



## Air Pollutant Emissions and Possible Health Effects Associated with Electronic Air Cleaners

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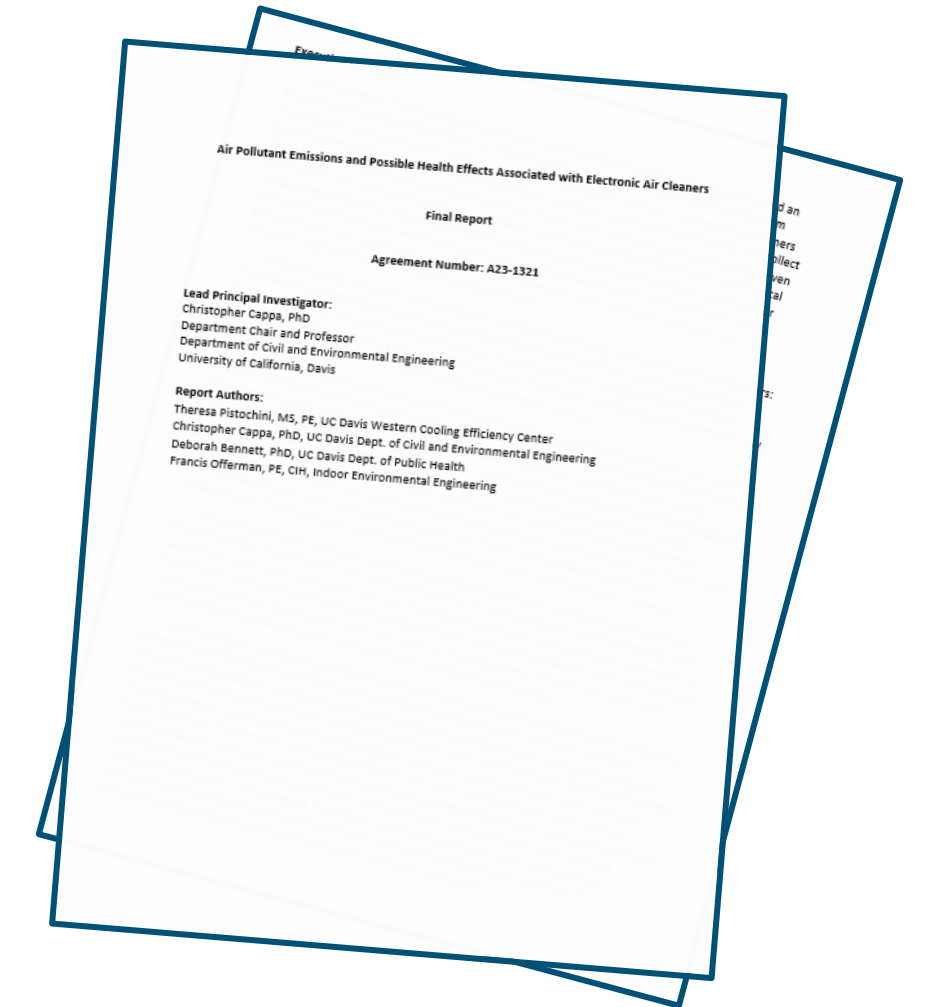
# Whitepaper Published

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<https://ww2.arb.ca.gov/sites/default/files/2023-09/CARB%20White%20Paper%20Final.pdf>

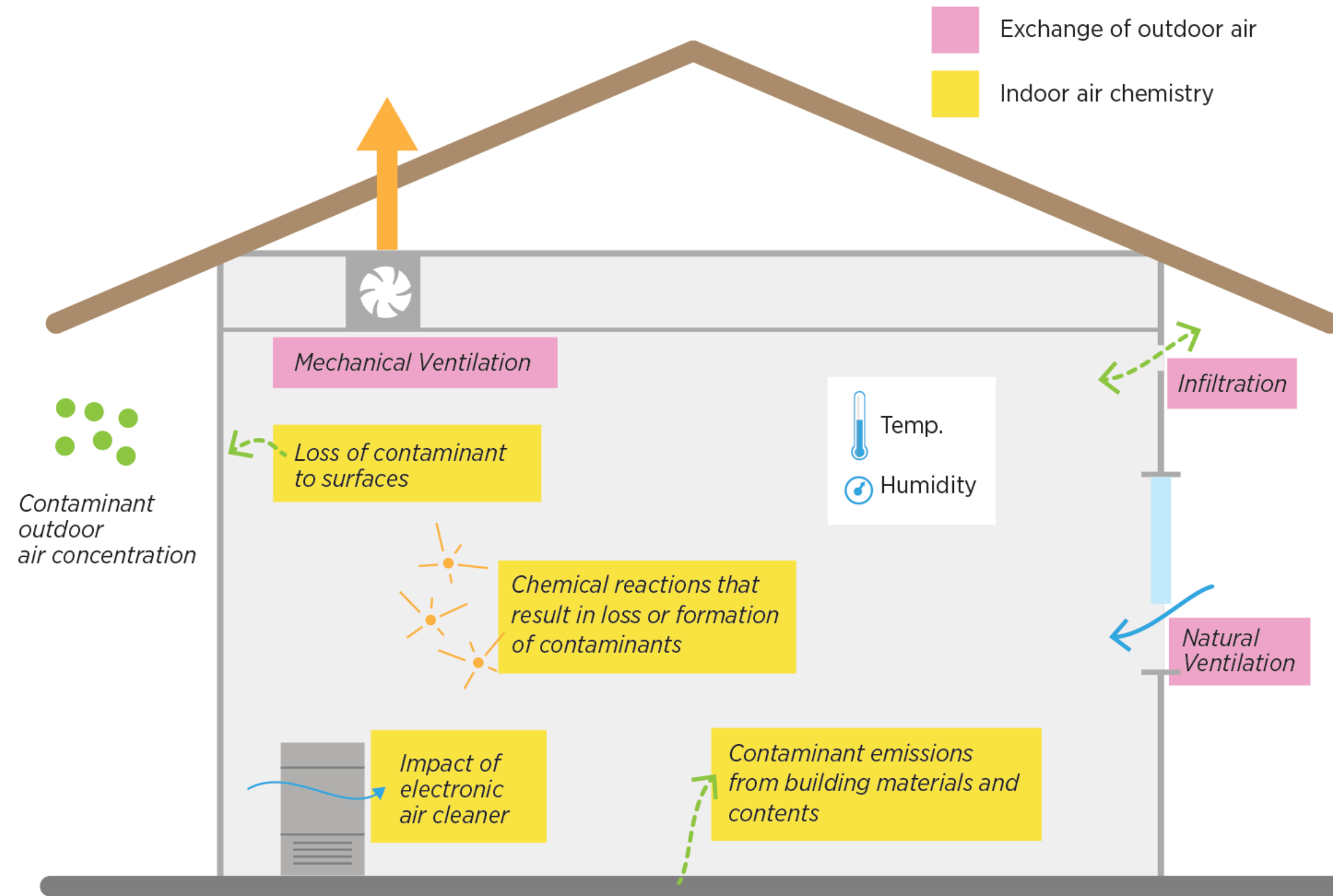
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# Purpose

