION 1	86. 4 3	5 7. 70	61. 91		
	O COMP	OUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	2 COMP	DUNDS OF CLASSIFICATION 3	102. 28	20.00	18. 07		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 5	173. 85	10.00	5. 38		
	O COMP	OUNDS OF CLASSIFICATION 6	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 7	78. 11	12. 30	14. 64		
	14 COMP	OUND COMPOSITE	92. 78	100.00	100. 00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: T. W. SONNICHSEN, KVB ENGINEER (REF. 2 AND 3)

C. APPLICABLE SCC CATEGORIES: 4-35-705-98

- 1. Considine, D. M., (ed.), "Chemical and Process Technology Encyclopedia," McGraw-Hill Book Co., 1974.
- 2. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 3. "Pesticide Use Report," Annual 1976, Department of Food and Agriculture, Agricultural Chemicals and Food.
- 4. Personal communication with Dr. Ming-yu Li, University of California Davis, Department of Food Protection and Toxicology Center.
- 5. Wiens, F. J., "Reactive Organic Gas Emissions from Pesticide Use in California," California Air Resources Board, December 1977.

6.35-12

26200-795

6-47 AREA SOURCE EMISSIONS, GEOGENIC

6-47-409 FORESTS

Process Description 1

The release of volatile organic substances through the aerial organs of plants is a well-known characteristic of the plant world. The prime source of the terpenes emitted from trees is believed to be the foliage. These emissions occur in conjunction with a plant's normal photosynthetic and respiratory exchange of atmospheric gases.

Emissions^{1,2}

Plant species release appreciable amounts of volatile organic substances to the atmosphere. The major hydrocarbon terpene compounds emitted as shown in profile $\bf 6$ -47-409 are mono-terpenes (C_{10}) like α -pinene, β -pinene, limonenes, and the hemiterpene (C_{5}) isoprene (Refs. 1 & 2).

Hydrocarbon emissions rate data provided by Zimmerman (Ref. 2) is presented in the following table. Local land management agencies and/or Government Forest Services should be contacted to obtain information on the type of vegetation and leaf biomass found for the area in question. An annual HC emission rate per acre per plant species can then be calculated using the following equation (Ref. 2):

		Leaf		Active-Dormant
Emission Rate		Biomass		Period
μg/g.hr	Х	10 ⁵ kg/km ²	х	hrs/yr

 \times 8.91 \times 10⁻⁶ = lb/acre.plant species.yr

TABLE OF FOREST NATURAL EMISSION RATES BASED ON LEAF BIOMASS

Southern	\mathtt{Active}^2	Dormant ²	2	Annual
California	Emission	Emission	\mathtt{Leaf}^2	Emission
Forest Type	Rate	Rate	Biomass	Factor
(% Composition)	μg/g·hr	μg/g•hr	10 ⁵ kg/km ²	lbs/acre.yr
<u>Hardwoods</u>	6 months	6 months	-	46
(60%) Oak	4	0	3	-
(40%) Maple	1	0	3	-
Douglas Fir	1	0	11	86
Mixed Conifer	-	-	-	150
(60%) Ponderosa Pine	3	1.5	11	
(40%) Douglas Fir	1	0	11	-
Pines	3	1.5	11	193
Pinjon Juniper	3	3	3	70
Brush	-	-	-	173
(85%) Sagebrush 1	12	4	3	_
(10%) Scrub Oak	4	0	3	_
(5%) Juniper	3	3	3	-

⁽Ref. 2

¹ Sagebrush equivalent to mesquite and chaparral in emissions.

²Consult local Forest Service for specific area information. Southern California, for example, was estimated to have a 6 month active and 6 month dormant period for some species. Leaf biomass for Southern California was also estimated.

Controls

Controls are not applicable to forest emissions.

Profile Basis 3

The primary source of data on natural emission factors was P. R. Zimmerman, Washington State University (Ref. 2). Zimmerman, in support of a national emission assessment, generated emission rates and biomass data based on tests performed in the northwest and east coast areas of the U.S. The values presented in the above table have been adjusted with the aid of Mr. Zimmerman and the National Forest Service to take into account the conditions found within Southern California.

Data Qualification

The HC emission factors presented in the above table are representative of forest conditions located within Southern California. Application of these values to areas other than Southern California may result in significant error. References 1 and 2 and local forest services should be consulted for emission factor data for other areas.

Profile **6**-47-409 may be used to characterize the volatile organic emissions from a forested area.

6.47 - 3

TABLE 6-47-409

AREA SOURCE EMISSIONS, GEOGENIC

FORESTS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0204

LINE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION
1	43123	TERPENES	136. 23	100.00	100. 00	2 OLEFIN
		TOTAL		100. 00	100. 00	
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00	
	1 COMP	OUNDS OF CLASSIFICATION 2	136. 23	100.00	100.00	
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00	
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 6	. 00	. 00	. 00	
	O COMP	DUNDS OF CLASSIFICATION 7	. 00	. 00	. 00	
•	1 COMP	DUND COMPOSITE	136. 23	100.00	100.00	

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST, DATA (REF. 2)

C. APPLICABLE SCC CATEGORIES: 6-47-409-99

- 1. Rasmussen, R. A., "What do the Hydrocarbons from Trees Contribute to Air Pollution," Journal of APCA, Vol. 22, No. 7, July 1972.
- 2. Zimmerman, P. R., "Determination of the Emission Rates of Hydrocarbon from Indigenous Species of Vegetation in the Tampa/St. Petersburgh Area," Interim Report for EPA Contract 68-01-4432.
- 3. Taback, H. J. et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I & II, KVB, Inc., Tustin, CA, June 1978.

6.47-5 26200-795

6-47 AREA SOURCE EMISSION, GEOCENIC

6-47-411 NATURAL SEEPS

Process Description 1

Natural seeps mainly consist of gaseous hydrocarbons and evolution of heavy oil and tar. Diffusion appears to account for the evolution of these gases and heavy oils.

A crude oil reservoir contains a mixture of water, oil and gas in the small pore spaces (holes) in the reservoir rock. Initially, the reservoir holds these fluids under considerable pressure, caused by the hydrostatic pressure of the ground water. At this pressure a large part of the gas is dissolved in the oil. These two fluids, the initial (connate) water and the gas in solution, combine to provide the driving force for moving the oil on a path of lease resistance through the ground.

Emissions^{1,2}

Studies (Ref. 2) conducted to quantify the emissions from two offshore seeps in Santa Barbara County rated the largest seep to have an organic compound emissions of $^{\circ}6$ tons per day. An emission factor per se is not feasible due to the very nature of such a process. The help of local petroleum agencies should be solicited if possible in an attempt to locate and quantify natural seeps.

In an attempt to speciate these natural seepage emissions, a nearby oil production field's raw petroleum gas was sampled and analyzed (Ref. 1). Profile 6-47-411 presents the results of this effort.

Controls

The presence of these seeps generally creates a local pollution problem due to "petroleum odors" and evolution of heavy oil and tar. A practical form of control is not available.

Profile Basis

Samples of raw petroleum gas believed to be representative were obtained from an oil field near Santa Barbara.

Samples were collected using glass gas collecting bottles and NIOSH type charcoal tubes.

Data Qualification

Profile 6-47-411 should be used with discretion to represent the gaseous hydrocarbon emissions from natural seeps until more applicable or updated information is made available.

AREA SOURCE EMISSIONS, GEOGENIC

PETROLEUM SEEPS

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0205

INE NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43122	ISOMERS OF PENTANE	72. 15	1. 60	. 69	1	PARAFFIN
2	43204	PROPANE	44. 09	29. 10	20. 70	1	PARAFFIN
3	43212	N-BUTANE	58. 12	14.00	7. 56	1	PARAFFIN
4	43214	ISOBUTANE	58. 12	6. 40	3. 45	1	PARAFFIN
5	43220	N-PENTANE	72. 15	1.20	. 53	1	PARAFFIN
6	43201	METHANE	16. 04	19.00	37. 16	6	METHANE
7	43202	ETHANE	30. 07	28. 70	29. 92	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	5 COMP	OUNDS OF CLASSIFICATION 1	49. 82	52. 30	32. 93		
	O COMP	DUNDS OF CLASSIFICATION 2	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 4	00	00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 5	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	19.00	37. 16		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	28. 70	29. 92		
	7 COMP	DUND COMPOSITE	31. 36	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA GC-MS ANALYSIS OF GRAB SAMPLE

B. REFERENCES: KVB TEST DATA (REF. 1)

C. APPLICABLE SCC CATEGORIES: 6-47-411-01

REFERENCES

- 1. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 2. Harrison, P. R. and Maas, S. J., "Monitoring of Natural Seeps in the Santa Barbara Channel Off Coal Oil Point," Meteorology Research, Inc., Report 76-R-1408, March 1976.

6-47 AREA SOURCE EMISSIONS, GEOGENIC

6-47-429 CITRUS GROVES

Process Description 1,2

Studies have shown that significant quantities of terpenes are emitted as part of the natural biological cycle of citrus trees (Ref. 1). The release of volatile substances through the aerial organs of plants is a well-known characteristic of the plant world (Ref. 2).

Emissions^{2,3}

Emissions occur in conjunction with a plant's normal photosynthetic and respiratory exchange of atmospheric gases. Freshly exuded bud resins and oleoresin blisters contribute significant quantities of monoterpenes to the surrounding air as to leaf, bark and wood tissues undergoing cellularlysis and decay. The major HC terpene compounds emitted as shown in profile $\mathbf{6}$ -47-429 are monoterpenes (C₁₀) like α -pinene, β -pinene, limonenes, and the hemiterpene (C₅) isoprene (Ref. 2,3).

An emission factor of 0.06 tons per year per acre of citrus trees was proposed by Zimmerman (Ref. 3). An annual emission rate can be calculated by multiplying the HC emission factor by the acreage of citrus trees contained within the study area.

Controls

VOC control equipment is not applicable to a citrus grove.

Profile Basis

The primary source of data on citrus grove emissions was Mr. Zimmerman, Washington State University. Zimmerman, in support of a national emission assessment, generated emission rates based on tests conducted in the Tampa/St. Petersburg Area (Ref. 3).

Data Qualifications

The 0.06 tons per year per acre of citrus trees and the corresponding organic emissions profile may be used in estimating emissions from citrus crops in general. The EPA should be contacted for any new information generated since the date of this publication.

DECEMBER 14, 1978

TABLE €-47-429

AREA SOURCE EMISSIONS, GEOGENIC

CITRUS GROVES

DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0199

NO.	SAROAD CODE	CHEMICAL NAME		MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CHEMICAL CLASSIFICATION	
1	43123	TERPENES		136. 23	100.00	100.00	2 OLEFIN	
		TOTAL			100.00	100.00		
	O COMP	OUNDS OF CLASSIFICAT	ION 1	00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICAT	ION 2	136, 23	100.00	100.00		
	O COMP	DUNDS OF CLASSIFICAT	ION 3	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICAT	ION 4	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICAT	ION 5	. 00	. 00	. 00		
	O COMP	DUNDS OF CLASSIFICAT	ION 6	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICAT	ION 7	. 00	. 00	. 00		
	1 COMP	DUND COMPOSITE		136. 23	100.00	100.00		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 3)

C. APPLICABLE SCC CATEGORIES: 6-47-429-99

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REFERENCES

- 1. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.
- 2. Rasmussen, R. A., "What do the Hydrocarbons from Trees Contribute to Air Pollution, " <u>Journal of APCA</u>, Vol. 22, No. 7, July 1972.
- 3. Zimmerman, P. R., "Determination of the emission Rates of Hydrocarbons from Indigeneous Species of Vegetation in the Tampa/St. Petersburg Area," Interim Report for EPA Contract 68-01-4432.

6.47-13 26200-795

6-49-999 SANITARY LANDFILLS

Process Description 1,2

One long-accepted method for residential and industrial waste disposal has been the use of sanitary landfills. The waste disposed of at these sites is comprised basically of refuse, domestic garbage, and inert construction material. Waste is generally laid in layers, compacted and covered by a thin layer of silt. Several studies (Refs. 1, 2) have shown that as a result of this procedure, appreciable amounts of methane-rich gas are generated due to the biological anaerobic decomposition of these wastes.

Emissions³⁻⁵

The production rate of landfill gas appears to be highly dependent on the type of refuse (organic content), moisture content, soil composition and permeability, and age of the landfill site. Gas production within a sanitary landfill is caused by the aerobic and anaerobic decomposition of organic material in the presence of moisture. The aerobic state predominates in new landfills until the consumption of oxygen exceeds the supply, at which time a transition to an anaerobic state takes place.

Methane gas (CH_4) generally first appears after the transition from an aerobic to anaerobic state has begun. An increase in CH_4 production with time can therefore be expected. Methane and carbon dioxide constitute approximately 99% of the gas emanating from a landfill. Reports on the quantitative nature of these gases vary widely due to the inherent differences existing at each site.

An empirical approach for estimating the rate of carbon escape is presented in a study by the California State Water Quality Control Board (Ref. 4). Data from this study shows that

$$r = \frac{177}{3.75 + 1.95t}$$

where:

r = rate of carbon escape, lb/ton refuse year

t = age of refuse, years

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Note that the carbon is released as both methane (CH_4) and carbon dioxide (CO_2) gas. To use this relationship, a gross estimate of the total quantity and age of wastes presently "alive" in an area is needed.

For this example, data obtained by KVB (Ref. 5) on Sanitary Landfill Sites in the California South Coast Air Basin will be used. Over 15 million tons of liquid and solid wastes are disposed of annually in the 45 major landfill sites distributed within the Basin. A summary of the resulting computation using the above formula and the assumed age and quantity of refuse "alive" in the Basin is given in Table 6-49. It was also assumed, for the purpose of this example, that the quantity of materials disposed of in landfills was constant over the last 75 years and was therefore proportional to the total population within the Basin. As shown in Table 6-49, 90% of the carbon emissions result from deposits made in the last 25 years. Assuming a 15% conversion by weight of the total carbon emitted is transformed to methane (Ref. 4), this would represent approximately 340,000 tons per year or about 930 tons per day of methane as shown in the following calculations.

The calculations performed to obtain the total tons per year of carbon released available as methane (CH_4) are as follows

$$(3401.23 \times 10^6 \frac{\text{1b of carbon}}{\text{one year}}) (\frac{1 \text{ ton}}{2000 \text{ lb}}) (15\%) = 255,092 \text{ tons/yr}$$

To obtain the actual weight of methane from carbon

$$255,092 \text{ tons/yr} \left(\frac{16 \text{ ton moles methane}}{12 \text{ ton moles carbon}} \right) = 340,123 \text{ tons/yr CH}_4$$
 then, dividing by 365 = 932 tons/day CH₄.

A more detailed approach utilizing actual field measurements of landfill gas can be found in Reference 4. $\frac{5}{1000}$

As can be seen from the previous discussion, sanitary landfills are a significant source of organic compound emissions. Table 6-49-999 presents the results of samples of landfill gases collected by KVB during its study for the California Air Resources Board (Ref. 5). As expected, the

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TABLE **6-**49. SUMMARY OF COMPUTATION OF TOTAL CARBON RELEASE FROM ACTIVE LANDFILL SITES DURING 1975*

Refuse Carbo Carbo Period ton refuse) 98 76.3 1566. 1970-75 2.5 20.52 98 76.3 1566. 1965-70 7.5 9.63 92 71.7 690. 1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 5.5 1.84 18 14.0 25.	n
Period t/yr ton refuse) Refuse (based on population) for (106 tons) Emissi in 19 (106 tons) 1970-75 2.5 20.52 98 76.3 1566. 1965-70 7.5 9.63 92 71.7 690. 1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	
Period t/yr (lb c/ ton refuse) (based on population) Period (10 ⁶ tons) in 19 (10 ⁶ tons) 1970-75 2.5 20.52 98 76.3 1566. 1965-70 7.5 9.63 92 71.7 690. 1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	ons
Period t/yr ton refuse) population) (10 ⁶ tons) (10 ⁶ tons) 1970-75 2.5 20.52 98 76.3 1566. 1965-70 7.5 9.63 92 71.7 690. 1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	3 F
1970-75 2.5 20.52 98 76.3 1566. 1965-70 7.5 9.63 92 71.7 690. 1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	
1965-70 7.5 9.63 92 71.7 690. 1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.)ns)
1960-65 12.5 6.29 83 64.7 406. 1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	49
1955-50 17.5 4.67 70 54.5 254. 1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	14
1950-55 22.5 3.72 55 42.8 159. 1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	68
1945-50 27.5 3.08 44 34.3 105. 1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	65
1940-45 32.5 2.64 35 27.3 71. 1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	38
1935-40 37.5 2.30 28 21.8 50. 1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	5 7
1930-35 42.5 2.04 23 17.9 36. 1925-30 47.5 1.84 18 14.0 25.	98
1925-30 47.5 1.84 18 14.0 25.	L7
	55
1020 25 52 5 1 67 12 12 10 1	30
1920-25 52.5 1.67 13 10.1 16.	91
1915-20 57.5 1.53 8 6.2 9.	53
1910-15 62.5 1.41 4 3.1 4.	39
1905-10 67.5 1.31 2 1.6 2.	04
1900-05 72.5 1.22 1 0.8 0.	<u>95</u>
Total 447.1 3401.	23

^{*}Reference 5

hydrocarbon portion of the ${\rm CO_2/CH_4}$ landfill generated gases that were primarily methane with trace quantities of various other materials. Considering the appreciable amounts of methane emissions, these trace quantities represent significant sources of organic compounds vented to the atmosphere in excess of one ton per day.

The control of hydrocarbons emanating from landfills consists of basically two methods:

- 1. Extraction plus combustion
- 2. Extraction, refining, and commercial sale

Both forms initially require the extraction of the landfill gas prior to its entering the atmosphere. A network of underground piping under a slight vacuum extracts the gas and transfers it to the prescribed control device. The rate of extraction is dependent on the site's ability to generate ${\rm CH}_4$ gas.

Profile Basis

Duplicate grab samples by gas collection bottles plus charcoal tubes were made at a Class II sanitary landfill site. This landfill was carefully selected to be representative of a typical Class II sanitary landfill site. Age of the site where the samples were taken was estimated at five to seven years.

Data Qualification

Profile **6**-49-999A applies to Class II sanitary landfill sites located in semi-arid regions. A Class II sanitary landfill prohibits the dumping of liquid sewage and hazardous waste material.

DECEMBER 14, 1978 TABLE 6-49-999A

AREA SOURCE EMISSIONS, SOLID WASTE LANDFILL SITE, CLASS II DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0202

LINE	SAROAD	CHEMICAL	MOLECULAR	PERCENT	PERCENT		CHEMICAL
NO.	CODE	NAME	WEIGHT	WEIGHT	VOLUME	CL	ASSIFICATION
1	43204	PROPANE	44. 09	10	. 03	1	PARAFFIN
2	43212	N-BUTANE	58. 12	. 20	. 05	1	PARAFFIN
3	43214	ISOBUTANE	58. 12	. 10	. 03	1	PARAFFIN
4	43220	N-PENTANE	72. 15	. 10	. 02	1	PARAFFIN
5	43242	CYCLOPENTANE	70. 14	. 20	. 05	1	PARAFFIN
6	43123	TERPENES	136. 23	. 10	. 02	2	OLEFIN
7	45102	ISOMERS OF XYLENE	106, 16	. 10	. 02	3	AROMATIC
8	45202	TOLUENE	92. 13	. 10	. 02	3	AROMATIC
9	43817	PERCHLOROETHYLENE	165. 83	. 30	. 03	5	MISCELLANEOUS
10	43201	METHANE	16. 04	98. 60	99. 69	6	METHANE
11	43202	ETHANE	30. 07	. 10	. 05	7	NON-REACTIVE
		TOTAL		100. 00	100. 01		
	5 COMP	OUNDS OF CLASSIFICATION 1	60. 68	. 70	. 18		
	1 COMP	OUNDS OF CLASSIFICATION 2	136, 23	. 10	. 02		
	2 COMP	OUNDS OF CLASSIFICATION 3	99. 15	. 20	. 04		
	O COMP	OUNDS OF CLASSIFICATION 4	. 00	. 00	. 00		
		OUNDS OF CLASSIFICATION 5	165. 83	. 30	. 03		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	98 . 60	99 . 69		
	1 COMP	OUNDS OF CLASSIFICATION 7	30. 07	. 10	. 05		
	11 COMP	OUND COMPOSITE	16. 23	100.00	100. 01		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA QC-MS ANALYSIS OF GRAB SAMPLES

B. REFERENCES: KVB TEST DATA (REF. 5)

C. APPLICABLE SCC CATEGORIES: 6-49-999-99

REFERENCES

- 1. Mery, R. C. and Stone, R., "Sanitary Landfill Behavior in an Aerobic Environment," Public Works, January 1966.
- 2. McFarlane, I.C., "Gas Explosion Hazards in Sanitary Landfills," Public Works, May 1970.
- 3. Dair, F. R. and Schwegler, R. E., "Energy Recovery from Landfills," Waste Age, March/April 1974.
- 4. "In-Situ Investigation of Movements of Gases Produced from Decomposing Refuse," Engineering-Sciences, Inc., California State Water Qualify Control Board Publication No. 31, 1965.
- 5. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," KVB, Inc., California Air Resources Board, Final Report, June 1978.

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6-49 AREA SOURCE EMISSIONS, SOLID WASTE

6-49-999 ANIMAL WASTE

Process Description 1

The decay of animal wastes consists of aerobic and anaerobic phases. The former occurs throughout excreta decomposition and in the surface drying of fecal matter. The odoriferous compounds released do not generally contain mercaptans or sulfides. The latter occurs after a crust has formed over the manure pile. The bacteria involved in this anaerobic digestion produce the more foul-smelling components which include sulfurous compounds.

Emissions²⁻⁵

In any animal production system, volatile compounds from decomposing manure are released into the atmosphere. In some cases these gases have an adverse effect on the animals themselves or possess such odors as to be judged a public nuisance.

Results from recent studies (Refs. 2-4) estimated animal wastes TOC emission factors to be:

Cattle	440	lbs	TOC/10 ³ head/day*
Chickens	7		u
Pigs	160		11
Horses	229		11
Sheep	33		11

Profile 6-49-999-B presents the organic species (>1%wt) estimated to be emanating from livestock waste in general (Ref. 5).

Local county or state agricultural reports are useful in obtaining information on livestock populations.

Controls 1-3

The control of organic gases generated from the decomposition of animal waste in confined areas generally take the form of:

. Maintenance of an aerobic environment in the waste-handling system (Ref. 3)

^{*} TOC--Total Organic Carbon

- . Recovery as a possible energy source (Ref. 2)
- . Wastelage--process of mixing fresh manure with hay (Ref. 1)
- . Combustion--catalytic oxidation
 --thermal incinerator
 --combustion air supply
- . Alkaline Scrubbing--odor control (Ref. 1)

Profile Basis⁵

The emission factors and profile presented are based on a survey and engineering evaluation of the literature cited (Refs. 1-5).

Data Qualification

The reported emission factors and corresponding profile may be used in estimating the total organic emissions from livestock waste in general. The effects of environmental factors such as rain, heat, and cold, however, were not considered. Their influence is estimated to be significant in estimating annual emission rates.

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AREA SOURCE EMISSIONS, SOLID WASTE

ANIMAL WASTE DECOMPOSITION DATA CONFIDENCE LEVEL: III

CONTROL DEVICE: NONE

PROCESS MODIFICATION: NONE

KVB PROFILE KEY 0203

INE NO.	SAROAD	CHEMICAL NAME	MOLECULAR WEIGHT	PERCENT WEIGHT	PERCENT VOLUME	CL	CHEMICAL ASSIFICATION
1	43551	ACETONE	58. 08	2. 00	. 65	4	CARBONYL
2	43302	ETHYL ALCOHOL	46. 07	2. 00	. 83	5	MISCELLANEOUS
3	43304	ISOPROPYL ALCOHOL	60. 09	2. 00	. 63	5	MISCELLANEOUS
4	43434	PROPYL ACETATE	102. 13	2. 00	. 38	5	MISCELLANEOUS
5	43721	ETHYLAMINE	45. 09	1.00	. 42	5	MISCELLANEOUS
6	43740	TRIMETHYL AMINE	59. 11	1, 00	. 33	5	MISCELLANEOUS
7	43201	METHANE	16. 04	70. 00	83. 96	6	METHANE
8	43202	ETHANE	30. 07	20, 00	12. 79	7	NON-REACTIVE
		TOTAL		100. 00	99. 99		
	O COMP	OUNDS OF CLASSIFICATION 1	. 00	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 2	. 0 0	. 00	. 00		
	O COMP	OUNDS OF CLASSIFICATION 3	. 00	. 00	. 00		
	1 COMP	OUNDS OF CLASSIFICATION 4	58. 08	2. 00	. 65		
	5 COMP	POUNDS OF CLASSIFICATION 5	59. 21	8.00	2. 59		
	1 COMP	OUNDS OF CLASSIFICATION 6	16. 04	70. 00	83. 96		
	1 COMP	POUNDS OF CLASSIFICATION 7	30. 07	20. 00	12. 79		
	8 COMP	POUND COMPOSITE	19. 23	100, 00	99. 99		

NOTES: A. METHOD: CALCULATIONS FROM COMPOSITE SURVEY DATA ENGINEERING EVALUATION OF LITERATURE DATA

B. REFERENCES: LITERATURE TEST DATA (REF. 1-5)

C. APPLICABLE SCC CATEGORIES: 4-49-999-98

REFERENCES

- 1. Bethea, Robert N., "Solution for Feedlot Odor Control Problems," Journal of ARCA, Vol. 22, No. 10, October 1972.
- 2. Horton, R. and Hawkes, D., "The Energy and Fertilizer Potential of Natural Organic Wastes," June 1976.
- 3. Aschbacher, P. W., "Air Pollution Research Needs Livestock Production Systems," Journal of APCA, Vol. 23, No. 4, April 1973.
- 4. Keller, R. M. and Cowherd, C., "Identification and Measurement of Atmospheric Organic Emissions from Natural and Quasi-Natural Sources," interim report for EPA Contract 68-02-2524, July 1977.
- 5. Taback, H. J., et al., "Control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin," Vol. I and II, KVB, Inc., Tustin, CA, June 1978.

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APPENDIX I

- I-A CHEMICAL FILE BY SAROAD CODE
- I-B CHEMICAL FILE BY CHEMICAL CLASSIFICATION
- I-C ORGANIC SPECIES BY SOURCES
- I-D NEDS SOURCE CLASSIFICATION CODES
- I-E AREA SOURCE CATEGORIES
- I-F BOILING POINT RANGE COMPOUNDS

APPENDIX I-A

CHEMICAL FILE BY SAROAD CODE

CHEMICAL FILE SORTED BY THE CHEMICAL CLASSIFICATION

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	CHEMICAL CLASSIFICATION
1	43000	UNIDENTIFIED HYDROCARBONS	86. 00	5 MISCELLANEOUS
2	43105	ISOMERS OF HEXANE	86. 17	1 PARAFFIN
3	43106	ISOMERS OF HEPTANE	100. 20	1 PARAFFIN
4	43107	ISOMERS OF OCTANE	114. 23	1 PARAFFIN
5	43108	ISOMERS OF NONANE	128. 25	1 PARAFFIN
6	43109	ISOMERS OF DECANE	142. 28	1 PARAFFIN
7	43110	ISOMERS OF UNDECANE	156. 30	1 PARAFFIN
8	43111	ISOMERS OF TRIDECANE	184. 36	1 PARAFFIN
9	43112	ISOMERS OF DODECANE	170. 33	1 PARAFFIN
10	43113	ISOMERS OF TETRADECANE	198. 38	1 PARAFFIN
11	43114	ISOMERS OF PENTADECANE	212. 41	1 PARAFFIN
12	43115	C-7 CYCLOPARAFFINS	98. 19	1 PARAFFIN
13	43116	C-8 CYCLOPARAFFINS	112. 23	1 PARAFFIN
14	43117	C-9 CYCLOPARAFFINS	126. 26	1 PARAFFIN
15	43118	MINERAL SPIRITS	114. 00	1 PARAFFIN
16	43119	LACTOL SPIRITS	114.00	5 MISCELLANEOUS
17	43120	ISOMERS OF BUTENE	56. 10	2 OLEFIN
18	43121	ISOMERS OF PENTENE	70. 13	2 OLEFIN

19	43122	ISOMERS OF PENTANE	72. 15	1 PARAFFIN
20	43123	TERPENES	136. 23	2 OLEFIN
21	43201	METHANE	16. 04	6 METHANE
22	43202	ETHANE	30. 07	7 NON-REACTIVE
23	43203	ETHYLENE	28. 0 5	2 OLEFIN
24	43204	PROPANE	44. 09	1 PARAFFIN
25	43205	PROPYLENE	42. 08	2 OLEFIN
26	43206	ACETYLENE	26. 04	5 MISCELLANEOUS
27	43207	CYCLOPROPANE	42. 08	1 PARAFFIN
28	43208	PROPADIENE	40. 06	2 OLEFIN
29	43209	METHYLACETYLENE	4 0. 0 6	5 MISCELLANEOUS
30	43212	N-BUTANE	58. 12	1 PARAFFIN
31	43213	BUTENE	56. 10	2 OLEFIN
32	43214	ISOBUTANE	58. 12	1 PARAFFIN
33	43218	1,3-BUTADIENE	54. 09	2 OLEFIN
34	43219	ETHYLACETYLENE	54. 09	5 MISCELLANEOUS
35	43220	N-PENTANE	72. 15	1 PARAFFIN
36	43223	3-METHYL-1-BUTENE	70. 14	2 OLEFIN
37	43224	1-PENTENE	70. 13	2 OLEFIN
38	43228	2-METHYL-2-BUTENE	70. 13	2 OLEFIN
39	43231	N-HEXANE	86. 17	1 PARAFFIN
40	43232	N-HEPTANE	100. 20	1 PARAFFIN

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41	43233	N-OCTANE	114. 23	1 PARAFFIN
42	43235	N-NONANE	128. 25	1 PARAFFIN
43	43238	N-DECANE	142. 28	1 PARAFFIN
44	43241	N-UNDECANE	156. 30	1 PARAFFIN
45	43242	CYCLOPENTANE	70. 14	1 PARAFFIN
46	43245	1-HEXENE	84. 16	2 OLEFIN
47	43248	CYCLOHEXANE	84. 16	1 PARAFFIN
48	43255	N-DODECANE	170. 33	1 PARAFFIN
49	43258	N-TR I DECANE	184. 36	1 PARAFFIN
50	43259	N-TETRADECANE	198. 38	1 PARAFFIN
51	43260	N-PENTADECANE	212. 41	1 PARAFFIN
52	43261	METHYLCYCLOHEXANE	98. 18	1 PARAFFIN
53	43262	METHYLCYCLOPENTANE	84. 16	1 PARAFFIN
54	43264	CYCLOHEXANONE	98. 15	4 CARBONYL
55	43301	METHYL ALCOHOL	32. 04	5 MISCELLANEOUS
56	43302	ETHYL ALCOHOL	46. 07	5 MISCELLANEOUS
57	43303	N-PROPYL ALCOHOL	60. 09	5 MISCELLANEOUS
58	43304	ISOPROPYL ALCOHOL	60. 09	5 MISCELLANEOUS
59	43305	N-BUTYL ALCOHOL	74. 12	5 MISCELLANEOUS
60	43306	ISOBUTYL ALCOHOL	74. 12	5 MISCELLANEOUS
61	43308	BUTYL CELLOSOLVE	102. 00	5 MISCELLANEOUS
62	43309	TERT-BUTYL ALCOHOL	74. 12	5 MISCELLANEOUS

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63	43310	METHYL CELLOSOLVE	76. 11	5 MISCELLANEOUS
64	43311	CELLOSOLYE	90. 12	5 MISCELLANEOUS
65	43320	DIACETONE ALCOHOL	116. 16	4 CARBONYL
66	43351	ETHYL ETHER	74. 12	5 MISCELLANEOUS
67	43367	GLYCOL ETHER	62. 07	5 MISCELLANEOUS
68	43368	GLYCOL	62. 07	5 MISCELLANEOUS
69	43369	PROPYLENE GLYCOL	76. 00	5 MISCELLANEOUS
70	43370	ETHYLENE GLYCOL	62. 07	5 MISCELLANEOUS
71	43390	TETRAHYDROFURAN	72. 10	5 MISCELLANEOUS
72	43404	ACETIC ACID	60. 05	5 MISCELLANEOUS
73	43432	METHYL ACETATE	74. 08	5 MISCELLANEOUS
74	43433	ETHYL ACETATE	88. 10	5 MISCELLANEOUS
75	43434	PROPYL ACETATE	102. 13	5 MISCELLANEOUS
76	43435	N-BUTYL ACETATE	116. 16	5 MISCELLANEOUS
77	43438	ETHYL ACRYLATE	100. 11	5 MISCELLANEOUS
78	43443	CELLOSOLVE ACETATE	132. 00	5 MISCELLANEOUS
79	43444	ISOPROPYL ACETATE	104. 00	5 MISCELLANEOUS
80	43445	METHYL AMYL ACETATE	140. 00	5 MISCELLANEOUS
81	43446	ISOBUTYL ACETATE	116. 16	5 MISCELLANEOUS
82	43450	DIMETHYLFORMAMIDE	73. 09	5 MISCELLANEOUS
83	43451	ISOBUTYL ISOBUTYRATE	144. 21	5 MISCELLANEOUS
84	43452	2-ETHOXYETHYL ACETATE	132. 00	5 MISCELLANEOUS

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85	43502	FORMALDEHYDE	30. 03	4 CARBONYL
86	43503	ACETALDEHYDE	44. 05	4 CARBONYL
87	43510	BUTYRALDEHYDE	72. 12	4 CARBONYL
88	43551	ACETONE	58. 08	4 CARBONYL
89	43552	METHYL ETHYL KETONE	72. 10	4 CARBONYL
90	435 5 9	METHYL N-BUTYL KETONE	100. 16	4 CARBONYL
91	43560	METHYL ISOBUTYL KETONE	100. 16	4 CARBONYL
92	43601	ETHYLENE OXIDE	44. 05	5 MISCELLANEOUS
93	43602	PROPYLENE OXIDE	58. 08	5 MISCELLANEOUS
94	43702	ACETRONITRILE	41. 05	7 NON-REACTIVE
95	43704	ACRYLONITRILE	55. 00	5 MISCELLANEOUS
96	43721	ETHYLAMINE	45. 09	5 MISCELLANEOUS
97	43740	TRIMETHYL AMINE	59 . 11	5 MISCELLANEOUS
98	43801	METHYL CHLORIDE	50. 49	7 NON-REACTIVE
99	43802	DICHLOROMETHANE	84. 94	5 MISCELLANEOUS
100	43803	CHLOROFORM	119. 39	7 NON-REACTIVE
101	43804	CARBON TETRACHLORIDE	153. 84	5 MISCELLANEOUS
102	43807	CARBON TETRABROMIDE	331. 67	5 MISCELLANEOUS
103	43811	TRICHLOROFLUOROMETHANE	137. 37	5 MISCELLANEOUS
104	43812	ETHYL CHLORIDE	64. 52	5 MISCELLANEOUS
105	43813	1,1-DICHLOROETHANE	98. 97	5 MISCELLANEOUS
106	43814	1, 1, 1-TRICHLOROETHANE	133. 42	7 NON-REACTIVE

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107	43815	ETHYLENE DICHLORIDE	99. 00	/ NUN-KEACIIVE
108	43817	PERCHLOROETHYLENE	165. 83	5 MISCELLANEOUS
109	43819	METHYLENE BROMIDE	173. 85	5 MISCELLANEOUS
110	43820	1, 1, 2-TRICHLOROETHANE	131. 66	7 NON-REACTIVE
111	43821	TRICHLOROTRIFLUOROETHANE	187. 38	7 NON-REACTIVE
112	43822	TRIMETHYLFLUOROSILANE	92. 00	5 MISCELLANEOUS
113	43823	DICHLORODIFLUOROMETHANE	120. 91	5 MISCELLANEOUS
114	43824	TRICHLOROETHYLENE	131. 40	5 MISCELLANEOUS
115	43860	VINYL CHLORIDE	62. 5 0	5 MISCELLANEOUS
116	45101	NAPHTHA	114. 00	3 AROMATIC
117	45102	ISOMERS OF XYLENE	106. 16	3 AROMATIC
118	45103	DIMETHYLETHYLBENZENE	134. 00	3 AROMATIC
119	45104	ISOMERS OF ETHYLTOLUENE	120. 19	3 AROMATIC
120	45105	ISOMERS OF BUTYLBENZENE	134. 21	3 AROMATIC
121	45106	ISOMERS OF DIETHYLBENZENE	134. 21	3 AROMATIC
122	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	3 AROMATIC
123	45108	ISOMERS OF PROPYLBENZENE	120. 19	3 AROMATIC
124	45201	BENZENE	78. 11	7 NON-REACTIVE
125	45202	TOLUENE	92. 13	3 AROMATIC
126	45203	ETHYLBENZENE	106. 16	3 AROMATIC
127	45207	1,3,5-TRIMETHYLBENZENE	120. 19	3 AROMATIC
128	45220	STYRENE	104. 14	3 AROMATIC

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129	45221	A-METHYLSTYRENE	118, 15	3 AROMATIC
130	45225	1, 2, 3-TRIMETHYLBENZENE	111.00	3 AROMATIC
131	45300	PHENOLS	94. 11	5 MISCELLANEOUS
132	45401	XYLENE BASE ACIDS	230. 00	5 MISCELLANEOUS
133	45801	CHLOROBENZENE	112. 56	3 AROMATIC
134	46201	1,4-DIOXANE	88. 12	5 MISCELLANEOUS

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APPENDIX I-B

CHEMICAL FILE BY CHEMICAL CLASSIFICATION

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C H E M I C A L F I L E SORTED BY THE CHEMICAL CLASSIFICATION

NO.	SAROAD CODE	CHEMICAL NAME	MOLECULAR WEIGHT	CHEMICAL CLASSIFICATION
1	43105	ISOMERS OF HEXANE	86. 17	1 PARAFFIN
2	43106	ISOMERS OF HEPTANE	100. 20	1 PARAFFIN
3	43107	ISOMERS OF OCTANE	114. 23	1 PARAFFIN
4	43108	ISOMERS OF NONANE	128. 25	1 PARAFFIN
5	43109	ISOMERS OF DECANE	142. 28	1 PARAFFIN
6	43110	ISOMERS OF UNDECANE	156. 30	1 PARAFFIN
7	43111	ISOMERS OF TRIDECANE	184. 36	1 PARAFFIN
8	43112	ISOMERS OF DODECANE	170. 33	1 PARAFFIN
9	43113	ISOMERS OF TETRADECANE	198. 38	1 PARAFFIN
10	43114	ISOMERS OF PENTADECANE	212. 41	1 PARAFFIN
11	43115	C-7 CYCLOPARAFFINS	98. 19	1 PARAFFIN
12	43116	C-8 CYCLOPARAFFINS	112. 23	1 PARAFFIN
13	43117	C-9 CYCLOPARAFFINS	126. 26	1 PARAFFIN
14	43118	MINERAL SPIRITS	114. 00	1 PARAFFIN
15	43204	PROPANE	44. 0 7	1 PARAFFIN
16	43207	CYCLOPROPANE	42. 08	1 PARAFFIN
17	43212	N-BUTANE	58. 12	1 PARAFFIN
18	43214	ISO-BUTANE	58. 12	1 PARAFFIN

19	43220	N-PENTANE	72. 15	1 PARAFFIN
20	43231	HEXANE	86. 17	1 PARAFFIN
21	43232	HEPTANE	100. 20	1 PARAFFIN
22	43233	OCTANE	114. 23	1 PARAFFIN
23	43235	NONANE	128. 25	1 PARAFFIN
24	43238	N-DECANE	142. 28	1 PARAFFIN
25	43241	UNDECANE	156. 30	1 PARAFFIN
26	43242	CYCLOPENTANE	70. 14	1 PARAFFIN
27	43248	CYCLOHEXANE	84. 16	1 PARAFFIN
28	43255	N-DODECANE	170. 33	1 PARAFFIN
29	43258	N-TRIDECANE	184. 36	1 PARAFFIN
30	43259	N-TETRADECANE	198. 38	1 PARAFFIN
31	43260	N-PENTADECANE	212. 41	1 PARAFFIN
32	43261	METHYLCYCLOHEXANE	85. 00	1 PARAFFIN
33	43262	METHYLCYCLOPENTANE	84. 16	1 PARAFFIN
34	43120	ISOMERS OF BUTENE	56. 10	2 OLEFIN
35	43121	ISOMERS OF PENTENE	70. 13	2 OLEFIN
36	43122	ISOMERS OF PENTANE	72. 15	2 OLEFIN
37	43203	ETHYLENE	28. 05	2 OLEFIN
38	43205	PROPYLENE	42. 08	2 OLEFIN
39	43208	PROPADIENE	40. 06	2 OLEFIN
40	43213	BUTENE	56, 10	2 OLEFIN

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41	43218	1. 3-BUTADIENE	54. 0 9	2 OLEFIN
42	43223	3-METHYL-1-BUTENE	70. 14	2 OLEFIN
43	43224	1-PENTENE	70. 13	2 OLEFIN
44	43228	2-METHYL-2-BUTENE	70. 13	2 OLEFIN
45	43245	1-HEXENE	84. 16	2 OLEFIN
46	45101	NAPHTHA	114. 00	3 AROMATIC
47	45102	ISOMERS OF XYLENE	106. 16	3 AROMATIC
48	45103	DIMETHYLETHYLBENZENE	134. 00	3 AROMATIC
49	45104	ISOMERS OF ETHYLTOLUENE	120. 19	3 AROMATIC
50	45105	ISOMERS OF BUTYLBENZENE	134. 21	3 AROMATIC
51	45106	ISOMERS OF DIETHYLBENZENE	134. 21	3 AROMATIC
52	45107	ISOMERS OF TRIMETHYLBENZENE	120. 19	3 AROMATIC
53	45108	ISOMERS OF PROPYLBENZENE	120. 19	3 AROMATIC
54	45202	TOLUENE	92 . 13	3 AROMATIC
55	45203	ETHYLBENZENE	106. 16	3 AROMATIC
56	45207	1. 3. 5-TRIMETHYLBENZENE	120. 19	3 AROMATIC
57	45220	STYRENE	104. 14	3 AROMATIC
58	45221	A-METHYLSTYRENE	118. 15	3 AROMATIC
59	45225	1, 2, 3-TRIMETHYLBENZENE	111.00	3 AROMATIC
60	45801	CHLOROBENZENE	112. 56	3 AROMATIC
61	43264	CYCLOHEXANONE	98. 15	4 CARBONYL
62	43320	DIACETONE ALCOHOL	116. 16	4 CARBONYL

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63	43502	FORMALDEHYDE	30. 03	4 CARBONYL
64	43503	ACETALDEHYDE	44. 05	4 CARBONYL
65	43510	BUTYRALDEHYDE	72. 12	4 CARBONYL
66	43551	ACETONE	58. 08	4 CARBONYL
67	43552	METHYL ETHYL KETONE	72. 10	4 CARBONYL
68	43559	METHYL N-BUTYL KETONE	100. 16	4 CARBONYL
69	43560	METHYL ISOBUTYL KETONE	100. 16	4 CARBONYL
70	43000	UNIDENTIFIED HYDROCARBONS	86. 00	5 MISCELLANEOUS
71	43119	LACTOL SPIRITS	114. 00	5 MISCELLANEOUS
72	43123	TERPENES	136. 23	5 MISCELLANEOUS
73	43206	ACETYLENE	26. 04	5 MISCELLANEOUS
74	43209	METHYLACETYLENE	40. 06	5 MISCELLANEOUS
75	43219	ETHYLACETYLENE	54. 09	5 MISCELLANEOUS
76	43301	METHYL ALCOHOL	32. 04	5 MISCELLANEOUS
7 7	43302	ETHYL ALCOHOL	46. 07	5 MISCELLANEOUS
78	43303	N-PROPYL ALCOHOL	60. 09	5 MISCELLANEOUS
79	43304	ISO-PROPYL ALCOHOL	60. 09	5 MISCELLANEOUS
80	43305	N-BUTYL ALCOHOL	74. 12	5 MISCELLANEOUS
81	43306	ISO-BUTYL ALCOHOL	74. 12	5 MISCELLANEOUS
82	43308	BUTYL CELLOSOLVE	102. 00	5 MISCELLANEOUS
83	43309	TERT-BUTYL ALCOHOL	74. 12	5 MISCELLANEOUS
84	43310	METHYL CELLOSOLVE	76. 1 1	5 MISCELLANEOUS

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85	43311	CELLOSOLVE	90. 12	5 MISCELLANEOUS
86	43351	ETHYL ETHER	74. 12	5 MISCELLANEOUS
87	43367	QLYCOL ETHER	62. 07	5 MISCELLANEOUS
88	43368	GLYCOL	62. 07	5 MISCELLANEOUS
89	43369	PROPYLENE GLYCOL	76. 00	5 MISCELLANEOUS
90	43370	ETHYLENE GLYCOL	62. 07	5 MISCELLANEOUS
91	43390	TETRAHYDROFURAN	72. 10	5 MISCELLANEOUS
92	43404	ACETIC ACID	60. 05	5 MISCELLANEOUS
93	43432	METHYL ACETATE	74. 08	5 MISCELLANEOUS
94	43433	ETHYL ACETATE	88. 10	5 MISCELLANEOUS
95	43434	PROPYL ACETATE	102. 13	5 MISCELLANEOUS
96	43435	N-BUTYL ACETATE	116. 16	5 MISCELLANEOUS
97	43438	ETHYL ACRYLATE	100. 11	5 MISCELLANEOUS
78	43443	CELLOSOLVE ACETATE	132. 00	5 MISCELLANEOUS
99	43444	ISOPROPYL ACETATE	104. 00	5 MISCELLANEOUS
100	43445	METHYL AMYL ACETATE	140. 00	5 MISCELLANEOUS
101	43446	ISOBUTYL ACETATE	116. 16	5 MISCELLANEOUS
102	43450	DIMETHYL FORMAMIDE	73. 09	5 MISCELLANEOUS
103	43451	ISOBUTYL ISOBUTYRATE	144. 21	5 MISCELLANEOUS
104	43452	2-ETHOXYETHYL ACETATE	132. 00	5 MISCELLANEOUS
105	43601	ETHYLENE OXIDE	44. 05	5 MISCELLANEOUS
106	43602	PROPYLENE OXIDE	58. 08	5 MISCELLANEOUS

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J7	43704	ACRYLONITRILE	55. 00	5 MISCELLANEOUS
108	43721	ETHYLAMINE	45 . 0 9	5 MISCELLANEOUS
109	43740	TRIMETHYL AMINE	59. 11	5 MISCELLANEOUS
110	43802	DICHLOROMETHANE	84. 94	5 MISCELLANEOUS
111	43807	CARBON TETRABROMIDE	331. 67	5 MISCELLANEOUS
112	43811	TRICHLOROFLUOROMETHANE	137. 37	5 MISCELLANEOUS
113	43812	ETHYL CHLORIDE	64. 52	5 MISCELLANEOUS
114	43813	1.1-DICHLOROETHANE	98. 9 7	5 MISCELLANEOUS
115	43817	PERCHLOROETHYLENE	165. 83	5 MISCELLANEOUS
116	43819	METHYLENE BROMIDE	173. 85	5 MISCELLANEOUS
117	43822	TRIMETHYLFLUOROSILANE	92 . 00	5 MISCELLANEOUS
118	43823	DICHLORDIFLUOROMETHANE	120. 91	5 MISCELLANEOUS
119	43824	TRICHLOROETHENE	13. 14	5 MISCELLANEOUS
120	43860	VINYL CHLORIDE	62. 50	5 MISCELLANEOUS
121	45300	PHENOLS	94. 11	5 MISCELLANEOUS
122	45401	XYLENE BASE ACIDS	230. 00	5 MISCELLANEOUS
123	46201	1,4-DIOXANE	88. 12	5 MISCELLANEOUS
124	43201	METHANE	16. 04	6 METHANE
125	43202	ETHANE	30. 07	7 NON-REACTIVE
126	43702	ACETRONITRILE	41. 05	7 NON-REACTIVE
127	43801	METHYL CHLORIDE	50. 49	7 NON-REACTIVE
128	43803	CHLOROFORM	119, 39	7 NON-REACTIVE

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129	43814	1,1,1-TRICHLORDETHANE	133. 42	7 NON-REACTIVE
130	43815	ETHYLENE DICHLORIDE	99. 00	7 NON-REACTIVE
131	43820	1. 1. 2-TRICHLORGETHANE	131. 66	7 NON-REACTIVE
132	43821	TRICHLOROTRIFLUOROETHANE	187. 38	7 NON-REACTIVE
133	45201	BENZENE	78. 11	7 NON-REACTIVE

APPENDIX I-C

ORGANIC SPECIES BY SOURCES

Table I-C is a breakdown of the organic species detected during the KVB, Inc. test program on control of Hydrocarbon Emissions from Stationary Sources in the California South Coast Air Basin. This table lists the organic specie, its chemical class, the concentration levels found and the sources that emitted that compound.

TABLE I-C-A. TEST RESULTS BY SPECIES, ALCOHOLS

Name	Chem. Cıass	Concentrations, ppm measured	Source Type
Methyl Alcohol	5	1 - 100	Appliance enamel, Flexograph ink, Landfill, Printed circuit stripper
Ethyl Alcohol	5	0.1 - 10	Appliance enamel, Flexograph ink
Isopropyl Alcohol	5	0.1 - 1.0	Lithograph ink inlet to control only (thermo burner)
n Butyl Alcohol	5	1 - 10	Appliance enamel
Isobutyl Alcohol	5	0.1 - 1.0	Appliance enamel

TABLE I-C-B. TEST RESULTS BY SPECIES, KETONES

Name	Chem. Class	Concentrations, ppm measured	Source Type
Acetone	4	0.1 - 100 ⁻	Adhesives, Appliance enamel, Flexograph ink, Landfill gas, Plastics coatings, Power plant combustion, Sewage gas, Water based paint
Methyl Ethyl Ketone	4	1 - 100	Appliance enamel, Plastics coatings
Methyl Isobutyl Ketone	4	1 - 10,000	Magnetic tape coating

TABLE I-C-C. TEST RESULTS BY SPECIES, ESTERS (ACETATES)

Name	Chem. Class	Concentrations, ppm measured	Source Type
Ethyl Acetate	5	0.1 - 10	Adhesives, Landfill gas, Water based paint
n Propyl Acetate	5	10 - 100	Flexograph ink
Isopropyl Acetate	5	0.1 - 100	Flexograph ink
n Butyl Acetate	5	10 - 100	Appliance enamel

TABLE I-C-D. TEST RESULTS BY SPECIES, HALO-COMPOUNDS

Name	Chem. Class	Concentrations, ' ppm measured	Source Type
Trichloro-fluoro- methane (Freon 11)	5	10,000 - 100,000	Refrigerant fill line gas
Dichloro-difluoro- methane	5	100,000 - 1,000,000	Refrigerant fill line gas
Methylene Chloride (dichloromethane)	5	1 - 10,000	Landfill gas, Lithograph ink, Printed circuit stripper solvent, Rubber masking paint
1,1,1-Trichloro- methane (methylchloroform)	7	1 - 1,000	Metal degreaser fluid, Printed circuit stripper
Vinyl Chloride	5	0.1 - 1.0	Landfill gas
Methyl Chloride	7	1 - 10	Printed circuit, Process Plant Background
1,2 Dichloro- ethylene (Acetylene dichloride)	5	1 - 10	Landfill gas
Perchloroethylene (tetrachloroethylene)	5	0.1 - 1,000	Adhesive, Appliance enamel, Landfill gas, Metal degreaser fluid, Rubber masking plant
Trimethylfluorosilane	5	0.1 - 10	Steel furnace gases

TABLE 1-C+E. TEST RESULTS BY SPECIES, AROMATICS

Name.	Chem.	Concentrations,	Sauras Series
Name	Class	ppm measured	Source Type
Benzene	7	1 - 10 10 - 100 100 - 1,000 1,000 - 10,000	Appliance enamel paint; Automotive water based paint; Coke oven gas; Crude oil, heavy API*, gas; Crude oil, light API* gas; Dip enamel paint; Flexograph ink; Gasoline; Lacquer paint, automotive; Landfill gas; Natural gas combustion; Oil field gas drier; Oil field sump; Paving asphalt; Rafinery process gas; Rafinery pump seal leak; Refinery sour water; Refinery stock for blending; Roofing tar; Rotogravure ink; Rubber sol- vent; Stripper solvent for printed circuits; Vinyl adhesive
Toluene	3	1 - 10 10 - 100 100 - 1,000 1,000 - 10,000	Appliance enamel paint, Automotive water based paint, Automotive lacquer paint, Dip enamel paint, Flexograph ink, Gasoline, Landfill gas, Natural gas pilot light combustion, Process gas com- bustion, Refinery blending stock, Refinery process gas, Refinery pump leak, Refinery sour water, Roofing tar, Rotogravure ink
Tylenes -	3	1 - 10 10 - 100 100 - 1,000	Appliance enamel paint, Chemical blanding process, Dip enamel paint, Flexograph ink, Gasoline, Landfill gas, Refinery bland stock, Refinery pump leak, Rotogravure ink
Sthylbenzene	3	1 - 10	Appliance enamel

TABLE I-C=F. TEST RESULTS BY SPECIES, ALDEHYDES

	Chem.	Concentrations, ·	
Name	Class	ppm measured	Source Type
Formaldehyde	4	1 - 100	Combustion Sources: Appliance enamel oven afterburner, Gas combustion, Gas turbine, Lithograph ink afterburner, Natural gas IC engine, Pilot burner gas, Power plant boiler oil, Refinery CO boiler, Refinery process heater, Sewage-sludge gas burning IC engine, Solvent based auto- motive paint oven afterburner-cata- lytic afterburner, Water based auto- motive paint after- burner, Water based automotive base coat spray booth, Water based auto- motive base coat fume incinerator

TABLE I-C-G. TEST RESULTS BY SPECIES, OLEFIN OXIDE

Name	Chem. Class	Concentrations, ppm measured	Source Type
1,4 Dioxane (Diethylene dioxide	5	0.1 - 1.0	Vapor degreaser solvent
	and the same (to provide any see		

TABLE I C-H. TEST RESULTS BY SPECIES, ACETYLENES

Name	Chem. Class	Concentrations, ppm measured	Source Type
Acetylene	5	1 - 10	Steel processing coke ovens, sintering plant

TABLE I-C-I. TEST RESULTS BY SPECIES, CYCLOPARAFFINS

Name	Chem. Class	Concentrations, ppm measured	Source Type
Cycloparaffins	1	1 - 100%	Appliance enamel paint; Automotive solvent based paint, primer, top coat; Crude oil, light API°, heavy API°, wet and dry gases; Dip enamel paint; Gasoline; Landfill gas; Paving asphalt; Refinery blend stock; Roofing tar; Rubber adhesive; Rubber solvent; Rotogravure ink

TABLE I-C-J. TEST RESULTS BY SPECIES, OLEFINS

			T
	Chem.	Concentrations,	
Name	Class	ppm measured	Source Type
Ethylene	2	l ppm to 100%	Appliance enamel paint; Automotive paint, solvent based primer oven, afterburner; Auto- motive paint, solvent based top coat oven, catalytic afterburner; Auto- motive paint, water based primer oven; Coke oven gas; Gasoline; Lithograph ink catalytic after- burner; Natural gas; Paving asphalt; Refinery process gas; Roofing tar; Sewage sludge gas, IC engines
Propylene Butene Pentene, etc.	2	1 ppm to 100%	Appliance enamel paint; Automotive solvent based primer paint; Coke oven gas; Crude oil, heavy API°, wet and dry gas; Flexograph ink oven; Gasoline; Landfill gas; Paving asphalt; Refinery blend stock, process gas, process gas heater; Sewage gas; Sinter- ing plant, Steel mill
Terpenes	5	10 - 100	Landfill gas

TABLE I-C-K. TEST RESULTS BY SPECIES, PARAFFINS

Name	Chem.	Concentrations,	Source Time
	C1433	ppm measured	Source Type
Methane	6	1 ppm to 100%	Adhesives; Appliance enamel; Asphalt processing; Automotive
Ethane Propane C ₁ - C ₃	7 1		solvent based primer paint oven and afterburner, top coat oven catalytic afterburner; Automotive water based basecoat paint, top coat paint and oven; Coke oven gas; Crude oil, light API*, heavy API*. wet gas, dry gas; Degreaser, Flexograph ink oven; Gas turbine, Gasoline; Landfill gas; Lithograph ink catalytic afterburner, thermo afterburner; Natural gas; Paving asphalt; Power plants; Refinery fugitives, process gas, process heaters, CO boiler; Roofing tar; Rotogravure ink; Sewage Gas; Stripping solvent for printed circuits
C ₄ and higher and their isomers	1	l ppm to 100%	Adhesive, vinyl; Appliance enamel paint; Automotive solvent based paint, primer, topcoat, catalytic afterburner, thermo afterburner; Automotive water based paints, undercoat, topcoat, ovens, afterburner incinerator; Automotive lacquer paint; Cleaning solvent, Stoddard; Crude oil, light API*, heavy API*. wet gas, dry gas; Degreaser; Flaxograph ink oven; Gasoline; Landfill gas; Lithograph afterburners; Paving asphalt; Rafinery blend stock, process gas, process heater; Rotogravure ink, Roofing tar; Rubber adhesive; Rubber solvent; Stripper solvent for printed circuit board

APPENDIX I-D

NEDS SOURCE CLASSIFICATION CODES

ENVIRONMENTAL PROTECTION AGENCY	SECTION	NEDS	SECTION 3	CHAPTER 7	SUBJECT 0
NATIONAL AIR DATA BRANCH	CHAPTER SUBJECT	Source Classification Codes and Emission Factors	1/3/76 PAGE		GE
VOLUME V. AEROS MANUAL OF CODES			1/3/10		.