- » Electronic air cleaners may emit reactive compounds or promote the formation of chemical byproducts which may have adverse health effects.
- » We will summarize for electronic air cleaners:
  - ✓ Types and operating principles
  - ✓ Reactive compounds and byproducts emitted or formed
  - ✓ Human exposure pathways and potential adverse health effects
  - ✓ Test standards and regulations regarding emissions
  - ✓ Gaps in understanding of the risks and recommendations for future research
- » Not included:
  - ⊘ Efficacy of electronic air cleaners for contaminant removal or inactivation of biological contaminants.



# Varied Nature of Indoor Environment



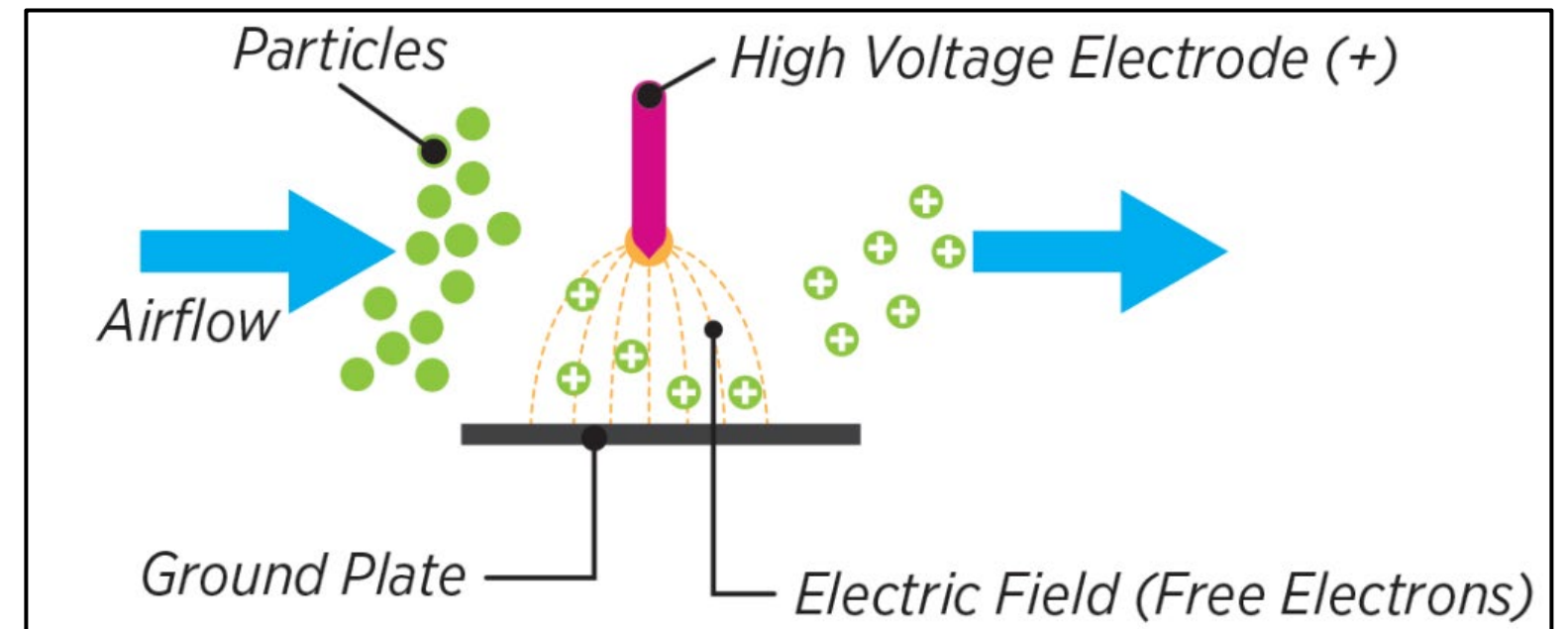
» Air cleaner emissions and byproducts impacted by the environment

# Electronic Air Cleaner (EAC) Types and Operating Principles

# Type: Ion Generator (IG)

*Also known as: unipolar/bipolar needlepoint ionization, corona discharge, plasma generator*

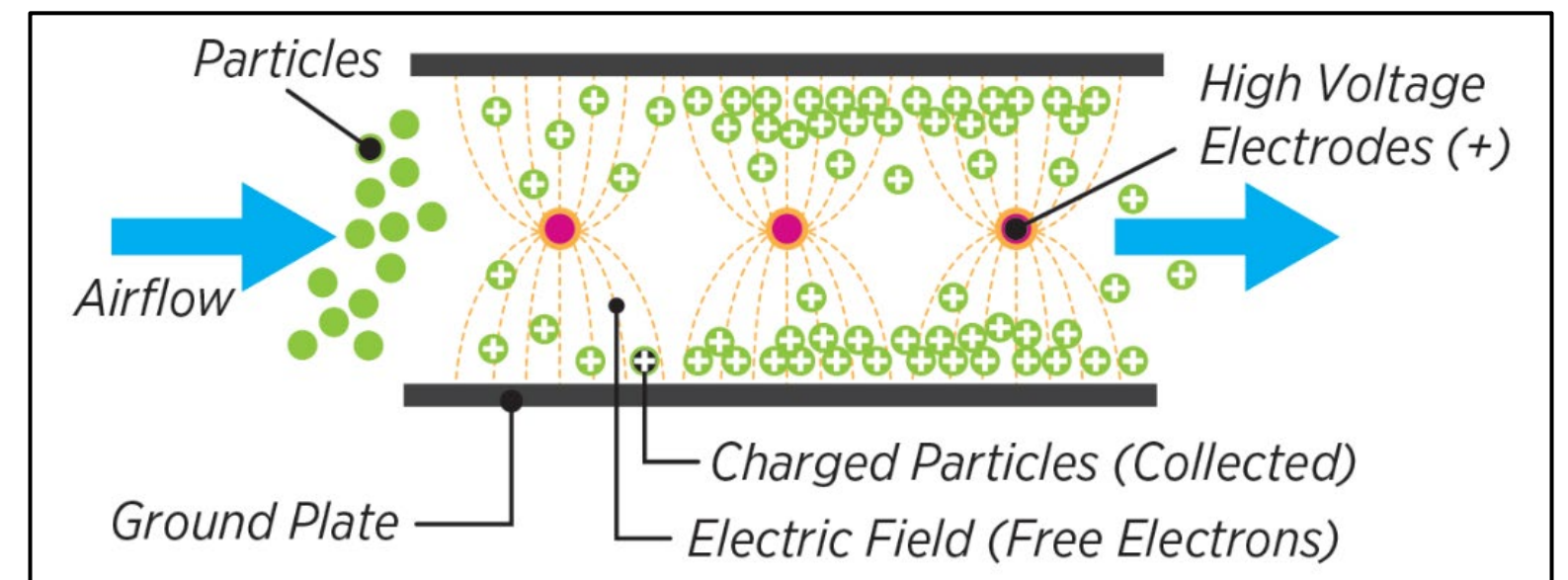
- » Impart a charge on gas molecules (i.e. ions) to charge particles and increase removal by filters or deposition onto room surfaces
- » High voltage to electrode creates an electric field that results in the air becoming electrically conductive
- » Positive electrode – positive ions. Negative electrode – negative ions
- » Ionization energy can also initiate chemical reactions in indoor air
- » Electrodes ideally designed to minimize ozone production



*Ion generator working principle*

# Type: Electrostatic Precipitator (ESP)

- » Working principle similar to an ion generator
- » High voltage electrodes in the form of wires
- » Active removal of charged particles → plates of opposite charge from the electrodes attract & collect charged particles
- » Plates need to be cleaned periodically
- » Ionization energy can also initiate chemical reactions in indoor air

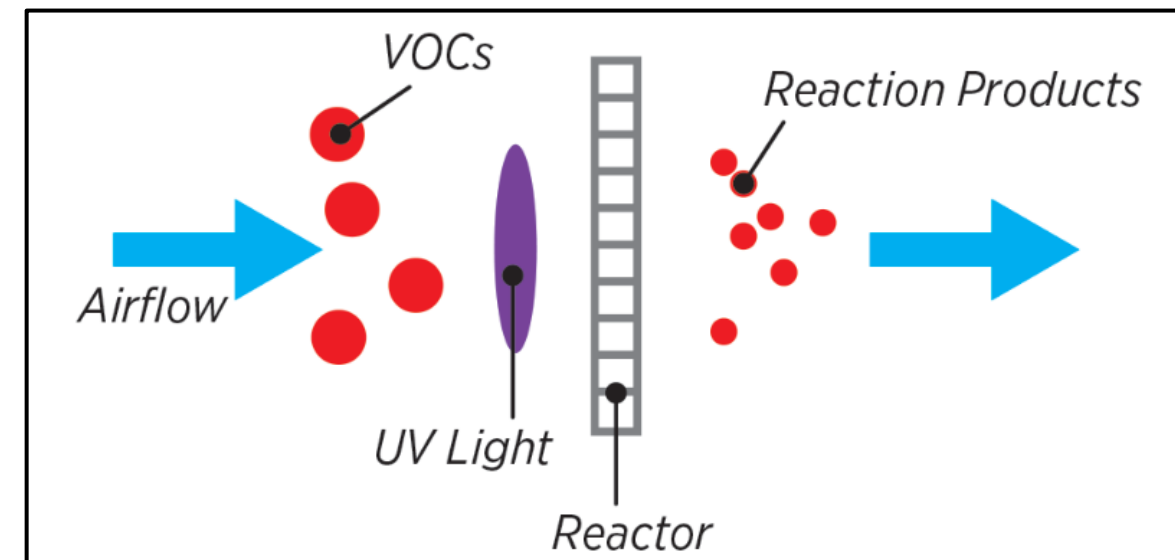


*Electrostatic precipitator working principle*

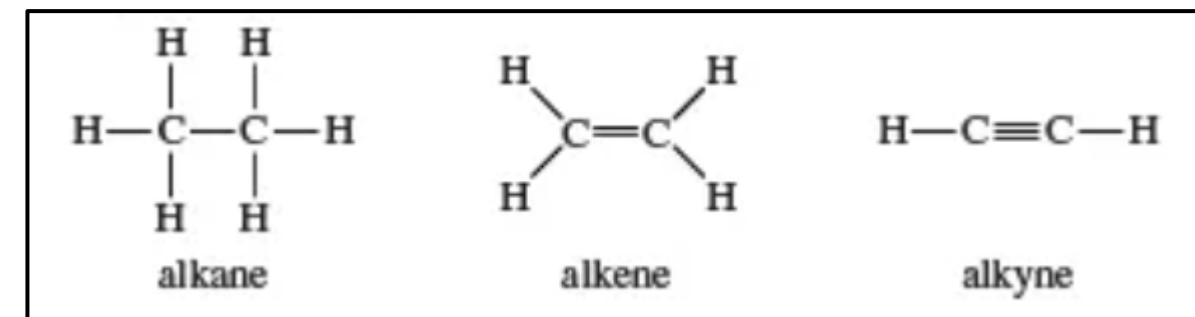
# Type: Photocatalytic Oxidation (PCO)

*Also known as: dry hydrogen peroxide technology*

- » Designed to break down volatile organic compounds (VOCs) (i.e., hydrocarbons)
- » Ultraviolet (UV) radiation shines on reactor (i.e., photocatalyst) to break down water in the air and form hydroxyl radicals ( $\bullet\text{OH}$ )
- » Hydroxyl radicals react with VOCs to decompose them to carbon dioxide ( $\text{CO}_2$ ) and water ( $\text{H}_2\text{O}$ ) in a complete reaction
- » Achieving a complete reaction is challenging and byproducts (partially decomposed VOCs) may be produced
- » The UV light can initiate photolysis reactions, including ozone production



*Photocatalytic oxidation working principle*



*Example VOCs (i.e. hydrocarbons). PCO aims to break down the HC bonds to  $\text{CO}_2$  and  $\text{H}_2\text{O}$ .*



# Type: Ultraviolet Germicidal Irradiation

- » UVC 100– 280 nm wavelength most common
- » Two types of approaches:
  - Enclosed devices that protect against human exposure
  - Devices that irradiate occupied spaces
- » Safety of human exposure and disinfection performance varies by wavelength and dose (power and contact time)
- » The UV light can initiate photolysis reactions, including ozone production.
- » Ozone produced at wavelengths  $\leq 240$  nm. Maximum production at 160 nm.