		P O U H D PART		TTEDPER			
ENTCORR ROILER	-ELECTRIC GENERATH	7.41	\$01	*91	×c	CO	,
AMTHRACITE COAL							
	PIGOMMETU PULYIZO	17.0 4	30.0		0.03	1.00	Tens BURNED
1-01-001-03	>100AMBTU STOKERS 10-100AMBTU PULVO	7.00 A	36.0		0.20 0.03	1.00	TOMS BURNED
PO-100-10	IO-100MRRTU STOER	8.00 4	38.0	10.5	0.50	6 , ND	TONS BURNED TONS BURNED
1-01-001-04	IO-190MMSTU PULVO IO-100MMSTU STOKE CIOMMSTU PULVIZEO CIOMMSTU STOKES OTMES/NOT CLASIFO	2.00 4	30.0	10.0	0.2n	10.0	TONS BURNED
1-0(-001-49	OTHER/NOT CLASIFD	17.0 4	30.0	4+00	0.20	1.00	TOWS BURNED
SITUMINOUS COAL							
1-01-002-01	>100mmstu Pulymet	13.0 A 17.0 A	38.0		0.30	1.00	TORS BURNED
1-01-001-03	>100MMBTU CTCLOME	2.00 A	38.0	\$5.0	0.30	1.00	TOMS BUPMED
1-01-002-05	>100mmetu SPDSTKR >100mmetu/HR OFSK	13.0 A 5.00 A	30.0		1.00	9.00	TONS BURNED TONS BURNED
1-01-002-04	10-100HRSTU PULMT 10-100HRSTU PULDT	13.0 A 17.0 A 8.00 A	38.0	30.0	0.30	1,00	TONS BURNED TONS BURNED TONS BURNED
1-0:-002-06			38.0	15.0	0.30	2,00	Tons BUPHED
	10-100MMBTU UFSTK <10MMBTU OFBTDEER	5.00 A	30.0	15.0	1.00	2.00	TONS BURNED TONS BURNED TONS BURNED
1-01-002-11	CLOMMBTU UPSTOKER	2.00 A	30.0	4.00	3.00	10,0	TOMS BURNED
	CIOMMBTU PULY-DRT OTHER/NOT CLASIFO	2.00 4 2.00 4 17.0 4 14.0 A	38.7	10.0	3.00 3.00 0.30 0.30	8.20	TONS BURNED TONS BURNED
LIGHITE			•••	•	*****	*****	
1-01-003-01	>1004#810 PULTET	4.50 A	30.0	13.0	0.30	1.00	TONS BURNED
1-01-00)-07	>100=MBTU PULYDRY >100=MRTU CYCLONE	6.80 A	30.0	13.0	8.30	1.00	TONS BURNED TONS BURNED
1-01-003-04	>100=#RTU OF STER	4.50 4	30.0	13.0	0.30 0.30	8,00	TONS BURNED
1-01-003-75	STORMSTU UF STER	4.80 4	30.0	13.0	0.30 0.30		TONS BUPNED TONS BUPNED
1-0:-003-57	10-100M=BTU DTPUL	4.50 4	30.0	13.0	0,30	1.00	TONS BURNED
1-01-003-08	10-100MHSTU WTPUL 10-100MHSTU OFSTR	4.50 A	30.0	13.0	0.30	1.00	TONS BURNED TONS BURNED
1-01-001-10	10-100-07: 1207-	4 80 4	30.0	13.0	1.00	2.00	TONS BURNED
1-01-003-11	CLORESTUSPOSTE	4.50 4	30.0	13.0	1.0n 3.00	18,0	TORS BURNED
1-01-003-13	CIOMMSTU OF STORR	4.50 4	30.0	13.0	3.00	10.0	TONS BURNED
1-01-003-14	ID-JODHNSTUSPDSTK CLORRSTU PULY DRY CLORRSTU DY STOKR CLORRSTU UF STOKR CLORRSTU UF STOKR CLORRSTU SPOSTOKR	6.80 A	30.0	13+0 13+0	3.00	10.0	TONS BURNED TONS SURMED
RESIDUAL DIL							
	>100mm87U/HR 6EHL	A . 00	187.		2.00	3.00	LODGEALLONS BURNED
	10-100MM8TU/HRENL <10MM8TU/HR EEHL	8,00 8.00	187.	105. 105.	2.00	3.00	loodeallows burnto
MISTILLATE DIL			*				
1-01-005-01		8.00	144, 1	106.	2.00	3.00	198964LLONS BURNES
1-01-006-03	D= COMMSTU/HRENL	8.00	144.	1 105. 1 105.	2.00 2.00	3.00	1000GALLOWS BURNED
HATURAL GAS	«IDPROTU/MR EE4L	A.00	1774	. 1055	2.00	*****	1980GALLONS OURNED
	>100MRRTU/HP	10.0	0.40	●00•	1.00	17.0	MILLION CUBIC FEET BURNED
1-01-004-07	10-100MHSTU/HR	10.0	n.40	£30.	1 + On	17.0	WILLION COBIC FEET BURNED
##01-004-03	<1098\$TU/HR	10.0	0.40	120.	1.00	17.0	MILLION CURIC PEET SURNED
1-01-007-01	>100MM87U/HR 10+100MM87U/HR	18.0		600. 6 230.	1.00	17.n 17.n	MILLION CUBIC FEET BURNED
	CID WESTU/HR	14.0	950.	120.	1.70	17.0	MILLION CUBIC PEET BURNED
CORF							
	>100mmBTU/H ^a	17.0 4	38.0	18+0	0.03	1+00	TORS BURNED
##00/84#E WASTE						_	•
1-0:-009-0:	ACCONDANC BOILES	75.0 27.5	1.50	10.0	2.00 2.00	2.00 2.00	TONS BURNED
	MOOD BOILER	10.0	1.50	10.0	6.00		TONS BURNED
0A6488E			_	A			
1=0 =0 =0 =0 =0 =0	DIGOMMETU/HR	22.0 22.0	o.	3.00	2.00 2.00	8.00	TOMS BURNED
1-01-011-03	10-100MHBTU/HR 410MHBTU/HR	27.0	0,	3.00	1.00	2.00	TONS BURNED TONS BURNED
8LD WASTE-SPECT	F1						•
	PIGG MARTU/HR						TONS SURNED
1-01-017-02	0-100 HR6TU/HR						TORS BURNED
							TONS BURNED

ENVIRONMENTAL	SECTION	NEDS	SECTION 3	CHAPTER 7	SUBJECT 0
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				C D P E R			
ERCORD BOILER WELE	CTRIC BENBRATH	PART	301	MOX	#C	C 0	U N 1 T S
LIG WASTE-SPECIFT							
01 \$0-610-10-1 01 \$0-610-10-1 01> \$0-610-10-1	100 ##8TU/#8						1000 EVITORS SAMED 1000 EVITORS SAMED 1000 EVITORS PADMED
STHER/HOT CLASIFD							
1-01-999-97 BP(1-01-999-98 SP(1-01-999-99 SP(ICIPY IN REMARK						MILLION CUBIC PEET BURNED 1800 SALLON ([18018] BURNED TONS BURNED (SOLID)
	DUST# AL			,			
AMTHRACITE COAL							
1-82-881-03 16- 1-82-881-04 16- 1-82-881-88 16- 1-82-881-88 41- 1-82-881-87 41-	SOMMSTU/NR ŠTER -100MMŠTU PULVO -100MMŠTU STER OMMŠTU/NR PULVO OMMŠTU/NR STER	17.0 A 8.00 A 17.0 A 2.00 A 2.00 A	30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$	18-0 10-5 16-0 15-5 18-0 4-00 3-00	0.03 6.26 0.03 0.20 0.20 0.30 2.50	1000 1000 1000 1000 1000 1000	TOMS BURNED TOMS BURNED TOMS BURNED TOMS BURNED TOMS RUNNED TOMS BURNED TOMS BURNED TOMS BURNED TOMS BURNED
BITURINGUS COAL							
[-0]=00]=03 > 1; 1-02=00]=04 1 1-02=00]=05 10 1-02=00]=07 10 1-02=00]=07 10 1-02=00]=08 10 1-02=00]=17 41; 1-02=00]=17 41; 1-02=00]=14 41; 1-02=00]=14 41;	OGMENTY PULYDRY DORNEYY CFCLOWE DOWNEYY UPSTE - CORNETY UPSTE	13.0 A 17.0 A 17.0 A 18.00 A 13.0 A 5.00 A 13.0 A 17.0 A	30.0 S	30.0 18.0 18.0 18.0 18.0 18.0 18.0 18.0 4.00 4.00 4.00 4.00 4.00	0.36 0.30 1.80 1.80 1.80 1.80 1.80 1.80 1.80 3.00 3.00 3.00	1.00 1.00 1.00 2.00 2.00 1.00 1.00 1.00	Tome outree
LIGHITE							
109:003:003 109:003:00 109:003:00 109:003:00 109:003:00 109:003:00 109:003:00 109:003:00 109:003:01 109:003:01 109:003:03 109:003:03 109:003:03 109:003:03 109:003:03 109:003:03 109:003:03 109:003:03 109:003:03 109:003:03	DOMESTU PULTWET DOMESTU PULTORY DOMESTU PULTORY DOMESTU PETTE DOMESTU PETTE 1:00mestu PETTE 1:	6.50 A	20.0 9 20.0 9 20	13.0 13.0 13.0 13.0 13.0 13.0 13.0 13.0	0.30 0.30 0.30 1.00 1.00 1.00 0.30 0.30 0.30 1.00	1 · · · · · · · · · · · · · · · · · · ·	TOMS SURMED
PEGIOUAL OIL							
1-01-004-03 10-	DONNSTU/NR - 1 ODNNSTU/NR - NNVUYSMMO	83.0	187. 8 187. 8 187. 8	40.0 40.0 40.0	3.00 3.00 3.00	9:00 9:00	1000 GALLENŽ ŽUPNED 1000 GALLENZ BURNED 1000 GALLONS BURNED
019714446 014							
1-68-008-08 10-	9949tu/HR •19949tu/HR 946tu/HR	18.0	149. 8 142. 8 148. 8	40.0 40.0 40.0	3.00 3.00 3.00	9.00 9.00	1000 GALLONS BURNED 1000 GALLONS BURNED
MATURAL CAS							
1-02-004-63 (9:	00mm8TU/MR - 00mm8TU/MR	10.0	0.60 0.60	60D. 230. 120.	3.00 3.00 3.00	17.0 17.0	MILLION CUBIC PEET BURNED MILLION CUBIC PEET BURNED MILLION CUBIC PPET BURNED

1 - 0 2 - 0 0 7 - 0 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	PINERY DIOD PINERY 10-100 PINERY 4:0 AST PMC DIOD AST PMC 10-100 AST PMC 4:0 CONTENT, 'S' INDICA	ITES THE SUL	rua Camtem	. BP TWE PUEL	ON A PERCENT	84518	RILLION CUBIC PET BURNES RILLION CUBIC PET BURNES BY WEIGHT!
				0-2			

ENVIRONMENTAL PROTECTION AGENCY	SECTION	NEDS	SECTION 3	CHAPTER 7	SUBJECT 0
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	POUNDS				- •
EXTCOMS BOILER - INDUSTRIAL	Part	801		+c co	U N 1 T S
PROCESS 645 CONTINUED					
1-01-007-07 COEE DYEN >100 1-01-007-08 COEE DYEN 10-100 1-01-007-09 COEE DYEN 410 1-01-007-99 OTHER/MOT CLASTED	,				MILLION CUBIC PEET BURNED MILLION CUBIC PEET BURNED MILLION CUBIC PEET BURNED MILLION CUBIC PEET BURNED
COKE					
1-02-008-02 10-100PH8TU/HR 1-02-008-03 <10PH8TU/HR	2.00 A	38.0 S	15.0 0	.20 2:no	TOMS BURNED Toms burned
WODD/BARK WASTE					
1-05-004-03 MOOD BOILES 1-03-004-01 MOOD\BPEK BOILES 1-05-004-01 BYNE BOILER	75.0 37.5 10.0	1.80	10.0	.00 2.00 .00 2.00	TONS BURNED Tons Burned Tons Burned
LIS PETROLEUM 645					
1-02-010-02 10-100##8TU/#R 1-02-010-03 410##8TU/#R	1.75	86.5 5	11.7 0	.30 1.88	FOODEWITCHS STEMED
9464950					
1-02-011-01 >100 MASTU/HR 1-02-011-02 10-100F=8TU/HR 1-02-011-03 410MBSTU/HR	22.0 22.0	0. 0.	2.00 2	.00 2.00	TOMS BURNED Toms burned Toms burned
BLD WASTE-SPECIFY					
1-02-012-01 >100 ===TU/MR 1-02-012-02 100-100 MM8TU/MR 1-02-012-03 <10 MM8TU/MR					TOMS BURNED TOMS BURNED TOMS BURNED
LIG WASTE-SPECIFY					
1-02-013-01 >100 MH8TU/HR 1-02-013-03 (10 MH8TU/HR 1-02-013-03 (10 MH8TU/HR					1000 GALLONS BURNED 1000 GALLONS SURNED
STHER/HOT CLESIFD					
1-02-999-97 SPECIFY IN REMAR 1-02-999-98 SPECIFY IN REMAR 1-02-999-99 SPECIFY IN REMAR	τ				MILLION CUBIC PEET SURVED 1000 BALLON BURNED (LIGUID) TONS BURNED (BOLID)
ESTCORB BOILER -COMMERCL-INSTUTM					
ANTHRACITE COAL					
1-03-001-05 10-100mM8TU PULW 1-03-001-06 10-100mM8TU PULD 1-03-001-07 10-100mM8TU PULD 1-03-001-08 (100m8TU PULV)12 1-03-001-07 (100m8TU STOEE 1-03-001-17 (100m8TU STOEE 1-03-001-17 (100m8TU STOEE 1-03-001-19 0TMER/M0T CLASIF	V 17.0 A E 13.0 A D 17.0 A	38.0 \$ 38.0 \$ 38.0 \$ 38.0 \$ 38.0 \$ 38.0 \$ 38.0 \$	4.00 0 19.0	.00	TOMS BURNED TOMS RURNED
ALTURINOUS COAL					
1-03-002-05 10-100m+07U PULW 1-03-007-06 10-100m+07U PULD 1-03-007-07 10-100m+07U PTT 1-03-007-08 10-100m+07U PTT 1-03-007-07 10-100m+07U PTT 1-03-007-17 10-100m-07U PTTORF 1-03-007-12 (10m+07U PTTORF 1-03-007-14 (10m+07U PTTORF) 1-03-007-17 070m+07U PTTORF	Y 67.0 A K 6.00 A K 6.00 A K 9.00 A R 30.0 A R 200 A R 200 A	30.7 9 30.0 5 30.0 5 30.0 5 30.0 5 30.0 5 30.0 5 30.0 5 30.0 5 30.0 5 30.0 5	18.0 E	1.00 .00 .00 .00 .00 .00 .00 .00	TONS BUPNED TONS BURNED
FIGHISE					
-03-003-05 10-100##87U PULW -03-003-06 10-100##87U PULD -03-003-07 10-100##87U PULD -03-003-07 10-100##87U PTFT -03-003-07 10-100##87U PULY-08 -03-003-12 (10##87U PTFTORE -03-003-12 (10##87U PTFTORE -03-003-13 (10##87U PTFTORE -03-003-14 (10##87U PTFTORE	7	30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ 30.0 \$ \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ 30.0 \$ \$ \$ 30.0 \$ \$ 30.0 \$ \$ \$ \$ 30.0 \$ \$ \$ \$ 30.0 \$ \$ \$ \$ \$ 30.0 \$ \$ \$ \$ \$ \$ 30.0 \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$	13.0 1 13.0 1 13.0 1 13.0 1 13.0 1 13.0 1 13.0 1	.00 3.00	TONS BURNED

	T	SECTION	NEDS				SECTIO	N	CHAPTER	SUBJECT
ENVIRONMENT	AL	22071011	.020				3		7.	0
PROTECTION AGI	ENCY	CHAPTER	Source	e (Classifi	cation	DATE		PA	GE
NATIONAL AI DATA BRANCI		SUBJECT	Codes		nd Emiss	sion	1/3/	76	•	4
VOLUME V AEROS MANUAL OF	CODES									
EXTCOMO BOTLER	-COMPERCL-1#5TUT	PART	D S E H \$0x		TPD PER	4 C 4 C	co		U W 1 T S	
PESIDUAL OIL							-• -			
1-03-004-03 1-03-004-03	>10-160mm8TU/MR 10-160mm8TU/MR <10mm8TU/MR	23.0 23.0 23.0	157. 157. 167.	;	40.0 40.0	3.00 3.00	4.00	1000	GTTTOMS BANGE GTTTOMS BANGE GTTTOMS BANGE	
D1511LL478										
1-03-008-03	>1000MBTU/HR 0-100MMBTU/HR < 0MMBTU/HB	16.0 16.0	192.	;	40.0 40.0	3.00 3.00	9.00	1000	EVITOME WARMED EVITOME SAUNED EVITOME WARMED	
MATURAL 619										
1-03-004-02	>100mm91U/MR 0+100mm87U/MR <10mm87U/MR	10.0	0.40 0.40	1	#30. 20. 80.0	8.00 8.00	10.0 10.0	MELL	ION CUBIC FEET BE ION CUBIC FEET BE ION CUBIC FEET BE	JRMED
PROCESS 645										
1-03-007-02	SERVEE 10-100 SERVEE 10-100 SERVEE 10-100	(R						FILL	ION CUBIC PEET BI ION CUBIC PEET BI ION CUBIC PEET BI ION CUBIC PEET BI	JRMEO JRMEO
#000/849£ ##\$TE										
1-03-009-02	MOOD BOILER MOOD/BISK BOILER	78.0 37.8 10.0	1.80		10.0 10.0 10.0	2.09 2.00 6.00	\$.00 \$.00	TONS	BURNED BURNED	
LIG PETROLEUR 6	49									
	10-100mm8TU/H# <10mm8Tu/H#	1 • 85 1 • 85	84.6		*.*0 *.60	0.75			GALLONS BURNED GALLONS BURNED	
SLO WASTE-SPECI	* 7									
1-03-012-02	PIOD MMGTU/MR 10-100 MMBTU/MR <10 MMBTU/MR							TONS	BURNED BURNED	
LIQ WASTE-SPECI	• •									
1-03-013-02	>100 MMSTU/MR 10+100 MMSTU/MR <10 MMSTU/MR							1000	EVITORS BRANCO EVITORS BRANCO	
OTHER/NOT CLASS										
1-03-999-98 1-03-999-99	SPECIPY IN REMAI	PK						1000	ION CUBIC PEET BE BALLON BURNEO (BURNED (BOLTO)	
FXTCOMB BOILES	-SPACE MEATER	••								
140UST#14L										
-05-001-02 -05-001-01 -05-001-05 -05-001-05 -06-001-06 -05-001-0 -05-001-97 -05-001-99 -05-001-99	PESIONAL DIL DISTILLATE DIL MATURAL GAS LIG PETROLEUM GO DIMER-SPECIPY OTMER-SPECIPY OTMER-SPECIPY	49						TOMS TOMS 1000 1000 FILL 1000 TOMS 1000	SURNED SURNED SURNED SALLOWS SURNED SALLOWS SURNED SON COME FEET S SALLOWS SURNED SALLOWS SURNED SALLOWS SURNED SALLOWS SURNED SON COMET FEET S	
[+05+n02+n]	ML AMTHRACITE COAL							••	. Bulance	
1-08-002-07 1-08-002-03 1-08-002-03 1-08-002-03 1-08-002-13 1-08-002-13 1-08-002-13 1-08-002-03	BITURINAUS CAN	••						TOMS TOMS 1000 1000 FILL 1000 TOMS 1000	i Burneo Burneo Burneo Gallong Burneo	

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NATIONAL AI DATA BRANCI		CHAPTER	Codes	ar	Classifi nd Emiss		1/3/		
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147E44FC048A2210M	ELECTRIC GENERAT	PART	0 \$ E M		7	- U N 1 T	co	, , , , , , ,	
DISTILLATE DIL									
2=01=001=01 2=01=001=02 44TUR4L 645	TURBINE RECIPROCATING	****	140.	1	67.8	\$,57	1514	1000 GALLONS BURNED	
2-01-002-01 2-01-002-01	TURBINE RECIPROCATING	14.0	94D.	•	413+	42.0	118.	MILLION CUBIC PEET MILLION CUBIC PEET	
DIESEL									
2-01-003-02	RECIPROCATING TURBINE	13.0 5.00	140.	;	370. 47.8	37.0 5.57	225. 15.4	THOUSANDS OF EALLONS IDNO EALLONS BURNED	
MESIQUAL DIL	****							-	
2-01-004-01	IONBINE		159.	8				1000 GALLONS BURNED	
JET FUEL	7.105.1 MP								
2-01-005-01	1049146		4.2					1000 EALLOWS BURNED	
CRUBE OIL				_					
2-01-004-01	TORBINE		194+	•				1000 GALLONS BURNED	
PROCESS 645									
2-01-007-01			*\$0.	\$				MILLION CUBIC FEET	
OTHER/HOT CLAS!		_						B11 1 100 PUBLIC POTO DUI	
3-01-444-48	SPECIFY IN REMAR	r r						MILLION CUBIC FEET BUI 1000 BALLONS BURNED	
14164#FGWBA3110#		•							
DISTILLATE OIL									
2-02-001-01 2-02-001-01	TURBINE RECIPROCATING	\$.00 33.5	190.	:	47.8 949.	\$ i \$ 7 37 : \$	15.4 102.	1000 SALLOWS BURNED	
MATURAL 645									
	TURBINE RECIPROCATING	14.0	**0.	,	413+	42.0	t 1 5 .	MILLION CUBIC PEET MILLION CUBIC PEET	
#ASOLINE				_				1000 0000 000 00000	
	RECIPROCATING	4.50	6.3	0	102.	161.	3,040.	1000 GALLOWS BURNED	
######################################	RECIPROCATING	33.5	144,	,	447.	37.6	102.	1000 GALLOWS BURNED	
1-03-004-05		5.00	190.	•	67.8	5.57	15.4	1000 GALLONS BURNED	
MESTOVAL DIL			18*.					1000 GALLONS BURNED	
2-02-005-01	ANSTHE		104.	•				1000 STEEDES BORNED	
JET FUEL			4.7	n				1000 GALLOWS BURNED	
2-02-006-01	1949 46		•••	•					
COURT OIL			144.	•				1000 GALLOWS SURWED	
2+07+007+01	1985 75		1701	•					
PROCESS 649 2-01-008-01	TURRINE		980.	,				MILLION CUBIC FEET	
2-02-008-02	BECIPHOCATING		*\$0.					MILLION CUBIC PEET BUR	460

OTHER/MOT CLASIFO

2-02-000-07 SPECIFY IN REMARK 2-02-000-08 SPECIFY IN REMARK MALLION CUBIC FEET BURNED 1000 GALLONS BURNED

I ENVIDONMENTAL :		NEDS			SECTIO	N. CHAPTER	SUBJECT
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NATIONAL AIR DATA BRANCH	CHAPTER	Codes as		fication ssion	DATE		GE 6
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AEROS MAROALO: 33333	7004	0 5 6 4 1 7	7		1		
147E4#FCOMBASTIOM -COMMERCF-1#810.	THL	401	HON	46	co	U N 1 T ¶	
DIESEL							
2-03-001-01 RECIPROCATING STMES/MOT CLASIFD	33.6	199. \$	•••.	27.5	102.	THOUSANDS OF GALLONS	
2mb3mpppmpp SPECIFY IN REMI 2mb3mpppmpp SPECIFY IN REMI						TILLION CUBIC PEET BE	NBMED
INTERMLECORBUSTION -ENGINE TESTING	•••						
AIRCRAFT							
73L08RUT 10-100-40-5	11.0	13.0	19+4	46.0	32.7	THOUSANDS OF GALLONS	ruei
ROCKET MOTOR							
2-04-002-01 SOLID PROPELLA	HT.					TORS OF FUEL	
STHER/HOT CLASIFS							
9-04-444-44 BPECIPY IN REN 2-04-444-44 BPECIPY IN REN 2-04-444-44 BPECIPY IN REN	ARK					MILLION CUBIC PERT BY 1000 GALLONS BURNED TONS BURNED	JAMED
INDUSTRIAL PROCES -CHEMICAL MPG	•••						
ADIPIC ACID PROD							
3-01-001-01 GEMERAL-CYCLOM		٠.	12+0	•:	n ;	TONS PRODUCED	
AMMONIA W/METHNTR							
3-01-003-01 PUREZ 618 3-01-003-03 STOP16E/L040[W	0. 0.	o.	7.	*0.0		TONS PRODUCED	
AMMONIA W/COADSRB	-						
3-01-003-01 REGEMERATOR EX		0.	٥.	٥.	200.	-	
3-01-003-03 PURRE 445 3-01-003-03 STOPARE/LOADIN 3-01-003-09 OTHER/NOT CLAS	0. 1F0	0.	0.	*0.0		TOMS PRODUCED TOMS PRODUCED TOMS PRODUCED	
AMMONIUM MITMATE							
2-01-004-09 DTHER/NOT CLAS CARBON BLACE	170	0.				TONS PRODUCED	
3-01-005-01 CHENNEL PROCES	\$ 2,300.	٥.	٥.	11.500.	33,800.	TONS PRODUCED	
\$-01-00%-02 THERMAL PROCES \$-01-00%-03 FURMACE PROC	9, 0,	0.	ů.	1,000.	\$,300,0	TONE PRODUCED	
3-01-005-09 PURMACE PROC 0	11.			400.	4,800.	TONS PRODUCED	
3-01-008-00 OTHER/HOT CLAS	70					TONS PRODUCT	
3-01-006-01 PTREL/018T1L/6							
3-01-004-TH OTHER/HOT CLAS	FO 400.			100.	320,	TONS PRODUCED TONS PRODUCT	
CHLORINE							
3-01-007-01 GENERAL 3-01-007-070 OTMER/NOT CLAS	1100	٠.				TONS PRODUCED	
CHLOR-SLEAL!							
3-01-008-01 L1QUIFTH-01APH 3-01-008-03 L1QUIFTH-01APH 3-01-008-03 L0AN-04-4	CEL	o. o.				100 TOMS CHLOSINE LI	QUEF1ED
3-01-008-03 LOADING THECON 3-01-008-08 LOADING THECON	VHT 0.	?: e:	2.	0.	D +	TOO TONS CHLORINE LE	guffitt
Sections 44 DIMESTHUL CFT	1142 0.	0.	٠.	0.	n•	100 TOMS CHEOSINE LI	
CLEANING CHEMICUS							
3-01-009-99 07HERS/NOT CLE 3-01-009-10 SPECIALTY CLES	978 90.0 HRS 870	0.				TONS PRODUCED TONS PRODUCT TONS PRODUCED	

A' INDICATES THE ASH CONTENT, 'S' INDICATES THE SULFUR CONTENT OF THE FUEL ON A PERCENT BASIS (BY MEIGHT)

	SECTION	NEDS			\$ECTIO	N	CHAPTER	SUBJECT
ENVIRONMENTAL PROTECTION AGENCY					3		7	0
NATIONAL AIR DATA BRANCH		Source Codes a	nd Emis		DATE 1/3/7	6	PA	GE 7
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	7 0 U H						- ,	
IMBUSTRIAL PROCES -CMENICAL MF6	PART	901	#0 x	#¢	c•		U # 1 T B	
EEPLOS; YES-THT								
3-01-010-01 MITHATION REG 3-01-010-02 MNO3 CONCTR		e. e.	140.	0.			PRODUCED PRODUCED	
3-01-010-03 M2504 REGENE 3-01-010-04 RED WATER INC	RATE O.	18.0	2.00	9;	ŏ.	-	PRODUCED	
S-DI-DID-OF BELLITE ERHAU	\$1 0.	0.70	0.	0.		TONS	PRODUCED	
3-01-010-99 BTMER/MOT CLA' MTDROCHLORIC ACID	170					TORS	PRODUCED	
3-01-011-0; \$TPRODUCTW/09:	CRUS	٥.				TONS	FINAL ACTO	
3-DI-DII-03 BTPRODUCT W/S		٠.					PINAL ACID	
MTDROFLUORIC ACID								
3-01-012-03 #072771Lww/05	RUSE D. CRUS D.						AC10 AC10	
3-01-012-03 GRIMO/DRY PLU 3-01-012-99 GYMER/MOT CLA	03PR 200.					PONS	PLUBR Š PAR AE1D	
MITRIC ACID								
3-0 -0 -0			62.5				PURE 4010 PROPU	
S-01-013-03 WITACO CONCTR TOMOS OSATIM PO-C10-10-6	0.0		8.00			TONS	PURE ACID PRODUC	eto Eto
3-01-013-08 UNCONTROLLED 3-01-013-84 W/C4TYL/COMBU						70#S	PURE ACID PRODUC	689
S-01-DIS-07 UMCONTROLLED S-01-DIS-08 W/ASSORRERS S-01-DIS-TP OTMER/NOT CLA						7048	PURE ACID PRODUCTION ACID PROD	660
941H7 HP4	3170					1848	AAME BEID AMOOD	680
S-DI-DIR-DI GENERAL	2.00			30.0			PRODUCED	
3-DI-DIR-DR PIERCHT KILM 3-DI-DIR-PP BTMER/HDT CLA	940						PROBUCT	
WARRISH HPG								
3-01-015-02 BLEORESIMOUS	SCHL D.			40.0 1 4 0.		Poug	PRODUCED	
3-01-018-03 ALKTO GENERAL 3-01-018-05 ACRYLIC GENERAL	8. 8.			10.0		TONS	PRODUCED PRODUCED	
\$=0 =DIS=99 GTHER/HOT CLA PHOS=ACID WETPROC	370					1983	PRODUCED	
3-01-016-01 MEACTOR-UNEON	110 0.						PHOSPHATE BOCK	
3-01-014-03 GTPSUM POMD 3-01-014-03 COMDEMSUNCO						TONS	PHOSPHATE BOCK	
3-01-916-99 GTHER/HOT CLA	970					Tetts	PRODUCED	
9008-4610 PMERMAL 3-01-017-01 GENERAL						TONS	PROSPHOROUS SUS	ieo
\$-DI-DIT-99 DTHER/HOT CLA	3 70					TOMS	PRODUCED	
, PLASTICS	25.0						PRODUCED	
\$=01=018=01	RAL 3.00					Paus	PRODUCED PRODUCE	
3-01-016-99 07MER/MOT CLA							PROBUCED	
PHYNALIC ANNYDRID				38.0			••••	
\$-01-019-03 UMCOHTSOLLED- Printing ing	****			3410		TONY	PRODUCED	
3-01-020-0: CODE 1 M4-46 MER	4L 0.			180.		Tons	PRODUCED	
3-0;-020-03	Esta D.			150,	,	Tens	PRODUCED PRODUCED	
3-01-020-09 EGGETME-ALETO 3-01-020-08 PIGMENT MISIN	5 0. 468* 2.00			100.	1	7049 7049	PRODUCED Planent	
3-01-020-79 STHER/MOT CLA SOCIUM CARBONATE	•••				,	7 (P) #1 (B)	PRODUCED	
Secions to 1 me and a second s	CYRY G.					Tans	PROBUCED	
3-01-021-03 96, var-manb(1)	HG 6.00 HG					Tong Tong	PRODUCT	
Secientii Promiedater Secientii Promiedater Secientii Promiedater Secientii Promiedater	HERL				•	Pens	PRODUCED PRODUCED PRODUCED	