*Low pressure mercury vapor*



*Light emitting diode (LED)*



*Pulse xenon arc lamp*



*Excimer glass lamp*

# Other types

## » Hypochlorous acid (HOCl)

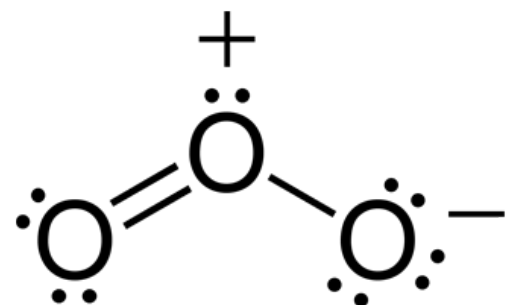
- Claim: HOCl generated from electrolysis of salt water.
- No peer-reviewed literature found evaluating claimed working mechanism.

## » Nano-confined catalytic oxidation (NCCO)

- Contains ozone generator upstream of a catalyst to decompose VOCs.
- Similar to PCO in working principle.
- Achieving complete reactions is challenging.
- Ozone emissions and byproducts (incomplete VOC breakdown) are a concern.

# Potential Emissions and Health Effects

# Ozone



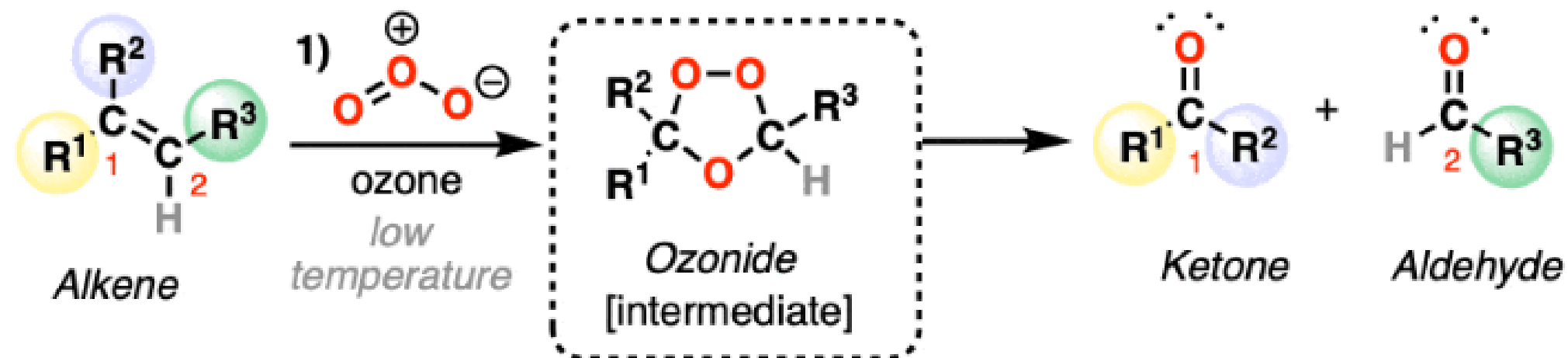
- » Main source indoors is transfer from outdoor air.
- » Outdoor ozone levels regulated by NAAQS (70 ppb 8-hr avg).
- » Indoors generally 20 to 70% of outdoors when no indoor ozone source present.

## » Health effects

- Respiratory tract inflammation, coughing, throat irritation, aggravation of asthma, chronic bronchitis and emphysema.
- Epidemiology studies show increase of outdoor ozone by 5 ppb associated with relative risk (RR) of 1.008 asthma-related ER visits.

# Ozone Reaction Products

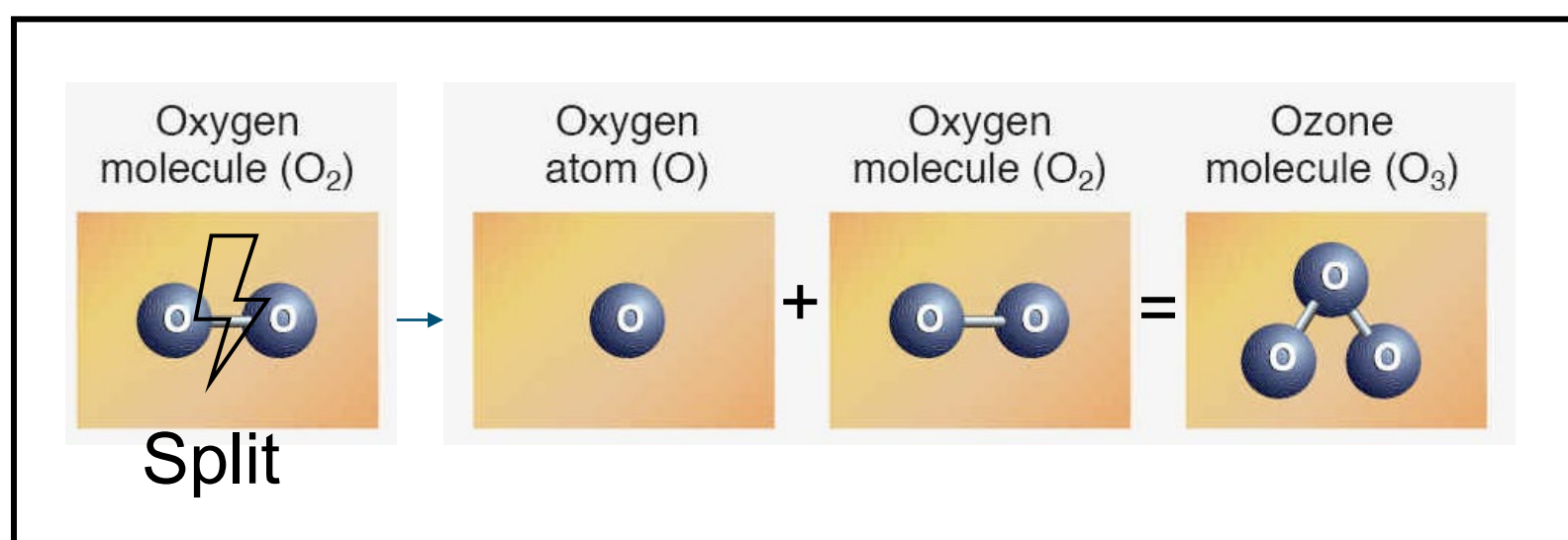
- » Ozone is a reactive molecule that drives complex reactions with VOCs.
- » Byproducts dependent upon what VOCs are in indoor air.
- » Generally, byproducts are VOCs that contain oxygen.
- » Two important byproducts are formaldehyde and acetaldehyde.
- » Increased building-related symptoms with higher outdoor ozone levels (cough, dry eyes, headache, respiratory and neurological symptoms).