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IMBUSTRIAL PROCES -CHEMICAL MF6	PART	90 n	NOX	# C	CO UNIT\$

M2504 -CHAMBER					
3-01-022-01 GENERAL				٥.	TOMS PURE ACTO PRODUCES
H2904-CONTACT					
3-01-023-01 44.7 COMYERSION	2.50	•.00			TONS PURE ACID PRODUCED
3-01-023-04 99.8 CONVERSION	2.50	7.00			TOWS PURE ACID PRODUCED TOWS PURE ACID PRODUCED
3-01-023-08 98.0 CONVERSION	2.50	27.0			TONS PURE ACID PRODUCED
3-01-023-10 97.0 CONVERSION	2.50	40.0			TONS PURE ACID PRODUCES
3-01-023-12 04.0 CGMYERSION	2.50	85.0			TONS PURE ACTO PRODUCED
3-01-023-14 95.0 CONVERSION 3-01-023-16 94.0 CONVERSION	2.50	70.0			TOMS PURE ACTO PRODUCED
3-01-023-18 93.0 COMVERSION	2.50	94.0			TONS PURE ACID PRODUCED
3-DI,-023-PP OTHER/HOT CLASFD					TOMS PRODUCED
STHTHETIC FIBERS					
3-01-024-01 WTLON GENERAL 3-01-024-02 DACRON GENERAL				7.00	TOMS FIBER TOMS FIBER
3-01-024-03 ORLOH				۰.	TONS PRODUCT
3-01-024-09 ELASTIC					TONS PRODUCT
3-01-034-04 TEPLON 3-01-034-04 POLTESTER					
3-01-034-08 MOMER					TONS PRODUCT Tons Product
3-01-024-10 ACRYLIC					TONS PRODUCT
3-0;-024-12 TYYEY					TONS PRODUCT
3-01-024-14 OLEFINS 3-01-024-44 DIMERS/NOT CLASFO					TONS PRODUCT Tons Produced
					1044 14000000
SENISTHTHICFIBR					
3-01-035-01 RATON GEMERAL				۰.	TONS FIRER
3-01-028-08 ACETATE				••	TONS PRODUCED
, 3-01-025-10 VISCOST 3-01-025-99 DIMERS/NOT CLASFO					TONS PRODUCED
					TONS PRODUCED
STHTHETIC RUBBER					
3-01-024-01 BUTADIENE-GENERAL					TONS PRODUCT
3-01-024-03 RETHTLPROPENE-GHL					TONS PRODUCT
3-D1-024-03 BUTTHE BEHERAL					TORS PRODUCT
3-01-034-04 PENTADIENE-GENEL 3-01-034-08 DINETHMEPTHE GENE					TONS PRODUCT
3-01-086-06 PENTAME-GEMERAL					TONS PRODUCT TONS PRODUCT
3-01-084-07 ETMAMEMITRILE-GEN					TOMS PRODUCT
3-01-034-08 ACRYLONITRILE-6EN 3-01-036-09 ACROLEIN-6ENERAL					TOMS PRODUCT
Jediedža.20 auto Tibts grups:					TONS PRODUCT
3-01-034-99 OTHER/HOT CLASED					TOWS PRODUCT
PERTILIT ADDNNITE					
3-01-027-01 PRILTWR-MEUTRLIZE	٥.		0.		TONS PRODUCED
3-01-027-02 PRILING TOWER 3-01-027-03 PRILING TOWER	12.0		••		TONS PRODUCED TONS PRODUCED
3-01-031-D4 GEVENTTI-MENITIE	٥.		0. 0.		TONS PRODUCED
3-01-087-05 GRANULATON 3-01-087-06 GRANULAT-ORTCOOLS	0.40		0.48		TONS PRODUCED
	7.00		3.00		Tans PRODUCED
FERTILIS-MSUPPHOS					
3-01-028-01 GRIND-DRY	7.00				
3-01-020-02 MAIN STACK	0.				TONS PRODUCED TONS PRODUCED
PERTILIE-TEPSPHOS					7043 74000000
3-01-029-01 RUM OF FILE	٥.				TOWS PRODUCED
3-01-034-03 GREMATE	٥.				TONS PRODUCED
PERTILIZ-DIAMPHOS					
9-01-030-01 DRYER-COOLERS					
3-01-030-02 AMONIAT-GRANULATE 3-01-030-00 GTMER/MOT CLASIFD	90.D				TOMB PRODUCED
3-01-030-09 STHER/HOT CLASIFD	4,00				TONS PRODUCED TONS PRODUCED
TEREPTHALIC ACID					
\$=\$1=\$11=\$1					
3-01-031-01 MMO3-PARAETLENGEN 3-01-031-99 OTHER/MOT CLASIFO			13.0		TOMS PRODUCED
					TONS- PRODUCED
SULFUR (ELEMENTAL)					
3-01-039-01 HED-CLAUS 337465		***			
		200.			TOMS PRODUCT TOMS PRODUCT
3-01-033-04, 03MEE/MD4 CF9214D 3-01-033-02 #00-CF978 48196E		144.			TONS PRODUCT
GAMESTAND CEASING					TONS PRODUCT

[&]quot;A" INDICATES THE ASH CONTENT, "S" INDICATES THE SULFUR CONTENT OF THE FUEL ON A PERCENT SASIS IDT WEIGHT:

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IMPUSTRIAL PROCES -CHEMICAL MPG	PART	\$0x	MOT	нс	co		U = 1 T S	
PESTICIPES								
3-01-033-01 MALATHION 3-01-033-04 OTHER/NOT CLASIFD ARINES/ARIDES							SHS OF PRODUCT PRODUCED	
3-01-034-01 GEMERAL/OTHER						10HS	PRODUCT	
P[442HT-140864H								
3-01-036-01 CALCIMATION 3-01-036-99 OTMER/NOT CLASIFO							OF PRODUCT	
SODIUM SULPATE								
3-01-036-01 GENERAL/OTHER 3-01-036-02 KILMS							PRODUCT PRODUCT	
SODIUM BULFITE								
3-01-037-01 S ÉMERÁL/OTHER 3-01-037-02 KILMS							PRODUCT PRODUCT	
SODIUM RICARS								
3-01-038-01 GEMERAL						1045	PRODUCT	
LITHIUM HTDROXIDF								
3+01-039-01 GEMEPAL						1045	PRODUCT	
PERTILITER UREA								
3-01-090-01 GENERAL						TONS	PRODUCT	
MITPOCELLULOSE								
3-01-041-01 REACTOR POTS 3-01-041-02 MESO CONCENTRYS 3-01-041-04 BOILING TUBS 3-01-041-94 DTMER/MOT CLASIFO	o. o.	1.30 45.0 0.	21+0 24+0 2+00	0	0: 0: 0:	TONS	PRODUCED PRODUCED PRODUCED PRODUCED	
admes i ves								
3-DI-020-DI GEMP\COMBND UNKAN						TONS	PRODUCT	
ACETATE PLAKE								
3+01+090-99						TOMS	PRODUCT	
ACETONE								
3-01-091-01 OTHER/HOT CLASED						***	PRODUCT	
MALEIC AMMYDRIDE								
3-01-100-01 GENERAL/OTHER						TORS	PRODUCT	
BOTAINT BABITION								
3-01-101-01 GENERAL/OTHER						TORS	PRODUCT	
SULPONIC ACTO/ATS								
3-01-110-01 GEMERAL/OTHER						TOMS	PRODUCT	
ASSESTOS CHEMICAL				•				
3-0j-ili-0; Cauleing 3-0j-ili-03 Stalants 3-0j-ili-03 Stalants 3-0j-ili-04 Fire Proof by 3-0j-ili-04 Others/Not Clased		0. 0. 0.	9. 0. 5.	0,	ñ;	TONS TONS TONS	PRODUCT PRODUCT PRODUCT PRODUCT PRODUCT	
FORMAL DEHTOE								
3-0;-120-0; SILVER CATALTST						TOHS	PRODUCT PRODUCT	
ETHTLEME DICHLEDE								
3-01-126-01 GITCHLOSINATION						TONS TONS	PRODUCT PRODUCT	
AMMONIUM SULFATE								
3-01-130-01 MM3-M2504 PROCES 3-01-130-03 CORE OVEN 57-PROD 3-01-130-03 CAPROLCTM ST-PROD						7045	PRODUCT PRODUCT	
AT THREE PART AND ARM COMPANY, FRE THE								

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				7 P D P E R			
14803181AL PROCES	-CHEH!CAL MF4	PART	50 z	40 X	**	C9	U # 1 9 S
WASTE GAS FLARE	9						
3-51-700-00 3-51-700-00	OTHER/ROT CLASIFS						#ILLION CUBIC FEET BURNED
3-01-99009	SPECIFY IN BEMARE						TONS PRODUCT
IMDUSTRIAL PROCES	-F008/4681CULTURAL						
ALFALFA DENTORA	TH						
3-02-001-01	GEMERAL BTHER/MOV CLASFD	40.0					TOMS MEAL PRODUCED
COFFEE ROADTING							
	DIRECTFIRE ROLSTA	7.40		0.10			TONS GREEN BEAMS
3-02-002-02 3-02-002-03	INDIACTPIREROASTA STONER/COOLER OTHER/NOT CLASPD	1.90		0.10			tomb green beams Tomb green beams Tomb product
COFFEE-1#874H7							
3-02-003+01	BPRAY DRIER	1.90		8.			Tene Green Beams
COTTON 6144146				-			
3-02-004-01	UNLOADING PAN	5.00	0.	6.	9;		BALES COTTON
3-02-004-03 3-02-004-03 3-02-004-03	CLEAMER BTICK/BURR MACHME OTHER/MOT CLASED	3.00	0,	0.	0.	0.	BALES COTTON BALES COTTON BALES COTTON
PEED/GRAIN TERM	PL						
3-02-005-01	************	1.00	٠,	9.	e.		TONO GRAIN PROCESSES
3-03-00\$-03 3-03-00\$-03 3-03-00\$-04	Transfer/Conveyns Bereening/Cleanns Drins	8.00 6.00	o.	0 •	0.	0.	TOMB GRAIN PROCESSED TOMB GRAIN PROCESSED TOMS GRAIN PROCESSED
PEED/GRAIN CHTR	78						
3-02-804-01	SHIPMS/RECEIVE		٠,	0.	0.	0.	TONS CHAIN PROCESSED
3-02-004-02	PRESPER/CONVEYNG	3.00	0.	8.	0,	0.	TORS GRAIN PROCESSED
3-02-004-01 3-02-004-01 3-02-004-01	BCREENING/CLEANNG DTYLNG DTMER/HDT CLABIPD	7.00	0.	٥.	0.	,	TONS AMAIN PROCESSED TONS ARAIN PROCESSED TONG ARAIN PROCESSED
GRAIN PROCESSIN	•						
3-02-007-01	CORN MEAL						TOMP SMAIN PROCESSED
3-02-007-02 3-02-007-03	STATELIANETICESM	7 • 00 0 • 20					TOME GRAIN PROCESSED TOME GRAIN PROCESSED
\$-0\$-007+04	MILO CLEAMER	9.95				,	TORE GRAIN PROCESSED
3-03-007-08 3-02-007-06	BARLETPLOUR HILL BET CORN PILLING	3.00	٠.			,	TOUS GRAIN PROCESSED TOUS OF PRODUCT
3-01-007-30	WHEAT PLOUR MILL		0.			1	TONS PRODUCT
PEED MANUFACTUR	97489/469 ELABPO					,	TORE PROCEESED
3-03-008-99	SARLET PEED-GENL STHER/HAT CLASPD	3.00					TOMO GRAIN PROCESORD TOMO PROCESSED
PERMENTATH-SEE!	1						
1-02-509-61 10-00-609-61	SHAIN MANDLING	3.00			o:		TONS SRAIM PROCESSED
8-02-004-03	DRYING SPUT GRAIN	8.00					TONS GRAIN PROCESSED PROUBLED OF GALLONG
3-02-004-48	OTHER/HOT CLASFD OTHER/HOT CLASFD						EALLONS PRODUCT TORS GRAIN PROCESSED
FER # 2 # 7 6 7 4 . WW] {							THE STATE PROCESSES
3-02-010-01	SHAIN MANDLING	3.00			0:	1	TONS ERAIN PROCESSED
	DRTING SPAT GRAIN	5.00 G.			10:0		TOPE STAIN PROCESSED
3-03-010-00	OTHER/NOT CLASED	- *			·		BALLONS PRODUCT
FEBRENTATHONINE							
3-03-011-01	erntaal	٥.			0.	•	BALLOMS PRODUCT
FISH NEAL							
3-63-613-61	COORERS-PRESHP18H COORERS-STALEF18H	٥.					IONS TION MEAL PRODUCED
30030013403	0.01285	0. 0.10				1	TORS FISH WEAL PROBUCED
347 E37431941 '4'	OTHER/HOT CLASIFO ASH CONTENT, 'S' INDI		PUB (6=+P=	AF THE PURI	OH 4 8192940		

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IMBUSTRIAL PROCES -FOOD/AGE	9497	0 8 E # 1 T T 202	E D P E R	U N 1 T	co	U # 1 T B
************	*******					
MEAT SHORING						
3-07-013-01 GENERAL Starch RFS	0.30			6.07	0:40 70	MS MEAT SMORED
3-01-014-01 GENERAL	.00				te	MS STARCH PRODUCED
BATTA CTHE SHOCES						
3-02-015-01 GENERAL 3-02-015-99 OTHER/NO	* CLASIFO					NS SUGAR PRODUCED NS PROCESSED
BUGAR BEET PROCES						
3-02-014-01 BRYER OM 5-02-014-07 BRYER/NO						NS RAW BEETS NS RAW BEETS
PEAMUT PROCESSING						
3-02-017-20 01L/m07 3-02-017-99 07MER/m0						NS PRODUCT NS PROCESSED
CAMBY/COMPECTMRT						
3-02-016-99 07HER/NO	7 CLASFO				70	NS PRODUCT
BAIRT PRODUCTS						
3-02-030-0: MILE SPR 3-02-030-99 STHER/HO						NS PRODUCT Ng PRODUCT
STHER/HOT CLASIFO						
3-02-777-78 BPECIFT 3-02-77-77 BPECIFY						NS PROCESSED (NPUT) NS PRODUCED (PINISMED)
INDUSTRIAL PROCES -PRIMARY						
ALUMINUM DRE-BAUT						
3-03-000-01 CRUSHING	/HAMDLING 4.00				70	HS OF BRE
AL ORE-ELECRORED						
3-03-001-01 PREBARE	CELLS 81.3					NE ALUMINUM PRODUCED
	SODERSA6 ***					MB ALUMINUM PRODUCED
	SODERSERS 78.4 S MANDLING 10.0					NB ALUMINUM PRODUCED
3-03-001-08 AMODE BA	EE PURHEE 3.00				76	ME ALUMINUM PRODUCED
3-03-001-44 6 7#ER/HO	7 CLABFD				76	MS ALUMINUM PRODUCED
AL BRE-CALC ALMID						
3-03-007-01 GEMERAL	200.				70	MR TEMBERS SHOPPICED
COKE MET STPRODUC						
3-03-003-01 GEHERAL	3.50	4.00	0.04	4.20		MS COAL CHARGED
3-03-003-01 BACH CHY		0.01	0.03	2.50	0.60 TO	MS COAL CHARGED
3-03-003-03 0464 Pug 3-03-003-04 QUE4CH1H				*****		MS COAL CHARGED
3-03-003-05 UML0401H					76	MS COAL CHARGED
3-83-803-06 UMDERFIR	146	4.00			Y	NS COAL CHARGED NS COAL CHARGED
3-03-003-07 COAL CRU 3-03-003-99 GTMER/HO	T CLASFD				10	MS COAL CHARGED
CORE MET-BECHIVE						
3-03-004-01 BENERAL	200.	0.	0.	8.00	lino Te	MS COAL CHARGED
COPPER SHELTER						
3-03-005-01 78746/48		1.250.				NS CONCENTRATED ORE
3-03-005-02 R04571H4		95.0 920.				NS CONCENTRATED ORE
3-83-008-03 84ELT!#4 3-83-008-04 68#¥ERT!		870.			70	NS CONCENTRATED ORE
3-03-005-05 REFINING	16.0	0.			#10	HS CONCENTRATED DOF
3-03-005-06 BRE DATE					70	NS OF ORE NS PRODUCED
3-03-005-08 FINISH 0 3-03-005-79 07HER/HO	T CLASED				76	MR CONCENTATION ONE
PERALLOT OPEN PRE						
3-03-004-0 BDs FEST	200.				10	MS PRODUCED
3-03-004-02 78s 78s;	319. 569.				19	NS PRODUCED
3-03-004-03 FCB FEST 3-03-004-04 BILICON	METAL 426.					MS PRODUCED
3-03-004-05 BILICOMA	HALMESE 196.				70	MS PRODUCED
"A" INDICATES THE ASH CONTE	INT, 'B' INDICATES THE	SULFUR CONTENT	OP THE PUEL	. OR A PERCENT	94518 (B	* WE16HT)

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INDUSTRIAL PROCES		PART	50 x	401	#C		U N ! T \$
FERROALLOY	CONTINUED						
	ORE DRYER LOUCARD CROREACTR STHER/HOT CLASFO			0.			TOMS PROCESSED TOMS PROCESSED TOMS PROCESSED TOMS PRODUCED
PERALOT SEMCOMP	HC						
3-03-007-01 3-03-007-02	FEROMANGAMESE General	45.0					TOMS PRODUCED
IRON PRODUCTION	1						
10-400-c0-c \$0-800-c0-c \$0-800-c0-c \$0-900-c0-c	BLAST FNC-ORECHE BLAST FNC-AGLCHE Simyering Beneral Ore-Crush/Handle	121. 44.0 42.0	o.	o. o.	o:	1,750. 99.0	TONS PRODUCED TONS PRODUCED TONS PRODUCED TONS OF ORE
3-03-008-08 3-03-008-06 3-03-008-07 3-03-008-08 3-03-008-99	SCARFING SAND HANDLING DPN HOLD OVENS SLAG CRUSM/HANDL	1.00	0.	0.	0.	0.	TOMS PROCESSED TOMS MANDLED TOMS SAND BAKED TOMS MANDLED TOMS PRODUCED
STEEL PRODUCTE	ON						
3-03-00*-02 3-03-009-03 3-03-009-03 3-03-009-03 3-03-009-11 3-03-009-12 3-03-009-20 3-03-009-20	BOF-SERERAL ELECT ARC W/LANCE ELECT ARC WOLANCE FINISM/PICKLIME FINISM/SOAK PITS FINISM/SAIMD,ETC FINISM/GRIND,ETC FINISM/OTHER	17.4 8.30 51.0 11.0 9.20				13T. 18•n 18•0	TOMS PRODUCED
LEAD SHELTERS							
3-03-010-01 3-03-010-02 3-03-010-03 3-03-010-04 3-03-010-05 3-03-010-09	BLAST FURNACE REVERS FURNACE ORE CRUSHING MATERIALS MANDLING	164. 278. 15.4 2.00 5.00	423. 34.7 0. 0.	0. 0. 0.	0.	0.	TOMS CONCENTRATED ONE TOWS CONCENTRATED ONE TOWS CONCENTRATED ONE TOWS OF LEAD PRODUCT TOWS CONCENTRATED ONE
#0L780€#U#							
3-03-01(-0) 3-03-01(-0) 3-03-01(-9)	MILLING-GENERAL			0.			MUNDREDS OF TOMS MINED TOMS PRODUCT TOMS PROCESSED
TITANIUM PROCE	95						
3-03-012-01	CHEORINATION STAT OTHER/MOT CLASIFD		σ.	0.	0.		TONS PRODUCT TONS PROCESSED
6010							
3-03-013-01	# 1 m 1 m 6 / PROCE 9 5 ! N 6				ο.		TONS OFE
9481UH							
3-03-014-01 3-03-014-03 3-03-014-03 3-03-014-44	REDUCTH RILH			0.			TONS PROCESSED TONS PROCESSED TONS PROCESSED TONS PROCESSED
BERTLLIUM ORE							
3-03-015-01 3-03-015-01 3-03-015-01	CRUSHING HELTINE		o.	ð. G.	0. 0.	n: 0•	TONS OF ORE TONS PROCESSED TOMS PROCESSED
3-03-014-04 3-03-014-08 3-03-015-06 3-03-015-07	STINDING SULFATION/DISSOLV		0.	0. 0.	0, 0, 0,	0.	TOMS PROCESSED TOMS PROCESSED TOMS PROCESSED TOMS PROCESSED
3-03-016-44	LEACH/FILTER OTHER/NOT CLASPD		e.	0 •	0.	0.	TONS PROCESSED TONS PROCESSED TONS PROCESSED
MERCURY RINING							
3-03-025-02 3-03-025-03	SURFACE GLASTING SURFACE ORILLING SURFACE HAMDLING MATURAL VAPOR STRIPPING	0.	0. 0.	0. 0. 0.	0.	0. 0.	TONS OF ORE TONS OF ORE TONS OF ORE TONS REMOVED
3-03-025-07 3-03-028-64	COMPET/HEULING UNLOSDING SSH CONTENT, 'S' IND	ICATES THE !	O. N. N. BULFUR CONTEN	0. 0. 0. 1 OF THE FUE	0. 0. L DN & P.#(0. 0. 2. FHT RASIS	TONS OF ORE TONS OF ORE TONS OF ORE (BY WEIGHT)

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				T T E D P E R			•	
IMPUSTFIAL PROCES		P4#1	\$0x	MOX	ĦC	co	U H 1 T S	
MERCUPY HINING	CONTINUED							
	CONVINAUL WASTE OTHER/HOT CLASFO CS	•	n.	0.	٥.		is of ore	
3-93-024-01	CRUSHING BOTART FURNACE		0.	••	0.		S PROCESSED	
	RETORT PURNACE		٥.	٠.	0	0: 10	S PROCESSED	
3-03-024-05	BURNT DEE BIN HOEING PROCESS		D.	ů. 0.	0.	0. TO	IS PROCESSED	
3-01-024-**	OTHER/HOT CLASF					70	S PROCESSED	
ZINC SMCLTING								
3-03-030-03 3-03-030-04 3-03-030-05 3-03-030-04	ROASTHE/HULTOHR	90.0 8.00 100. 00 3.00	1,100.		0.	10 10 10 10 10	PROCESSED PROCESSED PROCESSED PROCESSED PROCESSED PROCESSED PROCESSED	
OTHER/HOT CLASE	D							
3-03-*****	SPECIFY IN REMA	*c				70	IS PRODUCED	
INDUSTRIAL PROCES								
ALUMINUM OPERAT	N							
3-04-00 -02 3-04-00 -02 3-04-00 -04 3-04-00 -10 3-04-00 -11 3-04-00 -20 3-04-00 -20	SWEATINGPURNACE SMELT-CRUCIRLE SMELT-REVERS FM CHLORINATM STAT FOIL ROLLING FOIL CONVERTING CAN MANUFACTURE ROLL-DRAW-EXTRU OTMER/MOT CLASE	1.90 C 4.30 H 17.5	υ,	٥.	o:	70 70 70 70 70 70 70	PRODUCED SHETAL PRODUCED SHETAL PRODUCED SHETAL PRODUCED SPRODUCED PRODUCED PRODUCED PRODUCED PRODUCED	
BRISS/BROWZ FEL	•							
3-04-002-03 3-04-002-04 3-04-002-05 3-04-002-04	CRUCIBLE PHO CUPOLA FHO ELECT INDUCTION REVERB PHO	70.0 60.0				70 70 70 70 70	S CHARGE S CHARGE CHARGE S CHARGE S CHARGE S CHARGE S CHARGE S CHARGE S CHARGE	
GRAT IRON								
3-04-003-03 3-04-003-05 3-04-003-40 3-04-003-60	CUPOLA REYERS PHC ELECT INDUCTION ANNELLING DPERA HISC CAST-FARCT BRINDING-CLEMI SAND MANDL-SEML DTMER/NOT CLASI	T# N M6	σ.	٥.	o:	0: TO: 0: TO: TO: 0: TO:	MS METAL CHARGE MS METAL CHARGE MS METAL CHARGE MS METAL CHARGE MS PROCESSED MS PROCESSED MS MANDLED MS MANDLED MS METAL CHARGE	
LEAD SHELT SEC								
3-p004-02 3-04-004-03 3-04-004-03 3-04-004-03	POT PURMACE REVERS PMC BLAST/CUPOLA PM RCTARY REVERS P LEAD OXIDE MP6 OTMER/MOT CLASI	MC 70.0	0. 60.0 53.0 0.	0. 0. 0.	0:	0. To	IS HETAL CHARGED IS PROCESSED	
LEAD BATTERT								
10-200-P0-6 10-200-P0-6 P0-200-P0-6	TOTAL-SEMERAL CASTIME FURNACE PASTE RISER THREE PROCES OP OTHER/HOT CLASS	0.21 0.44	0. 0. 0.	0. 0. 0.	0; 0; 0;	0. TO	IS OF BATTERIES PROUS OF BATTERIES PROUS OF BATTERIES PROUS OF BATTERIES PROCESSED	00000
MAGNESIUM SEC								
	POT PUPHACE OTHER/HDT CLASI	***				70 70	NS PROCESSED NS PROCESSED	

^{&#}x27;A' INDICATES THE ASH CONTENT, 'S' INDICATES THE SULFUR CONTENT OF THE PUEL ON A PURCENT BASIS (OT MEIGHT)

	SECTION	NEDS	SECTION	CHAPTER	SUBJECT
ENVIRONMENTAL PROTECTION AGENCY			3	7	0
	CHAPTER	Source Classification	DATE	PAGE	
NATIONAL AIR DATA BRANCH		Codes and Emission Factors	1/3/76]	4
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IMPUSTRIAL PROCES -	8EC0mDary metals		30x	401	HC	co	U M 1 T S
STEEL FOUNDRY							
3-09-007-02	ELECTRIC ARC FHC OPEN HEARTH FMC	13.0		0.20			TOMS PROCESSED TOMS PROCESSED
3-04-007-04	OPEN MEARTH LANCO MEAT-TREAT PHE	10.0		۰.			TONS PROCESSED
3-04-007-04	SAND GRIND/HANDL	0.10	۰.	0.	o:		TONS PROCESSED TONS HANDLED
3-04-007-19	FIRISH/SOAK PITS FIRISH/HOT CLASFO OTHER/HOT CLASIFO						TONS PROCESSED TONS PROCESSED TONS PROCESSED
ZINC SEC							
3-04-008-02	RETORT PHC HORIZ MUPPLE PHC POT FURHACE RETTLE-SWEAT PHC	47.0 48.0 0.10 11.0					TOMS PRODUCED TOMS PRODUCED TOMS PRODUCED TOMS PRODUCED
3-04-001-05	GALVANIZING KETTL CALCINING KILN	\$.00 \$1.0					TONS PRODUCED
3-04-008-07	CONCENTRATE DRYER REVERB-SWEAT FAC OTHER/HOT CLASIFO	13.0					TONS PROCESSED TONS PRODUCED TONS PROCESSED
MALLEABLE 1804							
3-04-004-44	ARREALING OPERATH OTHER/HOT CLASIFD						TONS METAL CHARGE TONS METAL CHARGE
WICKEL							
	PLUE PURNACE DIMER/HOT CLASIFD						TONS PROCESSED
SINCONIUN	B4152 #11 W						**** ********
3-04-011-01	OTHER/HOT CLASIFD						TOMS PROCESSED TOMS PROCESSED
FURNACE ELECTROD	•						
3-04-010-01	HIZING		٥.	0.	o;	0:	TONS PROCESSED TONS PROCESSED
3-04-020-04	PITCH TREATING BARE FURNACES OTHER/MOT CLASIFD		٥.	••			TONS PROCESSED TONS PROCESSED TONS PROCESSED
#19C C49T6F48RCT	*						
	SPECIFY IN PEHARK						TOMS PRODUCED
OTHER/HOT CLASIF							
	SPECIFY IN REMARK						TONS PROCESSED
INDUSTRIAL PROCES -	######################################						
ASPHALT RODFING							
3-08-001-01	BLOWING OPERATION DIPPING ONLY	2.50			1.50		TONE SATURATED PELT PRODUCED TONS SATURATED PELT PRODUCED
3-05-001-03	SPRATING ONLY DIPPING/SPRATING OTHER/NOT CLASIFD	3.00			0.	٠.	TONS SATURATED PELT PRODUCED TONS SATURATED PELT PRODUCED TONS SATURATED PELT PRODUCED
ASPMALTIC CONCRE							
3-05-002-01 3-05-002-01	ROTARY DRYER OTHER SOURCES OTHER/ROT CLASIFO	10.0	7.	0.	0.	0.	TONS PRODUCED TONS PRODUCED TONS PRODUCED
BRICE MANUFACTUR							
1-05-281-61	DRTIME-DAW MYL	70.0		0.			TORS PRODUCED
3-03-003-03	STORAGE PAN HTL CURING BAS PINED	74.0		0.	0.00		TOMS PRODUCED
3-06-003-06 3-06-003-04	CURING OIL FIRED CURING COAL FIRED OTHER/MOT CLASIFD	0.07 0.07 1.30 A	0.02 6.00 \$ 7.60 \$	0.29 1.40 1.10	0.03 0.10 0.70	0.	TONS PRODUCED TONS PRODUCED TONS PRODUCED
CATCINE CUBBIOE							TONS PRODUCED
3-09-004-01	ELECTRIC PHE	30.0	3.00				TONS PRODUCED
3-08-004-03	CORE DATER PHC ROOM VENTS OTHER/NOT CLASIFD	2.00	0.				TONS PRODUCED TONS PRODUCED TONS PROCESSED

^{&#}x27;a' impirates the ask content, 's' indicates the sulpus content of the fuel on a percent sasis (of melont) $3.7.0{-}14$

ENVIRONMENTAL	SECTION	NEDS	SECTION	CHAPTER	SUBJECT
PROTECTION AGENCY	01145===	_	3		
NATIONAL AIR DATA BRANCH		Source Classification Codes and Emission Factors	1/3/76	PAGE 15	
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		P 0 U W 0 9					
	MINERAL PRODUCTS	PART	\$0 x	MOX	HC	60	U N 1 T S
-	14						
3-05-005-01	BEWHATL DRYER	30.0					TONS FEED HATERIAL
3-09-008-03	REWMATL CRUSH/PRC	120.					TONS FEED MATERIAL
3-05-008-03	ELECTRIC ARC MELT	50.0					TOWS PEED MATERIAL
	CURING DYEN MOLD/SHAKEOUT	0.20					TONS FEED MATERIAL TOWS FEED MATERIAL
	OTHER/HOT CLASIFD						TONS PEED MATERIAL
CEMENT RF6 DRY							
3-06-004-01	KILHS DRYERS/GRINDERETC	96.0 18.0	3.00	0.50			BARRELS CENENT PRODUCED
3-06-004-03	KILMS-OIL FIRED	246.	14.4	2.40	0;	n;	
3-06-004-04	KILMS-649 FIRED	245.	10.2	2.40	ŏ;	e.	TONS CEMENT PRODUCED
3-05-004-05	BILMS-COAL FIRED OTHER/MOT CLASIFD	296.	23.4	2.40	ь.	٥.	TONS CEMENT PRODUCED
CEMENT MFG WET							
3-05-007-01		43.0	3,00	0.50	o:	۰.	BARRELS CEMENT PRODUCED
3-06-007-02	DRYERS/GRINDERETC	6.00					BARRELS CENENT PRODUCED
3-05-007-03	KILMS-DIL FIRED KILMS GAS FIRED	226. 226.	10.2	2.40	٥;	٠,	TONS CEMENT PRODUCED
	EILMS-COAL FIRED	120.	23.0	3.60	0.	0	TONS CEMENT PRODUCED
	OTHER/NOT CLASIFO		•••		••	••	TONS CEMENT PRODUCED
CERAMIC/CLAT MP	•						
3-08-008-01	DRYING	70.0					TOMS INPUT TO PROCESS
3-05-008-02	SRINDING STORAGE	76.0 39.0					TOMS IMPUT TO PROCESS TOMS IMPUT TO PROCESS
3-06-008-44	STHER/HOT CLASIFD	34.0					TONS PRODUCED
CLAY/FLYASHSINT	EA						
3-05-009-01	FLTASH	110.					TORS FINISHED PRODUCT
3-05-009-02		55.0					TONS FINISHED PRODUCT
	DIMER/HOT CLASIFO	24.0					TONS FINISHED PRODUCT
COAL CLEAMING							
3-05-010-01	THERMIFLUID BED	20.0					TONS COAL DRIED
	THERM/PLASH	14.0					TORS COAL DRIED
3-05-010-03	THERM/MULTILOUVED BTHER/NOT CLASIFO	26.0					TOMS COAL CLEAMED
CONCRETE BATCH!	H6						
3-06-011-01		0.20					CUBIC TARDS CONCRETE PRODUCE
	ASSEST/CEMMT POTS	0.20	٥.	٥.	0;	0 4	TONS PRODUCT
1-06-011-21	#010 SUMPICE OTHER/HOT CLISPD		٥.	8.	, 6:	8.	TOMS PRODUCT
F19086LAS1 HF4							
1-84-619-81	AZYEBFHC-REGEHEX	3.00					TONS MATERIAL PROCESSED
3-06-013-03	REVERBENC-RECUPES	1.00					TONS MATERIAL PROCESSED
3-09-012-03		٠.					TONE HATERIAL PROCESSED
	PORMING LINE	50.0					TOMS MATERIAL PROCESSED
3-05-012-05	CURING DYEN OTHER/HOT CLASIFD	7.00					TONS MATERIAL PROCESSED
F#17 MP4							
	ROTARY FMC GENL	14.0					TONS CHARGE TONS CHARGED
3-05-013-77 6LASS NF4	OTHER/NOT CLASIFO						Suckato
							2000 di 400 da
3-05-014-01	SODALINE SEML FMC	2.00					TOMS GLASS PRODUCED Toms Processed
	847CH146/M[X]46		٥.	٥.	o:	n.	TONS PROCESSED
3-06-014-12	HOLTEN HOLD TARKS OTHER/HOT CLASIFD		٥.				TOMS PROCESSED TOMS PRODUCED
5-05-01 67PSUR #F6	ATHENNAN CPARTAD						
							••••
	RW WIL DRIER PRIMARY GRINDER	10.0					TONS THROUGHPUT PORS THROUGHPUT
	CALCINER CHINDER	P0.0					TORS THROUGHPUT
3-05-018-04	CONVETING	0.70					TONS THROUGHPUT
							······································
LINE MFG					o:	٥.	TONS PROCESSED
	PRIMARY CRUBUING	31.0	9.	٥.			
3-05-014-01	PRIMARY CRUBHING	\$1.0 \$.00	0. 0.	0.	ŏ.		TONS PROCESSED
3-04-014-01 3-05-014-02 3-05-014-03		2.00 9.00	0.	0.	٥.	٠.	TONS PROCESSED

ENVIRONMENTAL PROTECTION AGENCY	SECTION	NEDS	SECTION 3	CHAPTER 7	SUBJECT 0
NATIONAL AIR DATA BRANCH	CHAPTER	Source Classification Codes and Emission Factors	DATE 1/3/76	PA	GE L6
VOLUME V. AEROS MANUAL OF CODES	20D/ECI				

							· · · · · · · · · · · · · · · · · · ·	-
-INDUSTRIAL PROCES	-MINERAL PRODUCTS	PART	301	403 403	V # 1 T	¢0	U N 1 T S	
***************************************	***************************************							
LIME MFG	CONTINUED							
3-05-016-05	CALCIMATIC KILM	200.					TONS PROCESSED	
3-08-014-04	PLUIDIZO RED KILH HYDRATOR						TONS PROCESSED LINE PRODUCED	
3-08-014-99	OTHER/NOT CLASIFD						TOMS PROCESSED	
MIMERYL AGGL								
3-05-017-01 3-05-017-02	REVERS FHC	22.0 5.00	0.03				TOWS CHARGE TOWS CHARGE	
3-09-017-03	STOR CHTHSES	17.0					TONS CHARGE TONS CHARGE	
3-05-017-05	COOLER OTHER/HOT CLASIFO	3.00					TONS CHARACTONS PROCESSED	
PERLITE HFE								
	VERTICAL PHC SEN OTHER/NOT CLASIFO	21.0					TOMS CHARGE TOMS PROCESSED	
PHOSPHATE ROCK	0,464,40, 66451,0							
3-05-014-01	DRT1H6	18.0					TONS PHOSPHATE ROCK	
3-05-014-02	GR[HD]H6 TRAMSPER/STORAGE	30.0					TONS PHOSPHATE ROCK TONS PHOSPHATE ROCK	
3-05-014-04	OPEN STORAGE OTHER/MOT CLASIFO	40.0					TONS PHOSPHATE ROCE TONS PROCESSED	
STONE QUART/PRO								
	PRIMARY CRUSHING	0.50	٥.	0.	0.		TONS RAW MATERIAL	
3-08-020-03	SEC CRUSH/SCREEN TERT CRUSH/SCREEN	4.00	0.	0.	0,	0,	TONS RAW MATERIAL	
3-05-020-04 3-05-020-08	RECRUSH/SCREENING FINES MILL	6.00 6.00	o.	0 •	ě.	0.	TONS RAW MATERIAL TONS RAW MATERIAL	
3-05-020-04 3-05-020-07	SCREEN/COMYT/MHDL OPEN STORAGE	2.00 10.0	o.	0 • 0 •	0	0+	TONS PRODUCT TONS PRODUCT STORED	
3-05-020-08	CUT STONE-SEMERAL BLASTING-SEMERAL		0.	0.	0.	0.	TONS PROCESSED TONS PROCESSED	
	OTHER/HOT CLASIFD		••	••	••	••	TOMS PROCESSED	
SALT PINING								
3-05-021-01			۰.				TONS MINED	
POTASH PRODUCT:								
3-08-022-99	MINE-GRIND/DRY OTHER/NOT CLASIFO		••				TOMS ORE TOMS PROCESSED	
CALCIUM BORATE								
3-05-023-01 3-05-023-44	MINIME/PROCESSING OTHER/HOT CLASIFD				0.		TORS PRODUCT TORS PROCESSED	
ME CARBONATE								
3-05-024-01	MINE/PROCESS OTHER/HOT CLASIFO				0.		TONS PRODUCT	
\$440/4844EL								
3-05-025-01	CRUSHING/SCREEN OTHER/HOT CLASIFD	0.10	0.	0.	0.	•:	TONS PRODUCT	
01470M4C0U9ER7								
3-08-026-01	MANDLING OTHER/NOT CLASIFD		۰.	0.	0.	0.	TORS PRODUCT TORS PROCESSED	
CERANIC ELECT A	***							
	STHEP/HOT CLASIFO						TOMS PROCESSED	
ASSESTOS MINING								
3-02-03[-03	SURFACE BLASTING		o.	0.	o: o.	0.	TONS OF ORE	
3-08-031-03	LOSDING		0.	0.	0, 0,	0.	TONS OF ORE	
3-05-03 -06 3-05-03 -04	CONVET/HAUL MARTS		9.	6 •	0.	0.	TORS OF ORE	
3-06-031-02	UNLOSOING STRIPPING		٥.	0.	٥,	0.	TOMS OF DRE	
3-08-031-09	VENTILATION STOCEPILING		o. 0.	0.	0,	0.	TOMS REMOYED	
3-06-031-11	TAILINGS OTHER/HOT CLASED		o.	0.	0.	01	TIME OF MATERIAL	
							TONS PROCESSED	

	SECTION	NEDC			SECTION	CHAPTER	SUBJECT
ENVIRONMENTAL	J.C.I.O.	'AETO			3	7	0
PROTECTION AGENCY	CHARTER	C	G3 ! C		DATE		GE
NATIONAL AIR DATA BRANCH	SUBJECT	Codes a	and Emis		1/3/76		17
VOLUME V. AEROS MANUAL OF CODES	*OBJECT] -, -,		
	POUN		7 E D P E !				
impustrial proces omineral prof		\$6x	MOI	MC	co	U N İ, T S	
ASBESTOS MILLING					_		
3-05-037-01 CRUSHING 3-05-031-02 ORTING 3-05-037-03 RECRUSHING 3-05-037-05 FIRETIZING 3-05-032-05 FIRETIZING 3-05-032-07 OTHER/MOT CO		0. 0. 0.	0. 0. 0.	0.	0. TO 0. TO 0. TO 0. TO	MS PROCESSED	
MIM!MG-9PEC MATL		•		•	a. •	S OF MATERIAL	
3-08-090-01 OPEN PIT-BL 3-08-090-02 OPEN PIT-DL 3-08-090-03 OPEN PIT-CO 3-08-090-02 UNDERSED-SE 3-08-090-22 CONVET/MAUL 3-08-090-23 UNLOCIME 3-08-090-23 UNLOCIME 3-08-090-23 STREPPIME 3-08-090-25 STREPPIME 3-08-090-25 STREPPIME 3-08-090-35 OPENIMEN CONCENT 3-08-090-37 OPENIMEN CONCENT 3-08-090-38 ORE CONCENT 3-08-090-34 SCREENIME 3-08-090-34 TALLIME PIME 3-08-090-39 CAMER/MOT C	TLL ING BORNE HTTLAT HATL WASTE SHER RUSHER RATOR	0. 0. 0. 0. 0. 0. 0. 0.	0. 0. 0. 0. 0. 0. 0. 0.		0. Tol 0. Tol	TO OF MATERIAL	
STMER/HOT CLASIFO							
3-06-99-99 SPECIFY IN	RCMARK				70	19 PRODUCT	
impustrial proces -petroleum i							
PROCESS MEATER							
3-04-001-01 01L 3-04-001-02 645	0.02	0.43 \$	2,900.	0+03	0. 10 0, 10	DO BARRELS DIL BUR DO CUBIC FEET GAS	HED BUBHED
3-04-001-03 01L 3-04-001-04 448 PLUID CRACEERS	20.0	140. 9	230.	30.0	0, 10	DO BALLONS OIL SUB- LLION CUBIC PERT B	
3-04-002-01 GENERAL IFC	c) 242.	493.	71+0	220.	13,700. 10	OO BARRELS PRESH P	EED
HOVORED CATOCRACE							
3-04-003-01 GEHERAL (TC	c) 17.0	40.0	6.00	87.0	3,800. 10	00 BARRELS PRESH P	EEO
GLOW-DOWN STOTM							
3-04-004-01 A\COM180F3	,	o. o.	0. 0.	\$.00	0. 10 0. 10	DO BARRELS REFINER DO BARRELS REFINER	CAPACITY CAPACITY
PROCESS BRAIMS	٥.	٥.	٠.	8.00	0. 10	DO BARRELS WASTE W	
446AA4 418 80-80-8 403 048 439 80-80-8 446344 439 80-80-80-8		š:	•:	210.	ŏ. iŏ	DO BARRELS WASTE W	ATER
9-09-007-05 ALG COMINDE 9-09-008-01 ALCOMINDE	0.	0. 0.	0. 0.	130.	0. 10 0. 10	OO BYBEEFS AVCARM I	DISTILLATION DISTILLATION
COOLING TOWERS							
3-04-607-01	۰.	0.	••	4.00	0. 41	LLION BALLONS COOL	STAM DE
#18CEFF THEORS		_			-1 -		
3-04-008-01 PIPE/VALVE- 3-04-008-02 VESL RELIEF 3-04-008-07 PUMP SELEC 3-04-008-09 CEMPRESM SE 3-04-008-08 OTHER-SENL	AVER 0.	0. 0. 0.	0. 0. 0.	28.0 11.0 17.0 5.00 10.0	0. 10 0. 10	OO BARRELS REPINER DOGBARRELS REPINER DOGBARRELS REPINER DO BARRELS REPINER DO BARRELS REPINER	Y CAPACITY Y CAPACITY
PLARES							
3-04-007-07 MATURAL 448 3-04-007-77 GTMER/MOT C			٥٠			LLIONS OF CUBIC FE LLIONS OF CUBIC FE	
SLUDGE CONVERTER					•-	W4 \$8000000	
3-04-010-01 6 EHERAL					**	MS PROCESSED	

^{*}A* IMPICATES THE ASH CONTENT, *S* INDICATES THE SULFUR CONTENT OF THE FUEL ON A PERCENT BASIS (BY MEIGHT)

	,	SECTION	NEDS			SECTION	CHAPTER	SUBJECT
ENVIRONMEN	TAL					i 3	7	•0
PROTECTION AG	ENCY	CHAPTER	Source (Classificat	tion	DATE	PA	GE
NATIONAL A	IR I			nd Emission				
DATA BRANC	CH .	SUBJECT	Factors			1/3/76	18	3
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AEROS MANUAL O	F CODES							
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Impustatel PROCES		PART	90 t	W01	**	co	U 4 1 7 8	
***********	••••••	•						
ASPHALT EXTOISE						_		
3-04-01;-01 3-04-01;-99 Fluid Coxime	BEMERAL BYMER/MBT CLASIF	70					PROCESSED PROCESSED	
3-04-012-01		\$23.					-	
3-04-012-03	COOLING OPER TRANSPORTATION					1000	BARRELS PRESH PE BARRELS PRESH PE	ED
2-96-012-09 Catalitic Refo						1000	BARGELS PRESH FE	
3+04-013-01						1000	BARRELS PRESH FE	
OTHER/MOT CLASS							544665 FAESA FE	
	SPECIFY IN BEHAL	te.				Tave	PROCESSED	
	SPECIFY IN REMAI						EL9-PROCESSED	
INDUSTRIAL PROCES		•						
SULPATE PULPHS								
3-07-00;-01 3-07-00;-02	BLOWTHE ACCUMULT	'R 0.	o.			0. A1#-	DRY TONS UNBLEACH DRY TONS UNBLEACH	ED PULP
3-07-001-03	MULT-EFFECT EVAL		0. 4.00			0. 418*	DRY TOMB UMBLEACH	ID PULP
3-07-001-08 3-07-001-04	SMELT DISSOLV TO	1K 2.00	0.			0. 418-	DRY TOWS UMBLEACH DRY TOWS UMBLEACH	ID PULP
3-07-001-07 3-07-001-08	TURPENTINE CONDS	SR 0.	0.			0. 418-	DRY TONS UNBLEACH	10 Pulp
3-07-001-09	LIQUOR OFIDE TO	/R	••			0. AIR-	DRY TONS UNBLEACH DRY TONS UNBLEACH DRY TONS UNBLEACH	D PULP
BULFITE PULPING		•				-,		
	LIQUOR RECOVERY					4:0-	DRY TONS UNBLEACH	ED PULP
	BULFITE TOWER				۵:	AI#-	DRY TONS UMBLEACH	to Pulp
1-07-001-04	SHELT TANK EVAPORATORS				0	418-	DRY TONS UNBLEACH ORY TONS UNBLEACH	ED PULP
1-07-001-04	PULP DIGESTER OTHER/HOT CLASS				ě.	TONS	AIR DRY PULP	10 7007
	0.454,401 66111	•					TIP DAY POUR	
	P49E980480-4EN	0.				tons	FINISHED PRODUCT	
3-07-004-02	FIBERBOARO-4EH GTHER/NO? CLAS!	0.40				TOMS	PINISHED PRODUCT	
		•					7177766 7766661	
3-07-008-01	C#205072					***	OF WOOD T#EATED	
	STHEM/NOT CLASS	70					OF WOOD TREATED	
TALL011/R0914	****							
3-87-884-01 PLYW005/P18786						TONS	or Product	
	AND VEHEED DATER							
3-77-007-02	SAMBING SAMBING TEMERO BATER	o. 	o. o.	0.	0.	O. TOMS	PROCESSED	
SAUMILL OPERAT		•				TONS	PROCESSED	
	OTHER/NOT CLASS	73				•	PROCESSEO	
TICELSION MF4						1081	ACE 8340	
3-07-004-44	OTHER/HOT CLASS	70				****	PROCESSED	
CORE PROCESSIA						TORY	AFERICO	
3-07-010-44	07#24/H07 CL4514	70				TANE	_ PROCESSEO	
PURMITURE MPG						. 989		
3-07-020-49	OTHER/HOT CLASS	70				Tema	PROCESSED	
47W40/max 21 40								

3-67-999-99 SPECIFY IN BEHARE

TONS PROCESSED

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ENVIRONMENTAL					3	7	0
PROTECTION AGENCY NATIONAL AIR	CHAPTER	_	Classific nd Emissi		DATE		GE
DATA BRANCH		Factors		-A:	1/3/76	19	9
VOLUME V. AEROS MANUAL OF CODES	SUBJECT						
						- ,	
IMPUSTRIAL PROCES OMETAL PARRIC		\$01	401	HC	60	U N 1 T S	
IRON/STEEL							
3-09-001-01 MISC MARDWAL 3-09-001-02 FARM MACHINI 3-09-001-99 OTHER/MOT CL	T R Y	0.	0. 0.		701	S OF PRODUCT S OF PRODUCT S PROCESSED	
PLATING OPERATORS							
3-07-010-99 OTHER/NOT C	LASIFD				701	IS PLATED	
CAN MAKING OPPHS							
3-09-020-99 OTHER/NOT CI	48170				101	S PRODUCT	
MACHINING OPER							
3-04-030-01 DRILLING-SP 3-04-030-02 MILLING-SP		o. o.	0. 0.	0.	0. 701	S PROCESSED S PROCESSED	
3-07-030-03 REARING-SP : 3-07-030-04 GRINDING-SP	HATL	0. 0.	o. o.	0	a. for	PROCESSED	
3-09-030-05 gawing-gp m 3-09-030-06 moning-gp m	4 T L	o. o.	o.	0.	0. 70	PROCESSED	
3=07=030=99 OTHER-SP MA	TL .				701	S PROCESSED	
07MER/NOT CLASIFD 3-09-999-99 BPECIFY IN 1					700	s PANCESSED	
INDUSTRIAL PROCES -LEATHER PROI	DUCTS				. •	• **********	
OTHER/HOT CLASIFO							
3-20-999-99 \$PEC1PY IN 6	REHARK				TON	9 PROCESSED	
INDUSTRIAL PROCES -TEXTILE MF6							
GENERAL PAORICS							
3-30-00 -0 YARH PREP/BI 3-30-00 -02 PRINTING 3-30-00 -09 OTNER/NOT BI	-				701	S PROCESSED S PROCESSED	
RUBBERIZED PARRIC							
3-30-003-01 IMPRIGMATIO 3-30-002-02 WET COATING 3-30-002-03 NOT RELICED 3-30-002-07 OTMER/MOT SI	ATTH6				701 701	PROCESSED PROCESSED PROCESSED PROCESSED	
CARPET OPERATHS							
3-30-003-99 OTHE#/HOT \$	PECIFO				TO	. PROCESSED	
INDUSTRIAL PROCES - INPROCESS F	0						
SMTHRACTTE COAL							
3-98-801-99 GTHER/HOT C	LABIFD 0.	٥.	0.	0:	0: 10	S BURNED	
SITUMINUT COAL							
\$-90-002-0} CEMENT KILM \$-90-002-03 LIME KILM	/D#1E# 0.	o.	0.	0.	O. TON	9 BURNED 9 BURNED	
3-90-002-04 RADLIM KILM 3-90-002-04 BRICK KILM/	0.	0.	0.	0,	D. TOH	S BURNED S BURNED	
3-90-002-07 @YP\$UM KILM 3-90-002-06 CDAL DRYEMS	/ETC 0.	0.	0.	o. o.		S BURNEĎ	
3-70-002-09 BOCK/6R4YEL 3-70-002-09 BOCK/6R4YEL	DRYES 0.	ö. e.	ö. o.	0.	D. TON	S BURNED S BURNED	
MESIDAT OIT		••	••	••	J, 194		
3-90-004-01 ASPHALT DRY	ER O.	0.	0.	٥,	0. 100	C GALLONS BURNED	
3-90-004-02 CENENT ELN 3-90-004-03 LINE KILN	PORYER D.	0.	0.	0	0: 100 n: 100	O GALLONS BURNED	
3-90-004-04 KADLIN EILN 3-90-004-05 METAL MELT!	0.	0.	0.	ů. 0.	0. 100	0 GALLONS BURNED 0 GALLONS BURNED	
3-90-004-04 BRICK KILM/	DRY O.	0.	ö.	٥,	6. 160		
3-90-004-07 GYPSUM KILM 3-90-004-08 GLASS FURMA	ce o.	0.	D•	0;	D. 100	D SALLONS SURNED O SALLONS SURNED O SALLONS SURNED	
3#90+004+04 ROCK/SAAVEL 3#90+004+10 PRIT SMELTE		0. 0.	0.	0,	V 100	O SULLOWS SUBMED	
3-70-004-11 PERLITE FUR 3-70-004-30 PEED/GRAIN	MACE D.	e. 0.	0 ·	8.	9, 100	O SALLOWS SURNED	
'A' INDICATES THE ASH CONTENT,		BULFUR CONTE		, ON A PERI			

ENVIRONMENTAL PROTECTION AGENCY	SECTION	NEDS	SECTION 3	CHAPTER 7	SUBJECT 0
NATIONAL AIR DATA BRANCH		Source Classification Codes and Emission Factors	DATE 1/3/76	PA 20	
VOLUME V. AEROS MANUAL OF CODES	SUBJECT	SUBJECT			

•		P 0 U 4 D S			V M 1 7		
INDUSTRIAL PROCES	-IMPROCESS FUEL	PART	90 X	MOR	HC	co	U M 1 T S
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RESIDUAL DIL	CONTINUED						
1-90-009-31	F000-0RT/C00K/ETC	٥,	0.	0.	0;		1000 GALLONS BURNER
	PERTILIZER DRYING	ō,	0.	ě:	Λ:		1000 GALLONS BURNED
3-90-004-80	PULPBOARD-DRYERS	0.	0.	0.		0.	1000 SALLOWS BURNED
3-90-004-51	PLTWOOD-DRYERS	٥.	0.	0.	o,		1000 CALLONS BURNED
3-98-884-53	PULPARECOV BOILER	0.	0.	0.	٠.	0.	1000 EALLOWS BURNED
3-10-004-11	OTHER/HOT CLASIFD	0.	0.	0.	٥.	۰.	1000 BALLONS SURNED
SISTILLATE OIL							
	ASPHALT DRYER	0.	0.	۰.	0.	0.	1000 GALLONS BURNED
	CEMENT «ILM/DRYER	0.	0.	۰.	o,		1000 SALLONS BURNED
3-90-008-03	FIME KILM	0.	0.	٥.		۰.	1000 CALLONS BUPNED
3-90-005-04		٥,	٠.	0.		٥.	
3-90-006-04	METAL MELTIMS BRICE EILM/DRY	0.	o. o.	•		0.	
3-90-008-07		ě.	ě:		0,	ŏ.	
3-40-005-08	SLASS PURMACE	ě.	ö.			ě.	1000 SALLOWS BURNED
3-90-005-09	BLASS PURMACE ROCK/GRAVEL DRYER	o,	0.	ě.	ě:	٠.	1000 EALLOWS BURNED
3-90-008-10	PRIT SHELTER	0.	· ·	0.		٠.	1000 SALLOWS BURNED
3-90-005-11		٥,	۰.	٥.	۰.	٠.	1000 STFOMS BAKAKA
3-90-008-30	PEED/GRAIN DRYING	٠.	٥.	0.	0,	٠.	1000 EALLONS BURNED
3-90-001-31	F000-DRY/C00E/ETC	٠.	۰.	٠.	۰.		1000 GALLONS BURNED
3-40-008-31	PERTILIZER DRTING	٠.	•	٠.		۰.	1000 STFFORE SAMES
3-40-008-30	PULPSOARD-DPYERS PLYMOOD-DRYERS	ν,	o.	•	ě;	•	1000 SALLONS BURNED
1-00-005-51	BUI PARECON BOLLER	0.	ŏ.	0.	0.	0.	1000 GALLONS BURNED
3-90-005-99	PULP-RECOV BOILER OTHER/NOT CLASIFD	0. 0. 0. 0. 0. 0.	ŏ.	0. 0. 0. 0. 0. 0.	ě:	0.	1000 GALLONS BURNED
MATURAL 645				• •		••	
3-90-004-01	ASPHALT DRYER	٥.	٥.	٥.	0.	۸.	MILLION CUBIC FEET BURNED
3-90-004-02	CENENT EILN/DRYER	ŏ.	ŏ•	0.	ě;	•	MILLION CUBIC FEET BURNED
3-90-004-03		ě,	ñ.	ŏ.	0.	ŏ.	MILLION CUBIC FEET BURNED
3-70-004-04	EAOLIN EILN	٠.		ů.	ō.	0.	MILLION CUBIC FEET BURNED
3-90-004-05	METAL MELTING		o.	0.	۰.		MILLION CUBIC PEET BURNED
3470-004-06	BRICK KILH/DRYS	0.	0.	0.	٠.	0,	MILLION CUBIC PEET BURNED
3-90-004-07	STPSUM KILM ETC	0. 0. 0. 0.	٠.	0.	0	7.	WILLIAM CARIC LEEL BRENED
3-40-004-08	GLASS FURNACE ROCK/GRAVEL DRYER	٠.	٠.	٠.	٥.	0.	MILLION CUBIC PEET BURNED
3-90-004-10	PRIT SMELTER	0.	0. 0.	0.	٥.	•	MILLION CUBIC PERT BURNED
3-90-004-11	PERLITE FURNACE	٥.	0.	•	0,		MILLION CUBIC PEET BURNED
3-90-004-30	PEED/SRAIN DRIING	ě.	ŏ.	š.	0.		MILLION CUBIC FEET BURNED
3-90-004-31	PROD-ORT/CODE/ETC	ō.	0.	0.	D.	ŏ.	MILLION CUBIC FEET BURNED
3-70-004-32	FERTILIZER DATING	0.	0.	0.	9,	۰.	MILLION CUBIC PEET BURNED
3-70-004-50	PULPBOARD-DRYERS	٠.			٠.	0.	MILLION CUBIC PEET BURNED
3-70-004-61	PLTWOOD-DRYERS	٥.	0.	۰.	0.	0.	MILLION CUBIC PEET BURNED
3-70-006-52	PLTWOOD-DRYERS PULP-RECOV BOILER OTMER/NOT CLASIFD	8. 0.	o.	o. o.	0,	0.	MILLION CUBIC BEET BURNED
PROCESS 643		••	••	••	0.	0.	MILLION CUBIC PEET BURNED
	** ** **		_				
3-70-007-01	CO/OLAST FURNACE	0.	0.	0.	٥.	0.	MILLION CUBIC PEET BURNED
3-40-001-03	COEL OVEN 448	۰.	••	0.	0,	0.	MILLION CUBIC PEET BURNED
	STHER/HOT CLASIFO	0.	0.	••	0.	•	MILLION CUBIC PEET RURNED
COKE							
3-90-006-01	RINERAL WOOL FURM	٥.	0.	0.	0.	0.	TONS BURNED
3-70-006-99	STHER/HOT CLASIFO	٠.	0.	0.	٥.		TONS
w 0 00							
3-+0-004-++	OTHER/HOT CLASIFO	٠.		0.	o.		TORS BURNED
LTG PE7 648 (L)	•61						
3-90-010-99	OTHER/NOT CLASIFO	0.	0.	0.	o.;		1000 GALLONS BURNED
OTHER/HOT CLAS					•		
3-90-999-07	SPECIFT IN REMARE	_					
3-10-111-01	SPECIFY IN REMARK	0.	0.	۰.	۰,		MILLION CUBIC PEET BURNED
1-10-111-00	SPECIFY IN REMARE		٠.	0.	٥,		1000 EALLONS BURNED
		٥.	0.	۰.	0.	0.	TORS BURNED
IMDUSTRIAL PROCES	-0THER/MOT CL#51F0						
SPECIFY IN BENI	AT.						