*Example reaction of a VOC and ozone where an alkene transforms to a ketone and an aldehyde*

# Ozone Emissions from Air Cleaners

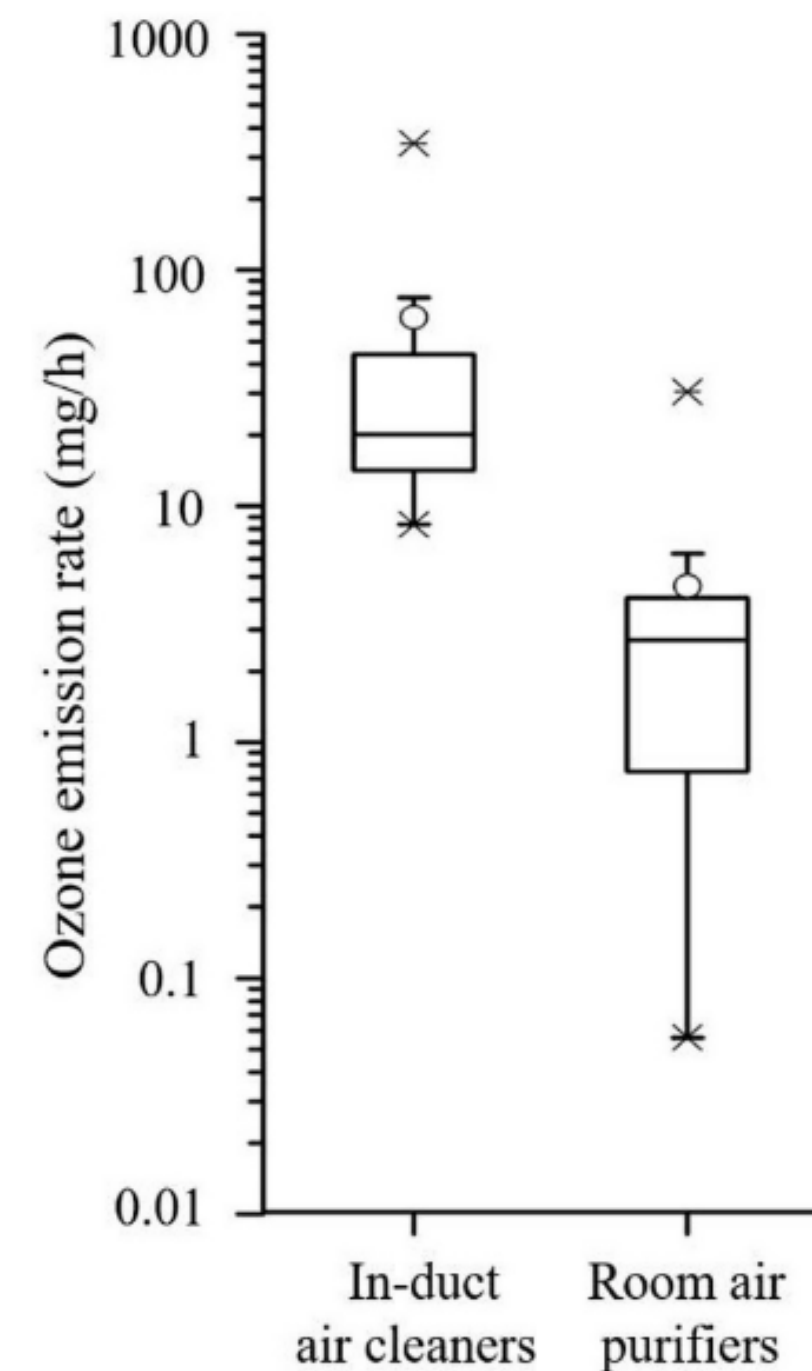
- » Can be generated from reaction of oxygen in the room initiated by UV light, ion generator, electrostatic precipitator
- » CARB requires electronic air cleaners be tested for ozone safety
- » Test standard (UL 867) requires increase of less than 50 ppb in the test chamber
- » Approx. equivalent to emission rate of 4 mg/hr
- » Device is tested when new



*Ozone formation*

# Ozone Emissions from Air Cleaners

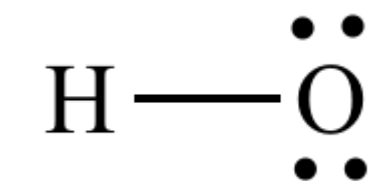
- » UL 867 will prevent the highest ozone generating air cleaners from being sold.
- » Passing devices can still generate ozone emissions that can meaningfully increase indoor concentrations.
- » CARB regulation includes in-duct air cleaners as of October 2020.



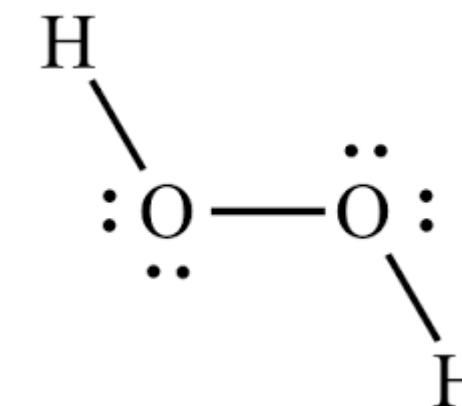
*Ozone emission rates  
reported by Guo et al. (2019)*

# Other Reactive Oxygen Species

- » Hydroxyl radicals and hydrogen peroxide are other highly reactive compounds derived from oxygen.
- » Hydroxyl radicals have a very short lifetime and are difficult to measure. Studies indicate indoor and outdoor concentrations are similar.
- » Hydrogen peroxide concentrations rarely reported in EAC studies.
  - Zeng et al. (2022) estimated 10 to 36 ppb (5 to 17 mg/hr) in testing an ion generator.
  - Well below 600 ppb observed when cleaning with hydrogen peroxide cleaning solution.
  - Well below National Institute for Occupational Safety and Health (NIOSH) 1 ppm 8hr exposure limit.