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TONS PROCESSED

	SECTION	NEDS	SECTION	CHAPTER	SUBJECT
ENVIRONMENTAL PROTECTION AGENCY		1	3	7	0
THOTEOTION ACERO!	CHAPTER	Source Classification	DATE	PA	GE
NATIONAL AIR DATA BRANCH		Codes and Emission Factors	1/3/76	23	L
VOLUME V. AEROS MANUAL OF CODES	SUBJECT		3, 3, 1, 5		

		P 0 U H D 9		TFO PE			,	
		PART	301	MOI	#C	CO	U 41 1	• •
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DRITCLE ANTHE								
4-01-001-01	PERCHLORETHYLENE	0. 0.	o.	0 + 0 +	210. 305.	0. 101	S CLOTHES	CLEANED
4-01-001-00	STOODARD SPECIFY SOLVENT	••	٧.	٠.		70	S CLOTHES	CLEAMED
DEGREASING								
4-01-001-01		0.	٠.	0.		a. To		USED
4-01-001-03	PERCHLOROETHTLENE					70	IZ BOFAEHL IZ BOFAEHL IZ BOFAEHL	03E0
4-01-003-04	METHYLENE CHLORDE THICHLORDETHYLENE					70	IS SOLVENT	USED
4-01-002-04	TOLUENE OTHER/MOT CLASIFO					78	S SOLVENT	USED
STHER/HOT CLASS	70						-	
4-01-979-99	SPECIFY IN REMARK					70	18 BOLVENT	U\$20
POINT SC EVAP	-BURFACE COATING							
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PA1H7								
4-03-001-01	GENERAL	٠	٠.	٥.	1.120.	0: 10	S COATING	
4-02-001-02 4-02-001-03	ACETONE ETHTL ACETATE				3.000. 2.000.	70	IS SOLVENT	IN COATING
4-02-001-04	ME E TOLUENE				2.000.	70	S BOLVENT	IN COSTING IN COSTING IN COSTING
4-02-001-**	SOLAEMA CEMENT				2,000.	70	S BOLVENT	IN COATING
448H13H/3HELLAC								
4-02-003-01					1,000.	70	5 COATING	
4-02-003-03	ETHTL ACETATE				2.000.	10	S BOLVENT	IN COATING
	TOLUENE				2.000.	701	S SOLVENT	IN COATING
4-01-001-44	BOLVERT GENERAL				2,000.	To	19 BOLVENT	IN COTAINE
LAQUER								
4-02-004-01	GENERAL ACCIONS				1.840. 2.000.	70 70	S BOLVENT	IN COATINE
4-02-004-03	ETHTL ACETATE				2,000.	10	SOLVENT	IN COATING
4-02-004-05	ISOPROPYL ALCOHOL MER				2,000.	10		IN COATING
4-02-804-04 4-02-004-07	TOLUENE				2,000.	70	SOLVENT	IN COATING IN COATING IN COATING
4-02-004-99	SOLVENT SENERAL				2,000.	70	S BOLVENT	IN COATING
EMAREL								
4-02-008-01	GENERAL Cellobolve acetat	۰.	۰.	••	940; 2,000.		19 COATING	
4-02-008-03	MEK				2.000.	10	IS SOLVENT	IN COATING
4-02-005-05	TOLUEME TYLEME				2,000.	10	S BOLVENT	IN COATING
	BOLVENT GENERAL				2,000.	70	IS BOLVENT	IN COSTING
es inte						••		
4-02-004-01 4-02-004-02	GENERAL MAPHTHA				1.320. 2.000.	70	IS COATING	IN COATING
	NTLENE MINERAL SPIRITS				2,000.	70	S BOLVENT	IN COATINA IN COATING IN COATING
4-01-004-08	TOLUENE				2,000.	ŶŌ	-	IN COATING
49HESIVE	BOLVENT SENERAL				.,	,,,		14 (01)146
4-87-007-01	454544					Tai	15 COATING	
9-02-007-02	MEK				2,000. 2,000.	700	S BOLVENT	IN COATING
4-01-007-04	TOLUEME BEMZEME				£.000.	70	S SOLVENT	IN COATING
4-07-007-05	BOLVENT' GENERAL				2.800.	70	IS BOLVENT	IN COATING
COATING DVEN	-							
4-02-000-01	GENERAL					70	-	
	BRIED < 1787 BAEED > 1787					70 70	S COATING	
4-01-008-**	OTHER/SPECIFY					10		

	SECTION	NEDS	SECTION	CHAPTER	SUBJECT
ENVIRONMENTAL PROTECTION AGENCY			3	7	0
PROTECTION AGENT	CHAPTER	Source Classification	DATE	PA	GE
NATIONAL AIR DATA BRANCH	Codes and Emission Factors	1/3/76	2:	2	
VOLUME V. AEROS MANUAL OF CODES	90,5000				

POUNDS ENTTYPOPER UNIT co POINT SC EVAP -SURFACE COATING 2.000. 2.000. 2.000. 2.000. 2.000. 2.000. 2.000. 2.000. 2.000. 2.000. TOMS SOLVENT \$. DOD . -4-DE-999-99 SPECIFY IN REHARK TONS COATING POINT SC EVAP -PEYROL PROD ST6 1000 GALLONS STORAGE CAPACITY
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^{&#}x27;4' IMPICATED THE ASM CONTENT, 'S' IMPICATES THE SULPUR CONTENT OF THE FUEL ON A PERCENT RASIS (BY MEIGHT)

	SECTION	NEDS		SECTION	CHAPTER	SUBJECT
ENVIRONMENTAL PROTECTION AGENCY	1			3	7	0
PROTECTION AGENCY	CHAPTER	Source (Classification	DATE	PA	GE
NATIONAL AIR	Codes and Emission			l		
DATA BRANCH	SUBJECT	Factors		1/3/76	2:	3
VOLUME V.						
AEROS MANUAL OF CODES						
	P G U N PART	D 8 E M 1 T	TEOPERUNIT	 	, 	
POINT SC EVAP -PETROL PROD	376	•••		••	• • • • • • • • • • • • • • • • • • • •	
WAR-VAPOR SPACE						
4-03-003-03 - work M6-6450 4-03-003-03 - work M6-6450		0. 0.	0. 10.2			UT
4-03-003-04 WORE M6-EER	SENE O.	٠.	0. 1.00	0. 1000	SALLOWS THROUGH	•U*
4-03-003-00 MOSKING-8EH)	ENE O.	o. o.	0. 1.00 0. 2.30	9. 1900	EALLDUS THROUGH	• U T
4-03-003-07 WORE!M6-CYC	OPENT D.	0. 0.	6. 2.40 8. 7.20	0. 1000	SALLONS THROUGH	*U T
4-03-003-07 WORK [ME-HEY	144E 0.	0. 0.	0. 1.40	0. 1000	GALLONS THROUGH	, u t
4-03-003-11 WORKING-1900	CTAME 0.	0.	0. 1.70	0, 1000	SALLONS THROUGH	· U •
4-03-003-12 WORKING-PEN'	7E47 0.	0. 6.	0. 17.6		GALLONS THROUGH	
4-D3-DD3-14 WORKING-TOL	JENE O.	0.	0. 0.73	01000	SALLONS THROUGH	TUT
STHER/HOT CLASIFD	.,,,			,,,,,		
4-03-174-77 BPECIFY IN 1	I E MARK			1000	SAL STORED	
POINT SC EVAP -MISC BREAMIN						
	•••••					
97MER/MOT CLASIFD 4-D4-001-99 SPECIFT IN 1				TAN	STORED	
POINT SC CVAP -PRINTING PRI						

DRYERS						
4-05-00]-0} SEHE-AL			••	TONI	SOLVENT	
4-06-002-01 BEHERAL			700.	TONI	NE	
4.05.001.02 EEROSENE			2,000.	TON	SOLVENT IN THE	
4-DS-002-03 MIMERAL SPI 4-DS-002-99 BOLVENT GEN	8;7 5 E 0 AL		2,000. 2,000.		SOLVENT IN INK	
PLEXOGRAPHIC						
4-05-003-0! GENERAL			1,300.		t ux	
4-09-003-02 CARBITOL			1.000. 1.000.	Tons	SOFAEMA IN THE	
3410801133	OL .		2,000.	TON	S SOLVENT IN INK	
4-05-003-05 180PROPTL AL	r Coxor		1,000. 2,000.	TON	BOLVENT IN INK	
9-86-003-07 WAPHTHA			1,000. 1,000.		BOTAENL IN INK	
4-06-003-44 86 L4EM7 6EM L1TM06819H1C			2,000			
• • • • • • • • • • • • • • • • • • • •			700.	TON) INC	
4-05-004-01 GEMERAL 4-05-004-02 MIMERAL SPI			2,000.	TON	SOLVENT IN INK	
4-85-004-03 ISOPROPYL A 4-85-004-99 SGLYENT GEN	LCOHOL Eral		2,000. 2,000.		BOTABUL IN THE PERSON OF THE PERSON IN THE	
gRavure				_		
4-06-005-01 GENERAL 4-06-006-02 BINETHTLFOR	MAMIDE		1,300. 2,000.	TON) INK BOLVENT IN INK	
4-04-004-03 ETHTL 4CETA	76		2,000.	TON	BOLVENT IN INK	
4-85-005-04 ETHYL ALCOM 4-05-005-08 180PROPYL A	F COMOF Of		2,000.	TON	BOTAENT IN INK	
4-06-006-04 MEK	··		1,000. 2,000.	TON:	SOLVENT IN THE	
4 08 008007 HISK 408800808 HINERAL SPI	9175		2.000. 2.000.	TON	SOFAEMA IN INC	
4-05-005-04 M-PROPTL AL 4-05-005-10 TOLUTHE	COMOL		3,000.	TOR	I BOLVENT IN INC	
4-05-005-FF BOLVENT GEN	ERAL		2,000.	TON	SOLVENT IN INK	

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VOLUME V. AEROS MANUAL OF CODES	SUBJECT				

	-PETAOL MRET-TRANS	PART	\$07	NOX	HE	CO	U N T T S
-	s						
4-94-901-01	L040(SPL49H)-6450	٥.	0.	٠.	12.4	0.	1000 GALLONS TRANSFERRED
9-04-001-08 9-04-001-03	L040(SPLASH)-CRUD L040(SPLASH)-JET	o.	o. o.	o.	10.4	0 ·	1000 GALLONS TRANSFERRED 1000 GALLONS TRANSFERRED
9-84-881-09	LOAD(SPLASH)-KERO	٥.	٥.	٥.	0.44	n.	1000 GALLONS TRANSPENSED
4-04-001-08	LOAD (SPLASH) - D ST LOAD (SUBM) - 6450LH	o. o.	0.	0.	0.*3 4.10	0.	1000 CALLONS TRANSFERRED 1000 CALLONS TRANSFERRED
4-04-001-27	LOAD (SUBH) - CRUDE	0.	0.	0.	3.80	0.	1000 GALLONG TRANSPERRED
4-04-001-28	14 13C+1486814907 HECASH+1466814907	o.	o.	o.	0.91 0.45	0.	1000 GALLONS TRANSFERRED 1000 GALLONS TRANSFERRED
9-04-001-30	LOAD(SUB#)-D157	۰.	٥.	0.	0.44	0.	1000 GALLONS TRANSFERRED
4-04-001-83	MATOVD-CANDE OIF	0. 0.	o. o.	0.	2.10 1.78	0 ·	1000 GALLONS TRANSFERRED
4-04-001-83 4-04-001-84	UMLOAD-JET FUEL UMLOAD-GEROSEME	0.	0. 0.	0.	0.48	0.	1000 GALLONS TRANSFERRED
4-04-001-55	UML040-0187 01L	ŏ:	٥.	o.	0.23	0.	1000 CALLONS TRANSFERRED 1000 CALLONS TRANSFERRED
9-04-001-98	LOAD(BPLSH)SPECFT LOAD(SUAM)SPECIFT UNLOAD-SPECIFT						1000 GALLONS TRANSFERRED 1000 GALLONS TRANSFERRED 1000 GALLONS TRANSFERRED
4-04-008-01	L040144-6450L1HE	٠.	٥.	٥.	2.96	••	
4-04-003-03	LOADING-CRUDE DIL LOADING-JET PUEL	o.	0.	0 • 0 •	2.50 0.40	0.	1000 GALLONS TRANSFERRED 1000 GALLONS TRANSFERRED
4-04-001-04	FOTD IME-KENDERE	ę.	٥.	٥.	0.27	0.	1000 CALLOWS TRAMSFERRED
	LOAD] WG-D137 O] L	0.	0.	0 ·	2.52	0.	1000 CALLONS TRANSFERRED 1000 CALLONS TRANSFERRED
4-04-002-27	UMLOAD-CRUDE OIL UMLOAD-JET FUEL	ę.	0.	0.	2.26	0,	1000 CALLONS TRANSFERRED
4-04-002-29	UNLOAD-KEROSENE	0.	0. 0.	0. 0.	0.83	0.	1000 SALLOWS TRAMSFERRED 1000 SALLOWS TRAMSFERRED
	UMLDAD-DIST DIL LOADING-SPECIFY	٥,	۰.	0.	0.25	0.	
4-04-002-49 UMDERARD 6-90 1	ÚMLOÁD-SPECIÉT						1000 SALLONS TRANSFERRED
		-	_				
4-04-003-03	SPLASH LOADING SUB LOAD-UNCONT	o.	0, 0.	0.	7.30	0.	1000 GALLONS TRANSFERRED 1000 GALLONS TRANSFERRED
4-04-003-03	SUB LOAD-OPH STS	٥.	٠.	٥.	0.60	0.	1000 SALLONS TRANSFERRED
4-04-003-04 4-04-003-05	SUB LDAD-CLS 879	o. o.	o.	o. o.	0. 1.00	0. 0.	
4-04-003-44	SPECIFY METHOD						1000 SALLOWS TRANSFERRED
PILL VEH GAS TA		_					
4-04-004-01 4-04-004-02 4-04-004-99	VAP DISP LOSS LIE SPILL LOSS OTHER LOSS	٥.	0. 0.	0.	11.0	0.	1000 GALLONS PUMPED 1000 GALLONS PUMPED 1000 GALLONS PUMPED
	-418C HC EATS						
OTHER/HOT CLASS	FB						
4-90-99-99	SPECIFY IN REMARK						TOMS PROCESSED
\$0L1D WASTE	-404ERNNENT						
-weicipal Incin	1						
6-01-001-01 6-01-001-02	NULTIFLE CHAMBER SINGLE CHAMBER	30.0 16.0	2.50	3.00 3.00	1.50 15.0	36.0	TORS BURNED TORS BURNED
99EH BUR41HE BU	PP						
#-01-001-01	LAMOSCAPE/PRUMIMA	14.0	1.00	4.00 2.00	30.0 20.0	40.0	TONS BURNED Tons gurned
CO-COO-10-2 ROTARSHISH:	46.7 TVEL					- •	MUNDREDS OF GALLONS
	PATHOLOGICAL	#.Da	9.	3.00	٠.		7089 BUBUES
\$-01-005-06 \$-01-00\$-07 \$-01-005-77	SLUGEE CONTCAL OTHER/HOT CLASIFO	20.0	1.00	5.00 5.00	1.00 20.0		TOMS BURNED TOMS DRY SLUDGE TOMS BURNED
AWE-FUEL/MD ENS							TONS BURNED
	RESIDUAL DIL						
8-01-400-05	DISTILLATE OIL	o.	o. o.	0.	0;	0.	1000 6411045
\$-01-900-04 \$-01-900-10	HATURAL GAS	٥.	۰.	9. 0.	9;	0.	MILLION CUBIC FEET
6-01-900-97	STHER/WOT CLARIED	0. 0.	?• 0•	o.	0.	ń.	1000 GALLOWS
5-01-900-98	OTHER/HOT CLASIFO OTHER/HOT CLASIFO	0.	0.	ö.	9:	0.	1000 GALLONS
	ASH CONTENT, 'S' INDI						
				.0-24		-,	

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		SUBJECT	Factors			1/3/7	76	2:	•
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SOLID WASTE	-C@##-!W\$T	P 0 U M PART	9 8 E H T	401	R W H 1 T	*			

\$-02-001-02 \$-02-001-03 \$-02-001-04	MULTIPLE CHANGER SIMBLE CHANGER CONTROLLED AIR CONTCAL-REFUSE CONTCAL-WOOD	7.80 16.0 1.40 26.0 7.00	2.80 2.80 1.80 2.00	3.00 3.00 10.0 5.00 1.00	3.00 19.0 6. 20.0 11.0	19.0 20.0 43.0 130.	TONS TONS TONS	Brates Brates Brates Brates Brates	
eren aukning									
0-01-002-01 0-01-002-03 4PARTMENT INCIN	REPUSE	17.0		\$.00	4.00	\$0. 0		BROWED GASHED	
8-01-003-01 2-01-003-01	PLUE 720 PLUE PED-MODIFIE	30.0	*.**	3.00 10.0	15.0	10.0		PANES PANES	
1 WC 1 WE 0 4 7 0 R 0 - 0 1 - 0 0 5 - 0 5 5 - 0 2 - 0 0 5 - 0 4	PATHOLOGICAL SLUDGE	#.00 180•	•. 1.00	3.00 6.00	#; • • •	•	1085	enance	
\$-03-005-97 Auz.puel/40 EMS	9THES/NOT CLASIF	•					7043	PANED	
	#ESIDUAL DIL	•.	•.	••	•;	••	1800	BALLORS	
0-02-700-05 0-02-700-04 0-02-700-77 0-02-700-98	SISTILLATE OIL MATURAL GAS LP4 OTHER/HOT CLASIFI STHER/HOT CLASIFI		0. 0. 0.	• • • • • • • • • • • • • • • • • • •			1000 1000 1000	OALLOWS 104 CUBIC PERT OALLOWS 10H CUBIC PERT SALLOWS	
	STHER/HOT CLASSF		••	••	••	••	,,,,,,		
14014888708	MULTIPLE CHAMBER	7.00	2.50	3.00	3.00	16;0	7045	OURRED.	
6-03-001-02	SINGLE CHAMBER	18.0	1.50	8.00	10.0	20.0		03##FD	
6-03-001-04	CONICAL REFUSE	20.0 7.00 13.0	2.80 8.10 8.10	0.00 1.90 4.00	80.0 11.0	130,0	TOUS	BURNED BURNED OF WASTE	
open ourning							•		
10-100-1008 20-100-1008 10-100-1008 200-1008	WOOD REFUSE AUTO BODY COMPTS COAL REFUSE FILE	17,0 14,0 100. 1 0.90	1.00 0. 1.10	2.00 4.00 4.00 0.10	30.0 30.0 9.50	125	Tous	OURMED OURMED OURMED C TAROS OF PILE	
AUTO BODY 19011	147								
0-03-003-03	W/O AFTERBURHER W/ AFTERBURHER	1.00		0.02	•••			S GAMMED S GAMMED	
BAIL COR BURNES							F	- Canade	
\$-03-004-01	976#						****	***************************************	
1 MC MERATOR 0-03-005-04 0-0 MOSL-00	SLUDGE STHER/HOT CLASIF	100.	1.00	1.00	1:00	•:		BRY SLUBAR BURNES	
AMI-PUEL/MD EM	-	•							
8-03-700-09 8-03-700-08 8-03-700-07 8-03-700-07 8-03-700-10	RESIDUAL DIL BISTILLATE DIL MATURAL GAS PRECESS GAS L P G OTHER/MOT CLAS!P GTHER/MOT CLAS!P STHER/MOT CLAS!P	0. 0. 0. 0.	0. 0. 0. 0. 0.	•. •. •. •. •.			#166 #166 #166 #166	OALLONG OALLONG ION CUOIC PEET ION CUOIC PEET OALLONG ION CUOIC PEET GALLONG	
	-PEDEL MONENTTER								
OTHER/HOT CLAS		•							
4-01-777-98	SPECIFY IN REMAR SPECIFY IN REMAR							ALLATIONS (EACH	1

[&]quot;A" IMPICATES THE ASH CONTENT, "S" INDICATES THE SULPUR CONTENT OF THE PUEL ON A PERCENT MASIS (OF MEIGHT)

APPENDIX I-E

AREA SOURCE CLASSIFICATION CODES

Area sources of hydrocarbons include industrial area sources, residential and commercial sources, highway vehicles both as area and line sources, railroads, river vessels, airports and off-highway mobile sources.

Table I-E is an expansion of the NEDS Area Source to EIS/P&R User-Point-ID 13 major categories reported in Section 5 of the "Comprehensive Data Handling System, Emissions Inventory/Permits and Registration Subsystem (EIS/P&R) Program Documentation and Users Guide" (EPA-450/3-74-045a, February 1975).

The Area SCC categories that appear after 91308500 have been provided by the California Air Resources Board for specific profiles included in this manual. Table I-E is necessarily incomplete because of the limited number of area source categories identified to date.

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	Category	EIS/P&R (USER-POINT-ID)	Fuel/Pollutant Source	(AREA) SCC
	Residential Fuel	901	Anthracite Coal	90100111
			Bituminous Coal	90100222
			Distillate Oil	90100330
			Residual Oil	90100440
			Natural Gas	90100500
			Wood	90100600
	Commercial and Insti-	902	Anthracite Coal	90200111
	tutional Fuel		Bituminous Coal	90200222
			Distillate Oil	90200330
			Residual Oil	90200440
			Natural Gas	90200500
AR			Wood	90200600
AREA	Industrial Fuel	903	Anthracite Coal	90300111
			Bituminous Coal	90300222
SOURCE			Coke	90300700
₹			Distillate Oil	90300330
Ħ			Residual 011	90300440
C			Natural Gas	90300500
CATEGORIES			Wood	90300600
છુ			Process Gas	9 0300800
)RI	On-Site Incineration	904	Residential	90401100
EZ.			Industrial	90401200
•			Commercial/Institutional	90401300
	Open Burning	905	Residential	90501100
			Industrial	90501200
			Commercial/Institutional	90501300
	Gasoline Fuel	906	Light Vehicle	90602100
			Heavy Vehicle	90602200
			Off Hiway	90602300
	Diesel Fuel	907	Heavy Vehicle	90702200
			Off Hiway	90702300
			Rail Locomotive	90702400

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		Category	EIS/P&R (USER-POINT-ID)	Fuel/Pollutant Source	(AREA) SCC
		Aircraft	908	Military	90803100
				Civil	90803200
				Commercial	90803300
		Vessels	909	Anthracite Coal	90900111
				Diesel Oil	90904230
				Residual Oil	90900440
I-E-3				Gasoline	9 0904430
		Evaporation	910	Solvent Purchased	91005100
	≻			Gasoline Marketed	91 005200
	AREA	Measured Vehicle Miles	911	Limited Access Roads	91106100
				Rural Roads	911 06200
	SOURCE			Suburban Roads	91106300
	Б			Urban Roads	91106400
		Miscellaneous Dust	912	Dirt Roads Traveled	91207100
		Ç			Dirt Air Strips
	AT)			Construction Land Area	91207300
	CATEGORIES			Rock Handling & Storing	91207400
		Miscellaneous Burning	913	Forest Fires	91308100
				Slash Burning	9 1308200
				Frost Control	91308300
				Structure Fires	91308400
				Coal Refuse Burning	91308500

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Category

Solvent Use

Geogenic

Solid Waste

EIS/P&R (User-Point-ID)

Fuel/Pollutant Source

Architectural Surface Coatings

Pesticides, Domestic & Commercial

Domestic Solvents, General

Landfill Site, Class II

Forests

Petroleum Seeps

Citrus Groves

Animal Waste

(Area)SCC

93510300

93570200

93570598

94740999

94741101

94742999

94999998

94999999

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APPENDIX I-F

BOILING POINT RANGE COMPOUNDS

INDUSTRIAL NAPTHAS (Ref. 1)

Refiners manufacture a large number of liquid hydrocarbon compounds in a narrow range of boiling points with a high degree of purity from natural gas and petroleum. The petroleum oils that result from this distillation process as a group are known as industrial napthas. The term "industrial napthas" includes solvents, thinners (for example, of the type used in the paints and varnish industries), and diluents (as used in the manufacture of pharmaceuticals and insecticides).

The original industrial naptha from petroleum was a selected naptha from the straight-run processing of certain crude oils. Today this classification of petroleum products also contains pentane, hexane, and heptane produced largely from natural gas and the aromatics-benzene, toluene, and xylene-now recovered from petroleum by modern refining methods.

As stated earlier, industrial napthas are a group of liquid hydrocarbon compounds produced by the distillation of petroleum over selected boiling point ranges. The liquid hydrocarbon compounds that result from this selective distillation by refiners are known individually as petroleum ether, rubber solvent, light spirits, VM&P naptha, mineral spirits, and Stoddard Solvent. Lactol spirits is a compound made from naptha and lactic acid (Ref. 2). These compounds are highly complex mixtures of hydrocarbons produced over a narrow range of boiling points, and except for the relatively pure compounds of hexane and toluene, do not lend themselves to scientific classification or speciation. Not only will the different hydrocarbon species in each boiling point range compound vary from crude to crude, from refinery to refinery and from operator to operator; but the range of variance for individual species within a boiling point range may be from 0 to 100 percent, again depending on these variables.

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for these reasons, KVB has elected not to provide a species breakdown for the boiling point range compounds known as 1) naphtha, 2) mineral spirits, 3) lactol spirits, and 4) Stoddard Solvent. The net result is that these compounds appear as a specie themselves and have not been further broken down into their individual components. However, there are two profiles which do provide a speciation for two of these compounds as used in specific processes. Profile 4-01-001 B lists the species for Stoddard Solvent and profile 4-02-006 J lists the species for mineral spirits. The speciation for these compounds, however, should be used with discretion for the reasons discussed above.

REFERENCES

- Bland, W. F., and Davidson, R. L., "Petroleum Processing Handbook," McGraw-Hill Book Co., 1967.
- Stecher, P. G., et al., "The Merck Index," an Encyclopedia of Chemicals and Drugs, Merck & Co., Inc., 1968.

APPENDIX II EMISSION PROFILE DEVELOPMENT

APPENDIX II

EMISSION PROFILE DEVELOPMENT BY MEANS OF FIELD TESTS, INDUSTRY QUESTIONNAIRES AND LITERATURE DATA

DESCRIPTION

A unique aspect of the KVB HC Test Program was the development of emission profiles, the identification of the organic compound species represented by the total hydrocarbon emission rates currently given in emission measurements. Only one other study (Ref. II-1) had previously attempted a breakdown into generic classes. That was done primarily for the purpose of dividing emissions into reactivity classes. The results of that previous study have been widely used in the California South Coast Air Basin.

The primary objective of the KVB test program was to identify the organic compound emissions for each stationary source type in the Basin and develop a data management system capable of applying this information to the total hydrocarbon emissions in order to calculate the emissions of the individual organic compounds. Thus an emission profile was formulated for each Source Classification Code (SCC) emitting organic compound species in the Basin. Both point and area sources were included. In certain instances a further breakdown was made into individual industries identified by Standard Industrial Codes (SIC).

Another objective of the KVB test program was to predict future emission trends. Satisfying this objective required emission profiles based on SCC number rather than individual plant profiles based on individual plant characteristics. All plant devices identified by the same SCC and SIC number were given the same emission profile. Conversely, it was important that profiles be truly representative of the device in general. Additional advantages of developing aggregate profiles by SCC number were that:

(1) estimations based on larger data samples were more statistically reliable than single data samples, (2) the profiles were compatible with the EIS

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concept by describing devices by the SCC number system, and (3) the volume of profile data was reduced to a more manageable level.

The initial intent was to provide a profile for each SCC listed in the data base. In many instances, however, an individual profile was found to cover several SCC and SCC/SIC combinations. The profile data base was therefore formulated and indexed by a KVB profile number. Separate profile numbers (with identical specie distributions) were given to specific SCC/SIC combinations to facilitate data management, specifically the segregation of emissions from devices with smaller SCC codes in two different industry classes into the appropriate ARB Application Categories.

In each profile the organic species were initially identified by their appropriate SAROAD code, ARB reactivity classification (3 class) and molecular weight. Each profile was also "tagged" with other identifiers to assist those who may wish to use or evaluate these data. Associated with each emission profile was an estimate of its relative error. This "Error Estimate" was strictly subjective and was included to give a relative level of confidence to the specific profile. No statistical significance have been or should be given to these error estimates.

The ARB reactivity scheme and Error Estimate reporting formulas have since, for the purposes of this report, been replaced with a seven group chemical compound classification system, both of which are discussed in the introduction. A profile therefore contains a subjective estimate of the level of confidence in the profile and also lists the SAROAD code, chemical name and weight percent contribution of each specie. The species are also summed by the seven group chemical compound classification system.

METHODOLOGY

Two general approaches were used to formulate the emission profiles, one where only one data point was available to characterize many sources and another where multiple data points were available. In cases where a profile was available from only one source and that source was believed to be representative of all such source types in the Basin, then that particular source emission profile was used. An appropriate error estimate was given to reflect the relative confidence level of these data. It was

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anticipated early in the program that a significant number of source types would fall into this category due to the limited amount of field tests available. Therefore, test locations were carefully selected on the basis of the representative nature of their emissions to all other devices of that particular type. In this way, data from this source could be correctly applied to other non-tested sources. Similarly, questionnaires were submitted to and received from selected solvent users. Follow-ups were made to assure that the data from these large and representative sources were obtained.

Two examples of formulating profiles based on one data point from a selected source are the following. The first, concerned with the emission profile typical of residual oil fuel combustion, was obtained by (1) recognizing that 95% of all residual oil combustion in the Basin occurs in utility boilers, (2) selecting a boiler that was "typical" of such devices in the Basin and finally (3) conducting a test on this unit. Multiple samples were taken and the profile was based on an average composition.

The second example involved the use of data from questionnaires. One source in the Basin, according to the South Coast Air Quality Management District files, was responsible for 90% of the emissions from adhesive use. A questionnaire was mailed to this source and follow-up contacts were made to assure that information from this source was received. The questionnaire contained a comprehensive breakdown of the solvent formulation and usage which formed the basis for the emission profile.

This approach of using one analysis to characterize a general source type also applied to profiles determined by inspection. For example, there were a few SCC's that specifically identify the solvent used in a coating operation or housed in a storage tank. In this case, a solvent identified by its SCC as toluene or xylene would be given an emission profile of 100% of that organic compound.

The second approach used was to develop emission profiles based on data from several sources within a particular source type. This involved (1) acquiring the data, (2) determining the relative magnitudes of each source compared to the total emissions from the source type, and

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(3) forming a composite profile by factoring the data from each source by an appropriate weighting factor. In this manner, emission profiles were developed for individual source types that in actuality represented the average emissions from sources of that category (SCC number).

An example of this approach was the formulation of a profile for "Miscellaneous Organic Storage" in the Basin. While SCC numbers had been assigned to storage tanks for gasoline, jet fuel, crude oil, various solvents, etc., the miscellaneous category covered all other petroleum products not listed. The following table presents a summary of the calculation procedures employed to determine this profile. Listed across the top are the various organic products identified and the fraction of the emissions from fixed roof tank storage for each based on information compiled from the SCAQMD file. Listed down the page are the various organic species that have been identified in the emissions from these products. The weight percentages of each specie associated with the product is listed in the appropriate column. The weight percentage for asphalt and Stoddard solvent were determined from KVB test The adhesive percentages came from questionnaire data. The remainder of the percentages were specified (e.g. 100% for acetone) or estimated based on contacts with industry (e.g. the breakdown of alcohols and ketone). weight percent of each organic compound in the composite profile was determined by multiplying the weight percents by the appropriate fractions and are listed in the right hand side of Table II-1.

Contained within the following three sections are a thorough discussion of the three sources of information which comprise the basis of all the developed profiles:

- . Field Source Test
- . Solvent Use Questionnaire, and
- . Literature Research

TABLE II-1. COMPOSITE PROFILE FOR MISCELLANEOUS PETROLEUM STORAGE (Fixed Roof Tanks)

Product Stored	Acetone	Adhesive	Alcohol	Asphalt	Perchloroethylene	Ethylene Dichloride	Pormaldehyde	Ketone	Stoddard	Xylene	Others	Composit
	Acetone	Addiestve	ALCOHOL	Aspliate	refchioroethylene	Dicatoride	rormildenyde	Ketone	Stoddard	Aylene	Others	Composit
Fraction of Emissions	0.163	0.022	0.084	0.078	0.051	0.004	0.004	0.191	0.071	0.057	0.275	
Organic Compounds												
Acetone	100.0	4.0										16.4
Perchloroethylene					100.0							5.1
Ethylene Dichlori de						100.0						0.4
Formaldehyde							100.0					0.4
MEK								65.0				12.4
MIBK								35.0				6.7
Xylene										100.0		5.7
Toluene		5.6									31.0	8.6
Ethane				1.0								0.1
Ethylene				2.0								0.2
Propane				13.0								1.0
N-Butane				18.0								1.4
I-Butane				8.0								0.6
N-Pentane				18.0								1.4
I-Pentane											12.5	3.4
Hexane		84.6		2.0							25.0	8.9
I-Hexane				12.0								0.9
Heptane				14.0								1.1
I-Heptan e				11.0								1.0
I-Octane				1.0					0.8			0.1
I-Nonane									27.3			1.9
I-Decane									69.4			4.9
I-Undecane									2.4			0.2
Ethyl Acetate		5.8									15.5	4.4
C-7 Cyclo- paraffins											15.5	4.2
Isopropyl Alcohol			40.0									3.4
Ethyl Alcohol			30.0									2.6
Isobutyl Alcohol			30.0									2.6

FIELD SOURCE TESTING

Background

The field tests conducted by KVB provided a realistic assessment of the organic emissions from stationary sources in the California South Coast Air Basin. In that standardized measurement procedures for organic emissions were as yet unestablished at the time this study was conducted, KVB felt it important to obtain a consensus of those active in the field of developing test plans and procedures. The California ARB, EPA, SCAQMD, and Western Oil & Gas Association (WOGA) were the agencies most involved with this effort.

In all, 618 field samples were taken by KVB and analyzed at Analytical Research Laboratories Inc. (ARLI) or KVB. All GC/MS analyses of the field samples were conducted at ARLI who also measured aldehydes and total organic content (TOC).

The KVB test crew consisted of two engineers and two technicians. On major tests all four worked together. These major tests required from two to ten working days at each test site. Occasionally the crew divided into two-man teams to collect from two to four samples on a special device or process that could not be obtained during a major test.

Test Methodology

The objectives of the KVB test program were to develop techniques and equipment as necessary to (1) determine the hydrocarbon emission rate from both ducted and fugitive sources, (2) collect and preserve representative samples of these emissions and (3) analyze the samples for their organic chemical composition. The general approach to emission rate determination was to either measure the emission rate or to determine it by calculations from process data or by experiment. From sources with stacks, emissions were determined by pitot traverse. Various techniques were used on fugitive emission sources. Where information was available on the amount of organic material lost from a process, this was used to determine emissions. Where the

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emissions were due to leaks or spills or other types of fugitive emissions, an attempt was made to either measure or estimate those emissions. In some instances, special experiments were conducted to obtain estimates of emission rates. An example of the type of experiments that were conducted is the determination of the amount of solvent which was emitted from an architectural coating as it was drying or curing. KVB's tests indicated that as much as 30 to 40 percent of the solvent is permanently retained in the paint after it is cured. Other experiments included emissions from open ponds, asphalt paving, auto gas tank filling, and domestic solvents.

For analytical purposes, samples of emission gases were collected in the following type of containers:

- . glass tubes filled with activated charcoal (NIOSH approved)
- . borosilicate glass gas collecting bottles
- . Tedlar bags
- . glass bulb containing 1% sodium bisulphite solution (aldehyde determinations)

The charcoal sorbent tubes were used to collect aliphatic organic compounds with boiling points above that of n-pentane and all other compounds from C₁ - up. The gas collection bottles and bags were used to collect aliphatic compounds with boiling points below that of n-pentane. On most major sources, a combination of sorbent tubes and either bags or bottles were used. Bags or bottles were used for the entire compound range when utilized for grab sampling.

All samples were analyzed using gas chromatography (GC) and mass spectrometry (MS) techniques on a tandem GC/MS apparatus. The bottle or bag grab samples were introduced directly into the apparatus while the samples collected on charcoal were first extracted with carbon disulfide. Because of the survey nature of the program only those GC peaks which contributed at least 1% of the total hydrocarbons were identified unless a substance of special importance was suspected to exist in the sample.

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Presented in the following sections are a detailed description of the field test and laboratory equipment, some explanation for their selection, the results of test runs using this equipment, and a detailed description of test procedures and data reduction techniques followed during the program.

Sampling

- A. Equipment Description--
- 1. <u>Sampling train</u>--KVB designed and built two identical portable sampling units that could:
 - . measure stack gas temperature and velocity
 - . filter out particulates larger than 2 microns
 - . collect samples in sorbent tubes, glass or polybags.

The general flow diagram, Figure II-1, illustrates all components of the assembly which are available to be swtiched into several sampling modes to conform to requirements dictated by the source to be tested. The components are:

- . a sample nozzle
- . a filter holder with 2.5 micron pore size glass fiber filter
- . a filter and line heater and thermostatic control
- . an impinger train containing LiOH crystals
- . a borosilicate (Pyrex) gas collection bottle
- . a sorbent tube train with thermometer and vacuum gauge
- . a Brooks flowmeter with needle valve flow control
- various interior and exterior valves and connectors as indicated in Figure II-1
- a meter connection to PD gas meter
- . a pressure gauge and pyrometer for use with a pitot tube

The above system was unitized within a portable aluminum closure. Its interior arrangement permitted significant freedom of directional orientation for rigging convenience. In addition to the packaged sampling unit, the following additional test equipment was used during the testing program:

- an "S: type pitot tube and a standard pitot for velocity measurements
- . two thermocouples for stack temperature measurements

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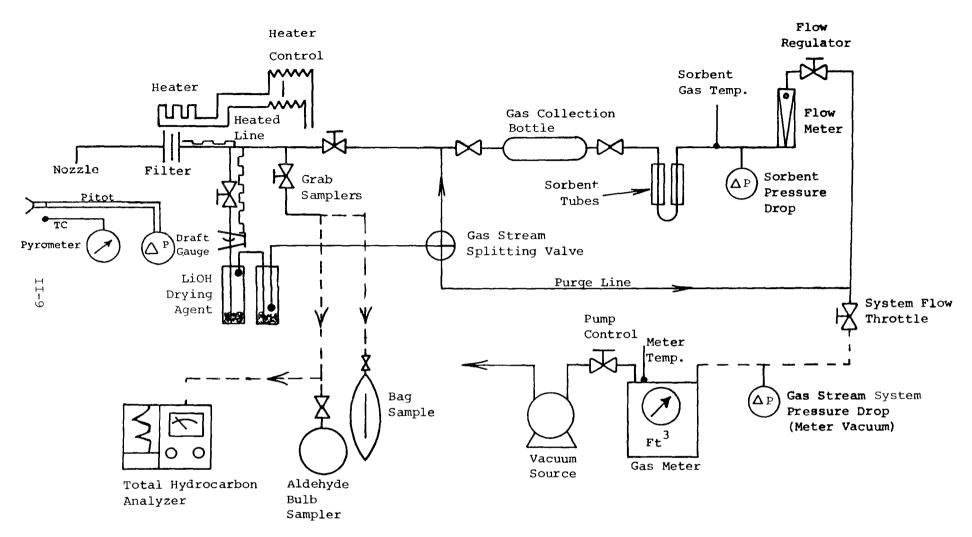


Figure II-1. Complete organic sampling train as set up for a hot combustion source (> $180\ ^{\circ}F$) (Mode 1 in Table II-4).

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- . three dry gas meters
- . additional glass sorbent tubes containing charcoal sorbent
- . two Gast vacuum pumps
- . six Spectrex diaphragm pumps
- . two squeeze bulb type hand pumps
- . an Orsat analyzer for CO, CO, O_2 , O_2 , and N_2 determination
- . a Draeger gas detector with detector tubes
- . a TLV sniffer wtih recorder (a total hydrocarbon tester with 0-10,000, 0-1,000, and 0-100 ppm range)
- . an anemometer
- . thermometers of various ranges

Typical test setup and configurations are discussed later under sampling methods.

2. Sampling Train Selection--

a. LiOH Impinger—The lithium hydroxide in the dry impinger train was selected for use based on experience gained on the Apollo space capsule. Initially an ice water impinger was considered for moisture, NOx, SOx, and CO removal. The problem with this approach was that it was felt that the alcohols and some other oxygenates would be highly water soluble and would not be easily separated for analysis. (The impinger solution was analyzed for hydrocarbons.) LiOH was used in the Apollo life support system to adsorb primarily CO₂. In the sampling train it neutralized NOx and SOx which would react with the hydrocarbons and adsorbed most of the condensed moisture. Furthermore according to Apollo data the LiOH does not adsorb hydrocarbons. A CS₂ extraction and a hydrocarbon analysis were made on the impinger contents and no hydrocarbons were found.

The probe, filter, line and valves leading to the impinger were maintained at less than 220 °F. Some light condensation was found downstream of the impingers in the collection bottles and sorbent tubes but this did not interfere with the hydrocarbon determinations. The water content of the exhaust gases was determined using a separate water knockout train, or aquasorb.

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b. Sorbent—The suitability of several different types of sorbent materials was investigated. The materials tested included: Tenax GC, Carbosieve B, activated charcoal, and XAD-2 resin. The criteria observed in the selection of the sorbent included quantitative retention and recoverability of every analyte possible. These qualities were dimensionalized by measurement of breakthrough volumes and recovery efficiencies. Table II-2 presents the breakthrough volumes of the sorbents (25 °C) for hexane and benzene. These analytes were considered to represent about the upper limit of materials that can be analyzed in gas grab samples. Carbosieve B and activated charcoal showed particularly high retention capacities.

Another important parameter in sorbent selection is the analyte recovery efficiency. Elevated temperature, thermal stripping (with a purge gas or in vacuo) or adsorbed components on Tenax, Carbosieve B and XAD-2 was considered but later rejected because the entire sample must be committed in a single determination. Recovery efficiencies using the thermal/purge-gas techniques also showed high molecular weight discrimination (see Table II-3).

TABLE II-2. RETENTION EFFICIENCIES OF VARIOUS SORBENTS

	Breakthrough Volume	es,* 1/g sorbent			
	Benzene	Hexane			
Carbosieve B	47	65			
Tenax GC	3	1.4			
XAD-2 Resin	12	20			
Activated Charcoal	30	43			

^{*}Measured as the volume of gas/grams of sorbent in cartridge to give a 0.1% FID response to gas stream containing 50 ppm of test component.

TABLE II-3. RECOVERY EFFICIENCY OF PURGE-THERMAL STRIPPING OF SELECTED ANALYTES

	TENAX	Carbosieve B	XAD-2
	% Recovery	% Recovery	% Recovery
Benzene	105		
n-C7 ^H 16	100	11	
n-C ₈ H ₁₈	99	<1	
n-C ₉ H ₂₀	94	<1	
n-C ₁₀ H ₂₂	72	<1	62
n-C ₁₁ H ₂₄	67	<1	60
n-C ₁₂ H ₂₆	67	<1	
n-C _{1.3} H ₂₈	58	<1	
n-C ₁₄ H ₃₀	56	<1	
n-C ₁₅ H ₃₂	61	<1	
n-C ₁₆ H ₃₄	46	<1	

Solvent stripping for analyte elution preparatory to chromatographic analysis was investigated. Carbon disulfide $({\rm CS}_2)$ was found to be an attractive solvent. Many of the other common solvents, such as methylene chloride $({\rm CH}_2{\rm Cl}_2)$, chloroform, hexane, benzene, etc., tended to swamp the chromatogram, obliterating any signals of components that have boiling points even decades higher.

Unfortunately, it was found that Tenax GC is soluble in ${\rm CS}_2$ as well as in ${\rm CH}_2{\rm Cl}_2$. Carbosieve B showed poor recoveries with solvents. Testing was therefore primarily focused on solvent extraction of activated charcoal with ${\rm CS}_2$ and XAD-2 resin extraction with ${\rm CH}_2{\rm Cl}_2$ (${\rm CS}_2$ also dissolved XAD-2). Table II-4 presents the results. Mueller and Miller (Ref. II-2) reported similar efficiencies for halogenated and oxygenated hydrocarbons using charcoal adsorption followed by ${\rm CS}_2$ elution. Based on the data they presented and the precedent set by the National Institute for Occupational Safety and Health (NIOSH) in the selection and published (Refs. II-3 to II-5) characterization of the charcoal/ ${\rm CS}_2$ analysis scheme, the use of coconutderived activated charcoal as supplied by Mine Safety Appliances or SKC, Inc. was selected as the material of choice for source sampling.

TABLE II-4. SORBENT RECOVERY EFFICIENCIES FOR NORMAL ALKANES USING SOLVENT ELUTION TECHNIQUES

n-Alkane	Activated Charcoal/CS ₂	Carbosieve B/CS ₂	XAD-2 Resin/ CH ₂ Cl ₂
n-C ₆	97	<1.0	Solvent Masked
n-C ₇	98	<1.0	Solvent Masked
n-C ₈	92	<1.0	Solvent Masked
n-C ₉	87	<1.0	Solvent Masked
n-C ₁₀	90	<1.0	100+
n-C ₁₁	90	<1.0	97
n-C ₁₂	90	<1.0	
n-C ₁₃	100+	<1.0	
n-C ₁₄	76	<1.0	

3. <u>TLV Sniffer</u>—The Bacharach TLV sniffer was selected for use on this program to (1) provide a preliminary estimate of total hydrocarbon emissions, (2) provide an indication of variations in hydrocarbon concentrations in the exhaust gas due to process changes and (3) assist in the quantifying of fugitive emissions. It also served as indicator check on the results attained by GC/MS analysis of fuel samples.

The TLV sniffer is an improved version of a lower-explosive-limit (LEL) detector of combustible organics with an improved sensor and an accuracy greater than the conventional LEL type instruments. It detects hydrocarbon emissions and quantitatively records them in ppm as hexane; however, this read-out can be converted to any specific hydrocarbon or LEL readings. Because it is fire marshall approved, it can be used in refineries or other locations where potentially explosive mixtures exist. It incorporates a contact mass sensor with resistance to catalytic poisonings, an explosion proof potentiometric recorder output, automatic voltage regulation, meter display, sampling pump and a rechargeable battery power source. The system

uses the heat of combustion of the gas-in-air mixture as hydrocarbon sensing. A relative response curve supplied with the instrument permits quantitative measurement of some individual gas species.

B. Sampling Method--

- 1. Train selection—The specific sampling train configuration to be used on a particular source depended on the following factors:
 - . the classes of organic compounds expected in the emissions
 - . the temperature of the emissions
 - . the water content of the emissions
 - . the type of emission flow (i.e., ducted or fugitive).

Table II-5 indicates the sampling equipment used for 17 different source types. For each ducted source the universal sampling train presented earlier in the section titled "Sampling" was adapted as indicated in Table II-5 by the "mode" numbers one through five. Figures II-1, II-2, and II-3 show the first three of these different adaption modes. Modes four and five involve the measurement of fugitive emissions. Figures II-4 through II-7 illustrate the sampling setups for a typical fugitive source, in this case a petroleum transfer line valve. In Figures II-4 and II-5 the setups for a cold valve are shown for two different leak rates while in Figures II-6 and II-7 the setups for a hot valve (T > 160°F) are shown.

Figure II-1 illustrates the train setup for high temperature combustion source sampling. The train filters out particulates at stack temperature, collects andehydes, collects moisture, NOx, SOx and CO on LiOH in two impingers, and collects hydrocarbons by entrapment in a bottle and by adsorption in sorbent tubes.

Figure II-2 illustrates the sample train as used sampling high and low temperature sources with insignificant water vapor content. The train filters out particles, collects andehydes, and collects hydrocarbons by entrapment and by adsorption. It records fluctuation in total hydrocarbon emissions using the TLV sniffer.

TABLE II-5. TEST AND SAMPLING TRAIN CONFIGURATIONS BY SOURCE TYPES

	SOURCE TYPE																
Sampling Train Components Filter and Lines	Refinery X Combustion	Fuel Combustion	x waste Disposal	X Coking X Operation	Catalytic X Burners	Metal (Smelting) Production	Heat Treated X Surface Coating	l O	g	Rubber, Adhesive Production	Hydrocarbon Storage	Degreasing, Stripping	Paint Shops	Dry	Fie duct	Dil, Solvent Transfer, Cold	l, Solvent ansfer, Hot
Unheated								Х	Х	Х	Х	Х	Х	Х	Х	Х	Х
Draeger Gas Indicator		Х	Х	Х	Х	Х	Х	Х	Х	х	Х	Х	х	Х	Х	х	х
Total Hydrocarbon Instr.	х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	Х	х	Х	Х	х	х
Orsat Analyzer		Х	Х	Х	Х	Х											
Aldehyde Bulbs		Х	Х	х	х	х	х		Х	х							
Impinger, LiOH		Х	х	Х	х	Х	Х										
Sorbent Tubes *	х	Х	х	Х	Х	Х	х	Х	Х	Х	х	Х	х	х	х	х	x
Gas Collection Bottle *	x	Х	Х	Х	Х	х	Х	Х	х	Х	Х	Х	х	х	х	х	х
Bags, Metered Flow					Х						х				х	х	
Bags, Rapid Fill	х		Х									х					
Minimum Sample Nos.	5	4	5	4	2 x 5	4	3	2	3	3	3	2	1	1	1	2	1
Velocity Measurement	Х	Х	Х	х	Х	Х	х	х	х	х		Х	Х	Х			
Meter	х	х	х	х	х	х	х	Х	х	х	х				х	х	
Gast Pump	х	Х	Х	Х	х	х	х	Х	х	Х							
Pyrometer (Source T)		Х	х	х	Х	х											
Thermometer (Source T)							Х	х	Х	Х	Х	х	х	х		х	х
Psychrometer							х	Х	Х	Х	Х	х	Х	Х			
Hand or Small Pump											х	х	х	х	х	х	Х
Mode	1	1	1	I	1	1	2	2	2	2	2	3	3	3	4	4	5
Type No.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17

^{*} Either or both.

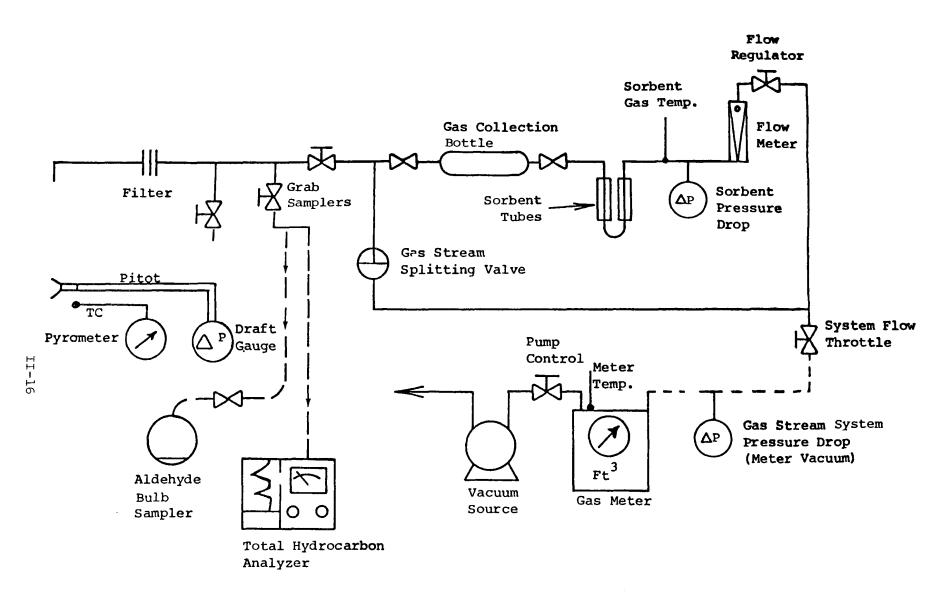


Figure II-2. Organic sampling train configuration for continuous solvent process related sources of high complexity (Mode 2, Table II-4).

Figure II-3 shows the configuration used in sampling cold solvent sources such as dry cleaning, degreasing and painting processes. The train filters particulates, monitors total hydrocarbon emissions fluctuations, and entraps hydrocarbons in gas collection bottles.

Figures II-4 and II-5 illustrate sampling setup for testing fugitive emission sources. The rate of emission is measured, total hydrocarbon concentrations monitored, and gaseous emissions are collected for analysis. In Figure II-4 the H/C leak rate is so great that the vapors fill the tent and drive the gas meter. In Figure II-5 a pump is used to draw purified air through the tent to pick up the emitted H/C vapors.

Figure II-6 and II-7 illustrate test setup for sampling a high temperature fugitive emission source. In Figure II-6 aluminum foil is substituted for polyfilm and rates are measured as Figure II-4 or II-5. When the foil cannot be used the setup in Figure II-7 is used. The temperature of the source is measured, a grab sample is obtained in a gas collection bottle, and the concentration of total hydrocarbons is measured. The leak rate is obtained by applying engineering judgments.

2. <u>Ducted sources</u>—Exhaust gas volumetric flow rates were determined by measurements using EPA Method #1 described in the Federal Register. These measurements were checked by material balance calculations if sufficient source information was available. Before testing, approximation of the gaseous hydrocarbon concentrations was made utilizing a Draeger gas detector with specific indicator tubes, or the TLV sniffer or both.

The ducted sources were sampled at an accessible point closest to the point of average gas velocity. An attempt was made to maintain an isokinetic sampling rate. Sampling time was adjusted according to hydrocarbon concentration to avoid breakthrough on the sorbent tube.

The test data and process data were recorded throughout the test. At the end of the test period the impingers were sealed, labeled and delivered to the laboratory. The sorbent tubes were removed from the train by disconnecting the flexible tubings from them, sealed with polyethylene end caps,

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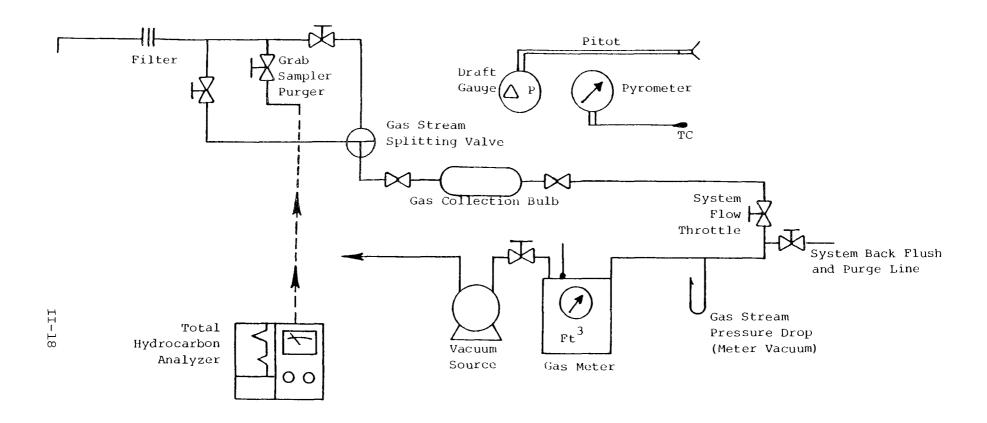


Figure II-3. Organic sampling train configuration for solvent operations in batch operations (Mode 3, Table II-4).

Figure II-4. Leak rate and concentration measurement of ambient temperature fittings. High leak rates. (Mode 4, Table II-4).

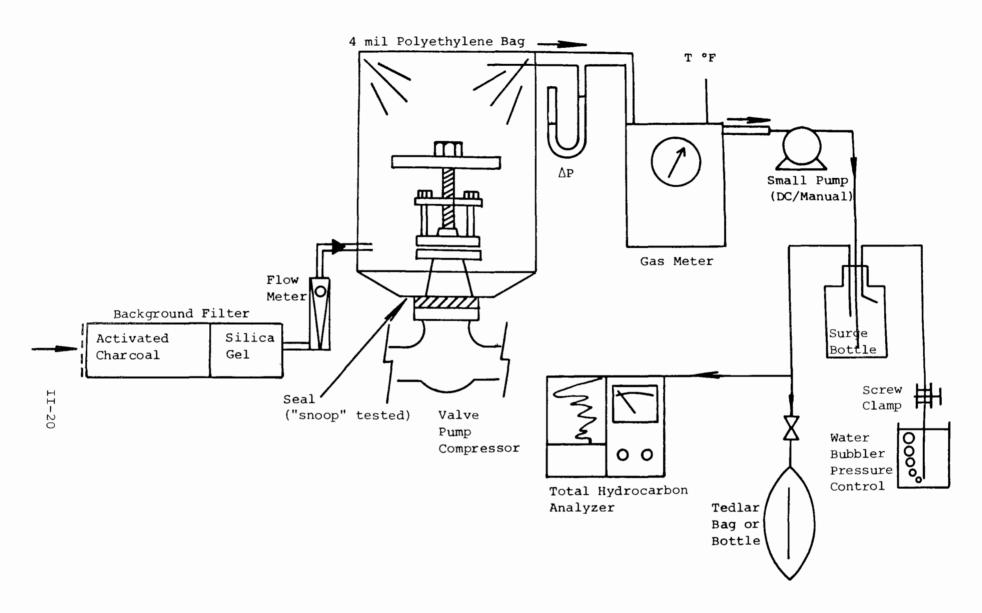


Figure II-5. Leak rate by dilution sweep and sampling of ambient hydrocarbon fitting. Low leak rates. (Mode 4, Table II-4).

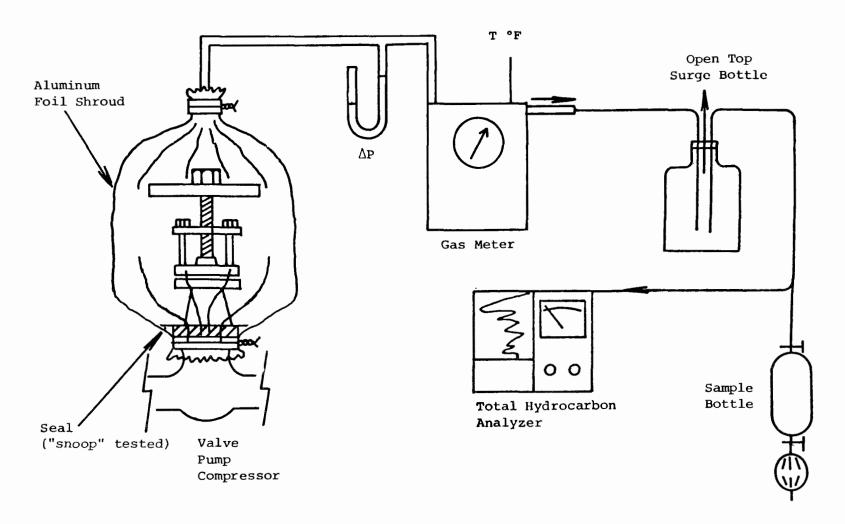


Figure II-6. Leak rate measurement and concentration measurement of high temperature fitting.

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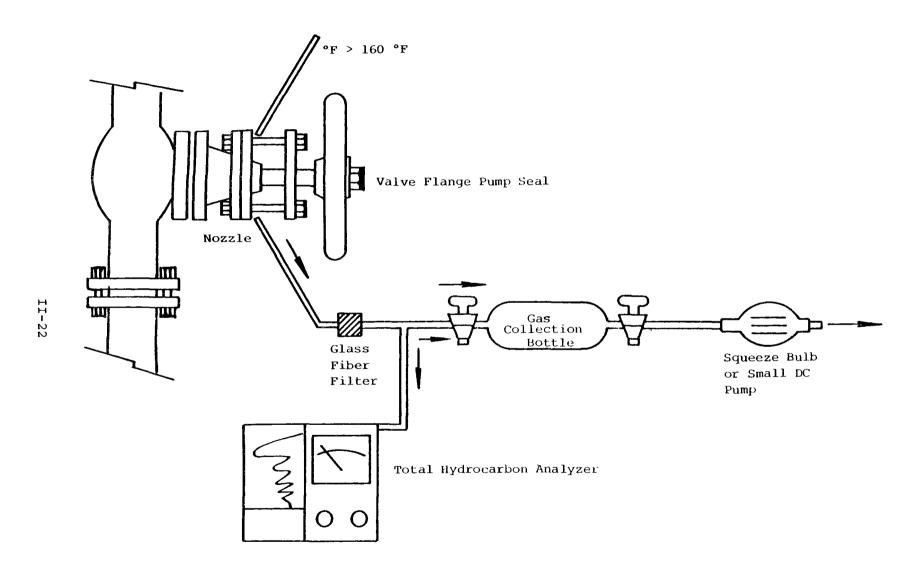


Figure II-7. Hydrocarbon sampling from hot oil or solvent transfer (Mode 5, Table II-4_.

labeled, identified and placed into a shipping container. The gas collection bulbs, bottles and bags were closed, labeled, identified and shipped to the laboratory for analysis.

Wherever possible, a small sample of the process feed and that of the product were obtained for analytical determinations, such as evaporation rate and vapor pressure. These data were used to obtain a material balance.

The TLV sniffer was used to indicate expected or unexpected process fluctuations.

3. Fugitive sources—The measurement of emission rates for non-ducted or fugitive emissions required ingenuity on the part of the test crew. As mentioned earlier, frequently these emissions were estimated or calculated on process data such as solvent make—up rates or on experimental data such as evaporation rates or emission factors for petroleum storage tanks. In certain cases, it was desirable to make selected measurements in order to estimate total emission rates. The most useful techniques for detecting and measuring leak rates involve the use of bubbling soap solutions and tenting with polyfilm sheeting. KVB used this approach in refineries, chemical plants, etc. where leakage losses could not be readily detected from the process flow rates.

The approach used was to usually check all of the accessible hydro-carbon-transfer fittings (valves, flanges, etc.) for signs of leakage (stains, etc.). Next the fittings were checked with soap solution, Figure II-8. Fittings showing leakage with soap solution were categorized as to their estimated leakage rate: low, medium, or high. Depending on the time available and the number of "leakers", a selected, representative number of leakers were tented and their emissions measured.

The test setup for measuring leakage rates is shown in Figure II-5.

The small Spectrex pump pulls a low rate of air through the polyfilm envelope.

The air drawn into the envelope is filtered to remove background hydrocarbon where necessary and is metered with a rotameter as shown. The outlet air

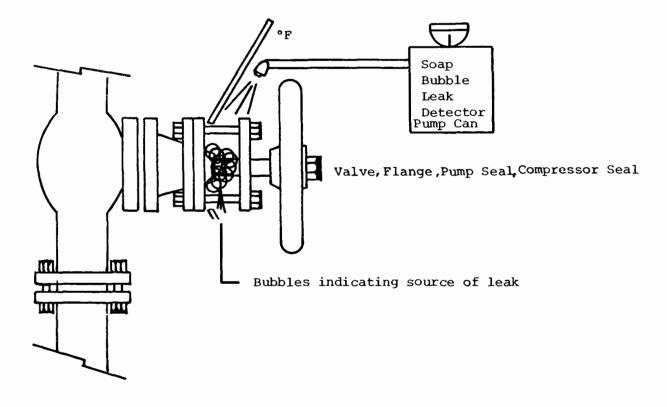


Figure II-8. Soap bubble detection and temperature evaluation of hydrocarbon fittings.

and hydrocarbon mixture is metered and delivered to the TLV analyzer where the total hydrocarbon level is measured continuously. When a steady state has been reached, the TLV analyzer reads a constant ppm level. Readings are taken for several minutes. Then a Tedlar bag of the emissions is taken. The total hydrocarbon leak rate is determined by the following calculations:

$$HC = 1.36 \times 10^{-5} \text{ ppm}_{TLV} \times \frac{ft^3}{t}$$

where

HC = hydrocarbon leak rate, lb/hr

ppm_TLV = parts per million total hydrocarbon concentration
 detected on TLV as hexane

ft³ = meter reading on gas meter corrected to 60 °F and 29.9 in. Hg, in cu. ft.

t = time in minutes ft was measured

1.36x10⁻⁵ppm =
$$\frac{1}{106}$$
 x $\frac{60 \text{ min/hr}}{379 \text{ ft}^3/\text{1b-mole}}$ x 86 (Mwt. of Hexane)

This calculation was checked with the data from the Tedlar bag. The volume of emissions collected in the bag and the filling time of the bag was measured and recorded. The total hydrocarbon content of the bag was determined by GC analysis in the laboratory as well as the specie breakdown and average molecular weight. From this information the total hydrocarbon emission rate was determined to check the results determined by the TLV. The percent composition determined by GC analysis was used to apportion the total hydrocarbon emission rate among the various species.

Based on these measured leakage rates, the leakage rates for other fittings were estimated on the basis of observing their performance during the soap-solution test. KVB also applied the use of the TLV sniffer to determine relative total hydrocarbon emissions from these types of fittings. This proved to be successful and it became particularly valuable for use on hot fittings and on pump seals.