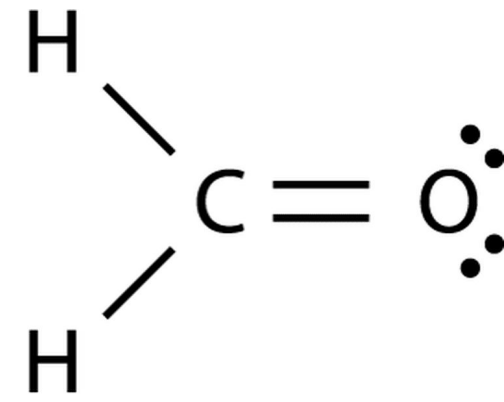
Hydroxyl radical



Hydrogen peroxide



# Formaldehyde

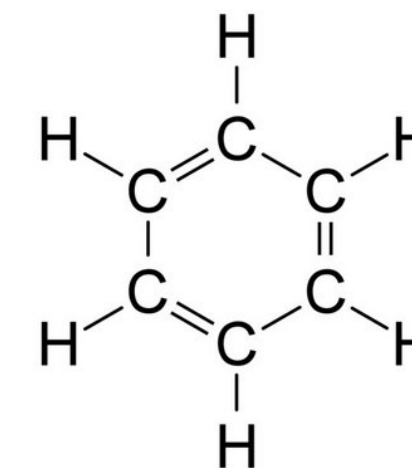


- » Carcinogen and respiratory irritant
- » California's Office of Environmental Health Hazard Assessment (OEHHA)
  - Safe chronic exposure limit of 7.3 ppb
  - 1 in a million cancer risk at 0.14 ppb
- » EPA's Integrated Risk Information System (IRIS)
  - Reference concentration of 5.7 ppb
  - 1 in a million cancer risk at 0.13 ppb
- » VOC commonly found elevated indoors
  - CA homes built 2002 to 2005 avg 35.0 ppb
  - CA homes built 2011 to 2017 avg 19.8 ppb (increased ventilation, lower emissions)
  - Still well above recommended chronic exposure levels

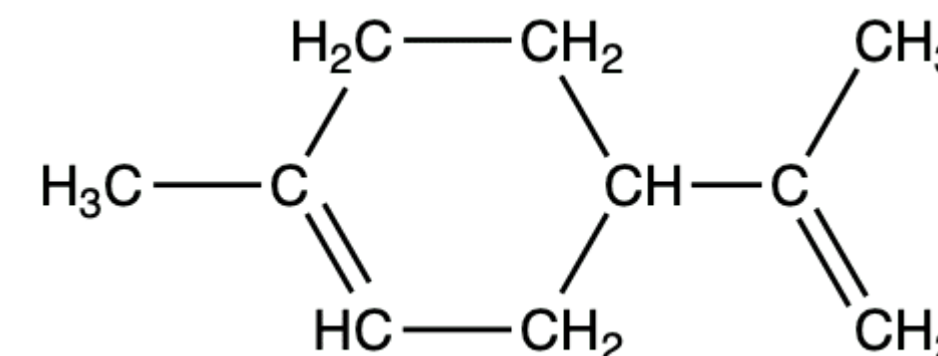
# Formaldehyde

- » Intermediate product formed in the breakdown of VOCs
- » Will be higher in situations where more VOCs are present
- » Formaldehyde production (i.e., conversion) generally observed in studies testing EAC
- » Whitepaper summarizes ~10 studies
- » Greater conversion (>50% on ppb basis) possible from EAC when non-aromatic VOCs are present in the room.

*Aromatic: Compound with carbon ring and alternating double and single bonds.*

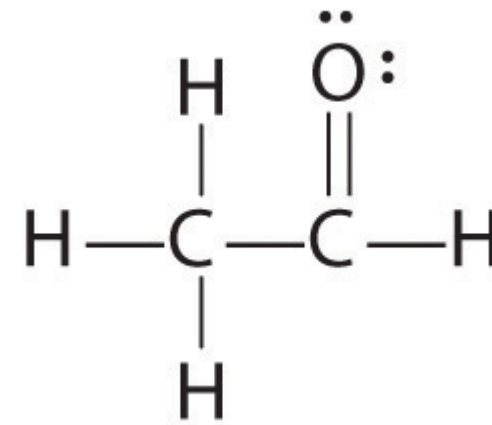


*Toluene, an aromatic VOC example. Less conversion to formaldehyde.*



*Limonene, a non-aromatic VOC example. Greater conversion to formaldehyde.*

# Acetaldehyde



- » Carcinogen and respiratory irritant
- » Commonly observed with formaldehyde (products of VOC breakdown)
- »  $\frac{\text{Acetaldehyde}}{\text{Formaldehyde}}$  production from EAC was 0.15 to 1.1 across four studies
- » California's Office of Environmental Health Hazard Assessment (OEHHA)
  - Safe chronic exposure limit of 77.7 ppb
  - 1 in a million cancer risk at 0.20 ppb
- » EPA's Integrated Risk Information System (IRIS)
  - Reference concentration of 5 ppb
  - 1 in a million cancer risk at 0.25 ppb

# Other VOCs

## » Acetone

- Observed in ppb range
- Eye nose and throat irritant at 250-1,000 ppm

## » Acrylonitrile

- Carcinogen observed emitted by one air cleaner in one study
- Likely an emission from air cleaner material (heated plastic) and not EAC mechanism, a concern for many consumer products

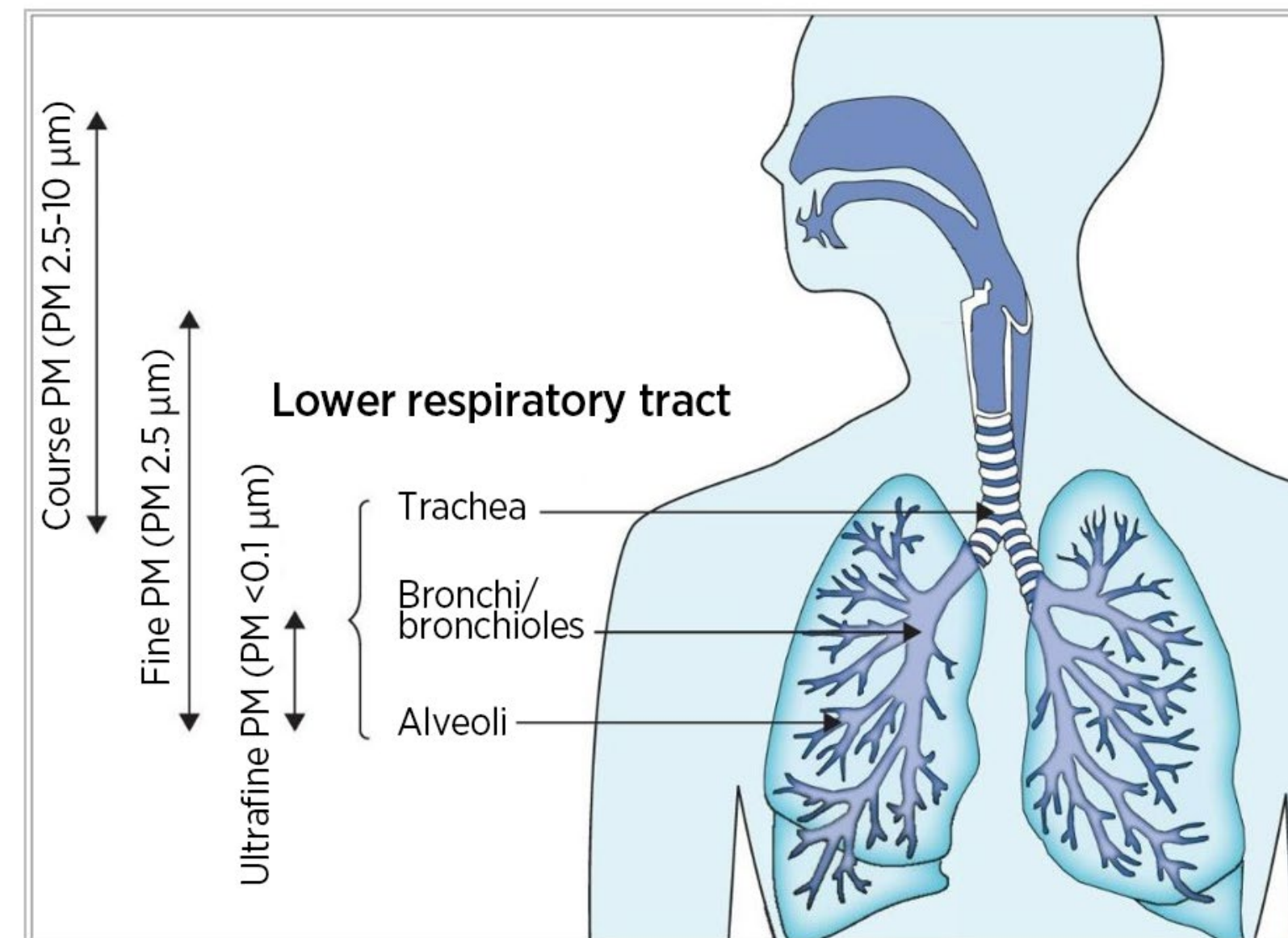
## » Propanediol, acetic acid, ethanol, and isopropanol

- Observed in nontarget analysis study by Ye et al.
- Any adverse health effects for these compounds are at concentrations several orders of magnitude above that observed in Ye et al.

*Limitation: You can only measure what your instruments are designed to detect*

# Ultrafine Particles (UFP)

- » Aerodynamic diameter  $\leq 0.1 \mu\text{m}$
- » Health effects: Pulmonary inflammation, heart rate variability, blood pressure cough, and worsen asthma. Linked to diabetes and low birth weight.



*UFP penetrates respiratory system and deposits in alveoli, where it can cross into bloodstream and diffuse into organ systems.*

*(Figure: Guarnieri and Balmes 2014)*

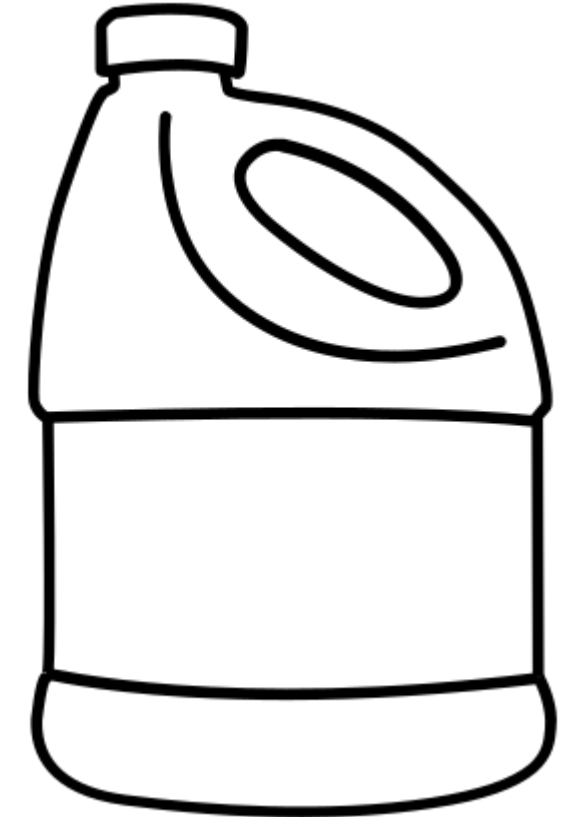
# Ultrafine Particles (UFP)

- » A byproduct of air cleaners that emit ozone
- » Ozone + terpenes (VOC) → UFP
- » Terpenes are in scented products like candles, air fresheners, and cleaning products
- » Yield rates of 8 to 39% terpene to UFP on a mass basis



# Hypochlorous Acid ( HOCl )

- » No data on HOCl concentrations from air cleaners
- » HOCl can reach ppb when cleaning with bleach
- » HOCl can react with surfaces to form toxic chlorine gas
- » Exposure to cleaning with chlorine bleach associated with respiratory and asthma symptoms, eye irritation, and sore throat
- » Health risks of this type of electronic air cleaner unknown