Sample Analysis

The primary analytical chemistry work on this program was performed by Analytical Research Laboratories Inc. (ARLI), Monrovia, CA. Their final report is presented in the appendix of Ref. 5. ARLI assisted KVB in the design of the sampling train, the selection of a sorbent and the design of a quality control system. This section is a summary of the equipment and methods used in analyzing field samples.

Samples received from the field included: 500 ml or 250 ml glass bottles, Tedlar bags, glass tubes containing charcoal sorbent and 100 ml flasks containing 1% sodium bisulfite solution. The bottles, bags and sorbent were analyzed for all organic species while the liquid in the flask was analyzed only for aldehydes.

Most of the gaseous samples in the bottles and bags were analyzed within 2-3 days following receipt, except for a small number that were processed as long as two weeks later. Several tests were made with synthetic samples to evaluate storage effects on the contents of capped charcoal sampling tubes. Recoveries did not change, within experimental error, between 24 hours and 30 days. Therefore, the charcoal samples could stand for longer periods without fear of losses, and were not usually analyzed until after the gas samples in the same sets had been analyzed. The charcoal eluates were usually run within an hour after the carbon disulfide was added to extract the sample components.

Initial analysis of all samples was conducted using a gas chromatograph (GC). Lower boiling component identifications were based on retention times established by repeated analyses of standards. If there were questions as to the positive identity of a GC peak,* the sample was rerun using GC/MS methods for the identification. This approach was often necessary because

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^{*}Conventional gas chromatograph data are recorded on a strip chart with a recording pen which moves literally in proportion to the concentration of the gas being emitted from the GC column. The resultant image on the chart is a peak-shaped trace whose area is proportional to the quantity of the gas present. Thus the term peak is used to refer to an indication of a component of the gas mixture being analyzed.

a number of chromatographic peaks contained at least two and sometimes three components. The mass spectra also provided a basis for determining ratios of the components in the GC peak being examined. These data were then used in making quantitative measurements of the contents of chromatographically unresolved but computer-integratable peaks. All peaks which contributed at lease one volume percent of the total organic vapor were identified and quantified.

A Beckman Model GC-55 equipped with a precision temperature-programmed, column oven and a flame ionization detector (FID) was used for most of the GC work performed on the program. The column was 1/8" O.D. by 6 ft. long stainless steel tubing containing a stationary phase of 100-200 mesh Poropak Q. Using the analytical conditions described below, this column furnished good resolution of the lowest boiling materials encountered while still eluting with good results the higher boiling hydrocarbons representing the top of the range of interest.

Analyses were performed using helium as the carrier gas at a flow rate of 30 cc/min. Detector gas flows were: $\rm H_2$ - 40 cc/min; air - 300 cc/min. The following conditions were used for GC analyses: 6 min. at 40 °C followed by temperature programming at 10 °C/min to 190 °C and holding at 190 °C for approximately one hour.

The effluent from the Beckman GC-55 gas chromatograph was split into two streams. One stream was directed to the FID of the GC, the other to a heated transfer line which carried the stream to a Flinnigan Jet Separator and into the mass spectrometer. The separator provided a twenty fold concentration of the material of interest in the gas stream.

The mass spectrometer used on this program was a Consolidated Electrodynamics Corporation (CEC) Model 21-104. This was a 180 degree magnetic sector instrument having an electron impact ion source and an electron multiplier detector system which permitted moderately highspeed mass scanning.

Multiple MS scans were taken when a GC signal was observed on the strip chart recorder. Multiple scan studies indicated that approximately 2 seconds were required for the maxima to be observed by the MS. Multiple scans were required to insure representative ion pair formation.

Mass spectra were interpreted manually using such reference works as:

- . "Compilation of Mass Spectral Data," Cornu, A. and R. Massot, Heyden & Son, Ltd., London, England, 1966.
- . "Index of Mass Spectral Data," AMD II, Americal Society for Testing and Materials, Philadelphia, 1969.
- . "Eight Peak Index of Mass Spectra," Atomic Weapons Research Establishment, Aldermaston, England, 1970.
- . "Atlas of Mass Spectra Data," Stenhagen, E., et al., Interscience, New York, NY, 1969.
- . "API Project 44 Selected Mass Spectra Data," Thermodynamics Research Center, Texas A&M University.

When an unknown peak could not be positively identified by this means, the spectrum was compared with the mass spectra of some 27,000 different compounds in the library of the Cyphernetics Corp. Mass Spectral Search System. This computerized search system was directly accessible on a time-shared basis. It was successfully used to verify assignments made during the earlier work on this program.

A spectrophotometric method similar to that specified by the NIOSH was used for the determination of aldehydes. The total volume of liquid in the aldehyde sample flasks was measured, and an aliquot taken for the determination. The sample was allowed to react with a modified Schiff's reagent prepared from rosanaline hydrochloride and sodium bisulfite. After a suitable development time, the adsorbance was read at 580 m μ against a reagent blank on a UV-vis spectrophotometer. Concentration was read from a calibration curve. The same determination was performed on a sample of the sodium bisulfite used for collecting/stabilizing the aldehydes and a 1 μ g/ml formaldehyde standard. Results were calculated and reported as total micrograms of formaldehyde equivalent in the sample. The minimum amounts of aldehydes that could be detected by this method were typically 1-3 μ g total (as formaldehyde).

Data Reduction

In the field, the total volumetric emissions from a source were measured. The laboratory analysis provided composition data in the form of weight of individual specie per unit volume (i.e., $\mu g/ml$).

The volumetric measurements of ducted sources were made using EPA Method 1 as described in the Federal Register, Volume 36, Number 159, August 17, 1971. Standard conditions used in all calculations were 60 °F and 29.95 in. of mercury pressure. Gas density correction factors were based on Chapter 3 of the "Source Testing Manual" published by the Air Pollution Control District of Los Angeles County, 1972 (now SCAQMD Metro Division).

Combustion source flow rates were measured by Method 1 and checked using Orsat analysis and combustion calculations based on fuel analysis and process data pertaining to the source tested. The method used is described in detail in Section 5.4 of the "Source Testing Manual".

The reported hydrocarbon concentrations were calculated from the laboratory results as follows:

The hydrocarbon concentration provided by the laboratory was on a dry, ${\rm CO}_2$ -free basis. This was converted to actual moisture and ${\rm CO}_2$ conditions at the source by the relation (page B of Table 3-7),

$$c_{HC} = (c_{HC})_d \frac{(100 - W.V. - CO_2)}{100}$$

where,

CHC = concentration of hydrocarbons at source (actual) conditions, parts per million by volume

(c_{HC})_d = concentration of hydrocarbons; dry, CO₂-free basis, from the analysis; parts per million by volume

W.V. = water vapor in source gases, percent by volume

CO₂ = carbon dioxide in source gases, stack conditions, percent by volume

The water vapor concentration was measured during the source test using an ice water impinger or Aquasorb tube. The carbon dioxide concentration was obtained from Orsat analysis (dry basis), converting to stack conditions by multiplying by the factor (1-M.W./100).

The reported emission rate of hydrocarbons was calculated by the general relation,

$$M_{HC} = 1.58 \times 10^{-7} c_{HC}^{Q(M.W.)}_{HC}$$

where,

 M_{uc} = emission rate of hydrocarbons, pounds per hour

Q = stack gas flow rate, standard cubic feet per minute (SCFM)

 $(M.W.)_{HC}$ = molecular weight of hydrocarbon

The 1.58×10^{-7} factor came from the following unit analysis:

$$M_{\text{HC}} \left(\frac{1b}{\text{hr}} \right) = c_{\text{HC}} \left(\text{ppm} \right) \times \left(\frac{\text{ppm}}{106} \right) \times Q \left(\frac{\text{ft}^3}{\text{MW}} \right) \left(\frac{60 \text{ min}}{\text{hr}} \right) \times \left(\text{M.W.} \right)_{\text{HC}} \left(\frac{1b}{\text{mole}} \right) \left(\frac{1 \cdot \text{mole}}{379 \text{ ft}^3} \right)$$

$$M_{HC} \left(\frac{1b}{hr}\right) = c_{HC} Q (M.W.)_{HC} [(1/10^6) \times (60/379) \times (1b/hr)]$$

= 1.58×10⁻⁷ $c_{HC} Q (M.W.)_{HC} (1b/hr)$

QUALITY CONTROL

A comprehensive quality control program was conducted as an integral part of the overall organic-emission field tests. The program featured:

- Calibration of field test instruments with ASTM methods and NBS standards
- Frequent response-factor calibration of laboratory instruments
- 3. Interlaboratory checks for accuracy

- 4. Concurrent samples taken from the same source with separate but identical trains for precision checks
- 5. Separate total organic content analysis to backup the GC analysis
- 6. Unannounced "blanks" of zero gas, calibration gas, etc.
- 7. An independent QC consulting team.

From the outset KVB engaged the services of three experts in the field of organic analysis, Drs. James N. Pitts, Jr., Daniel Grosjean and Barbara Finlaysen-Pitts working as a team from EcoScience Systems Inc. (ESS). This team participated in the initial evaluation of the sampling equipment and analytical methodology and defined a quality program with the above mentioned features. The special QC tests (duplicates, blanks, round robins, etc.) accounted for approximately 10% of the test budget and afforded an excellent assessment of measurement error which was as follows:

- 1. The total hydrocarbon emissions were good to within + 25%.
- 2. Values for the emissions of individual hydrocarbons, however, were less certain than that for total hydrocarbons.
- 3. The sum of the errors in sampling and analyses for individual alkanes probably was in the range of 25-50%.
- 4. The concentrations of oxygenates, aromatics and halogenates must be considered lower limits only with the possible error being a factor of three or more.

In addition to the above numerical assessment ESS concluded, "In summary then, the most feasible and reliable field sampling and laboratory analytical techniques were employed in this program to yield accurate source emissions data. The latter can be confidently applied to the development of a hydrocarbon emission inventory for stationary sources in the SCAB with 'state-of-the-art' accuracy and precision."

The ESS report is presented in the Appendix. Their report contains not only a discussion of selected data which lead to the above numerical error assessment but copies of comparative test data calibration data report forms documenting blank sample analyses results, etc.

Presented below are some selected data which illustrate the basis of the conclusions reached by ESS.

Round Robin and Sample Recovery Tests

Before beginning the field tests an analytical program was conducted to establish assurance in sampling and analysis for stationary pollution source studies. This included an evaluation of field sampling equipment, laboratory gas sampling, instrument variability, standardization of gas chromatographic analysis columns, accuracy and precision of data. To test these parameters, four calibration gases were procured in "K" bottles from Precision Gas Products, Inc. including selected (1) aliphatic hydrocarbons, (2) aromatic hydrocarbons, (3) oxygenated organics and (4) halogenated organic compounds. Concentrations were specified and controlled by the KVB program manager who retained certifications until analyses were performed.

Upon receipt of the four "K" bottles, three sets of samples were prepared in 250 ml gas collection bottles by KVB and delivered to ARLI, the SCAPCD laboratory in Los Angeles and the ARB laboratory in El Monte. (Only the aliphatic and aromatic samples were sent to the ARB.) The results of these analyses are presented in Tables II-6 through II-9. Added to each standard as a control compound was hexane selected because of its unreactive nature and low adsorptivity.

Two of the sampling trains shown in Figure II-1 were used to collect samples of the four calibration gases using the setup shown in Figure II-9. Results of the analyses of these samples are compared to other analyses of these calibration gases using various handling and analysis methods are presented in Tables II-10 through II-13 and discussed below.

Recovery studies using the aliphatic hydrocarbon standard indicated some limitations in sampling or analysis. For example, when the collection train was used as shown in Case 5 of Table II-10 32 liters of gas at the selected flow rate of 3 liter/minute showed a breakthrough of low molecular weight hydrocarbons on charcoal. This, of course, did not affect sample collection because the intended use of the adsorbents were for compounds

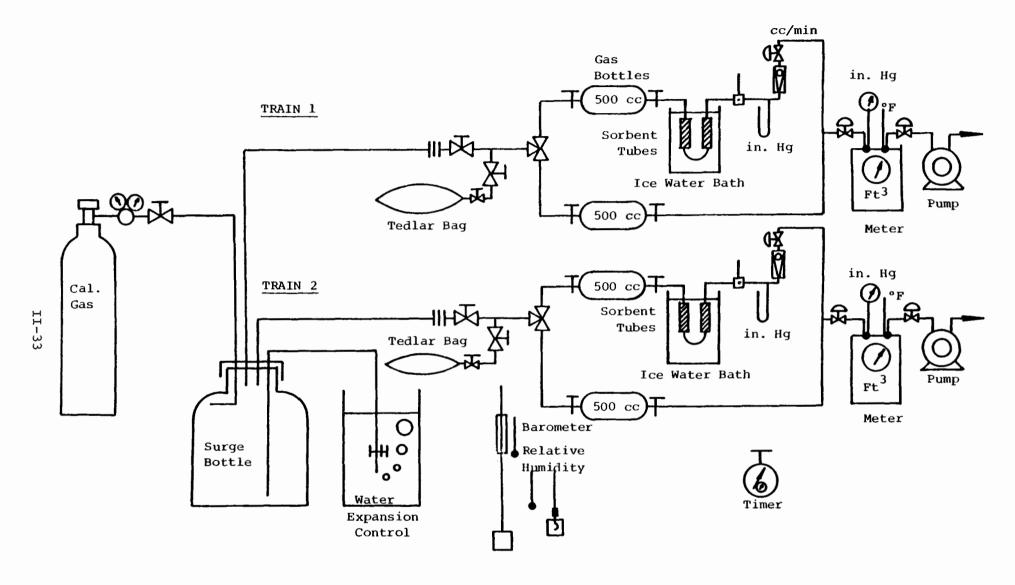


Figure II-9. Sampling trains in QC test with calibration gas.

TABLE II-6. QUALITY ASSURANCE ALIPHATIC STANDARD All data are reported in ppm in nitrogen.

	Certified	Analy	tical Resu	ılts
Compound	Contents*	ARLI	SCAPCD	CARB
Ethylene	51	50	21	D§
Propane	53	47	27	32
Propylene	53	57	32	30
1,3-Butadiene	51	3#	28 ⁺	28
Isobutane	NR	$_{ m NR}^{\ddagger}$	NR	0.4
Hexane	50	50	30	40
Heptane	51	4 6	24	D
Isooctane	50	43	26	D

^{*}Analytical information prepared by Precision Gas Products Co.

⁺ Identified as 1-butene

 $[\]S_D$ = Identified but not quantified (insufficient sample)

^{*}The ARLI recovery of butadiene was significantly below the certified composition. This was apparently caused during thermal desorption treatment of the sample bottle and transfer equipment. SCAPCD and CARB labs are believed not to have heated the sample bottles.

[†]NR = not reported

TABLE II-7. QUALITY ASSURANCE AROMATIC STANDARD

All data reported in ppm in nitrogen.

	Certified	Anal	Lytical Resu	ılts
Compound	Contents	ARLI	SCAPCD	CARB
Benzene	52	50	15	34
Toluene	48	46	11	34
Ethylbenzene	49	51	5	20
Chlorobenzene	50	51	3	NR
Hexane	49	50	20	35
Xylene	NR	NR	3	D

NR = Not reported D = detected but not quantified

TABLE II-8. QUALITY ASSURANCE HALOGENATED HYDROCARBON STANDARD

All data reported in ppm in nitrogen.

	Certified	Analytic	al Results
Compound	Contents	ARLI	SCAPCD
Freon 113	48	47	22
He xa ne	50	50	5
Methyl Chloride	55	48	NR
<pre>1,1,1 Trichloroethane (Methyl chloroform)</pre>	50	46	37
Chloroform	49	46	49

NR - Not reported

TABLE II-9. QUALITY ASSURANCE OXYGENATED ORGANICS STANDARD

All data reported in ppm in nitrogen.

	Certified	Analytic	al Results
Compound	Contents	ARLI	SCAPCD
Methanol	53	4	NR
Acetone	52	43	7
Isopropanol	NR	NR	9
Methyl Ethyl Ketone (2-Butanone)	51	45	2
Methyl Isobutyl Ketone	48	42*	20
Hexane	48	50	37

^{*} Identified as 2-hexanone

TABLE II-10. RECOVERY STUDIES USING THE ALIPHATIC STANDARD

Analysis by Gas Chromatography - Porapak Q Column Temperature Programmed Reported as ppm of Component in Nitrogen

Compound	Case 1	Case 2	Case 3	Case 4			Case 5		
					Charcoa	1 Tube		al Tube	
				Grab	Front	Back	Grab	Front	Back
Ethylene	51	50	46	33	0	0	29	0	0
Propane	53	47	47	37	14	0	32	12	8
Propene	53	57	48	30	14	0	27	10	9
1,3-Butadiene	51	3	33	8	25	0	8	24	(0.2)
n-Hexane	50	50	49	35	46	0	33	35	1
n-Heptane	51	46	49	36	43	0	36	44	0
Isooctane (2 Methyl Heptane)	50	43	<u>45</u>	_34	43	0	32	45	0
Total Recovery	359	296	317	213	1	85	197	1	98

- Case 1. Certified contents gravimetrically prepared.
- Case 2. Glass-Teflon grab sample, GC analyzed on receipt; sample bottle and transfer lines heated during sampling and analysis.
- Case 3. Glass-Teflon grab sample sampled and analyzed without heating during transfer process.
- Case 4. Glass-Teflon grab sample and activated carbon sorption tube using field sampling train first test. 27.7 liters of standard gas sampled.
- Case 5. Same sampling conditions as for Case 4 second test. 32 liters of standard gas sampled.

TABLE II-11. RECOVERY STUDIES USING THE AROMATIC STANDARD

Analysis by Gas Chromatography - Porapac Q Column Temperature Programmed
Reported as ppm of Component in Nitrogen

Compound	Case 1	Case 2	Case 3		Case 4	Case 5		
				Grab	Charcoal Tube	Grab	Charcoal Tube	
n-Hexane	49	50	40	43	45 <u>+</u> 2	44	45 <u>+</u> 3	
Benzene	52	50	43	47	45 <u>+</u> 2	47	46 <u>+</u> 4	
Toluene	48	46	26	28	45 <u>+</u> 1	23	47 <u>+</u> 3	
Ethylbenzene	49	51	23	25	46 <u>+</u> 1	20	49 <u>+</u> 3	
Chlorobenzene	50	51	45	48	37 <u>+</u> 1	46	39 <u>+</u> 3	
Total Recovery	248	248	177	191	218 + 2	180	226 + 3	

- Case 1. Certified contents gravimetrically prepared.
- Case 2. Glass-Teflon grab sample, GC analyzed on receipt; sample bottle and transfer lines heated during sampling and analysis.
- Case 3. Glass-Teflon grab sample sampled and analyzed without heating during transfer process.
- Case 4. Glass-Teflon grab sample and activated carbon sorption tube using field sampling train first test. 30 liters of standard gas sampled.
- Case 5. Same sampling conditions as for Case 4 second test. 28 liters of standard gas sampled.

TABLE II-12. RECOVERY STUDIES USING THE HALOCARBON STANDARD Analyzed by Gas Chromatography - Porapak Q Column Temperature Programmed Reported as ppm of Component in Nitrogen

Compound	Case 1 Case 2			Case 3		Case 4	Case 5	
			Grab	Charcoal Tube	Grab	Charcoal Tube		
Hexane	50	50	44	33	40	38	49 <u>+</u> 3	
Methylene chloride	55	48	<0.1*	<0.1	<0.1	<0.1	50 <u>+</u> 3	
Chloroform	49	46	15	28	19	31	33 <u>+</u> 1	
Methylchloroform (1,1,1-Trichloroethane)	50	46	<1	30	<1	34	41 <u>+</u> 2	
Freon 113 (1,1,2-Trichloro - 2,2,1-								
Trifluoroethane)	48	47	37	21	34	28	33 <u>+</u> 3	
								
Total Recovery	252	237		112		131	206 + 3	

^{*}MeCL, masked by solvent.

- Case 1. Certified contents gravimetrically prepared.
- Case 2. Glass-Teflon grab sample analyzed on receipt of standard sample bottle and transfer lines heated during sampling and analysis.
- Case 3. Glass-Teflon grab sample and activated carbon sorption tube using field sampling train first test. 24 liters of standard gas sampled.
- Case 4. Same as Case 3, second test. 28 liters of standard gas sampled.
- Case 5. Glass-Teflon grab sample analyzed approximately 45 days after sample receipt. Heated transfer lines and heated grab sample during analyses.

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TABLE II-13. RECOVERY STUDIES USING OXYGENATED ORGANICS STANDARD

Analysis by Gas Chromatography - Porapak Q Column Temperature Programmed Reported as ppm of Component in Nitrogen

Compound	Case 1	Case 2	Case 3	3 Case 4 Cas		Case 5	
				Grab	Charcoal Tube	Grab	Charcoal Tube
Hexane	48	42	Recovery*	*	39	*	47
Methyl Alcohol	53	4	Less	*	<1	*	<1
Acetone	52	43	Than	*	25	*	28
Methyl Ethyl Ketone (2-Butanone)	51	45	ppm Level	*	39	*	47
Methyl Isobutyl Ketone	48	42		*	_33	*	39
Total Recovery	252	176			136		161

^{*}The oxygenated material could not be displaced from the sampling container without heating because of adsorption or moisture and absorption.

- Case 1. Certified contents gravimetrically prepared.
- Case 2. Glass-Teflon grab sample analyzed by GC on receipt of standard mixture. All transfer lines and sampling system heated.
- Case 3. Same as Case 2 without heating.
- Case 4. Glass-Teflon grab sample and activated carbon sorption tube using field sampling train first test. 28 liters of standard gas sampled.
- Case 5. Same as Case 4, second test. 23.8 liters of standard gas sampled.

boiling above 80 °F. Low values reported for butadiene seemed to indicate polymerization or decomposition. The analytical data for Case 2 showed a recovery of only 3 ppm when the sampling container was heated to approximately 120 °F during transfer to the chromatograph. Case 3 of the gas sample was transferred to the GC loop without heating (33 ppm of butadiene was recovered). Because ARLI and the other referee laboratories found approximately 30 ppm of butadiene in all tests, it can be assumed that the gas phase of the Precision Gas Standard contains only this amount. Either wall adsorption or catalytic polymerization within the "K" bottle could explain the difference between the analyzed and gravimetrically prepared material. The accuracy observed for sorbent collected hydrocarbons above C_5 was within experimental error. It was also indicated that heating of the glass grab sample containers to 100 °F would maximize recovery of the lower hydrocarbons, $< C_A$.

Table II-11 shows the data obtained on the aromatic hydrocarbon gas standard sample. These data showed that accuracy could be achieved within the limitations of analytical repeatability. Warming the grab sample bottle of toluene and ethylbenzene appear to improve the yield. Unexplained retention of chlorobenzene on the charcoal was observed although 80% recoveries of higher molecular weight compounds are considered acceptable by most laboratories and government agencies. No corrections were made for hydrocarbon sampling to actual and projected stack emissions.

Table II-12 reflects the analytical studies made on halocarbon gas mixtures. These data are not consistent with the accuracy and reproducibility of the hydrocarbon data. It would appear that for the higher molecular weight halogenated materials, between 20 and 30% losses occurred simply on standing in the metal "K" bottle. Case 5 of Table II-12 illustrates this condition. The loss of methylene chloride in grab sampling appeared to be real although in ARLI's previous experience with analyzing for traces of methylene chloride there had been no problem. Methylene chloride elutes with carbon disulfide from a Porapak Q column. Therefore, no accurate measure of concentration could be made. The reported value of < 0.1 ppm appeared to be a small shoulder on the solvent peak and was probably an impurity in one or both of the compounds.

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Table II-13 presents the results obtained for oxygenated organic materials. If it can be assumed that the amount of methyl alcohol found in the "K" bottle by analysis, restandardization, and reanalysis (as was actually performed in the laboratory) were correct, the recoveries were within 20%. Methyl ethyl ketone elutes with hexane from the Porapak Q column. However, from other recovery data, standards and grab samples of the Precision Gas mixture, and response factors applied to the measured area, the calculated concentration reported under Case 4 and 5 agree within the experimental limits.

The oxygenated materials were strongly adsorbed on the glass bottle walls. This was apparent in the data present for the grab-train samples of Case 4 and 5 as well as the ambient grab sample of Case 3. All field grab samples that were expected to contain oxygenated materials were warmed and the transfer lines maintained at an elevated temperature injection into the GC.

Interlaboratory Field Sample Analysis

During the test of the Huntington Beach oil field emissions WOGA used a consulting firm, RETA, to monitor KVB's procedures. RETA collected duplicate samples of gas in grab bottles which were analyzed by the Union Oil Research Laboratory in Brea, CA. A comparison of the KVB-ARLI results with those of RETA-Union are shown in Tables 3-20 through 3-25, Ref. II-6.

Although these samples were taken sequentially rather than simultaneously, the results in these tables show good agreement for total hydrocarbons as well as for the two major components methane and ethane. For example deviations from the mean methane concentration ranged from -28% to +10% the average deviation being 13%. The agreement was consistent over a range of concentrations from under 700 to over 800,000 ppm. The greatest discrepancies percentagewise were in the low concentration species. This latter phenomenon was generally observed throughout the QC program.

Redundant Field Samples

On selected sources two or three independent trains were inserted into the source for simultaneous sampling. The analyses of the samples were performed without informing the laboratory of identical nature of the sources. The results of some of these tests are presented in Tables 3-26 through 3-30, Ref. II-6.

SOLVENT USE QUESTIONNAIRES

Solvent use questionnaires were processed primarily to obtain information on the quantities and specific types of solvents used in the South Coast Air Basin during 1975 for use in developing emission profiles and factors. Other information on process equipment, control devices, future plans for modifications or expansion were also requested for checking the EIS data and forecasting emission trends.

KVB prepared the questionnaires using modified formats from Reference II-7. Sections were specifically directed to the following solvent users:

- . Degreasing Operations
- . Dry Cleaning Operations
- . Protective and Decorative Coatings
- . Fabric and Rubberized Coatings
- . Miscellaneous Coatings
- . Ovens
- . Printing Operations

Data were requested on an individual device basis using actual process records or best estimates from total plant consumption. Provisions for confidential or proprietary information were made.

Individual questionnaires were mailed to approximately 200 plants within the Basin with reported total organic compound emissions from solvent use in excess of 25 tons per year. Only the questionnaire forms that would pertain to the individual source operation were included. The response to these questionnaires was only 25% despite follow-up by telephone.

Of those returned, the quality and completeness of the information as excellent providing detailed information on solvent composition and perating parameters previously not available. Data were obtained from a ide cross section of industries which added depth to the data base. These ata, along with updated solvent use patterns for those firms visited during he field test program have been incorporated into the development of the pplicable solvent emission profiles.

ITERATURE RESEARCH

profile data not obtainable through field source tests or solvent use uestionnaires was derived through an extensive survey of pertinent literature. here appropriate, discretion was used by KVB engineers in their engineering valuation of these data. Numerous consultations were also conducted with uthors of reports, industry representatives and various government agencies n the development of profile data from the literature.

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APPENDIX III PROFILE CALCULATIONS DEVELOPMENT

APPENDIX III

SAMPLE PROFILE CALCULATION FOR CALCULATING AVERAGE MOLECULAR WEIGHT

Given the species in any compound, the average molecular weight may be calculated knowing the individual specie's molecular weight and weight percent.

EXAMPLE: KVB PROFILE 0001

GIVEN:

<pre>n-butane with molecular weight =</pre>	58.12 and	14.00 weight %
hexane	86.17	5.00
formaldehyde	30.03	42.00
acetone	58.08	28.00
methane	16.04	11.00

Determine the number of mols of each component by dividing the weight (assume the mixture weighs 100 units so that 14% = 14.00 weight units) by the molecular weight.

n-butane	= 14.00/58.12	=	0.241 mols
hexane	5.00/86.17		0.058
formaldehyde	42.00/30.03		1.399
acetone	28.00/58.08		0.482
methane	11.00/16.04		0.686

2. The total weight of the mixture is 2.866 mols.

3. Since volume % = mol %, the volume percent of each component is determined by dividing the total weight of the mixture into each component's number of mols, or

```
n-butane = 0.241/2.866 \times 100\% = 8.41 \text{ vol.}
hexane 0.058/2.866 \times 2.02
formaldehyde 1.399/2.866 \times 48.81
acetone 0.482/2.866 \times 16.82
methane 0.686/2.866 \times 23.94
```

4. The average molecular weight of the paraffins is the sum of the volume percent of each individual component's molecular weight in that class.

5. The average molecular weight of the compound composite is the sum of the volume percent of each individual component's molecular weight in the compound.

	5	compound	compos	site	III-	2	34.90
methane			16.04	(0.2	2394)	=	3.84
acetone			58.08	(0.1	L682)	=	9.77
formaldehyde			30.03	(0.4	1881)	=	14.66
hexane			86.17	(0.0	202)	=	1.74
n-butane			58.12	(0.0	841)	=	4.89

TECHNICAL R	FPORT DATA	
(Please read Instructions on the	ne reverse before completing)	
1. REPORT NO. 2.	3. RECIPIENT'S ACC	ESSION•NO.
EPA-450/3-78-119		
4. TITLE AND SUBTITLE	5. REPORT DATE	
	<u>December 197</u>	8
Volatile Organic Compound (VOC) Species Dat	a Manual 6. PERFORMING OR	GANIZATION CODE
7. AUTHOR(S)	8. PERFORMING OR	GANIZATION REPORT NO.
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Research Triangle Park, NC 27711		
15. SUPPLEMENTARY NOTES		
EPA Project Officer: Frank M. Noonan		
16. ABSTRACT		
This manual contains tables of potential em	dissions of organic compour	ds for selected
source categories. The species profile tab	le format has been organiz	ed to be particu-
larly useful in preparation of emission inv	ventory inputs to photochem	nical modeling.
Accompanying each VOC profile table is a br	siof namnative that describ	nes process
Accompanying each voc profile table is a br	net and data quantification	n The chemical
emissions, controls and basis of source rep	ort and data quantification	les and ketenes
classifications include paraffin, olefin, a	romatic, carbonyi (aldenyo	ies and ketones),
methane, non-reactive other than methane, a	ind miscellaneous. Data co	ontidence levels
for each profile table have been assigned.	Reference lists for repor	rts, published
data and names and titles of personal conta	icts are provided for each	source category.
17. KEY WORDS AND DO	CUMENT ANALYSIS	
a. DESCRIPTORS	b.IDENTIFIERS/OPEN ENDED TERMS	c. COSATI Field/Group
Valadala Ourreda Compoundo (VOC)		
Volatile Organic Compounds (VOC)		
Chemical Species Distribution	 	
Emission Inventories (Air)		
Photochemical Model Input		
Stationary Sources		
Mobile Sources		
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