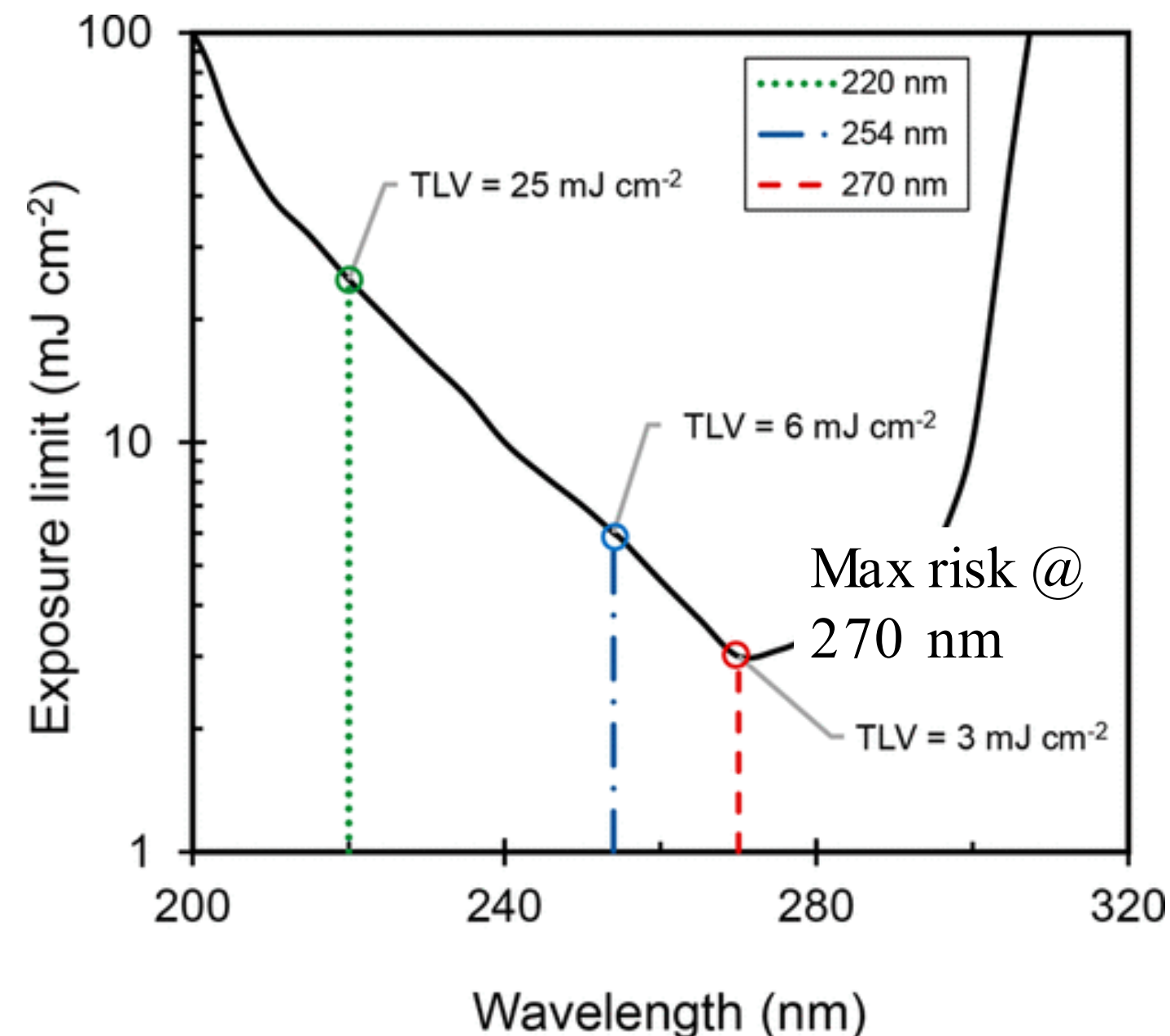
# Ions and Charged Particles

- » Gas or particle with an overall net electric charge
- » Generally elevated with electronic air cleaners
- » School children in China exhibited reduced heart rate variability (HRV) with exposure to electronic air cleaner and increased negative air ions. Lower HRV associated with increased risk of cardiovascular disease in adults.
- » Deposition rates of charged particles are generally higher than neutral particles. Unknown impacts to respiratory system.



# Direct UV Exposure

- » Risk of cancer of skin and eyes, sunburn and skin aging, cataracts and other eye diseases.
- » Guidelines developed for occupational health (does not consider children or sensitive population).
- » Some disinfection performance likely achievable @ 222nm within 23mJcm<sup>2</sup> exposure limit
- » Risk is negligible for enclosed UV lamps operated as designed



*Occupational 8-hour exposure limit  
(Raeiszadeh and Adeli, 2020)*

# Test Standards and Regulations

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## » Ozone

- UL 867 – chamber  $\leq 50$  ppb ( $\sim 4$  mg/hr)
- UL 2998 – chamber  $\leq 5$  ppb ( $\sim 0.4$  mg/hr)
- AHRI 681 – 50 ppb in a test duct

## » UV Safety

- IEC 62471:2006 – measures optical radiation (total energy) and irradiation (energy per unit surface area) and compares to exposure limits. Categorizes as low-risk, moderate-risk, or high-risk.

## » Chemical assessment

- AHAM-AC-4-2002 – removal rates of five gases. No byproduct measurements.
- ASTM-WK81750 – measurement of byproduct formed (ozone, formaldehyde, and UFP) when air cleaner challenged with chemical mix that includes ozone and VOCs.

# Conclusions and Recommendations

# Conclusions

- » Risks versus benefits of electronic air cleaners have been inadequately studied
- » No safety regulations other than ozone emission (CA only)
- » Byproducts that are compounds of clear concern: **ozone, formaldehyde, and ultrafine particles**
- » These compounds are an indicator of general byproduct production
- » Major knowledge gaps exist regarding byproduct emission rates, exposure level indoors, and health effects

# Knowledge Gaps

- » Quantification of human exposure to compounds of clear concern
  - Test standards for measuring byproduct formation. ASTM K81750 in process for ozone, formaldehyde, and UFP.
  - Increase in indoor air concentrations when using electronic air cleaners that emit byproducts at a quantified rate.
- » Quantification of human exposure to other compounds and health effects
  - Hydrogen peroxide, hypochlorous acid, ions and ionized particles
- » Health effects for direct exposure to UV light for the general population which includes infants, children, and other sensitive groups
- » Quantification of air cleaner benefits (e. test standards for contaminant removal) to enable comparison of benefit versus risk.

# Recommendations

## » Reduce ozone emissions

- End-users: Only use devices compliant with UL 2998 or mechanical filtration technology.
- Regulators: Consider requiring electronic air cleaners comply with UL 2998.

## » Fund research to address knowledge gaps

- Test air cleaners to ASTM K81750 and apply results to representative building models to estimate magnitude of increased exposure and resulting health impacts. Consider additional regulations if testing and modeling demonstrates unacceptable risk.
- Evaluate emission rates of
  - hydrogen peroxide from PCO air cleaners
  - HOCl from HOCl generating air cleaners
- Evaluate health effects at relevant exposure levels of:
  - hydrogen peroxide, HOC, ions and ionized particles
  - direct UV radiation considering general population (children and sensitive groups)

**Thank You**  
**Any Questions?**





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